

# FIRST PRINCIPLES STUDY ON THE FEATURES OF $\text{Ca}_x\text{Sr}_{2-x}\text{Ta}_2\text{O}_7$ AS PHOTOCATALYTIC MATERIALS

Marianela Gómez-Toledo, Khalid Boulahya and M. Elena Arroyo-de Dompablo

Departamento de Química Inorgánica, Universidad Complutense de Madrid, 28040, Madrid, Spain

**3rd International Electronic Conference on Applied Sciences**

Institutions that make possible this research:

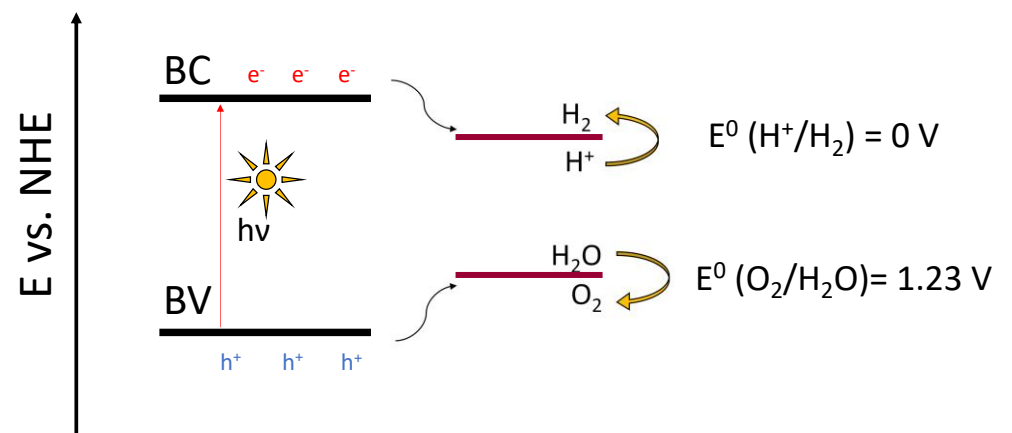
# INTRODUCTION

Hydrogen production through water splitting using sunlight might be vital for green hydrogen generation

Photocatalysts are semiconductors materials that are able to absorb **sunlight**, leading to the **promotion of electrons** from the valence band to the conduction band

Sunlight is composed by 7.2% UV + 45.6% IR + **47.2% visible**

For **H<sub>2</sub>** production through **water splitting**, **bands positions** are key



CB minimum should be **above** the H<sub>2</sub>O/H<sub>2</sub> reduction potential (0.0 eV)

VB maximum should be **below** the O<sub>2</sub>/H<sub>2</sub>O oxidation potential

**band gap** should be between 1.23 eV and 3.0 eV

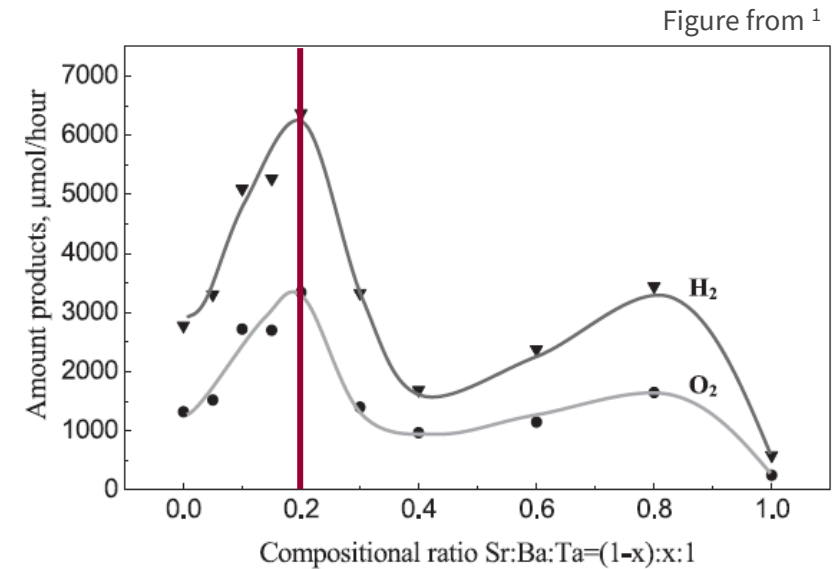
Among other photocatalysts, **Sr<sub>2</sub>Ta<sub>2</sub>O<sub>7</sub>** shows some exciting features, but its large band gap (4.6 eV) restricts the absorption to the UV region

## Characteristics that make $\text{Sr}_2\text{Ta}_2\text{O}_7$ an appealing compound for photocatalysis

$\text{Sr}_2\text{Ta}_2\text{O}_7$  shows a **large CB** position relative to  $\text{H}_2\text{O}/\text{H}_2$  reduction potential

Kim et al.<sup>1</sup> have demonstrated that the **partial substitution** of Sr with Ba in the  $\text{Sr}_{2-x}\text{Ba}_x\text{Ta}_2\text{O}_7$  ( $0 < x < 0.4$ ) family can lead to upgrading in photocatalytic activity.

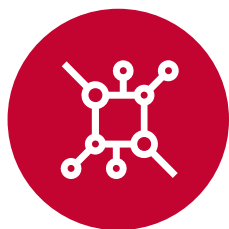
The photocatalytic activity in solid solution  $\text{Sr}_{2-x}\text{Ba}_x\text{Ta}_2\text{O}_7$  is **maximum** when  $x = 0.4$ .



Would partial substitution of Sr with **Ca** have the same effect?

This work focuses on the **computational study** through **DFT** calculations of the structural and electronic changes that the substitution of Sr with Ca can provoke in the  $\text{Sr}_{2-x}\text{Ca}_x\text{Ta}_2\text{O}_7$  family

# METHODOLOGY



## INPUT

Crystallographic  
models



## OUTPUT

Crystal and  
electronic  
structures

DFT

$$\hat{H}_{KS} = \hat{T}_0 + \hat{V}_H + \hat{V}_{ext} + \hat{V}_{xc}$$

In this work, **SCAN** meta-GGA XC-functional is used to study a photocatalysts' family

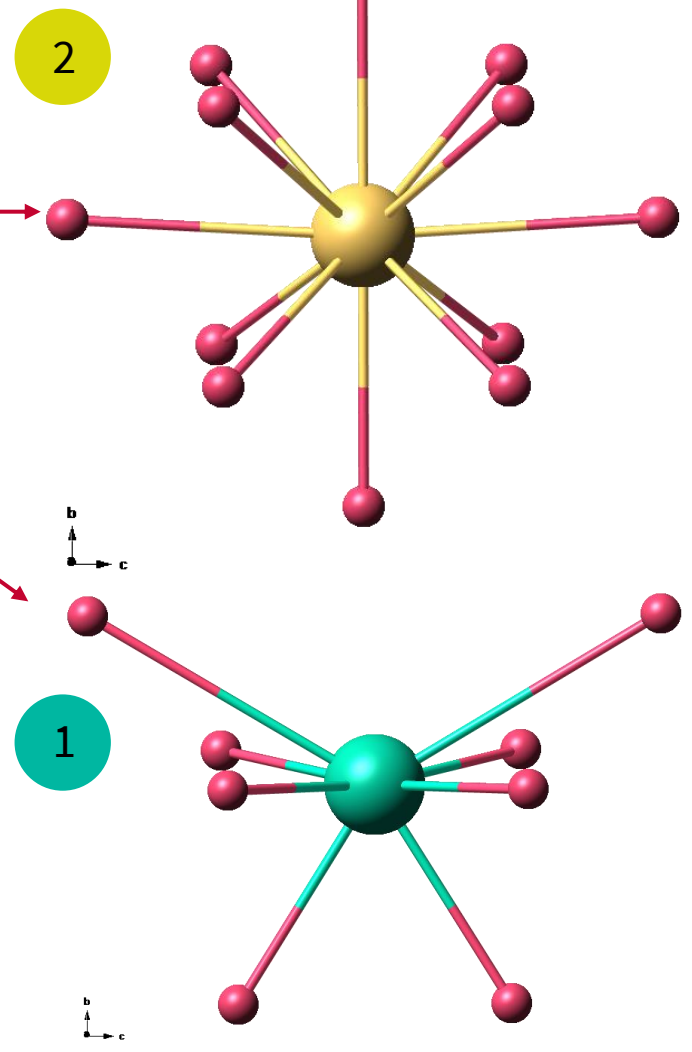
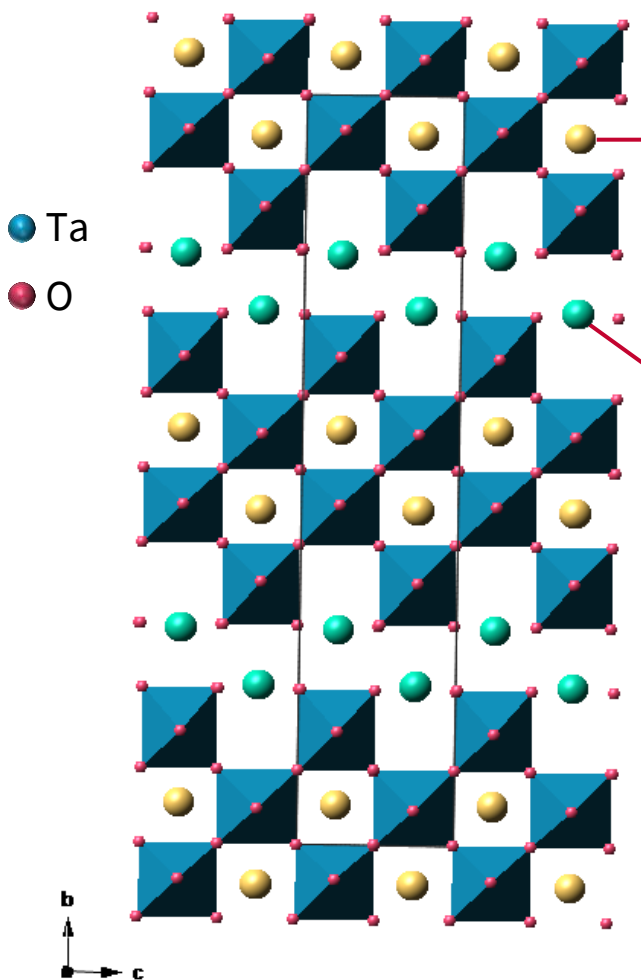
Calculations have been performed using VASP (Vienna ab-initio simulation package)

Interaction of core electrons with nuclei was described through the PAW (Projector Augmented Wave method)

Energy cut-off was set at 600 eV

K-point meshes were 6 x 2 x 6 in all calculations (Monkhorts-Pack scheme)

# CRYSTAL STRUCTURE



Ca substitution is studied using **two** different models

**Model 1**

Substitution of Sr(1) with Ca  
 $\text{CaSrTa}_2\text{O}_7(2)$

**Model 2**

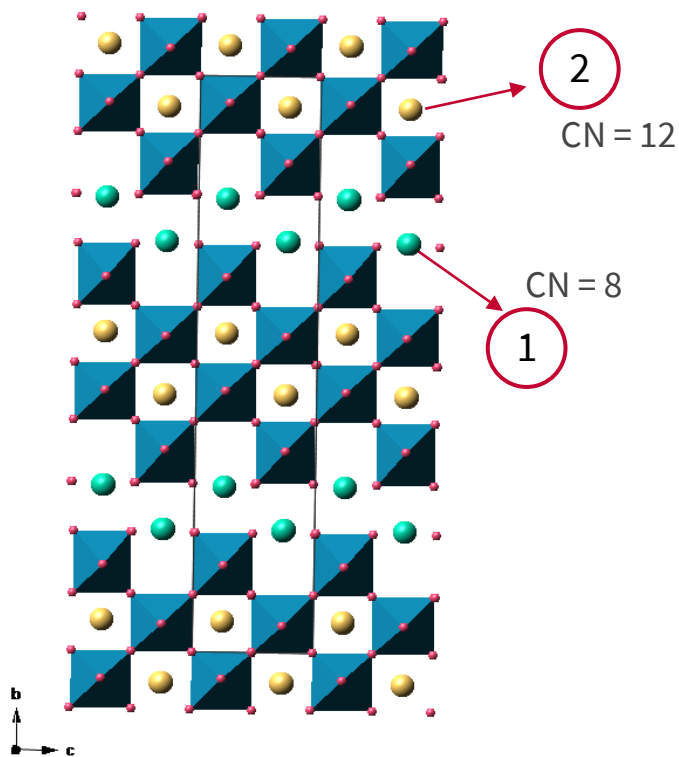
Substitution of Sr(2) with Ca  
 $\text{CaSrTa}_2\text{O}_7(2)$

# RESULTS



# CRYSTAL STRUCTURE

- SCAN functional correctly reproduces distances and lattice parameters of  $\text{Sr}_2\text{Ta}_2\text{O}_7$
- $\text{CaSrTa}_2\text{O}_7$  models preserves the initial structure of the parent  $\text{Sr}_2\text{Ta}_2\text{O}_7$  (SG  $C_{mcm}$ )



Lattice parameters (Å)					
$\text{Sr}_2\text{Ta}_2\text{O}_7$				$\text{CaSrTa}_2\text{O}_7(1)$	$\text{CaSrTa}_2\text{O}_7(2)$
	experimental	calculated	% error	calculated	calculated
a	3.9376	3.9537	0.4	3.9084	3.9273
b	27.1986	27.1894	0.03	27.0999	27.0855
c	5.6927	5.6951	0.04	5.6119	5.6646
Volume (Å <sup>3</sup> )	609.50	612.22	0.4	594.40	602.56

# CRYSTAL STRUCTURE

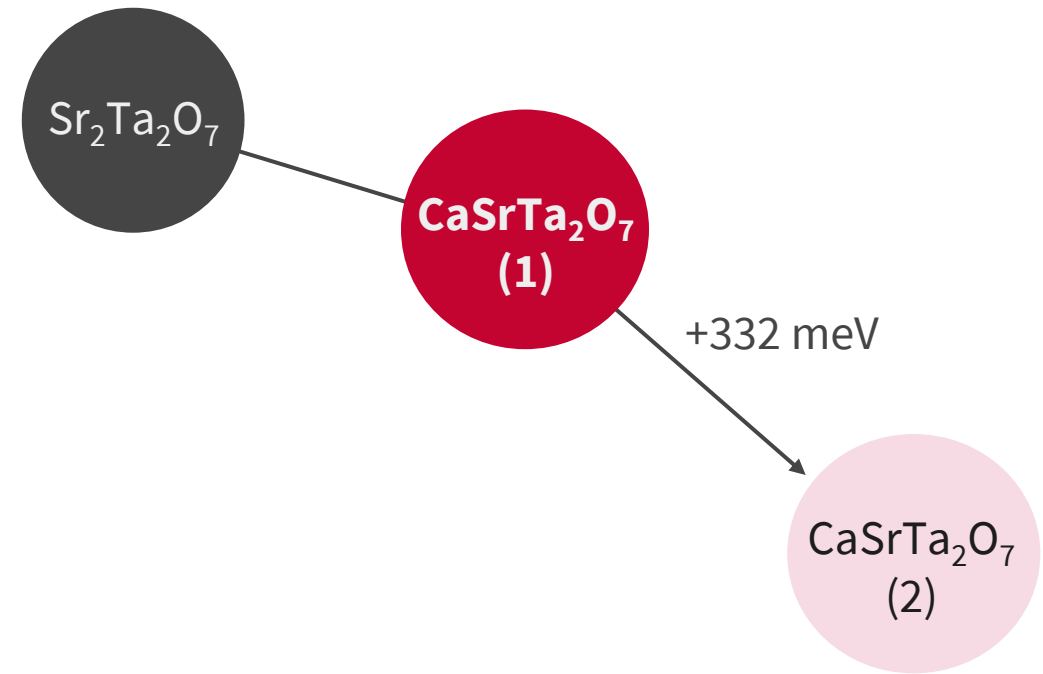
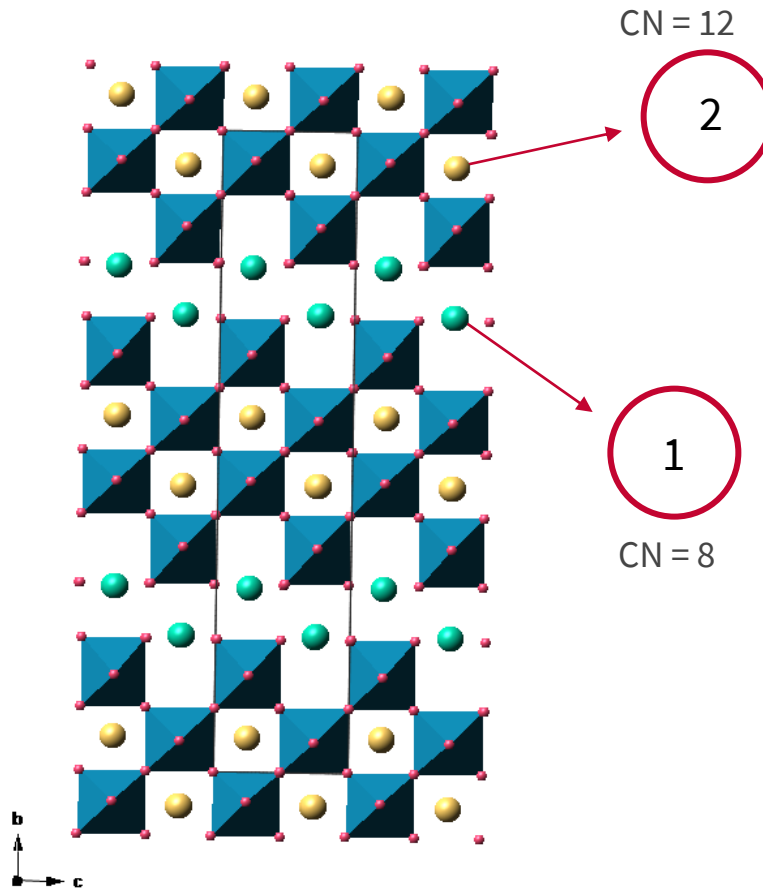


## Model 1

Substitution of Sr(1) with Ca

## Model 2

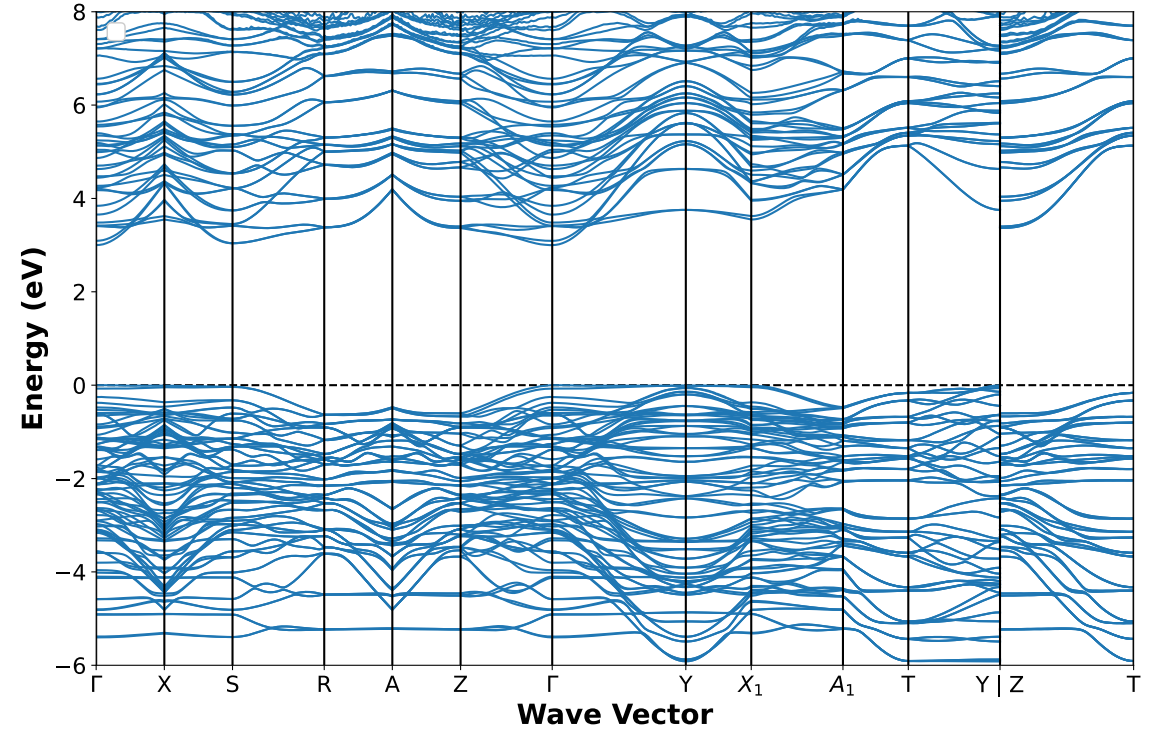
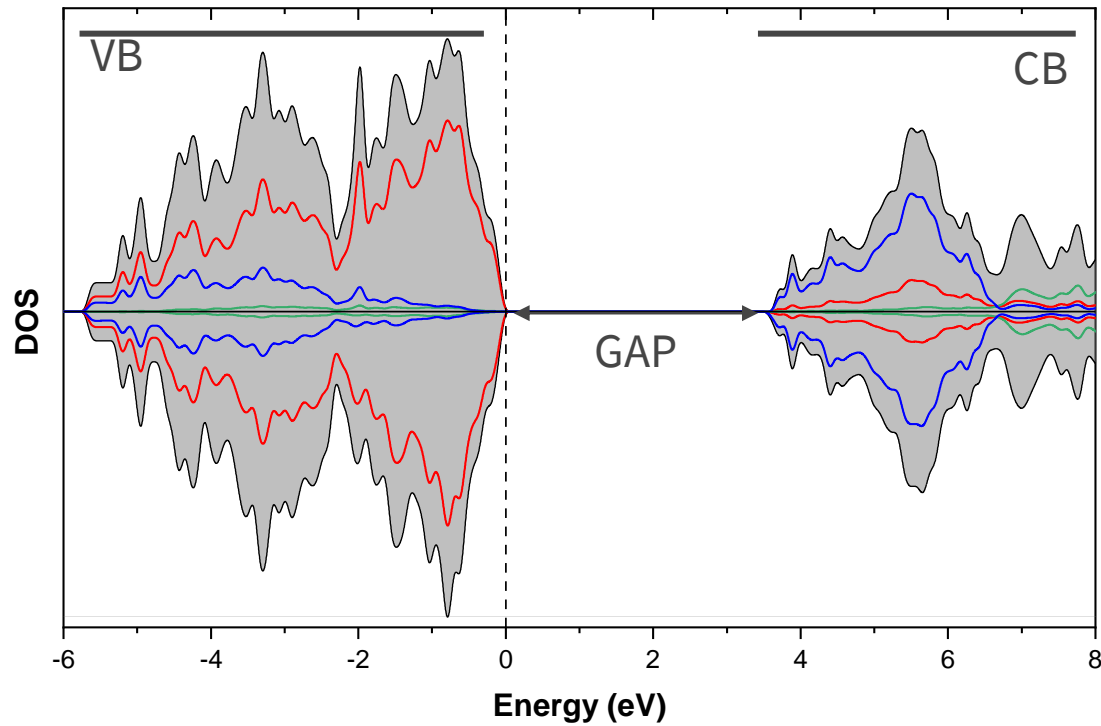
Substitution of Sr(2) with Ca



**Ca<sup>2+</sup>**  
 $r = 1.12 \text{ \AA}$  for CN = 8  
 $r = 1.34 \text{ \AA}$  for CN = 12

**Sr<sup>2+</sup>**  
 $r = 1.26 \text{ \AA}$  for CN = 8  
 $r = 1.44 \text{ \AA}$  for CN = 12

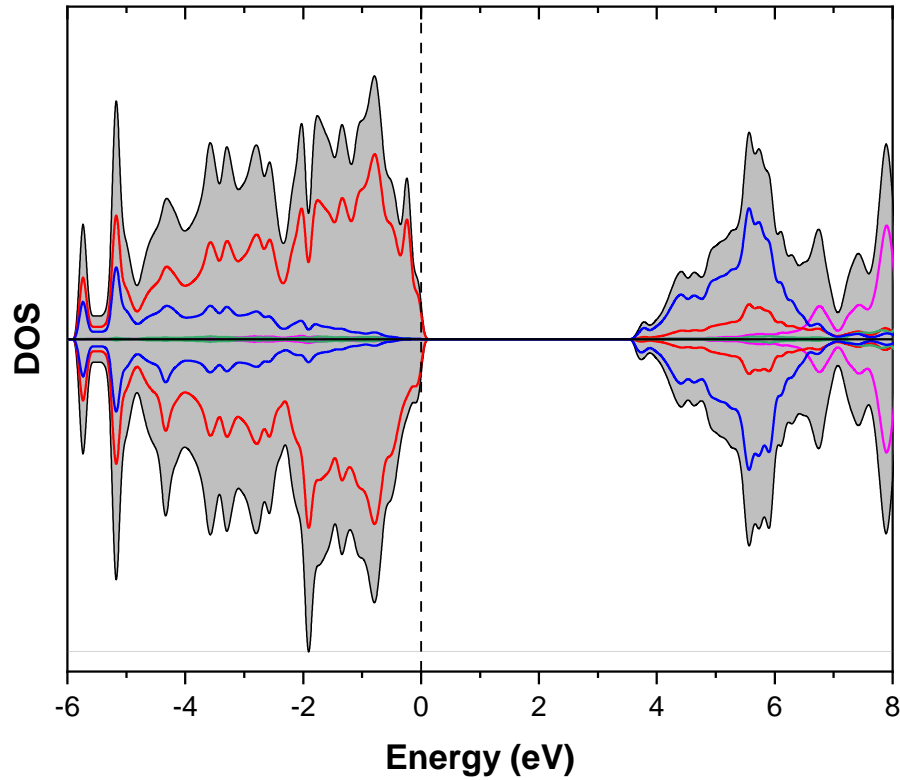
- **Model 1** has lower energy when compared to Model 2  $\rightarrow$  A more stable structure forms if the larger  $\text{Sr}^{2+}$  cation occupies the position inside perovskite blocks, i.e. position 2.



- Calculated band gap of 3.65 eV
- VB formed mainly by 2p O states while 5d Ta forms CB
- Hybridization exists in both bands

- Direct band gap

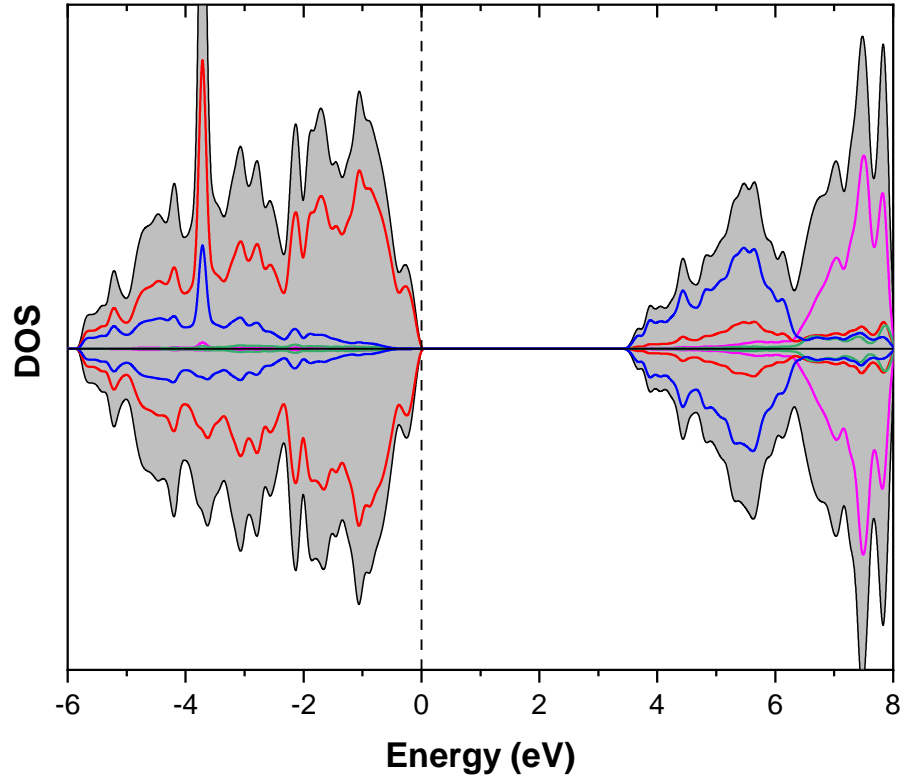
## CaSrTa<sub>2</sub>O<sub>7</sub> (1)



- Band gap of 3.55 eV

- VB and CB have essentially the same features as in Sr<sub>2</sub>Ta<sub>2</sub>O<sub>7</sub>

## CaSrTa<sub>2</sub>O<sub>7</sub> (2)



- Band gap of 3.50 eV

**Substitution of Sr with Ca does not lead to a significant change in band gap**

# CONCLUSIONS

## CONCLUSIONS

Computational investigations at the level of **Density Functional Theory (DFT)** permit predicting some basic and critical features in view of the photocatalytic activity of materials

- The  $x = 1$  phases of  $\text{Sr}_{2-x}\text{Ca}_x\text{Ta}_2\text{O}_7$  family preserve the perovskite-related structure of  $\text{Sr}_2\text{Ta}_2\text{O}_7$  parent compound
- Band-gap narrowing of 0.10-0.15 eV with partial substitution of Sr by Ca
- Absorption of sunlight is still limited to the UV region
- Work is in progress to investigate other substitutions