Photochemistry of 9-vinyl substituted anthracenes, of their reduced derivatives and of Diels Alder type adducts of 9-vinylanthracenes with activated dienophiles

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Introduction



T. Thiemann, C. Thiemann, S. Sasaki, V. Vill, S. Mataka, M. Tashiro, J. Chem. Res., **1997**, 21(S) 248 - 249; **1997**, 21(M) 1736 - 1750

Oligo-/polymerisation of bis(ethenyl)anthracenes

Cross-linking of ethenyl-substituted anthracenes on a solid support

Introduction (continued)

In 2010, A. Marrocchi et al.¹ have studied 9,10-diaryl-, 9,10-diarylethenyl, and 9,10diarylethynylanthracenes as potential semiconducting materials.



In recent times, 9,10-diaryl and 9,10-diarylethenyl substituted anthracenes have been forwarded as potential constituent molecules in light-harvesting devices for organic solar cell applications.²

So, it is of importance to understand whether such 9,10-substituted anthracenes are stable under photoirradiation for a longer time and what type of reactions they could undergo.

References: 1. A. Marrocchi et al., *J. Photochem. Photobiol.* **2010**, *211*, 162. 2. N. Belghiti et al., *Mat. Environ. Sci.* **2014**, *5*, 2191



Scheme 1. Possible photodimerisation products of anthranylacrylates



Scheme 2. Possible competing [4+4]-photodimerisation of anthranylacrylates



Preparation of the phosphorane reagents



Scheme 3. Synthesis of 9-aroylethenylanthracenes 8 and anthranylacrylates 2





Scheme 4. Bromination 9-substituted anthracenes 2a and 5



X = 4-CH₃O, H, 3-Cl, 3-NO₂, 4-C₂₂H₄₅O

Scheme 5. Suzuki cross-coupling reactions to extended anthracenes



UV band maxima of selected compounds in CH_2CI_2 and CH_3CN .

Fig. 1



Fig 2. Crystal packing of ethyl 3-(anthran-9-yl)acrylate showing intermolecular distances that are too large for photochemical dimerisation reactions in the crystal.



Fig 3. Crystal packing of ethyl 3-(9-bromoanthran-10-yl)acrylate showing intermolecular distances that are too large for photochemical dimerisation reactions in the crystal.



Scheme 6. Photochemical *cis-trans* isomerisation of 9-aroylethenylanthracenes

See also: H.-D. Becker, H. C. Becker, K. Sandros, K. Andersson, *Tetrahedron Lett.*, 1985, 26, 1589.
H. Bouas-Laurent, A. Castellan, J. P. Desvergne, and R. Lapouyade, *Chem. Soc. Rev.*, 2000, 29, 43 (review).
H. Bouas-Laurent, A. Castellan, J. P. Desvergne, and R. Lapouyade, *Chem. Soc. Rev.*, 2001, 30, 248 (review).



Scheme 7.

Slow build-up of a [2+2] photodimer in CH_2CI_2







Scheme 7. Plan to produce [2+2]-cycloadducts from the ethenylanthracenes: First "protect" the anthranyl unit in a Diels-Alder reaction, then carry out the photochemistry, then "deprotect" the anthranyl unit in a retro-Diels-Alder reaction.



Scheme 8. Facile Diels-Alder reactions of 9-substituted anthracenes with doubly activated dienophiles.

Fig 4. Spatial distribution and energies of HOMO – LUMO for representative anthranylacrylates and phenacyl and acetylethenylanthracenes. It can be seen that the HOMOs and the LUMOs, respectively, for the compounds are similar in energy and spatial distribution. All calculations were performed using the Gaussian 09W C01 program. Geometry optimizations for the each molecule were performed without symmetry constraints in the gas-phase using the B3LYP hybrid functional and 6-31G(d) basis set. Optimized structures were confirmed to be energy minima through vibrational frequency calculations Orbital energies and isosurfaces were calculated using the same functional and basis set.

	LUMO (eV)	HOMO (eV)
Str1	-2.17	-5.39
Str2	-2.05	-5.32
Str3	-2.02	-5.25
Str4	-2.12	-5.33



Scheme 9. *E-/Z*-isomerisation of **16** upon photoirradiation at λ = 254 nm. Arrows indicate disappearing ¹H and ¹³C NMR peaks due extreme broadening.



Scheme 10. Known photodimerisation of 9-formylanthracene in CH_2Cl_2 at $\lambda = 352$ nm.



Scheme 11. Hydrogenation of anthranyl acrylates and aroylethenylanthracenes with NaBH₄, AcOH, Pd/C



Scheme 12. Photodimerisation of ethyl anthranylpropionate (18-Et) at $\lambda = 352$ nm



Scheme 13. Photodimerisation of 4-methoxybenzoylethylanthracene (18-Ar-OMe) at λ = 352 nm see also:

H. Bouas-Laurent, A. Castellan, J. P. Desvergne, and R. Lapouyade,, Chem. Soc. Rev., 2000, 29, 43 (review).

H. Bouas-Laurent, A. Castellan, J. P. Desvergne, and R. Lapouyade, Chem. Soc. Rev., 2001, 30, 248 (review).



A view of the molecular structure of the molecule **19-Ar-OMe**, with non-hydrogen atom labelling. Displacement ellipsoids are shown at the 50% probability level. Disorder is clear at only one methoxyphenyl group.

Conclusions:

1.) A larger number of aroyl and ester substituted 9-ethenylanthracenes could be synthesized via solventless Wittig-olefination.



- 2.) X-ray crystal structures of two of them showed the molecules too far apart in the crystal for them to photodimerize.
- 3.) The keto substituted 9-ethenylanthracenes undergo photochemical E-/Z-isomerisation at λ = 254 nm, the ester substituted molecules undergo a slow polymerisation. The 9-aroylethenylanthracenes undergo a slow [2+2]-cycloaddition, but in very low yield.
- 4.) 9-Ethenylanthracenes were submitted facilely to Diels- Alder reactions with maleimides. The photoreaction of the cycloadducts, which still carry an α , β -enone or an α , β -unsaturated ester functionality, leads mainly to rapid *E*/*Z*-isomerisation.
- 5.) When photoirradiated at λ = 352 nm, 9-ketoethylanthracenes (ie., the hydrogenated 9ethenylanthracenes) undergo [4+4]-cycloaddition with the formation of head-to-tail cycloadducts.
- 6.) 9-Ethenylanthracenes are photoreactive under UV-irradiation. This can lead to aging of such ethenylanthracene containing material and should be taken into account when using these in organic solar cells.



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Thank you very much for your attention!