

Abstract



Hybrid Membrane Materials Based on Polybenzimidazole and Silica with Grafted Phosphonic Groups for Fuel Cell Applications ⁺

Anna Lysova 1,* and Igor Ponomarev 2

- ¹ N.S. Kurnakov Institute of General and Inorganic Chemistry of the Russian Academy of Sciences
- ² A.N. Nesmeyanov Institute of Organoelement Compounds of the Russian Academy of Sciences; gagapon@ineos.ac.ru
 - * Correspondence: ailyina@yandex.ru
 - Presented at the 1st International Electronic Conference on Processes: Processes System Innovation, 17–31 May 2022; Available online: https://ecp2022.sciforum.net.

Abstract: Owing to high thermal and chemical stability and good mechanical properties, polybenzimidazole (PBI) doped with phosphoric acid is a very promising material to be used as an electrolyte in the medium-temperature fuel cells. Their use at temperatures below ~160 °C is impeded by the leaching of the free H₃PO₄ from the membrane. In order to overcome this problem one of the possible approaches is the incorporation of inorganic particles capable to stabilize H₃PO₄ in PBI matrix. Surface-modified particles can be more efficient for this purpose. In this work we studied the properties of proton-conducting membranes based on PBI and silica particles surface-modified by propylphosphonic groups. Composite membranes were obtained by casting of polymer solution containing tetraethoxysilane and modified silane ((2-diethylphosphatoethyl)triethoxysilane) with next hydrolysis by HCl. The mass concentration of the dopant was 5 or 10 wt %, and the mole fraction of functional groups on the oxide surface was varied in the range of 0–100 mol % by changing the composition of the precursor mixture. All films were treated by 75% H₃PO₄. The resulting membranes have been characterized using transmission and scanning electron microscopy, IR spectroscopy, and impedance spectroscopy. Grafting of functional -PO₃H₂ groups onto the silica surface leads to a significant increase in the uptake of phosphoric acid by hybrid membranes, the content of which determines the conductivity of these materials. An increase in the number of -PO₃H₂ groups leads to both an increase in the degree of acid doping and ionic conductivity. The conductivity of the best samples obtained reaches 0.081 S/cm at 160 °C. The introduction of acid groups on the dopant surface is a promising approach from the point of view of reducing the amount of phosphoric acid required to maintain a high proton transport rate.

Keywords: proton conductive membrane; polybenzimidazole; hybrid membrane; fuel cell; proton conductivity; surface modified silica

Citation: Lastname, F.; Lastname, F.; Lastname, F. Title. *Proceedings* **2022**, *69*, x. https://doi.org/10.3390/xxxx

Academic Editor: Firstname Lastname

Published: date

Publisher's Note: MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Copyright: © 2022 by the authors. Submitted for possible open access publication under the terms and conditions of the Creative Commons Attribution (CC BY) license (https://creativecommons.org/license s/by/4.0/).