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Machine Learning-Predictive Modelling of Calcium Removal from Cooling Tower Water Using Amberlite IR120 Resins

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INTRODUCTION & AIM

The global increase in industrialization and quick development resulting from the Industrial Revolution has led to a continuous rise in wastewater output.^[1]The methods to remove heavy metals from aqueous solutions are reverse osmosis, ultrafiltration, chemical precipitation, adsorption and ion exchange. The ion exchange process (IEX) effectively reduces heavy metal concentration because it is environmentally friendly, economically viable, selective, and less sludge volume produced.^[2] The buildup of scale in cooling systems, especially evaporative cooling systems, is frequently a significant problem because of calcium (Ca) ions in raw or makeup water. As water evaporates, the concentration of these ions increases, leading to the formation of insoluble salts such as (CaCO₃).^[2]





Figure 1 shows challenges encountered in cooling water treatment. Machine learning provides the best predictive modelling with the highest accuracy, inspired by the brain's autolearning and self-improving capability to solve the study's complicated questions; therefore, it is beneficial for modelling transesterification processes.^[3]

Figure 1. The corrosion-deposition-biofouling triangle [source: ChemTreat]

The present study investigated the removal of Ca²⁺ from cooling tower water using Amberlite IR120 and predictive machine learning approaches. Response surface methodology (RSM), artificial neural network (ANN), and adaptive neuro-fuzzy inference system (ANFIS) were for predictive modeling of calcium removal.

METHOD

The experiments carried out in this study were conducted using Amberjet 1200. These experiments were carried out by varying the set points conditions of process variables as per experimental design in Table 1, noting the output as removal percentages Ca²⁺. Column setup was used as shown in Figure 2.



Figure 7 Actual and Predicted Adsorption Data for ANFIS

high predictive model efficiency was achieved



Figure 8 Comparison of experimental, RSM, ANN, and ANFIS removal percentage yield.

Table 2: Error metric	i ioni the eve				
Evaluation	RSM			evaluate the ef	
metrics		ANN	ANFIS	All the predic	
R ²	0.9777	0.9994	0.9903	rin the predic	
MSE	0.5727	0.017	0.3087	suitable to predi	
RMSE	0.7568	0.1303	0.5556	Ca^{2+} as there w	
MAE	0.5716	0.0677	0.2300		
MAPE	0.6135	0.0744	0.2414	shown in Figure	
٨DE	0.0061	0.0007	0.0024		

From the evaluation metrics used to effectiveness of the models, ictive modelling used are dict the removal efficiency of was not much difference as re 8 and as shown in Table 2.

Variables	-1	0	+1	Ca ²⁺ removal (%)
Contact time (min)	30	75	120	
рН	2	4	7	
Concentration (mg/L)	400	600	800	
Dosage (mL)	50	100	150	
Temperature (K)	273	308	343	

Removal (%)= $\frac{C_i - f}{C_i} \times 100$ (1)

The removal % was calculated using Equation 1, where C_i and C_f are the initial and final concentrations (mg/L), respectively.



Figure 3. The architecture of the ANN model

input variables: contact time, pH, Concentration, Dosage and Temperature. RSM was applied in Design Expert 13, and Neural Network Modular and Neurofuzzy were built with an NN toolbox using MATLAB 2021. 32 experimental data were randomly divided using the dividerand function into 70 % for training and 30% for validation and testing.

output

The ANFIS was generated using a grid partition and trained using a hybrid method; 80% was used for training, and

20% was used for checking.

0.0024 ANN outperformed with high R^2 and low ARE 0.0061 0.000/**MPSD** 11.106 1,3157 4,4689 error metrics.

CONCLUSION

Numerical optimisation yielded an optimal removal percentage of Ca2+ of 99.07% at 89.55 minutes, 4.17, 452.83 mg/L, 132.57 ml and 295.58 K. The developed predictive machine learning model fits the 3 machine learning models with regressions of 0.9777, 0.9994, and 0.9903 for RSM, ANN, and ANFIS, respectively. This study has shown machine learning to be an effective tool for removing Ca2+ from cooling water Amberlite IR120 resins.

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