

# Assessing the Likelihood of *Staphylococcus aureus* Contamination in Bottled Drinking Water Production<sup>†</sup>

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**Abstract:** This study's objectives were to evaluate the possibility of *S. aureus* contamination in bottled drinking water and to determine the elements that affected the level of *S. aureus* in raw water. In two drinking water treatment facilities, samples of raw water, soft water, reverse osmosis (R.O.) water, and finished water were taken at various stages. In addition, raw water samples and risk indicators like pH, temperature, and residual chlorine were gathered at the packaging facility during the washing process. For Factory A (small scale), the pH values for the raw water, soft water, R.O. water, and finished water samples were 7.17, 7.24, 6.69, and 5.92, respectively. For Factory B (medium scale), the pH values were 7.9, 7.44, 6.97, and 6.8. All water samples from Factory A (2 CFU/ml) and Factory B (1 CFU/ml) had *S. aureus* concentrations that were within the acceptable range for human consumption. All water samples from Factory A (2-26 CFU/mL) and Factory B (11-316 CFU/mL) contained total coliforms as well. Our study revealed that *S. aureus* contamination in water was mostly caused by the pH and processing time. To prevent pathogen contamination in bottled drinking water, it is recommended that raw and finished water be kept at a pH level between 6.5 and 7.5.

**Keywords:** Risk Estimation; Food Risk Assessment; Sensitivity Analysis

**Citation:** To be added by editorial staff during production.

Academic Editor: Firstname Lastname

Published: date



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## 1. Introduction

About 50–70% of a person's total weight is made up of water, which is essential for maintaining healthy cells, tissues, and organs (EPA, 2018) [1]. The source of drinking water may contain a variety of radionuclides, microorganisms, and chemicals. These dangers may result in gastrointestinal disorders, brain system or reproductive issues, as well as long-term conditions like cancer. Typhoid fever and cholera are examples of bacteria that cause waterborne disease. Diarrhea is one of the most typical negative effects of contaminated water and can be spread by drinking it [2]. One of the most frequent causes of food poisoning, *Staphylococcus aureus*, produces enterotoxin from an initial concentration of 100 organisms per milliliter of water. This pathogen has been found in a variety of places, including human nasal passages, skin, clothing, food, flies'

digestive tracts, dust, and moisture droplets. As a proven source of their origin in drinking water supplies, *S. aureus* can be discovered in rural drinking water. Residents exposed to contaminated water may become colonized by *S. aureus* from drinking water. [3]. Individuals are exposed to avoidable health risks when water and sanitation systems are improperly managed. According to some studies, the prevalence of *Escherichia coli* and *Staphylococcus aureus* was 54.17% and 16.67%, respectively, with an average concentration of 1.04 log CFU/ml and 0.26 log CFU/ml in the drinking water filtration dispenser toll machines water in the Mahasarakham province of Thailand [4]. The characteristics of the contaminants in the water source determine the treatment procedure for disinfecting drinking water to decrease the prevalence of waterborne diseases [1]. Reverse osmosis (R.O.), charcoal filtration, and UV light are used to treat the water. According to the [5], R.O. can eliminate 99% of the germs in tap water as well as organic and inorganic pollutants. Drinking water may become quickly re-contaminated by other factors, particularly poor process hygiene brought on by erratic maintenance [6]. This can result in the contamination of the water with hazardous chemicals and dangerous microbes. In order to establish the high-risk parameters of processing steps as part of risk assessment, probabilistic modeling and sensitivity analysis of crucial control points in food processing and food safety systems are used [7]. Risk assessors can also utilize these models to help them make management decisions that will lower the risk of contracting foodborne illnesses [8]. In order to clarify the current state of the risk parameters for bottled drinking, it is required to apply modeling of the probability of *S. aureus* in terms of risk assessment of drinking water.

## 2. Materials and Methods

### 2.1. Water Sample

#### 2.1.1. Sample collection

The physicochemical and microbiological tests were performed on water samples from one packing house (n = 5) and bottled drinking water (n = 3 at each step) from two factories (Factory A represents a small-scale factory and Factory B represents a medium-scale plant). The bottled drinking water processing facility took samples of raw water, soft water, reverse osmosis water, and finished water. The samples were maintained in iceboxes after sample collection and transported to the lab (Department of Medical, Regional Medical Medical Sciences Center5, Samut Songkhram) within two hours. In triplicate, each sample was taken and examined.

#### 2.2. Physicochemical analysis

The AOAC official technique was used for the physicochemical evaluations of all water samples (AOAC, 2005) [9]. The packing house gathered and evaluated the water sample's pH, residual chlorine (ppm), temperature (°C), and processing time. Using a pH meter (Hanna Instruments, USA), water samples from bottled drinking water were analyzed to determine the pH at each stage.

#### 2.3. Microbiological analysis

All samples were collected for measurement of the quantification and risk of *S. aureus* and total coliforms by following the FDA BAM, 2019 (Chapter 12) [10] and FDA BAM (Appendix 2) [11] procedures.

##### 2.3.1. Plating media method of *S. aureus* and Total Coliforms

A quantity of 25 ml of samples was measured, rinsed in a 250 mL bottle containing 225 mL of sterile 0.1% peptone water, and then the suspension diluted from 1:10–2 to 1:10–4. A volume of 0.1 mL of each dilution was spread on Braid Parker agar plates (BPA) (Merck, Germany) for the enumeration of *S. aureus* and Five tubes should be used for each dilution of the multiple-dilution MPN series used to enumerate of total coliforms

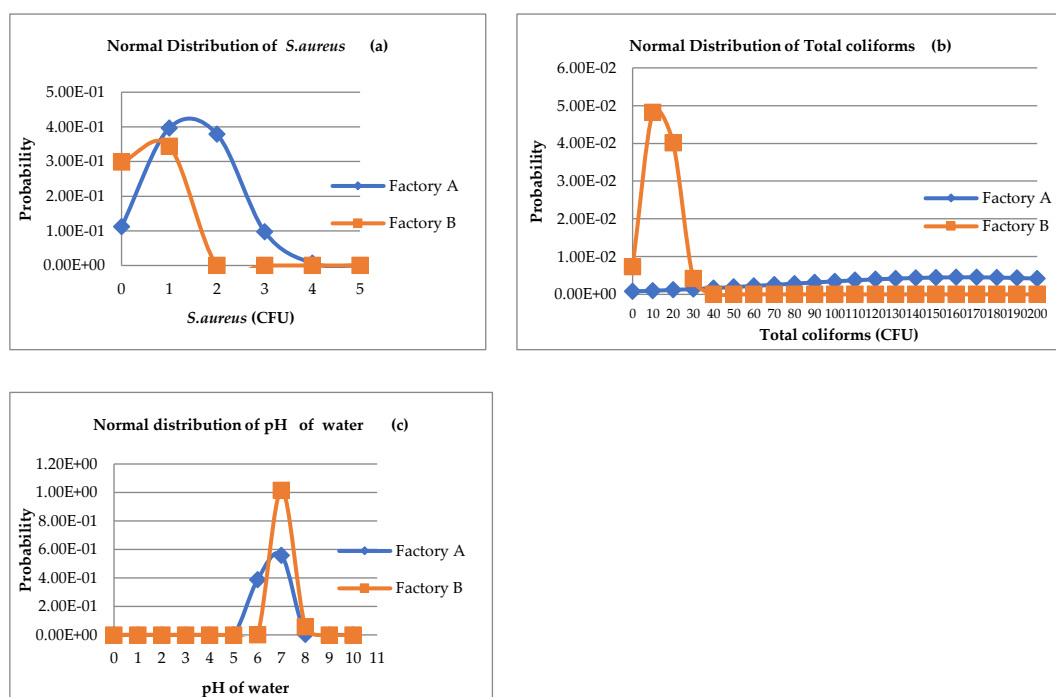


Temperature (°C)	17.3–19.1	-	-	-	-	-	-	-	-
pH	10.33–10.81	7.17±0.25	7.24±0.26	6.69±0.21	5.92±0.32	7.60±0.33	7.44±0.29	6.97±0.22	6.80±0.21
Residual chlorine (ppm)	0.23–1.67	-	-	-	-	-	-	-	-
Total coliform (CFU)	-	189±8.60	316±44.22	30±44.40	145±7.72	26±3.89	2±3.13	20±1.48	13±1.47
<i>S. aureus</i> (CFU)	3.40 LogCFU	3±1.00	1±0.00	1±0.00	1±0.00	1±0.00	1±0.00	1±0.00	1±0.00

Note: The numbers are means±S.D. from three independent replicates.

### 3.2. Probabilistic distribution

The pH of the water and the concentration of *S. aureus* and total coliforms in the water of Plants A and B were simulated to have probability distribution values of Normal(1.5,1.0) and Normal(1,0) and Normal(163.09,89.10) and Normal(13.97,6.70), respectively. Water from Plants A and B had normal pH distributions of 6.75, 0.61 and 7.20, 0.38, respectively. The concentration of *S. aureus* and total coliforms in water samples of factory A and B was in the range of 0-4 CFU and 0-2 CFU and 0-316 CFU and 0-30 CFU, respectively, with the highest probabilities of *S. aureus* and total coliforms contamination in the water of Plant A and Plant B being 0.42 and 0.35 and 4.47E-03 and 0.0514, respectively. The pH values in the water of Plants A and B had the maximum probabilities of 0.6 at pH ranging from 5-8 and 1 at pH ranging from 6-8, respectively.



**Figure 1.** (a) The probability distribution functions of *S. aureus* concentrations in water (b) The probability distribution functions of total coliforms concentrations in water samples. (c) The probability distribution functions of pH values of water samples.

### 3.3. Modeling with multiple regression equations

The sensitivity analysis of water was developed using the equation for multiple regression, and the equations are as follows for the multiple regression of *S. aureus* contamination in water (eq. 3) and *S. aureus* contamination in bottled drinking water (eq. 4):

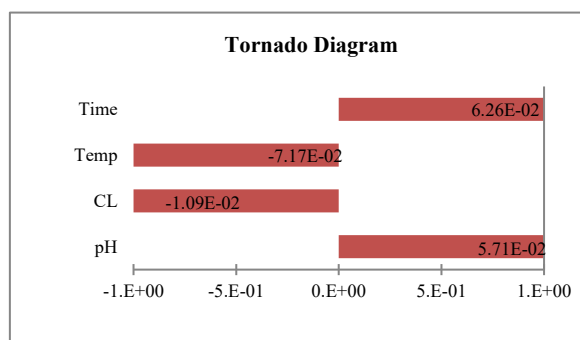
$$Y = -233.9910 + (1.8471 \times \text{Time}) - (9.1343 \times \text{Temperature of Water}) - (5.0746 \times \text{Residual Chlorine in Water}) + (40.5970 \times \text{pH of Water}) \quad (3)$$

$$Y = -0.1260 + (0.1972 \times \text{pH of Water}) \quad (4)$$

The model's input parameters included the water's temperature, pH, residual chlorine content, and temperature. The model's result was the detection of *S. aureus* water contamination.

### 3.4. Sensitivity Analysis

The Excel Platform was used to do the sensitivity analysis. For each input variable, horizontal bars on a tornado chart were drawn, with the length of the bars indicating the strength of the association with the output variable.



**Figure 2.** Sensitivity analysis of the risk factors of *S. aureus* contaminated in water.

The sensitivity analysis results (Figure 2) show that the temperature of the water and residual chlorine concentration were the conditional parameters affecting a decrease in the number of *S. aureus* in the water and an increase in the concentration of *S. aureus* in the water. Water temperature and residual chlorine concentration were the main intervention measures that helped control the effect on the water.

## 4. Discussion

To assess the health risk of drinking water, the study of Wibuloutai and colleagues, which assesses the microbiological quality of drinking water produced by drinking water filtration dispenser toll machines (DFTMs) was compared to this study. 210 samples in total were chosen at random from 70 DFTMs that were spaced 500 meters apart from Mahasarakham University. The DFTM water had high levels of *Escherichia coli* and *Staphylococcus aureus*, with prevalence rates of 54.17% and 16.67% and average values of 1.04 log cfu/ml and 0.26 log cfu/ml, respectively. The risk assessment used the @Risk tool to calculate the likelihood of exposure as 1.67 E-01 and the odds of getting sick from *S. aureus* and *E. coli* as 2.08 E-03 and 1.58, respectively (@Risk is a software tool to generate a distribution of possible outcomes and estimate the probability of different levels of pathogen exposure). [4] It has been established that the findings of our investigation may be safely consumed. The suggested interventions keep the water's pH between 6.5 to 7.5 since doing so could help to lessen the amount of microbial contamination in the water [15]. According to Pratum and Khananthai (2017) [14], the quality of drinking water also depends on the tap water or water source used throughout the manufacturing process. The local government must issue local laws for the control of such businesses in the area of responsibility in accordance with the Ministry of Public Health's notification on hazardous health activities [4].

## 5. Conclusions

This study was successful in identifying the potential risk of *S. aureus* contamination in water as well as the recommended pH levels for processing bottled drinking water at small- and medium-sized facilities. The findings of the sensitivity study revealed that the pH of the water samples played a significant role in the procedure. This result implies that preserving the pH of the water throughout the procedure, as shown by this study, is

a helpful technique for lowering the risk of contamination. However, harmful bacteria have the potential to taint finished goods if the water quality is inadequate.

**Author Contributions:** “Conceptualization, P.C.,N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; methodology, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; software, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; validation, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; formal analysis, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; investigation, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; resources, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; data curation, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; writing original draft preparation, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; writing—review and editing, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; visualization, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; supervision, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; project administration, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; funding acquisition, P.C., N.C.,P.K.,K.W.,P.K.,T.L.,T.L.,P.P., S.W., W.J., S.S.,and S.O.; All authors have read and agreed to the published version of the manuscript.”

**Funding:** “This research received no external funding” or “This research was funded by the Department of Medical Sciences, Regional Medical Sciences Center 5, 7, 11, Ministry of Public Health, Thailand, grant number 006 FM 0117”

**Acknowledgments:** The Department of Medical Sciences at Samut Songkhram (Thailand) is acknowledged. Thank you to Regional Medical Sciences Center, Ministry of Public Health, Thailand for the insightful advice and useful details you supplied for this study. The Department of Medical Sciences, Samut Songkhram Regional Medical Sciences Center, Ministry of Public Health, Thailand, financed this study. & Chiang Mai University's Faculty of Agro-Industry in Chiang Mai, Thailand. Chiang Mai University provided some funding for this study.

**Conflicts of Interest:** “The authors declare no conflict of interest.”

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