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# Proceedings Fabrication and Characterization of PVA/PEO/CB Nanocomposite Films

Birhanu Alemayehu <sup>1\*</sup>, Akash Kota <sup>1</sup>, Lenin Wung Kum <sup>1</sup>, Amy T. Neidhard-Doll <sup>1</sup>, Vamsy Chodavarapu <sup>1</sup>, and Guru Subramanyam <sup>1</sup>

- <sup>1</sup> Department of Electrical and Computer Engineering, University of Dayton, Dayton, OH, USA; alemayehub2@udayton.edu, kotaa1@udayton.edu, leninwungk1@udayton.edu, aneidharddoll1@udayton.edu, vchodavarapu1@udayton.edu, and gsubramanyam1@udayton.edu
- \* Correspondence: alemayehub2@udayton.edu
- + Presented at the title, place, and date.

Abstract: We report the fabrication of polyvinyl alcohol (PVA) and polyethylene oxide (PEO) 11 blended polymer nanocomposites (PNCs) films loaded with different percentages of carbon black 12 (CB) using the stencil printing method. Effect of blend polymer composition, weight ratio and load-13 ing on the electrical and morphological properties of the PNC films were studied using a surface 14 profilometer, SEM, and four-point probe method. SEM analysis showed homogenous dispersion of 15 CBs in the blend matrix as well as the formation of CB agglomerations. An electrical conductivity 16 of 0.417 S/m was achieved with 14wt% CB loading. In general, the CB-polymer composite films with 17 improved morphological and electrical properties were fabricated and characterized for potential 18 sensing applications. 19

Keywords: PNC; PEO; PVA; CB; Film

# 1. Introduction

Polymer nanocomposites (PNCs) have recently gained widespread attention owing 23 to their intriguing mechanical, electrical and optical properties [1]. Exploring the structural and electrical properties of PNC materials have been the focus of many researchers 25 due to their potential applications as sensing materials or layers in sensor devices [2,3]. 26

Polymer nanocomposites typically comprise of polymers as a host matrix and nano-27 particles as conductive fillers. A wide variety of polymers have been used as host matri-28 ces. However, in recent years, non-toxic and water-soluble polymers such as PVA and 29 PEO are gaining attention due to their promising applications in humidity sensing devices 30 [4,5]. As conductive fillers, carbon-based nanofillers [6,7], 2D-materials [8], metal oxides 31 [9] and metal nanoparticles [4] have been widely used to synthesize PNCs. Among the 32 carbon-based nanofillers, carbon black (CB) is a promising candidate for use in preparing 33 PNCs because of its unique properties, including high electrical conductivity, high ther-34 mal stability, and desirable wetting characteristics [10]. CB based PNC films have been 35 fabricated and utilized as sensing layers in sensor devices. For instance, [11] and [12] re-36 ported the potentials of CB based PNC films on efficiently sensing humidity and volatile 37 organic compounds. 38

Nowadays, instead of using a pristine polymer as a host matrix, multiple polymers 39 are blended to prepare a new host matrix by adapting the useful properties of the pristine 40 polymers [13]. In this regard, PNC films based on a variety of polymer blends have been 41 reported. For instance, PNC films based on PVA and polyvinylpyrrolidone (PVP) blended 42 host matrices loaded with different weight ratio of alumina and zinc oxide have been fabricated, and the corresponding morphological, thermal, and electrical properties of the 44

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**Copyright:** © 2021 by the authors. Submitted for possible open access publication under the terms and conditions of the Creative Commons Attribution (CC BY) license (https://creativecommons.org/license s/by/4.0/). PNC films have been studied [14,15]. Similarly, PNC films based on PEO-PVP blend ma-1 trix loaded with gold nanoparticles were also demonstrated [16]. In general, polymer 2 blends are doped with various kinds of nanofillers to prepare PNCs and advanced func-3 tional materials. Nevertheless, the topographical, morphological, and electrical properties 4 of PNC films based on CB loaded PVA-PEO blend matrix are yet to be studied to confirm 5 their suitability as potential sensing materials. 6

In this work we present the preparation and characterization of PNC films. The films 7 were prepared from PEO-PVA blends loaded with different concentrations of CBs. Stencil 8 printing method was used to print PNC films of different thicknesses. High-resolution 9 scanning electron microscope (HRSEM) and four-point probe resistance measurement 10 techniques were applied to characterize the morphological, structural, and electrical prop-11 erties of the PNC films. The topographic properties such as the surface roughness and 12 thickness of the PNC films were obtained using a surface profilometer. The findings indi-13 cated that the PNC films exhibited good structural, morphological, and electrical proper-14 ties which make them suitable for various sensing applications. 15

# 2. Materials and Methods

#### 2.1. Materials

PEO (molecular weight (MW) ≈ 600,000 g/mol) and PVA (MW ≈ 89000-98000 g/mol 18 and 99+% hydrolyzed) powders were purchased from Sigma-Aldrich, USA. Vulcan 19 XC72R carbon black was generously donated by Cobat inc. These reasonably high MW 20 materials are appropriate to prepare polymer blends owing to their film forming charac-21 teristics. Different thicknesses of stencil sheets were utilized to create the desired microstructures. Glass was utilized as a substrate. DI water was used as a solvent. 23

# 2.2. Methods

#### 2.2.1. Synthesis of PNC films

To prepare the PVA-PEO blend matrix, 10wt% PVA and 5wt% PEO polymers were 26 used. The 10wt% PVA polymer was prepared by dissolving 1 g of PVA in 9 g of distilled 27 water. To get a homogenized solution, first the PVA solution was stirred on a magnetic 28 stirring plate at 500 rpm for 30 mins at room temperature. After 30 mins, the PVA solution 29 was stirred at 90°C for 2 additional hours. Similarly, 5wt% PEO polymer was prepared by 30 dissolving 0.5 g of PEO in 9.5 g of distilled water. The PEO solution was stirred for 3 hours 31 at room temperature on a stirring plate at 500 rpm and then stirred at 55°C for an addi-32 tional hour. During the PEO and PVA stirring process air bubbles were generated in the 33 polymers. After the stirring process, the air bubbles were removed by exposing the poly-34 mers to the nitrogen gas for 10 mins. 35

The PVA-PEO polymer blends were prepared with different weight ratios of PVA t0 36 PEO, see Table 1. The polymer blends were then mixed for 10 mins in an AR-100 condi-37 tioning mixer. Blend polymer nanocomposites were prepared by adding 10wt% CB to the 38 PVA-PEO polymer blend matrices. The PNCs were mixed for 15 mins in the AR-100 con-39 ditioning mixer to obtain homogeneous viscous solutions. In a similar way, with a fixed 40 PVA-PEO polymer blend ratio of 1:1, various PNCs were prepared with different weight 41 percentages of CBs. 42

Table 1. Different weight ratio of PVA-PEO loaded with CBs.

<b>PVA:PEO</b> ratio	<b>CB (wt%)</b>
1:1	$\mathbf{x}^*$
3.5:1	10
1:3.5	10

\* Different wt% of CBs content.

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Using the PNC solutions, the PNC films were fabricated on the surface of the substrates (7.5 cm×2.5 cm) as follows. To fabricate the PNC films in a rectangular shape, the stencils were engraved with the desired pattern using a CO<sub>2</sub> laser (Boss LS1416). Then, the conventional screen-printing technique was utilized to obtain the PNC films, illustrated in Figure 1. 5



**Figure 1.** The conventional stencil-printing technique: (a) Placing of the stencil on the top of the substrate; (b) Pouring of the PNC solutions onto the opening of the stencil; (c) Spreading of the composites using a squeegee; (d) The resulting film obtained after curing.

#### 2.2.2. Characterizations

Various characterization techniques were used to study the electrical, morphological, 12 and structural properties of the films. The morphology and structure of films were ana-13 lyzed using a HRSEM (Hitachi S-4800) and a surface profile (Dektak 6M). The HRSEM 14 was carried out using two different accelerating voltages of 5 and 10 kV. The samples were 15 mounted on the specimen stubs and coated on a carbon tape to obtain the SEM images. 16 Surface profilometry was conducted to extract the topographical properties from the sur-17 face of the films. It provides quantitative data regarding the roughness of the surface of a 18 film as well as the corresponding thickness by scanning the surface of the films as a dia-19 mond-tipped stylus moves from the substrate to and over the surface of the films. 20

The current-voltage (I-V) characteristics or electrical conductivity of the films were 21 analyzed using a four-point probe resistance measurement method (Ossila) at room temperature. The measurements were performed by recording the current on application of 23 voltage, and vice versa. 24

### 3. Results and Discussion

#### 3.1. Effect of the weight ratio of PEO-PVA blends

PVA-PEO polymer blends loaded with a fixed 10 wt% of CBs were prepared with 27 different weight ratios of the pristine polymers as presented in Table 1, followed by screen 28 printing to fabricate the PNC films on glass substrates. Visual inspection was performed 29 to observe the underlying physical properties. For instance, the formation of cracks on the 30 surface of the films was observed on the PNC films which were prepared from 3.5:1 31 weight ratio, see Figure 2a. While the PNC films which were prepared from 1:3.5 weight 32 ratio were peeled off from the substrates, see Figure 2b. It can be inferred that these weight 33 ratios were not enough to form stable blend polymers (dual-phase), as the formation of 34 the dual-phase is strongly influenced by the weight ratio of the parent polymers [17]. 35 However, the formation of cracks and peeling off were not observed on the PNC films 36 with a weight ratio of PVA to PEO of 1:1. The PNC films revealed a very good strength, 37 rigidity, and toughness. This might be due to the formation of dual-phase in the polymer 38 matrix. Henceforth, the 1:1 weight ratio was used to prepare the blended polymers. 39

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**Figure 2.** Effect of the weight ratios of PVA-PEO blends on the physical properties of the films: (**a**) 3.5:1; (**b**) 1:3.5.

#### 3.2. Topographical analysis

Surface profilometry was carried out on the surface PNC films to obtain the 3 topographical properties, such as thickness and the surface roughness parameters. For 4 each surface profilometer scan, a z-axis which was perpendicular to the surface of the 5 film was considered as the scanning axis, where z = 0 corresponded to a position on the 6 surface of the substrate. Then, the z-direction values were recorded. All the scanning of 7 the film surfaces were performed using a 12.5  $\mu$ m diamond-tipped stylus for a scan length 8 of 2000 µm scanned in 50 seconds. The surface profiles of the PNC films which contained 9 8, 10, 12 and 14wt% CB are shown in Figure 3. The surface roughness (RA) and 10 thickness (t) values of the films loaded with different weight percentages of CBs were 11 summarized in Table 2. 12

Table 2. The RA and t values for different wt5 of CBs.

CB Content (wt%)	R <sub>A</sub> (μm)	t (µm)
0	$0.2\pm0.001$	$2 \pm 0.11$
8	$1.9 \pm 0.042$	$57 \pm 0.24$
10	$1.5 \pm 0.015$	$76 \pm 0.02$
12	$1.2 \pm 0.031$	$92 \pm 0.05$
14	$1.2 \pm 0.020$	$120 \pm 0.32$

From Table 2, it is evident that the thickness of the PNC film increased by increasing 15 the wt% of the CB. This was expected because when the conductive filler wt% was 16 increased in a polymer matrix, its dry thickness also increased. On the other hand, the  $R_A$ 17 of the PNC films decreased with the increase of the wt% of CBs. For instance, when the 18 CB concentration was increased from 8 to 14wt%, the R<sub>A</sub> decreased from 1.9 to 1.2  $\mu$ m. 19 This might be due to change in the viscosity of the PNC solution caused by the increasing 20 of the wt% of CBs. The RA values indicated the flatness or regularity of the surface of the 21 films. 22

# 3.3. Morphological analysis

The morphology of the PNC films was studied using the HRSEM. The topographic 24 HRSEM images of PNC films which contain 12 and 10wt% CBs at different magnifications 25 were shown in Figure 4. From Figure 4, it can be observed that the CB particles were 26 uniformly dispersed in the PVA-PEO blend matrix. This indicated the presence of strong 27 cohesion bond between the CBs and the blend matrices. The SEM images further revealed 28 the presence of porous structures on the surface of the films. Apart from the pores, the 29 formation of CB agglomerations in the PVA-PEO blend matrix was also observed, see 30 Figure 4c. Due to this phenomenon, the increase in particle sizes was noticed. For instance, 31 the smallest particle size detected for the cases of 12 wt% CB containing film was 41.7 nm, 32 while the largest particle size detected was 437.2 nm. In general, a homogeneously 33 dispersed conductive nanofillers in a blend matrix provides large surface area and greater 34

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Figure 3. The surface profiles of the PNC films with different wt% of CBs content.

number of active sites which in turn plays an important role in improving the both 3 mechanical strength and the sensing performance of the PNC films [15].



**Figure 4.** The SEM images of PNC films loaded with: (**a**) 12wt% CB at 10 μm; (**b**) 12wt% CB at 1 μm resolution; (**c**) CB agglomerations; (**d**) 10wt% CB at 10 μm.

#### 3.4. Electrical properties

The I-V characteristics and the electrical conductivity of the PNC films were obtained 9 using the four-point probe system at room temperature. The I-V characteristics were 10 obtained for two different cases. In the first case, the CB content was fixed to 10wt%, and 11 PNC films with different thicknesses were prepared by using 100, 150 and 200  $\mu$ m thick 12 stencils. The thicknesses of the films after curing were 53.7, 74.1, and 96.6 µm, respectively. 13 Figure 5a illustrates the variation of the in-plane I-V characteristics of the films with the 14film thickness. From Figure 5a, it can be observed that the I-V curves were linear and the 15 slope of the I-V curves increased with the increase of the film thicknesses. For instance, 16 for the PNC films having thicknesses of 53.7, 74.1 and 96.6 µm, the corresponding slopes 17 were 0.091, 0.132, and 0.167 A/V, respectively. In the second case, the CB content was 18 varied from 8 to 14wt%, and the thickness of the films was fixed. Figure 5b shows the 19 variation of the in-plane I-V characteristics of the films with wt% of CB contents. Here as 20 well, the I-V curves were linear and the slope of the I-V curves increased with the increase 21

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of the CB contents. For instance, for the PNC films having CB contents of 8, 10, 12 and 1 14wt%, the corresponding slopes were 7.5, 10.2, 11.3 and 17.4 A/V, respectively. 2

Figure 5. I-V characteristics of the PNC films showing the effect of: (a) Thickness; (b) CB contents variation.

The electrical conductivity of a PNC film can be calculated using the relation  $\sigma = 1/(R_s \times t),$ (1)

where  $\sigma$  represents the electrical conductivity,  $R_s$  represents a sheet resistance which 7 can be obtained by taking the inverse of the slope of the I-V characteristic curve, and t8 represents the thickness of the film. Figure 6a shows the effect of CB contents variation on 9 the electrical conductivity of the PNC films. From Figure 6a, it can be observed that the electrical conductivity of the PNC films increased with the increasing CB contents. For instance, the electrical conductivity increased significantly from  $1.56 \times 10^{-9}$  to 0.176 S/m 12 with the introduction of 8wt% of CBs. Then, the electrical conductivity of the films grad-13 ually increased and reached to 0.471 S/m as the CB contents increased from 8 to 14wt%. 14 The change in the  $\sigma$  value of a PNC film might be attributed to the formation of charge 15 transfer complexes (CTCs) in the polymer matrix after loading it with CBs. The CTCs re-16 duce the interfacial barrier between the tapping sites and provide a conductive path 17 through the amorphous regions of the PVA-PEO blend matrix thus enhancing the con-18 ductivity [18,19]. 19

Further, the electrical conductivity behavior of conductive composites can be ration-20 alized using the classical percolation theory [20]. The classical percolation theory equation 21 can be expressed as: 22

$$\sigma = \sigma_o (\varphi - \varphi_c)^a, \tag{2}$$

where  $\varphi$  is the volume fraction of the fillers,  $\varphi_c$  is the volume percolation threshold and  $\alpha$ 23 is the critical exponent which represents the dimension of the conductive network. For 24 PNC films with CB contents and  $\varphi > \varphi_{c_r}$  the linear fitting of log  $\sigma$  vs log ( $\varphi - \varphi_c$ ) was per-25 formed to extract the percolation threshold and critical exponent by varying the value of 26  $\varphi_c$  until the best fit was obtained. Hence, based on the data in Figure 6b, the percolation 27 threshold of PEO-PVA-CB nanocomposites was estimated and was around 0.2 vol% with 28  $R^2$  of 0.98. As reported in literature, the percolation threshold of many composites with 29 conductive nanofillers was in the range of 3 -15 wt% [21]. The result showed that the per-30 colation threshold reported in the present study was comparable. Also,  $\alpha$  value was esti-31 mated to be approximately 1, which indicated the presence of a two-dimensional conduc-32 tive network [22]. 33

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Figure 6. Effect of CB content on electrical conductivity: (a) The electrical conductivity as a function of CB content; (b) The linear fit of the films using the percolation theory.

# 4. Conclusions

The PNC films based on PEO-PVA blend loaded with different wt% of CB contents were prepared. The PNC films were characterized to obtain the topographical, morpho-5 logical, and electrical properties. The weight ratio of PVA to PEO had an impact on the 6 presence of dual-phase in the blend polymer matrices. The SEM images of the PNC films 7 confirmed the uniform dispersion of CBs in the PVA-PEO blend matrices. The formation 8 of CB agglomerations was also observed. The effect of the variation of CB content and 9 thickness on the electrical properties of the PNC films was demonstrated. The PNC film 10 changed from being as an insulator to a conductive with addition of 8wt% of CBs which 11 in turn helps the formation of CTCs in polymer matrix. The critical exponent value indi-12 cates the existence of a 2D conductive network. An enhancement was observed in both 13 morphological and electrical properties of the films. In general, the method reported 14 herein opens a route to fabricate blend polymer-CBs based PNC films and their suitability 15 for future sensor applications, including humidity and voc sensors. 16

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### References

- 1. Nezakati, T.; Seifalian, A.; Tan, A.; Seifalian, A.M. Conductive Polymers: Opportunities and Challenges in Biomedical 26 Applications. Chem. Rev. 2018, 118, 6766-6843, doi:10.1021/acs.chemrev.6b00275. 27 2. Zamiri, G.; Haseeb, A.S.M.A. Recent trends and developments in graphene/conducting polymer nanocomposites 28
- chemiresistive sensors. Materials (Basel). 2020, 13, doi:10.3390/ma13153311. 29
- Naveen, M.H.; Gurudatt, N.G.; Shim, Y.B. Applications of conducting polymer composites to electrochemical sensors: A 3. 30 review. Appl. Mater. Today 2017, 9, 419-433, doi:10.1016/j.apmt.2017.09.001. 31
- Yao, W.; Chen, X.; Zhang, J. A capacitive humidity sensor based on gold-PVA core-shell nanocomposites. Sensors Actuators, 4. 32 B Chem. 2010, 145, 327-333, doi:10.1016/j.snb.2009.12.021. 33
- 5. Wang, S.; Xie, G.; Su, Y.; Su, L.; Zhang, Q.; Du, H.; Tai, H.; Jiang, Y. Reduced graphene oxide-polyethylene oxide composite 34 films for humidity sensing via quartz crystal microbalance. Sensors Actuators, B Chem. 2018, 255, 2203-2210, 35

2 3

1

4

17 18 19

24

	doi:10.1016/i.spb.2017.09.028	1
6	Kang NK : Jun TS : La DD : Oh LH : Cho XW : Kim XS Evaluation of the limit of detection canability of carbon black-	1
0.	nolymer composite sensors for volatile breath biomarkers Sensors Actuators B Chem 2010 147 55-60	2
	doi:10.1016/i sph.2010.03.025	4
7	Khan MW $\cdot$ Asif SU $\cdot$ Ur Rehman KM $\cdot$ Uddin W $\cdot$ Muhasher Ahmed S $\cdot$ Khan EU $\cdot$ Tagliaferro A $\cdot$ Iagdale P $\cdot$ Fakhar-	5
	e-Alam. M. The electrical behavior of functionalized multiwall carbon nanotubes decorated with polymer nanocomposites	6
	<i>Phys. B Condens. Matter</i> <b>2019</b> , 556, 17–21, doi:10.1016/j.physb.2018.12.029.	7
8.	Vucai, N.: Ouinn, M.D.I.: Baechler, C.: Notley, S.M.: Cottis, P.: Hojati-Talemi, P.: Fabretto, M. V.: Wallace, G.G.: Murphy, P.I.:	8
	Evans, D.R. Vapor phase synthesis of conducting polymer nanocomposites incorporating 2D nanoparticles. <i>Chem. Mater.</i>	9
	<b>2014</b> , <i>26</i> , 4207–4213, doi:10.1021/cm5014653.	10
9.	Khairy, Y.; Elsaeedy, H.I.; Mohammed, M.I.; Zahran, H.Y.; Yahia, I.S. Anomalous behaviour of the electrical properties for	11
	PVA/TiO2 nanocomposite polymeric films. <i>Polym. Bull.</i> <b>2020</b> , 77, 6255–6269, doi:10.1007/s00289-019-03028-y.	12
10.	Baby, K.C.; Fikri, U.; Schwesinger, N. Resistive characterization of soft conductive PDMS membranes for sensor applications.	13
	SAS 2016 - Sensors Appl. Symp. Proc. <b>2016</b> , 344–349, doi:10.1109/SAS.2016.7479870.	14
11.	Kim, J.; Cho, J.H.; Lee, H.M.; Hong, S.M. Capacitive humidity sensor based on carbon black/ polyimide composites. <i>Sensors</i>	15
	<b>2021</b> , <i>21</i> , 1–11, doi:10.3390/s21061974.	16
12.	Mallya, A.N.; Kottokkaran, R.; Ramamurthy, P.C. Conducting polymer-carbon black nanocomposite sensor for volatile	17
	organic compounds and correlating sensor response by molecular dynamics. Sensors Actuators, B Chem. 2014, 201, 308–320,	18
	doi:10.1016/j.snb.2014.04.056.	19
13.	Weder, C. Functional polymer blends and nanocomposites; 2009; Vol. 63; ISBN 9781482239270.	20
14.	Choudhary, S. Structural, morphological, thermal, dielectric, and electrical properties of alumina nanoparticles filled PVA-	21
	PVP blend matrix-based polymer nanocomposites. Polym. Compos. 2018, 39, E1788–E1799, doi:10.1002/pc.24793.	22
15.	Choudhary, S.; Sengwa, R.J. ZnO nanoparticles dispersed PVA-PVP blend matrix based high performance flexible	23
	nanodielectrics for multifunctional microelectronic devices. Curr. Appl. Phys. 2018, 18, 1041-1058,	24
	doi:10.1016/j.cap.2018.05.023.	25
16.	Morsi, M.A.; Abdelghany, A.M. UV-irradiation assisted control of the structural, optical and thermal properties of PEO/PVP	26
	blended gold nanoparticles. Mater. Chem. Phys. 2017, 201, 100–112, doi:10.1016/j.matchemphys.2017.08.022.	27
17.	Graebling, D.; Muller, R.; Palierne, J.F. Linear viscoelasticity in incompatible polymer blends in the melt in relation with	28
	interfacial properties. J. Phys. <b>1993</b> , 3, 1525–1534, doi:10.1051/jp4:19937238.	29
18.	Devi, C.U.; Sharma, A.K.; Rao, V.V.R.N. Electrical and optical properties of pure and silver nitrate-doped polyvinyl alcohol	30
	films. Mater. Lett. 2002, 56, 167–174, doi:10.1016/S0167-577X(02)00434-2.	31
19.	Bhajantri, R.F.; Ravindrachary, V.; Harisha, A.; Ranganathaiah, C.; Kumaraswamy, G.N. Effect of barium chloride doping on	32
	PVA microstructure: Positron annihilation study. Appl. Phys. A Mater. Sci. Process. 2007, 87, 797–805, doi:10.1007/s00339-007-	33
	3923-у.	34
20.	Kilbride, B.E.; Coleman, J.N.; Fraysse, J.; Fournet, P.; Cadek, M.; Drury, A.; Hutzler, S.; Roth, S.; Blau, W.J. Experimental	35
	observation of scaling laws for alternating current and direct current conductivity in polymer-carbon nanotube composite	36
	thin films. J. Appl. Phys. 2002, 92, 4024–4030, doi:10.1063/1.1506397.	37
21.	Li, J.; Ma, P.C.; Chow, W.S.; To, C.K.; Tang, B.Z.; Kim, J.K. Correlations between percolation threshold, dispersion state, and	38
	aspect ratio of carbon nanotubes. Adv. Funct. Mater. 2007, 17, 3207-3215, doi:10.1002/adfm.200700065.	39
22.	Wang, Y.; Hao, J.; Huang, Z.; Zheng, G.; Dai, K.; Liu, C.; Shen, C. Flexible electrically resistive-type strain sensors based on	40
	reduced graphene oxide-decorated electrospun polymer fibrous mats for human motion monitoring. Carbon N. Y. 2018, 126,	41
	360–371, doi:10.1016/j.carbon.2017.10.034.	42