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# Proceeding Paper Supercritical Fluid CO<sub>2</sub> Extraction Technology to Produce an Innovative Healthy Product from Almond Wastes \*

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+ Presented at 2nd International Electronic Conference on Processes, online, 17-31 May 2023.

Abstract: In this work, we studied the potential of supercritical fluid CO<sub>2</sub> technology to extract almond wastes and obtain a fiber product rich in minerals and phenolics without the use of extraction co-solvent. The analysis of phenolics in the resulting extracted product was performed by 13 liquid chromatography tandem mass spectrometry (LC-MS/MS) and showed vanillin, catechin and the acids dihydroxybenzoic, vanillic and syringic as main phenolic compounds (PC). In addition, the analysis of minerals carried out by Inductively Coupled Plasma Optical Emission spectroscopy (ICP-OES) showed a wide range of macroelements like Magnesium (Mg) and Potassium (K) in quantities up to 1.7g/kg (Mg) and 6 g/kg (K), so that represent a value matrix to be integrated into functional drinks targeting sporty people while promoting the circular economy and the food up- cycling. 20

Keywords: Supercritical Fluid CO2 Extraction; Almond by-products; Functional foods; Liquid-21 chromatography mass spectrometry; Phenolics; Minerals; Inductively coupled plasma optical emis-22 sion spectrometry 23

#### 1. 1. Introduction

Citation: To be added by editorial staff during production.

Academic Editor: Firstname Lastname

Published: 19 May



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The world production of almond Prunus dulcis (Miller D.A.Webb) and their derived prod-25 ucts such as almond oil, has increased in recent years, due to its important nutritional <sup>26</sup> characteristics [1]. The Food and Agriculture Organization of the United Nations (FAO) 27 reports that world almond production stood at 3,214,522 tons for the year 2020 [2], leading by the United States and followed by Spain, reaching an estimated amount of 371,460 tons 29 in 2021, according to Spanish Ministry of Agriculture, Fisheries and Food [3]. Consequently, the increase of almond production is accompanied by a parallel increase of al-31 mond residues (shell, shell, skin, downstream). It is estimated that the almond oil industry 32 generates up to 52% of waste weigh, mainly almond cake, from the shell, in relation to the 33 material used [4]. Almond cake is a valuable source of bioactive compounds such as phe-34 nolic compounds, fatty acids, minerals, tocopherols, steroids and volatile compounds, 35 which have demonstrated important biological activities both *in vitro* and *in vivo* [5,6], including prebiotic, antimicrobial, antioxidant, anti-inflammatory, anticancer, hepatopro-37 tective, cardiometabolic, nootropic, anxiolytic, sedative-hypnotic, and nervous system-38 enhancing effects [6–8]. In fact, the use in traditional medicine of the almond and its different botanical parts is reported in the treatment of some brain disorders, respiratory and urinary tract problems [8]. Extraction of bioactive compounds from almond cake can be 41 challenging due to their low solubility in conventional solvents and the presence of un- 42 desirable compounds [4]. In this sense, the use of supercritical fluid CO<sub>2</sub> extraction (SFE- 43 CO<sub>2</sub>) is a promising technique to obtain high quality and purity compounds, with unique 44 characteristics and significant advantages. One of the main benefits of using SFE-CO<sub>2</sub> is 45

its ability to obtain products with a minimal presence of residual solvents compared to 1 conventional extraction methods [9]. In addition, SCFE-extracted compounds may have 2 unique organoleptic and functional properties, making them suitable for use in the for- 3 mulation of functional foods, dietary supplements, or cosmetic products [10]. Carbon di- 4 oxide (CO<sub>2</sub>) is the most widely used supercritical fluid in the extraction of bioactive com- 5 pounds due to its low cost, low toxicity, non-flammability, and its ability to be easily re- 6 moved from the final product. Furthermore, supercritical CO2 has a high dissolving ca- 7 pacity, which makes it suitable for extracting a wide range of bioactive compounds from 8 almond residues [10,11]. In the present work, the effect of SFE-CO<sub>2</sub> in the phenolic, min- 9 eral, and fatty acid profile of the resulting almond cake products is evaluated with the aim 10 of obtaining ingredients rich in bioactive compounds which can be used in various appli- 11 cations in the food, nutraceutical and cosmetic industry (Figure 1).



Figure 1. Schematic representation of work objectives

# 2. Material and Method section

# 2.1. By-products sample preparation and Supercritical Fluid Extraction (SFE-CO<sub>2</sub>)

Almond press cake were provided by Spanish nut oil processing industries in 2021. Once 15 in our facilities, the press cake was submitted to a dehydration process in an evaporator 16 concentration of 800 L of capacity, coupled to a vacuum equipment. Thus, it was obtained 17 a final dried product with a humidity below to 10%, which was submitted to an SFE-CO2 18 extraction process to obtain an oily extract and a fibre ingredient which were nutritionally 19 evaluated because of their content of phenolics, minerals and fatty acids. SFE-CO<sub>2</sub> exper-20 iments were performed in a HA220-40-48 System (HuaAn Supercritical Extraction Co., 21 Ltd. Nantong, China), composed of conditioner, pumps for CO<sub>2</sub>, filters, heaters and 2 ex-22 tractor cells of 24 L of volume each. The temperature of the extraction system was fixed at 23 40 °C, the CO<sub>2</sub> flow was 210 L/h and the extraction time was 45 min. Two pressures were 24 tested, 20 and 24 MPa. The most important parameters in a SFE-CO2 extraction are the pressure and temperature inside the cell. The equipment allows an extraction pressure up to 40 MPa and an extraction temperature up to 85°C, respectively. In this work, the pressure and temperature were adjusted by a pressure regulator and a temperature controller to experimental conditions. Moreover, characteristics of matrix to be extracted such as its degree of humidity was controlled by a humidity analyzer (PCE Instruments). 25

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Liquid chromatography-mass spectrometry (LC-MS/MS) was used for the analysis of phenolic compounds as reported by the previous study [12]. Briefly, the amount of 1 g of 2 sample were extracted with 30 mL of MeOH: W (80:20) in a continuously magnetic agita- 3 tion (300 rpm) for 1 hour at 40°C. Then, the extracts were centrifuged, and the extraction 4 was repeated twice. After that, the samples were dried, frozen, lyophilized and dissolved 5 in 2 mL of MeOH prior LC-MS/MS analysis. High Performance Liquid Chromatography- 6 Mass Spectrometry (HPLC-MS, 1260 Series, Agilent) coupled to a compact Mass Detector 7 equipment (TRIPLE QUAD 3500; AB SCIEX INSTRUMENTS) was used for the analysis s in a C18 column (PHENOMENEX LUNA, 150 mm  $\times$  2 mm and 3  $\mu$ m) at 40°C. The flow 9 rate was 0.3 mL min<sup>-1</sup> and the injection volume was 10 µL. Mobile phase was composed 10 of 0.1% formic acid in water (A), and 0.1% formic acid in acetonitrile (B), using a gradient.  $_{11}$ Initial conditions (98% A and 2% B) were held for 4 min before ramping to 20% B at 7 min 12 and 90% B at 14 min. Then, initial conditions were kept from the minute 15 and held for 6 13 min. Instrument parameters were as follows: curtain gas (CUR), 25 psi; collision gas 14 (CAD), 7 psi: ion spray voltage (IS), -4500 V; temperature (TEM), 400°C; ion source gas 1 15 (GS1), 55 psi; ion source gas 2 (GS2), 55 psi; interface heater, on. Phenolic compounds were identified and quantified employing standard solutions and constructing the calibration curves for each compound. The transition used for quantification were vanillin (VA, m/z 166.7 > 122.9), cinnamic acid (CA, *m*/*z* 147.0 > 103.0), dihydroxibenzoic acid (DA, *m*/*z* 152.9> 109.0), ferulic acid (FA, *m*/*z* 195.0 > 176.9), *p*-coumaric acid (*p*-CA, *m*/*z* 162.0 > 119), phallic acid (PA, *m*/*z* 164.8 > 77.0), syringic acid (SA, *m*/*z* 199.0 > 140), *m*-toulic acid (M-TA, *m*/*z* 134.9 > 91.0), luteolin (LU, *m*/*z* 285.0 > 133), syringaldehyde (SY, *m*/*z* 183.0 > 77.0), quercetin (QE, *m*/*z* 301.0 > 150), vanillin (VN, *m*/*z* 150.7 > 108), rutin (RU, *m*/*z* 609 >300), tyrosol (TYR, *m*/*z* 137.0 > 106.0), hydroxytyrosol (HTYR, *m*/*z* 153.0 > 123.0), ligstroside (LIG, *m*/*z* 522.8 >360.0), oleacin (OLE, *m*/*z* 318.8> 195), oleuroside (OLS, *m*/*z* 538.9 > 307) and oleuropein (OLP, *m*/*z* 538.9 >377). FA, SA and SY were detected in positive mode, remaining phenolics in negative mode. Results were reported as mg/kg.

#### 2.3. Ash content and analysis of minerals by ICP-OES

The ash content was determined thermogravimetrically [13]. Briefly, 10 mg of water and 18 volatile matter-free samples were ignited at 900°C beneath a flow of an O2-rich gas (30 19 mL/min) until constant weight. Minerals were analyzed by inductively coupled plasma 20 optical emission spectrometry (ICP-OES) using a Perkin-Elmer Optima 4300 DV spec-21 trometer (Shelton, CT, USA), equipped with an AS-90 autosampler, axial system, a high 22 dynamic range detector and a cross-flow type nebulizer for pneumatic nebulization. The 23 ICP-OES assessment was performed following the procedure described by Millos et al. 24 (2009) [14]. Briefly, 0.25g of sample were digested with nitric acid and hydrogen peroxide 25 using a Multiwave 3000 oven (Anton Paar, Graz, Austria), equipped with eight digestion 26 vessels. For quantification, standard stock solutions with the addition of internal standard 27 were used to construct the corresponding calibration curves. Results were reported as 28 mg/kg. 29

#### 2.4. Fatty acid analysis

FAs were analyzed using a GC-FID system (Agilent Technologies, Loveland, CO 80537, 31 USA). The amount of 1g of sample was extracted and derivatized according the method 32 of Miller and Berger as mentioned in Otero et al. [15]. The column used was an Agilent 33 HP-5MS UI capillary column (30 m × 0.250 mm × 0.25 um). The caried gas was Helium at 34 flow 1.8 mL/min. Oven temperature started at 50°C, increased to 210°C at 20°C increase 35 per min and hold for 18 min. Then, temperature was further increased to 230°C at 20°C 36 increase per min and kept at 230°C for 13 min. The injection volume was 1µL in splitless 37 mode. Inlet temperatures was set at 260°C and MS ion source and interface temperatures 38 were230°C and 280°C respectively. Data were acquired in a full scan from 40 to 500 m/z 39 and results are expressed as a relative percentage (%). 40

#### 3. Results and discussion

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### 3.1. Analysis of phenolic family compounds

Table 1 shows the phenolic profile analysed in almond cake and the resulting products43obtained after SCFE. Almond cake were predominantly composed of polyphenols like 3,4-44dihydroxybenzoic acid, vanillic acid, syringic acid, protocatechuic acid and in lesser ex-45tent, we also found salicylic, ferulic, *p*-coumaric and phthalic acids. These results concord46

with previous studies in which hydroxybenzoic, vanillic, protocatechuic and syringic ac-1 ids are the predominant phenolic acids in almond wastes [4,16]. In this sense, almond cake 2 here analysed show 54.4 mg/kg of 3,4-hydroxybenzoic acid, 7.06 mg/kg of vanillic acid 3 and 2 mg/kg of syringic acid and 0.64 mg/kg of protocatechuic acid. In addition, almond 4 cake also showed flavonoids like rutin, quercetin, vanillin and luteolin, from which van-5 illin was the most predominant with quantities of 13.5 mg/kg and aldehydes like syrin- 6 galdehyde in quantities of 5.57 mg/kg. Almonds are a rich source of phenolics [17] and 7 consequently their wastes can be used for the development of new nutraceuticals while s promoting the circular economy and the food upcycling. Almond skins present between 9 70 and 100% of the total phenols present in the whole almond fruit [4]. The characteriza-  $_{10}$ tion of polyphenolics carried out in 7 varieties of almond hulls (Prunus dulcis L.) showed 11 that chlorogenic acid, catechin, and protocatechuic acid were the most important poly- 12 phenols in almond hull and up to 220 mg/kg of PCA is reported [18]. Comparing data 13 before and after SCFE, we observed that phenolic acids, flavonoids, and aldehydes re- 14 mained in the solid part called fibre ingredient. While the quantity of these bioactive com- 15 pounds analysed in the oily extracts is scarce or very low. And comparing pressures, in 16 general terms, high amount of phenolic family compounds is obtained when SCFE is car-17 ried out at the lower pressure tested of 20 MPa instead of 24 MPa. So far, information about extraction of compounds from almond wastes is scarce and only focused on oil ex- 19 traction from the fruit. There is no information about the use of almond press-cake for 20 high-end markets. 21

Table 1. Main phenolics found in almond wastes before and after SCFE process (mg/kg)							
			SCFE of Almond cake				
COMPOUND	ABREV.	Almond Cake	Fibre Ingredient	Fibre Ingredient	Oily Extract	Oily Extract	
			20 MPa	24 MPa	20 MPa	24 MPa	
Phenolic acids							
3,4-dihydroxybenzoic acid	DA	54.40	57.40	35.00	1.65	4.18	
Vanillic acid	VA	7.06	7.41	5.06	6.88	6.15	
Syringic acid	SA	1.99	2.01	1.60	0.35	0.10	
Protocatechuic acid	PTA	0.64	0.62	0.48	nd	nd	
Salycilic acid	SAA	0.22	0.24	0.17	nd	nd	
Ferulic acid	FA	0.25	0.25	0.21	0.34	0.44	
<i>p</i> -coumaric acid	P-CA	0.13	0.13	0.09	0.30	0.63	
Phalic acid	PA	0.25	0.24	0.16	0.04	0.04	
Flavonoids							
Rutin	RU	0.54	0.554	0.422	nd	nd	
Quercetin	QE	0.12	0.12	0.13	0.16	0.05	
Vanillin	VN	13.50	16.49	9.23	0.55	0.39	
Luteolin	LU	0.01	0.01	0.02	0.04	0.01	
Aldehydes							
Syringaldehyde	SY	5.57	6.34	4.04	1.54	0.81	

\*SCFE: Supercritical Fluid Extraction

#### 3.2. Analysis of ash content and minerals

First, the content of ash in almond cake and in the obtained fibre ingredients after SCFE 2 at both pressures 24 MPa and 27 MPa was calculated. Results showed the content of Ash 3 were homogenous for three products tested and in the range of 2.63-2.74 % (table 2). Then, 4 the identification and quantification of macroelements and microelemtnos in all almond 5 wastes was carried out by ICP-OES and results are included in table 2. Almond cake con-6 tains high amounts of essential minerals. The content of macroelements in g/kg, decrease as follows: K (6.00) > Ca (2.84) > P (2.80) > Mg (1.63) > Na (0.34). And the amount of micro- s elements in mg/kg kept the following order: Fe (290.2) > Zn (22.7) > Mn (18.8) >Cu (15.5). 9 These results are in line with others found in the bibliography which show that almond 10 contain around 1.2-2.7 g/kg of Mg, 1.9-5.2 g/kg of P, 5.2-7.6 g/kg of K and up to 0.053 g/kg 11 of Fe [19]. Comparing data before and after SCFE, we observed that minerals found in the 12 almond cake remained in the fibre ingredient after the SCFE process. The content of those 13 minerals in the oily extracts is much lower. And comparing pressures, higher amounts of 14 minerals were obtained when SCFE is carried out at the lower pressure of 24 MPa. 15

Table 2. Ash content (%) and elements in almond cake before and after SCFE processes							
		SCFE of Almond cake					
MINERAL	Almond Cake	Fibre Ingredient	Fibre Ingredient	Extract	Extract		
		20 MPa	24 MPa	20 MPa	24 MPa		
Ash (%)	2.63	2.74	2.72	-	-		
Macroelements (g/kg)							
Ca	2.84	3.02	1.64	0.030	0.012		
K	6.00	6.08	5.91	0.001	0.001		
Mg	1.63	1.74	1.46	0.001	0.003		
Р	2.80	2.95	1.39	0.001	0.007		
Na	0.34	0.32 0.46		1.11	0		
Microelements (mg/kg)							
Mn	18.8	19.2	0.9	0.05	0.14		
Fe	290.2	299.7	299	1.83	8.96		
Cu	15.5	11.9	0.2	0.29	0.29		
Zn	22.7	23.3		0.82	0.72		

\*SCFE: Supercritical Fluid Extraction

#### 3.2. Analysis of fatty acids

Next, the analysis of fatty acids was carried out by GC-FID. Each fatty acid was quantified 19 with their respective standard and then, expressed as the summary of total saturated fatty 20 acids (SFAs), total monounsaturated fatty acids (MUFAs) and total polyunsaturated fatty 21 acids (PUFAs) for each almond by-product in mg per kg (table 3). Results showed that 22 fibre ingredient obtained from almond cake contain up to 4 times less of FA content. For 23 example, almond cake showed 81.29 mg/kg of FA while 19.61 mg/kg were found in the 24 fibre ingredient after SCFE at 24MPa and 28.92 mg/kg in that obtained at 20MPa. It is 25 worthy to mention that the content of SFA was also reduced in both fibres obtained after 26 SCFE up to 4 times (10.22 mg/kg FA, 24 MPa).

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Table 3. Fatty acid quantities in almond cake before and after SCFE process (mg/kg).						
		SCFE of Almond cake				
FA	Almond Cake	Fibre Ingredient	Fibre Ingredient	Oily Extract	Oily Extract	
		20 MPa	24 MPa	20 MPa	24 MPa	
Total SFAs	42.53	15.17	10.22	676.91	633.45	
Total MUFAs	27.93	10.09	6.91	453.65	437.22	
Total PUFAs	10.83	3.66	2.48	153.00	119.77	
Total FA	81.29	28.92	19.61	1283.56	1190.44	

\*SCFE: Supercritical Fluid Extraction. SFA: Saturated fatty acids. (MUFAs: monounsaturated fatty acids. PUFA: polyunsaturated fatty acids.

#### 4. Conclusions

In this work, the use of almond wastes is proposed to obtain ingredients from an alternative source through technologies that will enable upcycling companies to diversify their product ranges and make higher profits. After drying the almond cake, solid fraction was extracted by SFE-CO2 employing two pressures. As a result of this process oily extracts and a defatted fibre ingredient were obtained. Due to the prevalence of lifestyle diseases and the growing geriatric population, consumers across the globe are becoming health conscious. Also, the increasing number of health and fitness clubs drive dominance of sports and energy drinks in functional foods and beverages. In this sense, the SCFE technology allowed to obtain a high-quality product from almond press cake, *i.e.* a defatted and natural antioxidant fibre ingredient with up to 4 times less fat than the almond cake, with high content of minerals and phenolics, which could be used in novel fortification food applications. The SCFE technology has several advantages over traditional extraction technologies, in that they can be more efficient and more cost effective. In addition, there is also increasing concern regarding the use of solvents that, despite being food grade, can leave chemical residues.

Author Contributions: Conceptualization, P.O. and J. S-G.; methodology, PO and M.A.P.; formal 20 analysis, P.O.; investigation, F.C, J. E, M.A.P, J. S-G and P.O..; resources, J. S-G..; writing-original 21 draft preparation, F.C, P.O. and J.E.; writing – review and editing, P.O.; visualization, F. C and P.O.; 22 supervision, P.O.; project administration, P.O.; funding acquisition, J.S-G. All authors have read and 23 agreed to the published version of the manuscript. 24

Funding: Please add: This research was funded by the Bio Based Industries Joint Undertaking (JU) 25 under grant agreement No 888003 UP4HEALTH Project (H2020-BBI-JTI-2019), Ibero-American Pro-26 gram on Science and Technology (CYTED-AQUA-CIBUS, P317RT0003), The JU receives support 27 from the European Union's Horizon 2020 research and innovation program and the Bio Based In-28 dustries Consortium. The research leading to these results was supported by the European Union through the "NextGenerationEU and supported by MICINN through the Ramón y Cajal grant (RYC-2017-22891). The project SYSTEMIC Knowledge hub on Nutrition and Food Security, has received funding from national research funding parties in Belgium (FWO), France (INRA), Germany 32 (BLE), Italy (MIPAAF), Latvia (IZM), Norway (RCN), Portugal (FCT), and Spain (AEI) in a joint action of JPI HDHL, JPI-OCEANS and FACCE-JPI launched in 2019 under the ERA-NET ERA-34 HDHL (nº 696295). 35

Institutional Review Board Statement: Not applicable"

Informed Consent Statement: Not applicable.

Data Availability Statement: Suggested Data Availability Statements are available in section "MDPI Research Data Policies" at https://www.mdpi.com/ethics.

Conflicts of Interest: The authors declare no conflict of interest.

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