On the Mechanism of the Acid-Catalyzed Stereoselective Chroman Cyclization Reaction

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Naturally occurring tocopherols and tocotrienols are single-isomer vitamin E compounds. $(2R,4'R,8'R)-\alpha$ -Tocopherol (4) as a prominent example is of high commercial interest due to its biological and antioxidant properties.^[1] Although the stereospecific cyclization reaction of intermediates and precursors such as **1a/2a** to chromans **3/4** under carefully controlled acidic conditions (Scheme 1) is known for a long time^[2,3] and has been used as a key step in many total syntheses,^[1,4] the mechanism of this transformation is unknown.



Scheme 1. Stereospecific acid-catalyzed chroman ring formation.

Several proposals are documented in literature. The Roche Nutley research group favoured a chroman ring formation via a hemiketal followed by rearomatization (Scheme 2) or, alternatively, a cyclization via a redox process (Scheme 3).^[3] Chroman ring formation by double inversion (Schemes 4 and 5) was also postulated.^[5,6]



Scheme 2. Chroman ring formation via hemiketal.^[3]



Scheme 3. Chroman ring formation via redox cyclization.^[3]



Scheme 4. Proposed chroman ring formation by double inversion.^[5]



Scheme 5. Proposed chroman ring formation via epoxide.^[6]

We investigated the course of the acid catalyzed ring closure reaction by starting from doubly ¹⁸O-labelled derivative **1b** (synthesized via stereoselective bishydroxylation). Chromans **3** and **4** (via intermediate **2b**) obtained by applying standard literature procedures^[4,7] showed complete (>95%) chirality transfer as well as ¹⁸O-incorporation (Scheme 6). Loss of the tertiary hydroxy group and double-inversion as sketched in Schemes 4 and 5 can be ruled out.





The results of this study corroborate the mechanistic pathway of this key reaction applied in various total syntheses of optically active vitamin E compounds such as (2R,4'R,8'R)- α -tocopherol (4) and other stereoisomers.

References:

[1] T. Netscher, *Vitam. Horm.* 2007, *76*, 155-202; M. Eggersdorfer, D. Laudert, U. Létinois, T. McClymont, J. Medlock, T. Netscher, W. Bonrath, *Angew. Chem. Int. Ed.* 2012, *51*, 12960-12990. [2] H. Mayer, W. Vetter, J. Metzger, R. Rüegg, O. Isler, *Helv. Chim. Acta* 1963, *50*, 1168-1178. [3] N. Cohen, R. J. Lopresti, C. Neukom, *J. Org. Chem.* 1981, *46*, 2445-2450. [4] See e.g. C. Rein, P. Demel, R. A. Outten, T. Netscher, B. Breit, *Angew. Chem. Int. Ed.* 2007, *46*, 8670-8673, and references cited therein. [5] E. Mizuguchi, K. Achiwa, *Synlett* 1995, 1255-1256. [6] H.C. Shen, *Tetrahedron* 2009, *65*, 3931-3952. [7] J. Hübscher, R. Barner, *Helv. Chim. Acta* 1990, *73*, 1068-1086.