



Development of A Gas Sensor for *Eucalyptol* **Supervision: Supporting Tool for Extreme Wildfire Management** ⁺

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Abstract: Recent research on volatile organic compounds (VOC) released by the heated vegetation has shown that, under specific conditions (e.g., extreme heat, humidity, wind, and topography), VOC might foster wildfire ignition sources and explain sudden changes in fire behavior, particularly in the most susceptible and flammable forests (eucalypt forests). This work aims to develop an electronic nose (e-nose) based on a sensor's array to monitor the concentration of eucalyptol, the major VOC compound of the *Eucalyptus globulus* tree. The detection of this target compound was achieved by measuring the impedance spectra of layer-by-layer developed thin films based on polyethyleneimine, poly(allylamine hydrochloride), and graphene oxide , by injecting in a custommade vacuum chamber system the analyte. The obtained results were analyzed by principal component analysis method. The developed e-nose sensor was able to distinguish different concentrations in a range from 411 to 1095 ppm.

Keywords: wildfires; volatile organic compounds; eucalyptol; electronic nose; impedance spectroscopy

1. Introduction

Extreme wildfires cause the loss of various human lives and have a significant impact 28 on the biodiversity of ecosystems. These phenomena are, still, not yet fully understood. 29 Recent studies have proposed a new theory, suggesting that flammable gases generated 30 from heated vegetation, in particular, Volatile Organic Compounds (VOC) common in 31 Mediterranean plants may, under some topographic and wind conditions, accumulate in 32 locations where, after the arrival of the ignition source, it rapidly burst in flames as in 33 explosions [1–2]. VOC can exhibit a flammable nature, enabling fire ignition sources and 34 sudden changes in the fire behavior[3] 35

The electronic nose (e-nose) system, which comprise an array of sensors with partial 36 specificity and an appropriate pattern recognition system, can recognize complex gases. 37 Moreover, the e-noses have shown favorable efficiency in monitoring applications, mak-38 ing it a potential tool for the study of VOC [4-6]. Although the e-nose exhibit high sensi-39 tivity, the forest environments consist of a complex mixture of gases and therefore the 40 used sensors must be able to detect, classify and quantify the target compound. To im-41 prove the sensor's sensitivity, different materials can be used as coatings, and thus en-42 hancing the chemical and physical properties of the sensor [7–9]. 43

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The purpose of this work was the development of a custom-made measuring system 44 attached to an e-nose system to monitor eucalyptol in a range of concentrations, from 411 45 to 1095 ppm. Thus, different layer-by-layer (LbL) thin films were developed in order to 46 attain the best combinations for the monitoring of the target compound. The thin films 47 were produced with polyethyleneimine (PEI), poly(allylamine hydrochloride) (PAH), and 48 graphene oxide (GO), namely, (PEI/GO) and (PAH/GO). These films have been already 49 described in [10–12] and, thus, may have an interesting potential to monitor VOC, namely, 50 eucalyptol. 51

2. Materials and Methods

The developed e-nose consists of an array of sensing devices based on ceramic solid supports with deposited gold interdigitated electrodes (IDE), comprising eight "fingers" each, with dimensions of $22.8 \times 7.6 \times 0.7$ mm and each "finger" has 200 µm of width. These solid supports were acquired from DropSens (Oviedo, Asturias, Spain) [13].

The thin films were deposited on the sensors IDE by the layer-by-layer (LbL) technique, which consists of the alternate deposition of polyelectrolytes layers with opposite electrical charges, to obtain several bilayers. The polyelectrolytes used to built-up the thin films layers were polyethyleneimine (PEI), poly(allylamine hydrochloride) (PAH), and graphene oxide (GO), all purchased from Sigma-Aldrich, Steinheim, Germany. The aqueous solutions of the polyelectrolytes were prepared with a 10⁻²M concentration of each polyelectrolyte. Each of the aqueous solutions were prepared with ultrapure water, obtained in a Milli-Q ultrapure water system (Millipore GmbH, Billerica, MA, USA). This process was carried out by alternated adsorption of the positive PEI or PAH polyelectrolytes and the negatively charged GO molecules.

After each adsorption of the polyelectrolyte layers, the solid support was immersed in water in order to remove any polyelectrolyte molecules that were not completely adsorbed. The immersion time, in which the adsorption of the molecules takes place, was of 60 s for each of the polyelectrolytes used and 30s for the washing process. After the adsorption of each bilayer, the thin film was dried using nitrogen gas stream (99% purity, Air Liquide, Algés, Portugal). Thus, thin films of (PEI/GO) and (PAH/GO), with 5 bilayers each, (PEI/GO)⁵ and (PAH/GO)⁵, were produced.

The eucalyptol (99%) used for the experiments was purchased from Sigma-Aldrich, Steinheim, Germany. To test the response of the sensor when exposed to the eucalyptol, a range of concentrations from 411 to 1095 ppm was evaluated.

The measurements of each sensor were performed inside a custom-made vacuum chamber system designed by the author's team, which is depicted in Figure 1.

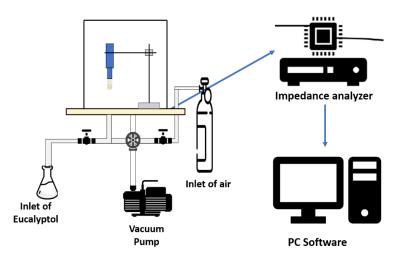


Figure 1. Schematic illustration of the experimental setup, including the custom-made vacuum chamber.

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1.00E+10

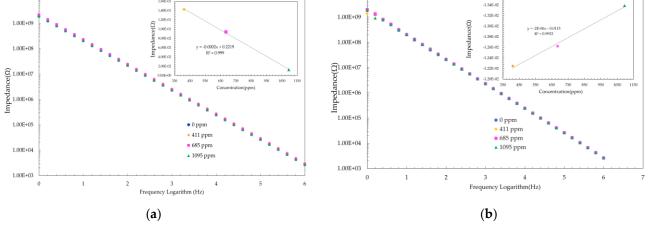
A sample holder containing the sensor was placed inside the chamber which presents 82 an approximate volume of 58 L. A rotatory vacuum pump was also connected so that 83 primary vacuum could be achieved and maintained inside the chamber during the tests, 84 enabling a "clean" environment for the measurements. The sample holder was connected 85 to Solartron 1260 Impedance Analyzer (Solartron Analytical, AMETEK scientific instru-86 ments, Berwyn, PA, USA), in order to measure the impedance spectra at the IDE terminals. 87 The chamber also has two inputs for the inlet of the eucalyptol and compressed air in the 88 chamber, allowing the interaction between the target compound and the sensor. 89

In order to start the testing process, the sample holder with the sensor was placed inside the chamber and connected to the Impedance Analyzer. After that, the vacuum pump was switched on to achieve a pressure of 1.3×10^{-3} mbar. Following this, the eucalyptol would be evaporated into the chamber by the opening of a needle valve that connects the chamber with the mixture of eucalyptol. Subsequently, and after the evaporation of the compound and reaching a certain pressure level, the inlet was open to inject the compressed air until the pressure of 1.3×10^{-3} mbar was attained. Afterward, the electrical measurements were conducted in a frequency range of 1 Hz to 1 MHz, and an AC signal voltage of 25 mV. This process was then repeated for each eucalyptol concentration.

The electrical impedance spectra data features were treated by the Principal Component Analysis (PCA) method to reduce the data size and to obtain a new space of orthogonal components in which different concentration patterns can be observed. The principal component analysis (PCA) plots were obtained by performing the normalization (Z-Score normalization (value - μ)/ ϑ , being μ and ϑ , the mean value and the standard deviation of the samples, respectively) of the impedance spectroscopy data.

3. Results and Discussion

Figure 2 shows the electric impedance spectra of the sensors, coated with thin films 106 of (PEI/GO)⁵ and (PAH/GO)⁵, when in contact with different concentrations of eucalyptol, 107 for different frequencies. 108



1.00E+10

Figure 2. Impedance spectra measured with (PEI/GO)⁵ (a) and (PAH/GO)⁵ (b) LbL films inside of the chamber filled with different eucalyptol concentrations. In the insets are shown the evolution of the impedance at 63 and 10kHz as a function of concentration, respectively.

In order to have a more thorough analysis, a normalization was performed on the data, using the following equation [12]: 113

$$\frac{PP(C) - PP(0ppm)}{PP(0ppm)} \tag{1}$$

where PP(C) corresponds to a physical property at a given eucalytol concentration, and 114 PP(0ppm) to a measure at a reference concentration, in this case when there was 0 ppm of 115

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eucalyptol. In each of the cases represented in the insets of Figure 2 (a) and (b) the physical 116 property used was the impedance, at a constant frequency, where the effects of the euca-117 lyptol are better represented. In both cases it may be observed that an increase in the 118 concentration of eucalyptol results in the decrease of the sensors impedance values. 119

Furthermore, a preliminary analysis of the electronic nose concept was performed by 120 mean of the Principal Component Analysis (PCA). Thus, the data of each one of the sen-121 sors, (PEI/GO)5 and (PAH/GO)5, when in the presence of different tested concentrations 122 of eucalyptol was plotted, as shown in Figure 3. 123

15 1095 ppm 10 685 ppm 5 F2 (35.81 %) 0 -5 💧 0 ppm -10 411 ppm -15 5 -10 -5 10 -20 -15 0 15 20 25 F1 (52.19%)

Figure 3. PCA plot for both sensors used with thin films of (PEI/GO)5 and (PAH/GO)5, for a range of eucalyptol from 0 to 1095 ppm.

By the PCA analysis it can be observed that the e-nose assembled, with two sensors, 128 each one coated with thin films of (PAH/GO)5 and (PEI/GO)5, can distinguish between the 129 blank and the different concentrations of eucalyptol that they interacted with. 130

In fact, the choice of the GO as the upper layer seems to increase the efficiency and 131 discrimination of the measurements, as the molecules of the thin film may react with the 132 target compound enabling a better adsorption, since it possesses many functional groups. 133

The electronic nose, consisting of an array of two sensors coated with (PEI/GO)5 and (PAH/GO)⁵ thin films, built-up with the LbL technique, was able to detect eucalyptol and distinguish three different concentrations levels, as the PCA technique has shown. The use of graphene oxide as thin film bilayer increased the interaction between the custommade chamber's head space and the thin film coating, and consequently the efficiency of the impedance measurements and capability of the e-nose to distinguish different eucalyptol concentration.

The present work presents a novel and preliminary study, with an in-deep effort in the built-up of the custom-made chamber. It should be also noted that, this project is under developing, thus, it is expected the built-up of different coatings, e.g., different combination of polyelectrolytes and/or other sensing materials, to improve the e-nose performance.

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