

# Comparative Analysis of the Interaction of Cytochrome c with Supported Lipid Films and DNA Aptamers Using QCM-D Method <sup>†</sup>

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<sup>†</sup> Presented at the 3rd International Electronic Conference on Biosensors (IECB), 8-21 May 2023; Available online: <https://iecb2023.sciforum.net>.

**Abstract:** Cytochrome c (cyt c) is important indicator of the cell apoptosis and can be therefore used for diagnosis of cancer. We performed comparative analysis of cyt c detection at surface of lipid films or monolayer of 11-mercaptopundecanoic acid (MUA) with immobilized specific or nonspecific DNA aptamers. The quartz crystal microbalance with dissipation monitoring (QCM-D) in a multi-harmonic mode has been used to study the interaction of cyt c with various surfaces. For this purpose, the changes of resonant frequency,  $\Delta f$ , and dissipation,  $\Delta D$ , were determined. We have shown that the strongest interaction of cyt c has been observed with sensor based on specific DNA aptamers that was accompanied by decrease of frequency and increase of dissipation. The limit of detection (LOD) for this aptasensor has been established as  $2.89 \pm 0.12$  nM. The interaction of cyt c with supported lipid films also resulted in decrease of resonant frequency, but its significant changes occurred only in  $\mu\text{M}$  concentration range of cyt c. Changes in dissipation were much lower in comparison with aptamer-based surfaces, which suggest weaker contribution of cyt c adsorption to the viscosity.

**Keywords:** cytochrome c; biosensor; lipid films; DNA aptamers; QCM-D

**Citation:** Tatarko, M.; Spagnolo, S.; Csiba, M.; Šubjaková, V.; Hianik, T. Comparative Analysis of the Interaction of Cytochrome c with Supported Lipid Films and DNA Aptamers Using QCM-D Method. *Eng. Proc.* **2021**, *3*, x. <https://doi.org/10.3390/xxxxx>

Published: 12 June 2023

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## 1. Introduction.

Cytochrome c (cyt c) belongs to the most essential proteins in living organisms. This relatively small hemoprotein (molecular weight of 12 kDa, diameter 5 nm) has a significant role in electron transport in mitochondria. It is positively charged and revealed redox properties [1], which is often used in electrochemical studies [2]. Cyt c plays also important role in cell apoptosis, which is accompanied by release of cyt c from mitochondrial membrane to the cytoplasm [3,4]. Therefore, detection of cyt c can serve as suitable tool for evaluation of the effectivity of chemotherapy. Monitoring of cell apoptosis during chemotherapy can prevent not desirable damage of the healthy tissues [5,6].

Standard methods of cyt c detection such as flow cytometry, enzyme linked immunosorbent assay (ELISA) and high-performance liquid chromatography (HPLC) are commonly used for cyt c detection [7]. However, these methods are time consuming, require expensive instruments and well-trained personnel. Current trends for cyt c detection are focused on application of biosensors based on DNA aptamers as receptors [8]. Aptamers are relatively short single-stranded RNA or DNA (typically around 30–80 bases) that are selected by combinatorial chemistry known as SELEX (Systematic Evolution of Ligands by EXponential enrichment) [9,10]. Using this method, the DNA aptamers specific to cyt c were developed. However, because cyt c is a positively charged, interaction with the negatively charged DNA aptamers can be also non-specific due to electrostatic binding. DNA aptamers, as an alternative to more expensive and less stable antibodies [7], were



unilamellar liposomes (diameter approximately 20 nm) were prepared by sonication of the lipid solution in a PB. For this purpose, 8 mg mixture of phospholipids (equimolar ratio of DMPC: DMPG) was dissolved in small volume of chloroform in a glass round shaped flask. Chloroform was then evaporated by gentle stream of nitrogen and constant circular motion of flask caused deposition of lipid film on its wall. Aliquot containing PB was then added and incubated for the 30 minutes. Incubation was followed by the 20-minute ultrasonication at the room temperature in the sonicator (Bandelin Sonorex RK31, Berlin, Germany) [15]. Final liposome solution with a concentration of 0.5 mg/mL was then applied by a flow on the surface of DDT-modified gold layer of quartz crystal. MUA layer on a quartz crystal surface was prepared by chemisorption from 2 mM MUA during 16 hours incubation. The crystals were then washed with MiliQ water and incubated for 35 minutes in 20 mM EDC and 50 mM NHS mixture to activate carboxylic group of MUA. The aptamer layers were formed by addition of 1  $\mu$ M aptamers to the surface of MUA activated by NHS/EDC. This resulted in covalent immobilization of the aptamers.

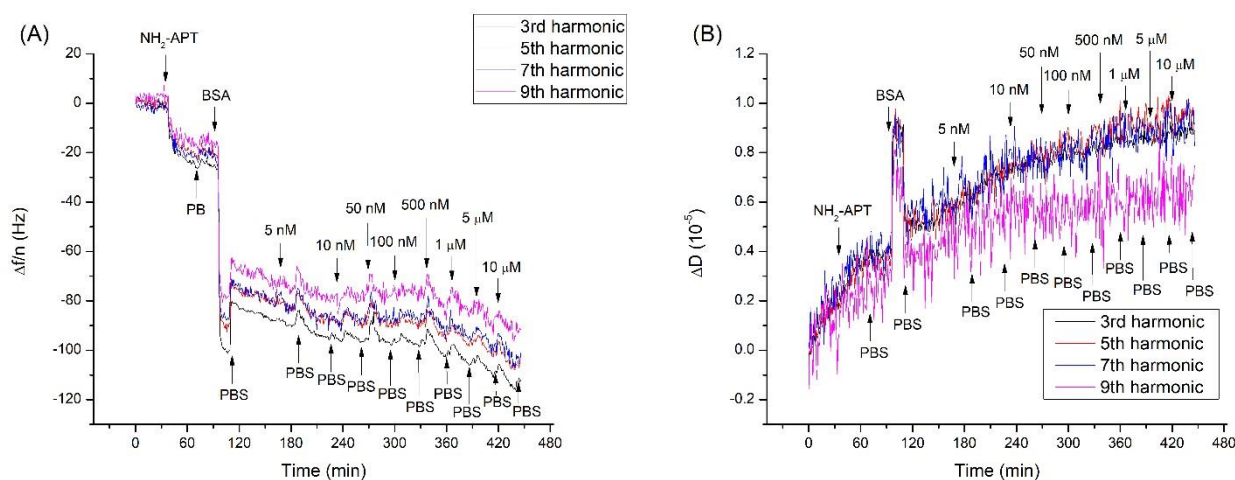
Cyt c has been added in the flow mode (flow rate 50  $\mu$ L/min) to the surface of lipid films or aptamer layers at various concentrations. Changes of resonant frequency,  $\Delta f$ , caused by adsorption of cyt c were evaluated by Sauerbrey equation [17]

$$\Delta f = \frac{-2nf_0^2}{\sqrt{\rho_q \mu_q}} \frac{\Delta m}{A} \quad (1)$$

where  $n$  represents harmonic number,  $f_0$  is fundamental resonance frequency,  $\Delta m$  is mass change,  $A$  is working area of crystal (0.2 cm<sup>2</sup>),  $\mu_q$  is shear modulus of elasticity ( $2.947 \times 10^{11}$  g·cm<sup>-1</sup>·s<sup>-2</sup>), and  $\rho_q$  is density of the crystal (2.648 g·cm<sup>-3</sup>).

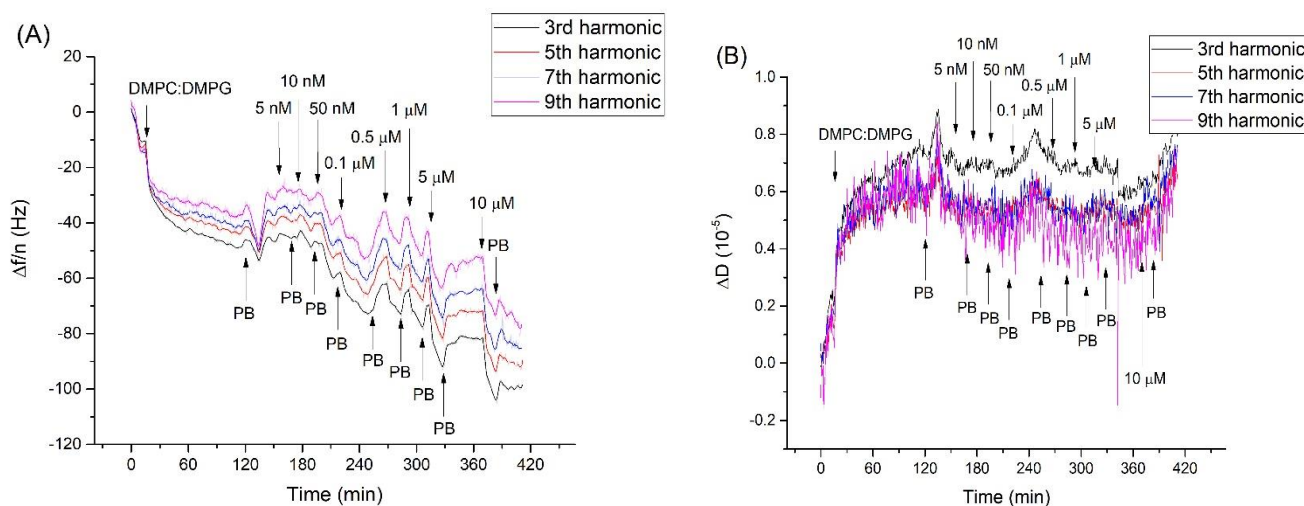
### 3. Results and Discussion

In the first series of experiments, we studied the changes of the frequency,  $\Delta f$ , and dissipation,  $\Delta D$ , following addition of cyt c to the layers formed by NH<sub>2</sub>-aptamers covalently immobilized at MUA layer. Remaining amino-reactive MUA sites were blocked by BSA. The kinetics of the frequency and dissipation changes following addition of BSA and cyt c for 3rd to 9th harmonics are shown on the Figure 1. The decrease of the resonant frequency and increase of dissipation can be seen.



**Figure 1.** The kinetics of the 3rd to 9th harmonic frequencies,  $\Delta f/n$ , normalized by harmonic number (A) and dissipation  $\Delta D$  (B) following addition of specific DNA aptamers (NH<sub>2</sub>-APT), BSA and cyt c in a concentration range 5 nM–10  $\mu$ M on MUA layer chemisorbed on the thin gold layer of piezo-crystal. The moments of addition of aptamers, BSA, cyt c, and PBS wash are shown by arrows.

We also performed similar experiments on supported lipid membranes. In the experiments we used PB contained 2 mM MgCl<sub>2</sub>. The kinetics of the changes for normalized frequency and dissipation following addition of various concentrations of cyt c are shown on Figure 2.

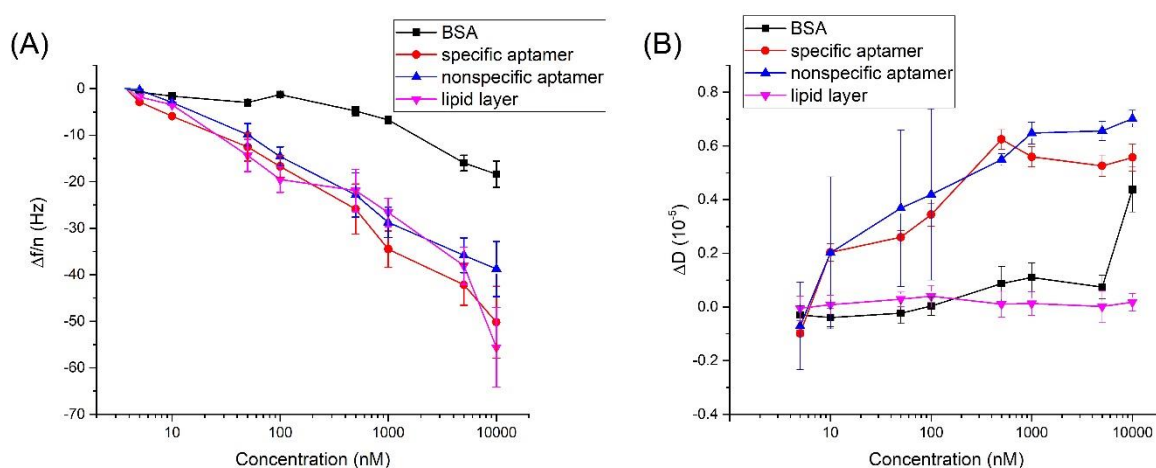


**Figure 2.** The kinetics of the 3rd to 9th harmonic frequencies,  $\Delta f/n$ , normalized by harmonic number (A) and dissipation  $\Delta D$  (B) following addition of DMPC:DMPG liposomes (0.5 mg/ml), cyt c in a concentration range 5 nM–10  $\mu$ M on chemisorbed layer of DDT on the gold surface of the piezo-crystal. The moments of addition of liposomes, cyt c and PB wash are shown by arrows.

As it can be seen, the substantial changes of the acoustic values occurred following addition of cyt c in  $\mu$ M concentrations. We also tested interaction of cyt c with the surface formed by nonspecific DNA aptamers and BSA immobilized at MUA monolayer. Specific and nonspecific interactions of cyt c with the aptamers resulted in similar changes of the acoustics parameters. Much lower changes in frequency and dissipation were observed for interaction of cyt c with BSA covalently immobilized at MUA monolayer. The plot of the changes of frequency (A) and dissipation (B) vs. concentration of cyt c for various surfaces studied are shown on Figure 3. The obtained results suggest that there is the increased adsorption of cyt c on the surface formed by specific DNA aptamers. Cyt c interacted also with surfaces formed by nonspecific aptamers or BSA, but the frequency changes were lower. Estimated limit of detection (LOD) for cyt c by specific aptamers was  $2.89 \pm 0.12$  nM. It is higher than those reported by Poturnayova et al. [18] ( $0.50 \pm 0.05$  nM). However, in this article the sensor has been formed by biotinylated aptamers attached to the neutravidin layer. It can be also seen that with increased concentration of cyt c the dissipation substantially increased for aptamer-based sensors, while practically no changes of this value occurred for lipid layers as well as for BSA. In later case, however, sharp increase of dissipation occurred at rather high cyt c concentration (10  $\mu$ M). Substantial changes of dissipation for interaction of cyt c with aptamer layers can be due to influence of the cyt c on conformation and flexibility of aptamers. At the same time, cyt c at the surface of lipid films forms probably tightly packed rigid protein layer that practically does not contribute to the surface viscosity.

Using Sauerbrey equation, we also compared the surface density of cyt c at different surfaces and at the maximally applied cyt c concentrations, 10  $\mu$ M. Results of calculations are presented in the Table 1. From frequency changes, we firstly calculated surface density and then number of molecules that were adsorbed on the surface. As it can be seen the surface density of cyt c at specific aptamers has been 3.7 times higher than the surface density of these aptamers. Considering that the specific aptamer should have probably one binding site for cyt c, the larger number of adsorbed cyt c molecules can be also due

to non-specific interactions between positively charged cyt c and negatively charged aptamers. It can be also seen that the surface density of cyt c is only slightly higher for specific aptamer in comparison with non-specific one. This is additional confirmation of existence non-specific interactions between cyt c and aptamers. The adsorption of cyt c at the lipid membrane was most extensive and resulted in the cyt c surface density of  $(18.73 \pm 2.88) \times 10^{12}$  molecules.cm<sup>-2</sup>. Considering that changes of dissipation for lipid films was rather small, one can assume that Sauerbrey equation can be more correctly used in comparison with the sensing layers based on aptamers. In later case viscosity contribution on the frequency changes can cause overestimated frequency decrease. This means that the real surface density of cyt c at such surface can be lower in comparison with those showed in Table 1.



**Figure 3.** The plot of the changes of resonant frequency  $\Delta f/n$  (A) and dissipation  $\Delta D$  (B) vs. cyt c concentration for surfaces based on physically adsorbed BSA on a gold surface (black), specific aptamer (red), nonspecific aptamer (blue) and on mixed lipid layer (purple). Measurements were performed in PB buffer. The results are mean  $\pm$  S.D. obtained at least for 3 independent measurements in each system.

**Table 1.** The frequency changes,  $\Delta f$ , mass density,  $\Delta m/A$ , and surface concentrations of molecules,  $\sigma$ , calculated by Sauerbrey equation for the systems studied. For calculation we used molecular weight of BSA (66 kDa), specific aptamer (26.165 kDa) nonspecific aptamer (10.647 kDa) and cyt c (12.327 kDa). The surface density of the molecules has been calculated as mass density  $\times (N_A/Mw)$ , where  $N_A$  is Avogadro's number ( $6.02205 \times 10^{23}$  mol<sup>-1</sup> and  $Mw$  is molecular weight).

Adsorbed molecule	$\Delta f$ (Hz)	$\Delta m/A$ ( $\mu\text{g}\cdot\text{cm}^{-2}$ )	$\sigma$ ( $10^{12}\cdot\text{cm}^{-2}$ )
BSA (on MUA)	$-47.67 \pm 2.4$	$0.33 \pm 0.02$	$3.00 \pm 0.15$
Specific apt. (on MUA)	$-28.75 \pm 4.14$	$0.20 \pm 0.03$	$4.57 \pm 0.66$
Non-specific apt. (on MUA)	$-10.15 \pm 4.40$	$0.07 \pm 0.03$	$3.96 \pm 1.72$
Cyt c (on BSA)	$-18.38 \pm 2.80$	$0.13 \pm 0.19$	$6.20 \pm 0.94$
Cyt c (on specific apt.)	$-50.21 \pm 7.70$	$0.35 \pm 0.05$	$16.92 \pm 2.59$
Cyt c (on non-specific apt.)	$-38.75 \pm 5.96$	$0.27 \pm 0.41$	$13.06 \pm 2.01$
Cyt c (on lipid film)	$-55.58 \pm 8.55$	$0.38 \pm 0.59$	$18.73 \pm 2.88$

#### 4. Conclusion

In this study we demonstrated that cyt c interacts with both specific and non-specific DNA aptamers covalently attached to MUA layers. The surface density was, however, slightly higher for specific aptamers. The fact that cyt c interacts rather strongly with non-specific aptamers is challenging. This phenomenon requires further analysis. The mixed

DMPC/DMPG monolayers also revealed high adsorption of cyt c. This effect can be used for further study of detection of cyt c at lipid surface for example by nanowires modified by DNA aptamers.

**Author Contributions:** Conceptualization T.H.; formal analysis, M.T., S.S. and T.H.; investigation, M.T., S.S., M.C. and V.Š.; methodology, M.T., S.S., V.Š. and T.H.; validation, M.T., M.C. and V.Š.; funding acquisition, T.H.; project administration, T.H.; supervision, T.H.; writing—original draft, M.T., S.S., V.Š. and T.H.; writing—review and editing, M.T. and T.H. All authors have read and agreed to the published version of the manuscript.

**Funding:** This work has received funding from the Science Grant Agency VEGA, project number: 1/0445/23.

**Data Availability Statement:** Not applicable.

**Conflicts of Interest:** The authors declare no conflict of interest.

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