

Proceeding paper

Defects Formation of Light-Emitting Particles during the Synthesis of a Hierarchical Porous Surface of ZnO/SiO₂/Si Structures [†]

Rashid Zhapakov ^{*}, Mykhail Begunov, Tatyana Seredavina, Danatbek Murzalinov [‡], Abay Serikkanov, Elena Dmitriyeva, Sultan Zhantuarov and Sayara Ibraimova

Institute of Physics and Technology, Satbayev University, Almaty 050013, Kazakhstan; email1@email.com (M.B.); email2@email.com (T.S.); email3@email.com (D.M.); email4@email.com (A.S.); email5@email.com (E.D.); email6@email.com (S.Z.); email7@email.com (S.I.)

^{*} Correspondence: r.zhapakov@ncste.kz

[†] Presented at the 2nd International Electronic Conference on Processes: Process Engineering—Current State and Future Trends (ECP 2023), 17–31 May 2023; Available online: <https://ecp2023.sciforum.net/>.

[‡] Project supervisor.

Abstract: The formation of structures of different sizes on the surface of a single sample and the study of the mechanisms of their excitations and energy transitions is a promising area of research. ZnO/SiO₂/Si heterostructures were formed by two-stage electrochemical etching of silicon wafers and synthesizing zinc oxide nanoparticles using sol-gel technology. By scanning electron and atomic force microscopies, various pores and clusters were identified. The presence of five levels of surface hierarchy provided the synthesis of light-emitting particles with different properties. Light emission mechanisms are associated with the recombination of excitons and the hyperfine structure of charged particles trapped on oxygen vacancy.

Keywords: ZnO; photoluminescence; exciton; oxygen vacancies; porous silicon

Citation: Zhapakov, R.; Begunov, M.; Seredavina, T.; Murzalinov, D.; Serikkanov, A.; Dmitriyeva, E.; Zhantuarov, S.; Ibraimova, S. Defects Formation of Light-Emitting Particles during the Synthesis of a Hierarchical Porous Surface of ZnO/SiO₂/Si Structures. *Eng. Proc.* **2023**, *37*, x.

<https://doi.org/10.3390/xxxxx>
Published: 17 May 2023



Copyright: © 2023 by the authors. Submitted for possible open access publication under the terms and conditions of the Creative Commons Attribution (CC BY) license (<https://creativecommons.org/licenses/by/4.0/>).

1. Introduction

The advantage of developed surface morphology is associated with a high ratio of surface area to volume and enhanced roughness. These features can be applied when choosing ways to obtain sensors for various gases. Complicating the surface structure increases the sensitivity of the sensor, which is accompanied by the formation of particles with an uncompensated charge [1–3].

Since silicon is frequently used in semiconducting devices, zinc oxide nanostructures (Ncs) synthesized on Si-based substrates are of great interest in recent years. Nevertheless, the discrepancy between the coefficients of thermal expansion and the crystal lattice between ZnO and Si substrate precludes the ZnO layer from being formed or deposited directly on the substrate. In contrast, porous silicon (PS) has a spongy structure, a high internal surface-to-volume ratio, controllable roughness that decreases mismatches between Si substrates and ZnO Ncs. It provides a suitable substrate for the formation of ZnO nanostructures without the metal catalysts [4–7].

Nevertheless, in order to reduce the degradation of devices, it is necessary to enhance the stability of porous silicon. There are several methods to solve this problem, and the metal oxide coating on PS is the most promising [8].

One of the basic defects of zinc oxide is the vacancy of oxygen. Current models of methanol synthesis on the ZnO surface, as well as new theoretical calculations, show that

when methanol synthesis begins, oxygen vacancies synthesized on the surface of ZnO crystals may be the active centers for CO chemisorption [9].

Simultaneously, if oxygen vacancies catch a charge, the process of particle adsorption changes. Moreover, when the vacancy concentration increases, the sample's sorption capacity is increased.

Vacancies are also involved in the processes of radiative recombination of particles, for example, when catching an electron or a hole.

The role of vacancies in the process of spatial separation of charges is attractive. This process is necessary to save the charges and prevent their recombination. Nearby vacancies of different types (one cationic and the other anionic) capture an electron and a hole and as a result, these particles lose mobility and autolocalize. However, the energy connection between them is preserved.

In our previous work, by the method of spectroscopy electron paramagnetic resonance (EPR) was confirmed the possibility of accumulation of charges of the same type with the growth of nanostructures of ZnO deposited on the surface of PS by spray pyrolysis. An important feature of this process is the spatial localization of paramagnetic centers on fragments or clusters localized at the interface of the walls of the surface macropores and the main surface, as well as in micropores [10].

The effectiveness of the formation of nanostructures on the surface of a substance is associated with the presence of particles with an uncompensated charge on it. The formation of a hierarchical porous surface is interesting because there is a synthesis of structures with different sizes and properties.

The novelty of these studies is the formation of the sample structure, which includes five levels of the pores hierarchy and clusters of matter.

The purpose of these studies is to determine the mechanisms of formation of light-emitting ZnO/SiO₂/Si structures of various scales with increased photoluminescence intensity.

2. Materials and Methods

2.1. Materials and Synthesis

2.1.1. Synthesis of Porous Silicon

By electrochemically etching n-Si (100), having a resistance of 12 Ω·cm, porous silicon layers were produced. The electrolyte was an aqueous alcohol solution of hydrogen fluoride. Hydrofluoric acid 45.00%. Isopropanol SPIRT-9805.F01080. For 10 min, there was a 10 mA/cm² anodizing current density.

2.1.2. Synthesis and Deposition of a ZnO Coating on a Substrate

Sol-Gel Method

The film-forming solution was synthesized using the sol-gel technique. The sol solution was produced by mixing 0.1 M zinc acetate dihydrate (Zn(CH₃COO)₂·2H₂O) with 9 mL isopropanol (C₃H₇OH) as a solvent and 1 mL monoethanolamine (C₂H₇NO) as a stabilizing agent. Within an hour of the synthesis of the solutions, ZnO films started to be deposited on the substrate.

Deposition of the Film Layers

Substrates on the heating element are sprayed with liquid solutions using a pneumatic airbrush. The distance between the airbrush nozzle and the substrates ranged between 20 and 30 cm. The spray solution flow was most consistent when the pressure was adjusted to 1.4 bar.

The substrate's temperature ranged from 350 to 400 °C. In this case, spherical and hexagonal zinc oxide films were produced.

Based on the fact that in the spray pyrolysis technology, one layer of ZnO is extremely thin, it was assumed that when 20 and 25 layers are deposited on a porous surface, volumetric nanostructures are formed.

2.2. Characterization Methods

Scanning electron microscope (SEM) JSM-6490LA (“JEOL”, Akishima, Japan). The take-off angle for the JSM-6490LA is 35°, with an analytical working distance of 10 mm. The microscope has a high resolution of 3.0 nm.

EPR spectrometer “JEOL” (JES-FA200, Japan). Measurements in the ranges ~9.4 GHz (X-Band) and ~35 GHz (Q-Band). Microwave frequency stability ~ 10⁻⁶. Sensitivity—7 × 10⁹/10⁻⁴ Tl. Resolution—2.35 μT. Output power—from 200 mW to 0.1 μW, quality factor (Q—factor) 18,000.

Atomic force microscope (AFM) (Japan, “JEOL” JSPM-5200). The operating temperature range is -100 °C to 500 °C, and the vacuum depth is up to 10⁻⁶ mm Hg.

The spectrophotometer Cary Eclipse (Agilent, US) was used to detect photoluminescence (PL) from 200 to 800 nm. The spectral width of the slit in this device is changeable, ranging from 0.5 to 2.4 nm. As a radiation source, a tungsten-halogen lamp is used for visible measurements and a deuterium lamp is used for UV measurements.

3. Results and Discussion

3.1. Investigation of Surface Morphology by Scanning Electron Microscopy

To study the process of surface structure formation, images of the initial sample and samples coated with ZnO were obtained using a scanning electron microscope. The macroporous structure of the samples is clearly represented in the images at a magnification of 430 times (Figure 1a).

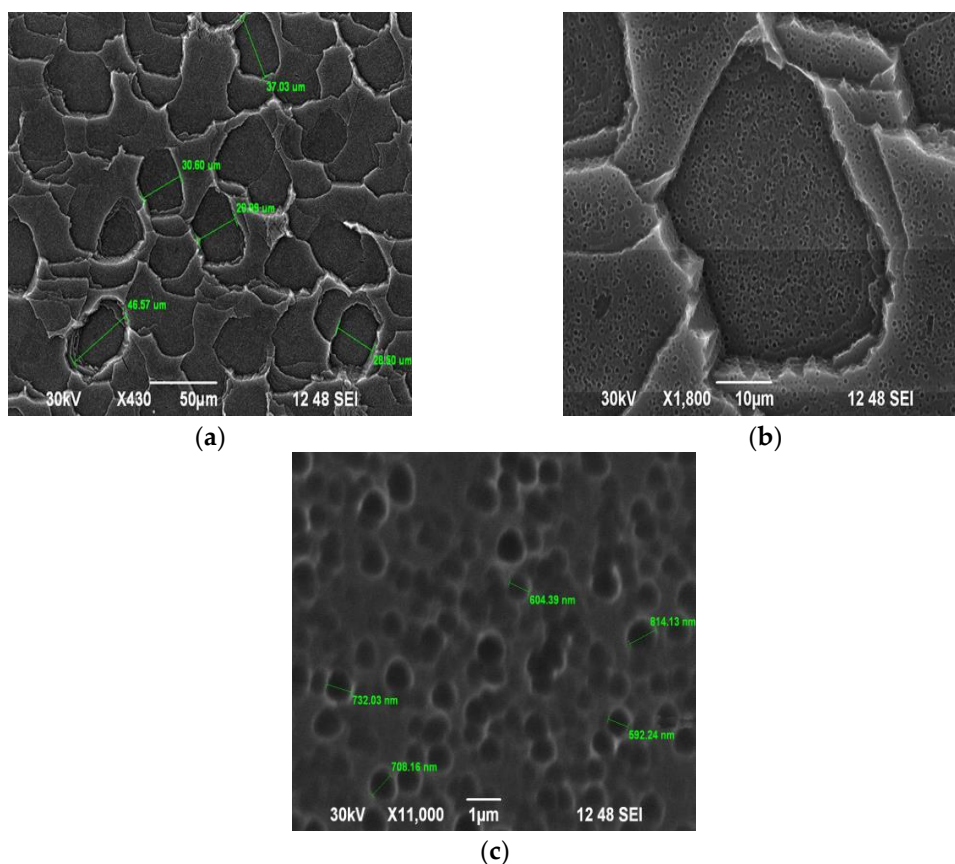


Figure 1. SEM images of samples without ZnO coating: (a) at magnification in ×430; (b) at magnification in ×1800; (c) at magnification in ×11,000.

Figure 1b,c show the results of the SEM study of surface morphology at a higher resolution (at magnification in $\times 430$ and in $\times 11000$). The images show that the surface structures have a step-like nature of porosity, from macro to micro-dimensional. It is also possible to determine the developed structure of the pore walls.

The deposition of coating layers led to a decrease in the diameter of the macropores. Thus, with an increase in the number of layers, the surface became more finely porous and at the same time the porosity is more uniform in size. This is because the coating is localized at the boundaries of the pores. Moreover, with an increase in the number of coating layers, the walls of the macropores became smoother, and the size of the micropores decreased.

The more developed the structure of the samples, the more particles with uncompensated charge they contain. This is due to the high chemical activity of the surface. When zinc oxide is deposited to PS, the mechanical stability of the structures increases and light-emitting particles appear.

3.2. Investigation of Surface Morphology Using Atomic Force Microscopy

For a detailed study of the structures inside the macropores, atomic force microscopy images were obtained.

Figure 2a shows the structure between the micropores. It is clearly seen that in addition to regularly spaced pores, the clusters of matter are noticeable. The clusters have similar shape and the distances between them are identical.

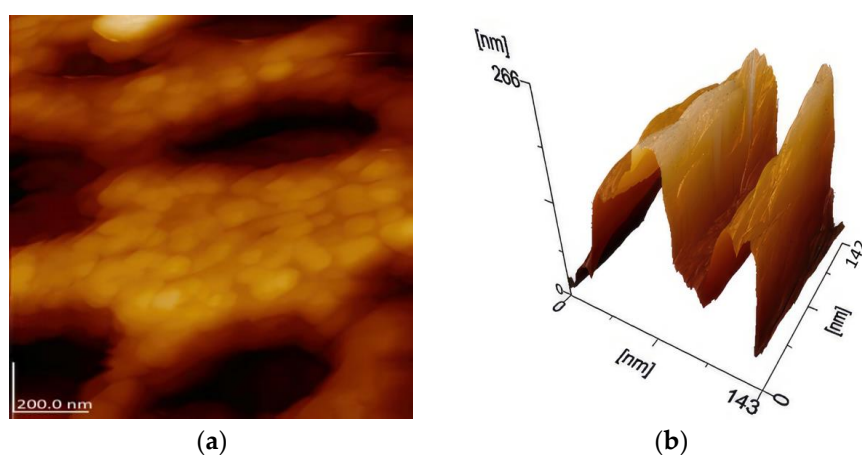


Figure 2. (a) The structure of the surface inside the macropores; (b) the micropore structure.

The following Figure 2b shows the structure of one micro pore. The structure inside and outside the deepening is noticeably developed, which is associated with the penetration of the film-forming solution in different directions. This is important when using a porous surface as a substrate. ZnO crystals could be formed due to porous silicon substrates providing nucleation centers.

Thus, a hierarchical porous structure of the sample surface was formed, including macro and micro pores, nanocrystals between micro pores and ridges on the outer and inner sides of micro pores. The morphology of the surface is complicated by the developed structure of the walls of macropores. The complexity was increased through the synthesis of multilayer ZnO/SiO₂/Si.

3.3. Photoluminescence Studies

At the same time, the smaller the size of the structures, the more effective photoluminescence they have. Figure 3 shows the PL spectra recorded at room temperature, for the initial sample and coated samples (excitation at 320 nm).

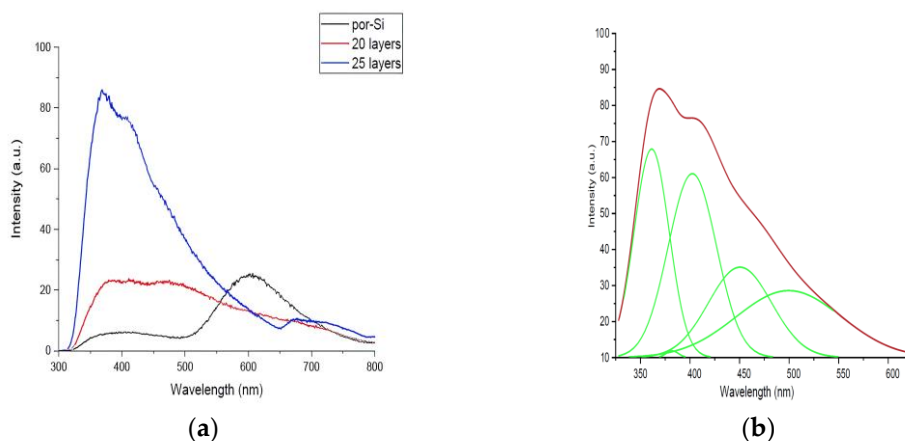


Figure 3. (a) Photoluminescence spectra of the initial sample and samples with deposited zinc oxide; (b) decomposed photoluminescence spectra of a sample with 25 layers of zinc oxide.

The spectrum of the initial sample has a low intensity, which disappears after the films are deposited. This is due to passivation of the sample surface, that is, a decrease in the number of particles with broken bonds.

The complex spectrum of PL after coating is associated with the presence of structures with different sizes. The most characteristic peak for zinc oxide crystals at 380 nm is noticeable on the spectrum of a sample with 25 layers. Its nature is related to the recombination of excitons. The band at 400 nm is caused by radiative transitions near the edge of fundamental absorption. The broadband of ~500 nm is associated with recombination of charges on oxygen vacancies during the formation of nanocrystalline ZnO films. The boundaries of nanocrystals have the structure of a gradual transition from one type of particle to another.

3.4. EPR Studies

To clarify the mechanisms of interaction of particles with uncompensated charge during the formation of nanostructures, EPR studies were carried out.

Figure 4a,b shows the EPR spectrum of a sample with 25 coating layers. In the central part of the spectrum (highlighted in blue), a signal with a G-factor of ~1.99 is observed. Noticeably that its shape differs from the graph of the first derivative. This is due to the presence of a weak anisotropy of a signal, which can be seen when the EPR signal is saturated (Figure 4b). Its nature is associated with trapped electrons on oxygen vacancies, probably at the boundaries of micropores, possibly F-centers in a small concentration.

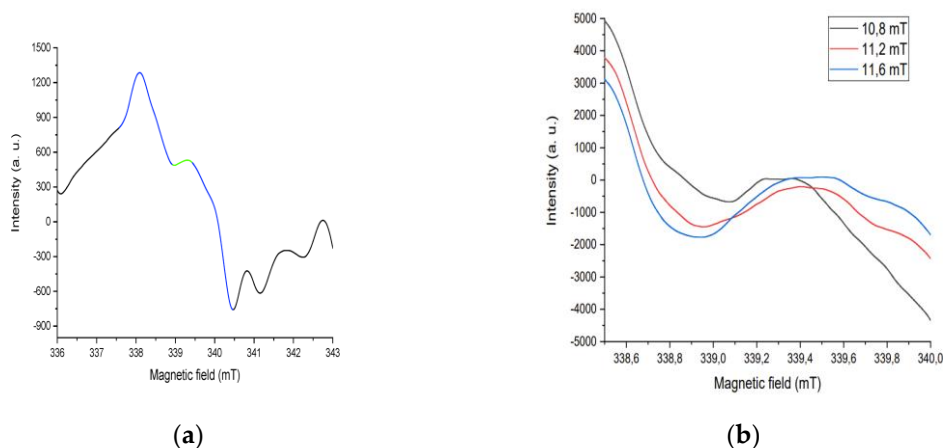


Figure 4. (a) EPR spectra of the sample with 25 layers of zinc oxide; (b) saturation of the EPR signal of the sample with 25 layers of zinc oxide at microwave power: 10.8 mT, 11.2 mT, 11.6 mT.

4. Conclusions

A fractally hierarchical structure of the sample surface was formed by electro-chemical etching of silicon with subsequent deposition of ZnO layers. As a result of the deposition of zinc oxide layers, a more homogeneous surface (than without deposition) as well as stable and efficient light-emitting particles were formed. A complex EPR signal from the captured charge of all coated samples was identified. With an increase in the number of ZnO layers to 25, the intensity of PL has significantly increased, the nature of which is associated with the capture of charge carriers on oxygen vacancies within formed nanocrystals of matter. The most intense PL signal is related to the recombination of excitons.

Author Contributions: Conceptualization, D.M. and M.B.; methodology, R.Z.; software, E.D.; validation, S.Z.; formal analysis, D.M. and T.S.; investigation, D.M.; resources, A.S.; data curation, S.I.; writing—original draft preparation, R.Z. and T.S.; writing—review and editing, T.S. and D.M.; visualization, S.Z. and M.B.; supervision, R.Z.; project administration, D.M.; funding acquisition, A.S. All authors have read and agreed to the published version of the manuscript.

Funding: This research was funded by the Science Committee of the Ministry of Science and Higher Education of the Republic of Kazakhstan (Grant No. AP09058002 «Investigation of the properties of dynamic memory based on Si₃N₄/Si and the formation of silicon nanoclusters with increased photoluminescence intensity»).

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Conflicts of Interest: The authors declare no conflict of interest.

References

1. Martínez, L.; García-Salgado, G.; Morales-Morales, F.; Campillo, B.; Hernández, A.G.; Karthik, T.V.; Campos-Álvarez, J. ZnO Films Incorporation Study on Macroporous Silicon Structure. *Materials* **2021**, *14*, 1–10.
2. Paliwal, A.; Sharma, A.; Tomar, M.; Gupta, V. Carbon monoxide (CO) optical gas sensor based on ZnO thin films. *Sens. Actuators B Chem.* **2017**, *250*, 679–685.
3. Cheng, I.-K.; Lin, C.-Y.; Pan, F.-M. Gas sensing behavior of ZnO toward H₂ at temperatures below 300 °C and its dependence on humidity and Pt-decoration. *Appl. Surf. Sci.* **2021**, *541*, 148551.
4. Zhang, H.; Jia, Z. Application of porous silicon microcavity to enhance photoluminescence of ZnO/PS nanocomposites in UV light emission. *Optik* **2017**, *130*, 1183–1190.
5. Ghosh, R.; Giri, P.K.; Imakita, K.; Fujii, M. Photoluminescence signature of resonant energy transfer in ZnO coated Si nanocrystals decorated on vertical Si nanowires array. *J. Alloys Compd* **2015**, *638*, 419–428.
6. Makarona, E.; Skoulikidou, M.C.; Kyrasta, T.; Smyrnakis, A.; Zeniou, A.; Gogolides, E.; Tsamis, C. Controllable fabrication of bioinspired three-dimensional ZnO/Si nanoarchitectures. *Mater. Lett.* **2015**, *142*, 211–216.
7. Al-Salman, H.S.; Abdullah, M.J. Preparation of ZnO nanostructures by RF-magnetron sputtering on thermally oxidized porous silicon substrate for VOC sensing application. *Measurement* **2015**, *59*, 248–257.
8. Pavlenko, M.; Myndrul, V.; Gottardi, G.; Coy, E.; Jancelewicz, M.; Iatsunskyi, I. Porous silicon-zinc oxide nanocomposites prepared by atomic layer deposition for biophotonic applications. *Materials* **2020**, *13*, 1987.
9. Ischenko, V.; Polarz, S.; Grote, D.; Stavarache, V.; Fink, K.; Driess, M. Zinc oxide nanoparticles with defects. *Adv. Funct. Mater.* **2005**, *15*, 1945–1954.
10. Murzalinov, D.; Kemelbekova, A.; Serejavina, T.; Spivak, Y.; Serikkanov, A.; Shongalova, A.; Mukhamedshina, D. Self-Organization Effects of Thin ZnO Layers on the Surface of Porous Silicon by Formation of Energetically Stable Nanostructures. *Materials* **2023**, *16*, 838.

Disclaimer/Publisher's Note: The statements, opinions and data contained in all publications are solely those of the individual author(s) and contributor(s) and not of MDPI and/or the editor(s). MDPI and/or the editor(s) disclaim responsibility for any injury to people or property resulting from any ideas, methods, instructions or products referred to in the content.