# MOLECULAR EMITTERS AS A TUNABLE LIGHT SOURCE FOR OPTICAL MULTISENSOR SYSTEMS

<u>Anastasiia Surkova</u><sup>1,2</sup>, Aleksandra Paderina<sup>1</sup>, Andrey Legin<sup>1</sup>, Elena Grachova<sup>1</sup>, Dmitry Kirsanov<sup>1</sup>

<sup>1</sup> Institute of Chemistry, St. Petersburg State University, St. Petersburg, Russia
<sup>2</sup> Samara State Technical University, Samara, Russia

# **OPTICAL MULTISENSOR SYSTEM (OMS)**

- <u>Multisensor system</u> an analytical device
  - composed of two or more sensors
  - optimized for a particular analytical task
  - employs chemometrics to maintain accuracy
- OMS composed of
  - light sources
  - photodetectors
  - optical fibers
  - 3D-printed parts, etc.

- Advantages
  - inexpensive
  - portable
  - on-site/on-line implementation



## PURPOSE OF THIS STUDY

- To develop OMS prototype based on molecular emitters as a tunable light source
- To study a real-life applicability of the developed prototype

### **MOLECULAR EMITTERS AS A LIGHT SOURCE**

- Properties
  - laser diode excites the emission of molecular emitters
- Requirements to choose emitters
  - absorption spectrum must not overlap with emission spectrum
  - excitation radiation must fit into the absorption maximum
  - brightness (quantum yield of emission)
  - emission wavelength in solid phase
- Advantages
  - high versatility
  - short analysis time
  - adjustment of the emission wavelength for a specific application

#### 5

Cu(I) complexes



[Cu(MePPy<sub>3</sub>)I]<sub>2</sub>[Cu<sub>2</sub>I<sub>4</sub>] (1) [Cu<sub>4</sub>I<sub>4</sub>(py)<sub>4</sub>] (2) [Cu(Tpdp)I] (3) [CuCl(PPh<sub>3</sub>)<sub>2</sub>(py)] (4) [Cu(PPh<sub>3</sub>)<sub>3</sub>(4-Mepy)]Br (5) [Cul(PPh<sub>3</sub>)<sub>2</sub>(4-Mepy)] (6)

Ir(III) complexes



 $[Ir(dfppy)_{2}(bpbpy)]PF_{6} (1)$  $[Ir(ppy)_{2}(bpbpy)]PF_{6} (2)$  $[Ir(pybt)_{2}(bpbpy)]PF_{6} (3)$  $[Ir(mpqc)_{2}(bpbpy)]PF_{6} (4)$ 

### **EXAMPLES OF MOLECULAR EMITTERS**

#### **MOLECULAR EMITTERS AS A LIGHT SOURCE**



### **EXPERIMENTAL**



	(1) Ir(III)-based OMS <sup>a</sup>	(2) Cu(I)-based OMS
Sample volume	0.15 mL	4 mL
Sample placement	glass cup (1 cm in diameter)	polystyrene Petri dish (3.5 cm in diameter)
Excitation source	laser diode ( $\lambda_{exct}$ = 365 nm)	laser diode ( $\lambda_{exct}$ = 385 nm) / UV flashlight ( $\lambda_{exct}$ = 365 nm)
Detector	fiber-optic UV-vis spectrometer AvaSpec-ULS2048CL-EVO	

<sup>a</sup> Gitlina A.Y., Surkova A., Ivonina M.V., et al. Dyes Pigments. 2020; 180:108428. doi: 10.1016/j.dyepig.2020.108428

## **RESULTS FOR MODEL SOLUTIONS**

Individual calibration series for Co(II) and Cu(II) nitrates



excitation source: a laser diode,  $\lambda_{exct}$  = 365 nm; b laser diode,  $\lambda_{exct}$  = 385 nm / UV flashlight,  $\lambda_{exct}$  = 365 nm

## **PRACTICAL APPLICATION**

- Calibration sample set for  $PO_4^{3-}$ 
  - 9 samples
  - 0-0.96 mg/L with 0.12 mg/L step



- Calibration sample set for F<sup>-</sup>
  - 11 samples
  - 0-0.4 mg/L with 0.04 mg/L step



 Test set: 5 samples from tap, rivers (Neva, Volga, Tatyanka) and lake Kaban

Samples were colored in accordance with the procedures described in GOST 18309-2014 and GOST 4386-89.

## **PRACTICAL APPLICATION**

Determination of fluoride and phosphate in surface water



excitation source: UV flashlight ( $\lambda_{exct}$  = 365 nm); interval: <sup>a</sup> 600–750 nm; <sup>b</sup> 450–800 nm

# CONCLUSIONS

- Ir(III) and Cu(I)-based complexes are suitable light sources for OMS
- Ir(III) luminescent complexes have bright controlled emission
- Cu(I) complexes are easier to produce, cheaper, and environmentally friendly
- Using UV flashlight instead of laser diode as excitation source is more convenient
- Developed OMS allows determination of fluoride and phosphate in surface waters with high accuracy

# ACKNOWLEDGEMENTS

- Saint Petersburg State University
  - PostDoc program
- RSF project #19-79-00076
- Samara State Technical University
- CSAC2021 organizers

