

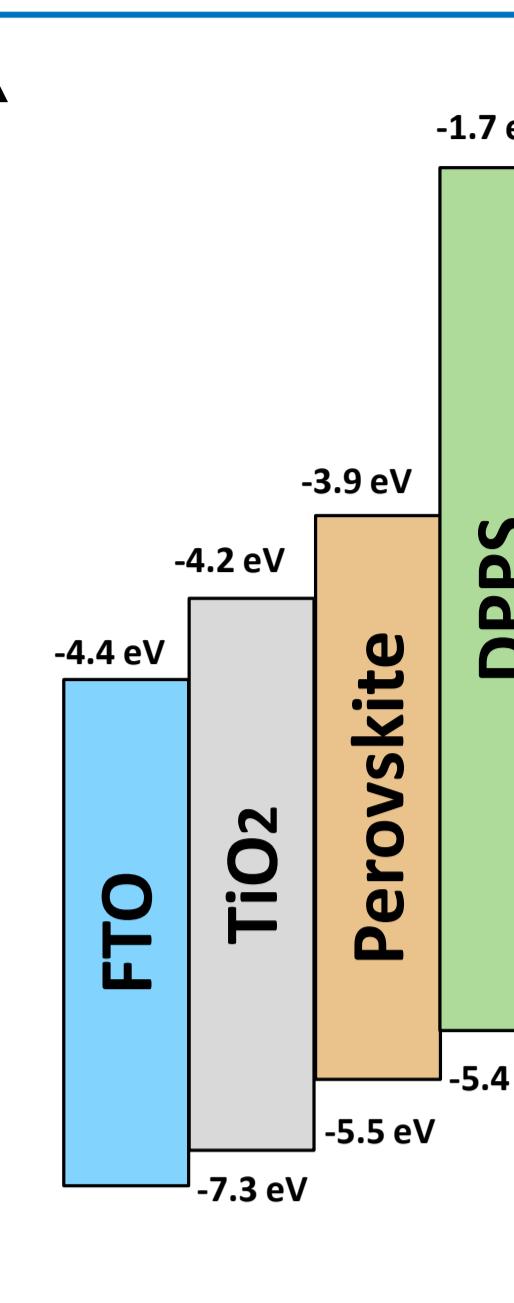
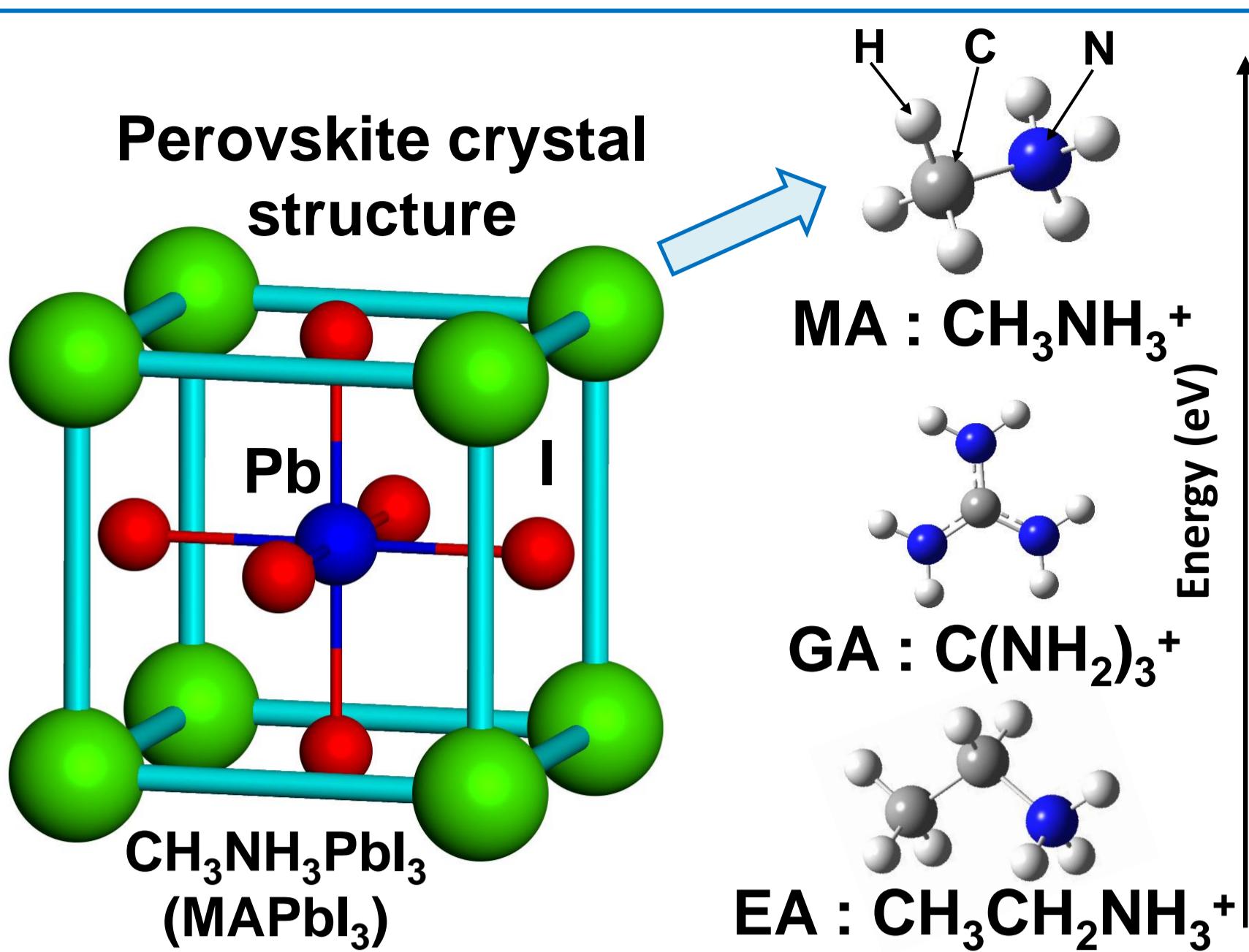
Effects of halogens and alkali metals on guanidinium/ethylammonium doped perovskite photovoltaic devices

THE UNIVERSITY OF SHIGA PREFECTURE ¹The University of Shiga Prefecture, OSAKA GAS CHEMICALS ²Osaka Gas Chemicals Co. Ltd.
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Introduction

Perovskite solar cell

- Advantage
 - Inexpensive and easy
 - High efficiency
 - Wide variety
- Problem
 - Low stability
 - large cell



Tolerance factor *t*

Crystal structure strain and stability indicators.

$$t = \frac{r_A + r_X}{\sqrt{2}(r_B + r_X)} \quad r : \text{Ion radius of each site}$$

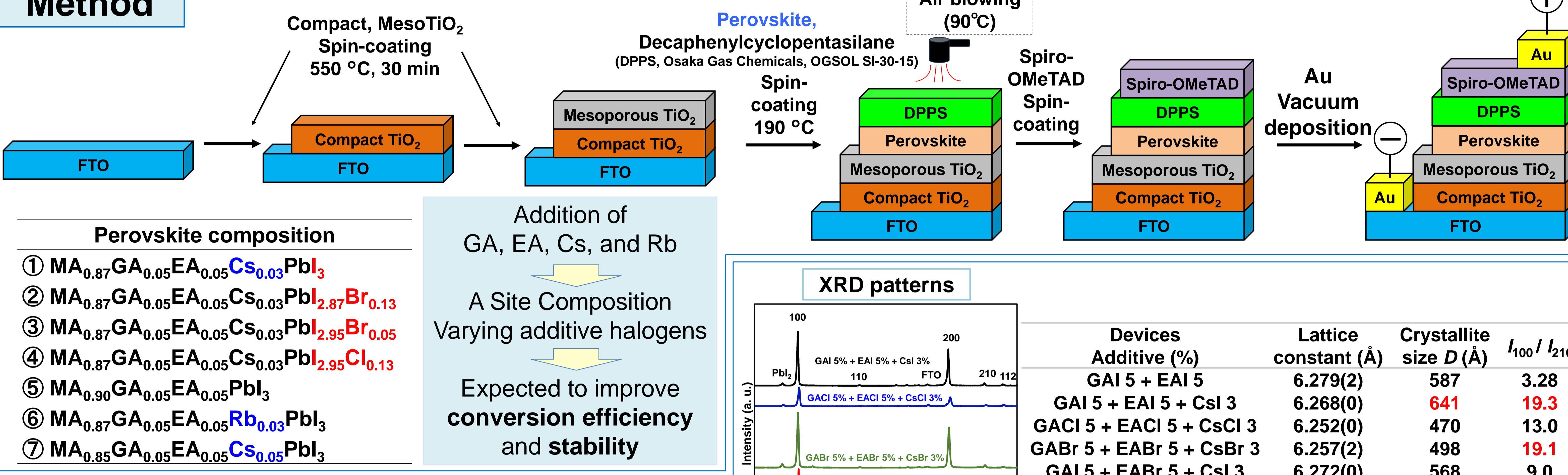
MAPbI_3 : 0.911 → Made this time : 0.920~0.924

A Site	B Site
MA : 2.17 Å	Pb : 1.19 Å
GA : 2.78 Å	X Site
EA : 2.74 Å	I : 2.20 Å
Rb : 1.72 Å	Br : 1.96 Å
Cs : 1.88 Å	Cl : 1.81 Å

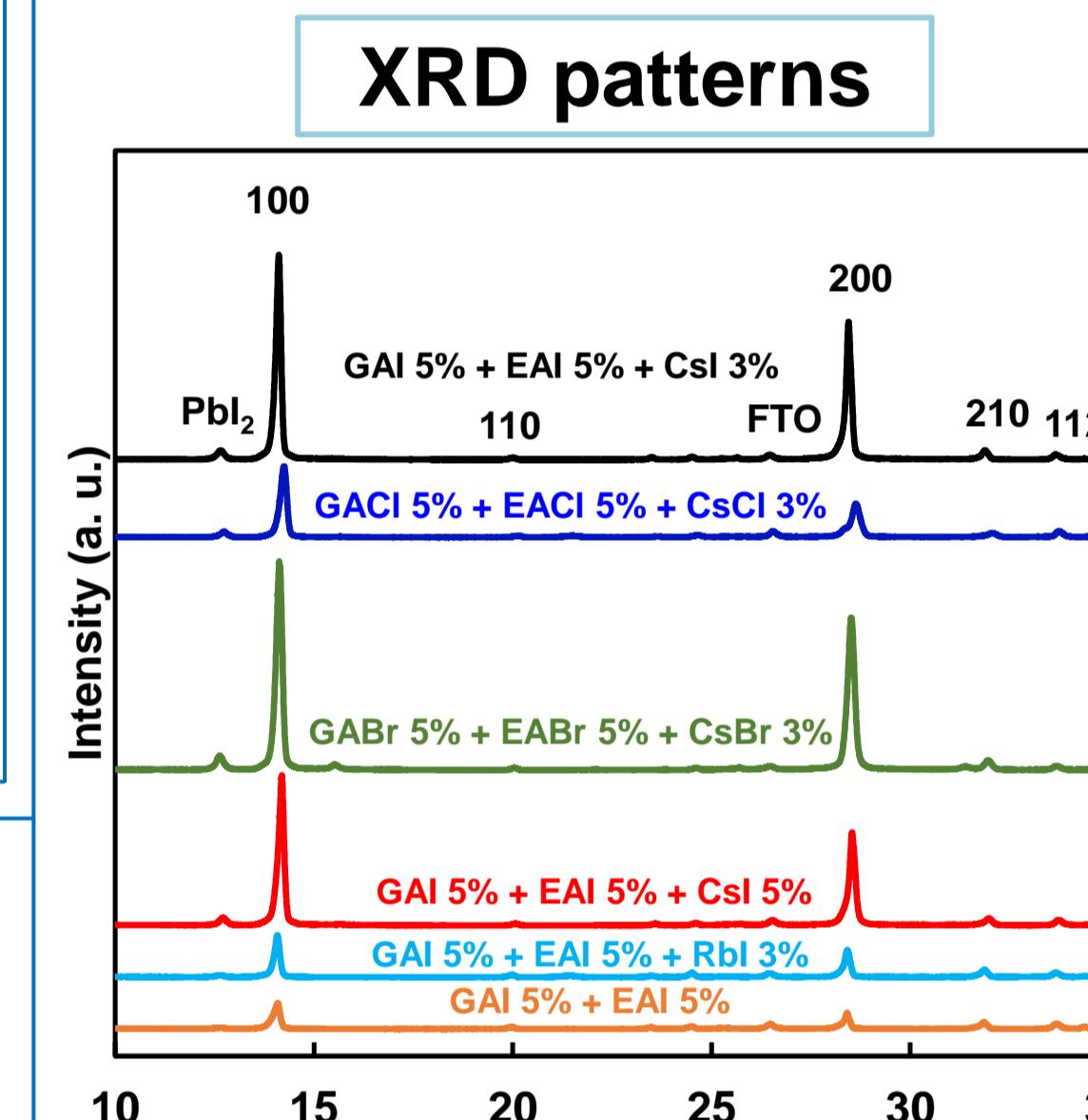
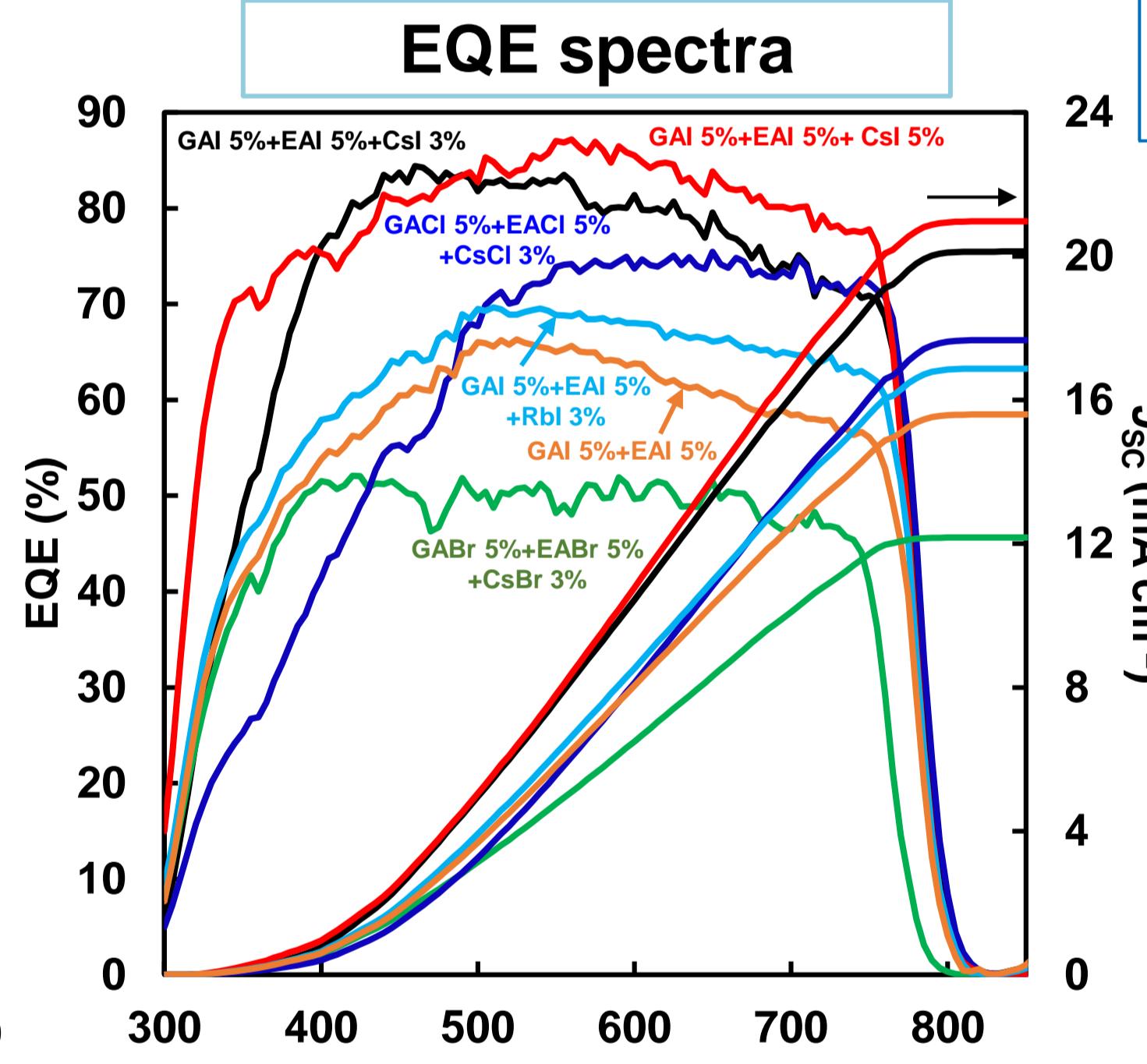
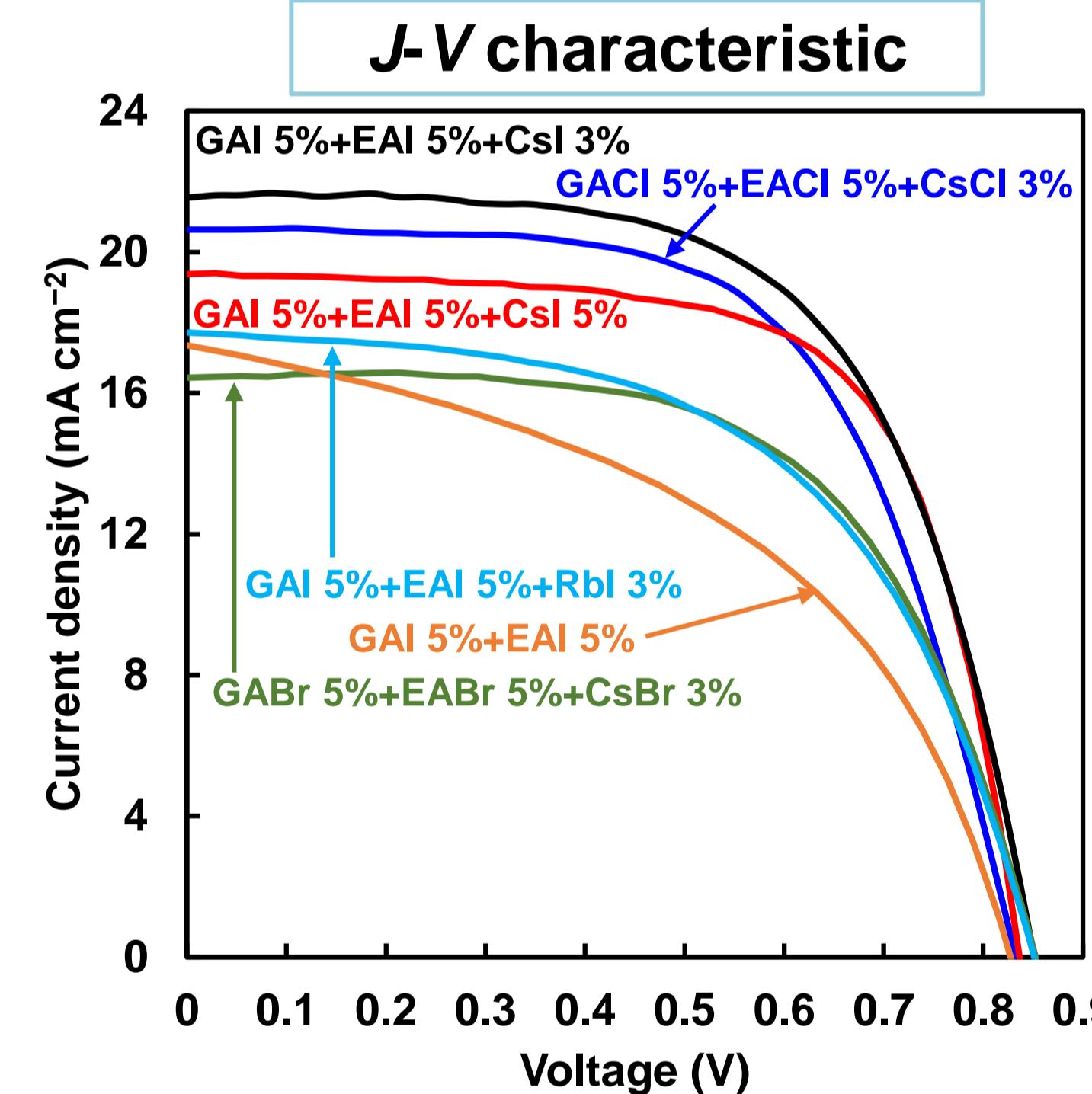
Purpose

- Improved power generation efficiency and durability.
- Fabrication and characterization of Perovskite solar cells with diverse compositions.

Method



Results

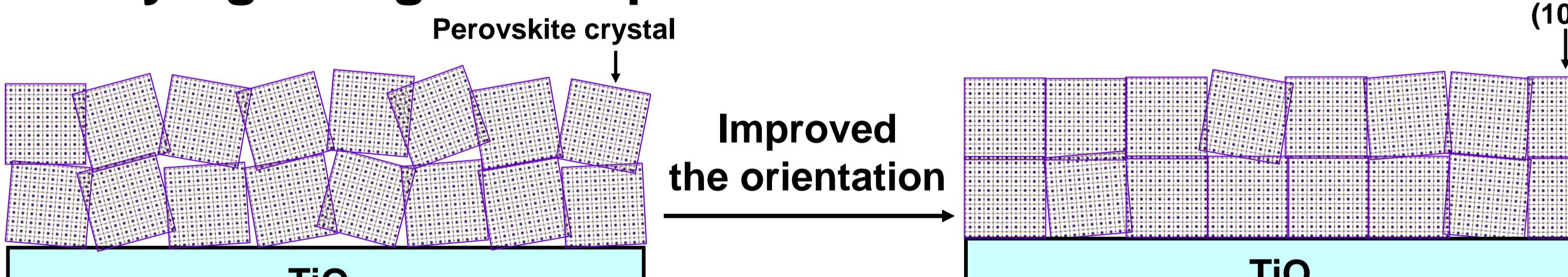


Devices Additive (%)	Lattice constant (Å)	Crystallite size <i>D</i> (Å)	I_{100}/I_{210}
GAI 5 + EAI 5	6.279(2)	587	3.28
GAI 5 + EAI 5 + Csl 3	6.268(0)	641	19.3
GACI 5 + EACI 5 + CsCl 3	6.252(0)	470	13.0
GABr 5 + EABr 5 + CsBr 3	6.257(2)	498	19.1
GAI 5 + EABr 5 + Csl 3	6.272(0)	568	9.0
GAI 5 + EAI 5 + Csl 5	6.265(1)	617	18.8
GAI 5 + EAI 5 + Rbl 3	6.278(1)	504	5.01

Devices Additive (%)	V_{TFL} (V)	Trap Density ($\times 10^{15} \text{ cm}^{-3}$)
GAI 5 + EAI 5	0.57	8.4
GAI 5 + EAI 5 + Csl 3	0.53	7.8
GACI 5 + EACI 5 + CsCl 3	0.54	8.0
GABr 5 + EABr 5 + CsBr 3	0.58	8.6
GAI 5 + EABr 5 + Csl 3	0.55	8.1
GAI 5 + EAI 5 + Csl 5	0.51	7.6
GAI 5 + EAI 5 + Rbl 3	0.51	7.6

$$N_t = \frac{2V_{TFL}\epsilon_0\epsilon}{eL^2} \quad V_{TFL} : \text{Trap fill limit voltage}$$

Varying halogen composition of additives



Improved orientation → decreased grain boundary
(Scattering of electrons by grain boundaries → electrical resistance (R_S))

$$\eta_n = J_{SC}V_{OC}FF, I = I_{SC} - I_0 \left[\exp \left\{ \frac{q(V+R_S I)}{nkT} \right\} - 1 \right] - \frac{V+R_S I}{R_{Sh}}$$

Doping alkali metal cations to A site

Metal ion doped → high conversion efficiency
⇒ Desorption ease : $\text{MA}^+ > \text{Metal cation (Cs}^+, \text{Rb}^+)$
⇒ Stabilization of structure

Improved structural stability and orientation
→ Improved short-circuit current density and conversion efficiency

Cs^+, Rb^+ 3% addition → V_{OC} improvement
(introduced at defects and grain boundaries)

Conclusion

Reference: H. Shimada, T. Oku, I. Ono, R. Okumura, K. Kuroyanagi, A. Suzuki, T. Tachikawa, T. Hasegawa, S. Fukunishi, *Hybrid Advances* 6 (2024) 100252.

Changed halogen composition of additives

- Effects of halogen : I > Cl > Br
- Addition of Br : stability, E_g increase is affected.

→ Effects of ions added in addition to GA / EA : $\text{Cs}^+ > \text{Rb}^+ > \text{none}$
(∴ Ease of detachment : $\text{MA}^+ > \text{Alkali metal cations (Cs}^+, \text{Rb}^+)$)

- Cs addition : 3% is considered more appropriate.