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Synthesis of 5-amino-3-(het)aryl-1*H*-1,2,4-triazoles *via* cyclization of (het)aroylaminoguanidines in aqueous medium

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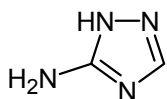
Abstract

The effective and clean procedure for the preparation of 5-amino-3-(het)aryl-1,2,4-triazoles (**3a-c**) *via* cyclization of (het)aroylaminoguanidines (**2a-c**) in water was reported. Two tautomeric forms, namely 5-amino-3-(het)aryl-1,2,4-triazoles (**A**) and 3-amino-5-(het)aryl-1,2,4-triazoles (**B**) were found to exist in tautomeric equilibrium. Form **A** was a predominant tautomer in DMSO solution ($K_T = 9-33$).

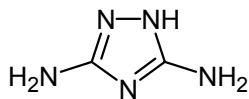
Key words: 1,2,4-triazoles, acylaminoguanidines, cyclization, aqueous medium, tautomerism.

Introduction

The 3(5)-amino-1,2,4-triazoles are known to be biologically active compounds. For example, the 5-amino-1,2,4-triazole itself has been used as the pesticide Amitrole and 3,5-diamino-1,2,4-triazole (Guanazole) is an antitumor drug that inhibits ribonucleotide reductase and DNA synthesis.



Amitrole



Guanazole

The 3(5)-amino-1,2,4-triazoles play a very important role as amidine type synthons in heterocyclic chemistry [1,2]. They are particularly useful for the syntheses of fused ring systems, such as imidazo[1,2-*b*][1,2,4]triazole, imidazo[2,1-*c*][1,2,4]triazole, 1,2,4-triazolo[1,5-*a*]pyrimidine and 1,2,4-triazolo[1,5-*a*][1,3,5]triazine possessing a variety of biological effects.

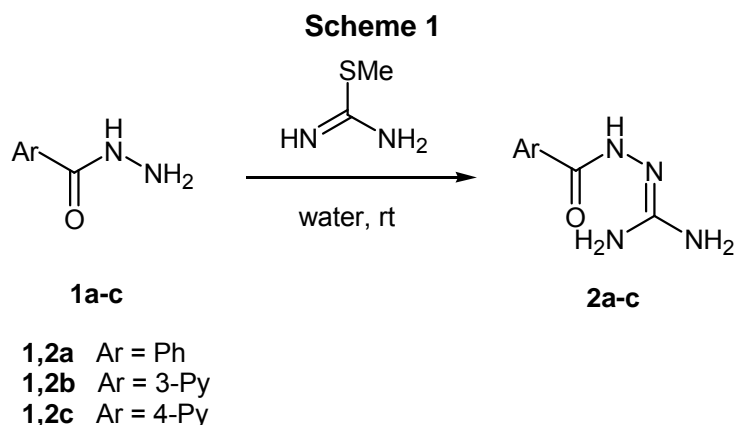
Heating several (het)aroylaminoguanidines, including **2a-c**, at 220-250 °C has been reported to give 5-amino-3-(het)aryl-1,2,4-triazoles as cyclization products [3-5]. The intramolecular cyclization of **2a** with refluxing sodium ethoxide has been found to afford the formation of 5-amino-3-phenyl-1,2,4-triazole (**3a**) in low yield [3].

It has been shown that using water as a solvent may provide several advantages in many organic chemistry processes [6,7].

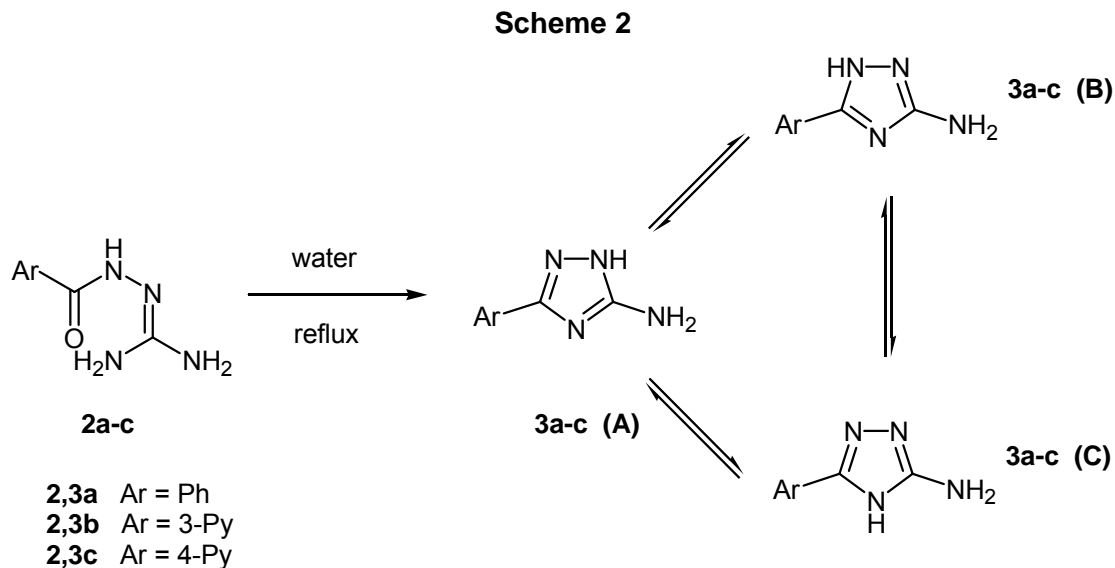
In this report we describe a new effective and clean method of preparation of 5-amino-3-(het)aryl-1,2,4-triazoles (**3a-c**) from **2a-c** in aqueous medium. The annular tautomerism in the obtained triazoles (**3a-c**) was investigated using NMR spectral data.

Results and discussion

The reaction of benzhydrazide (**1a**) with methyl isothiourea (Scheme 1) according to the reported method [3] was used for the preparation of benzamidoguanidine (**2a**). Nicotinamidoguanidine (**2b**) and isonicotinamidoguanidine (**2c**) were synthesized analogously from the appropriate hydrazides (**1b,c**).



When the (het)arylamino-guanidines (**2a-c**) were heated in water (Scheme 2), 5-amino-3-(het)aryl-1,2,4-triazoles (**3a-c**) were obtained in almost quantitative yields (97-98%). The reaction was found to be clean and afforded products (**3a-c**) with satisfactory purity. Interestingly, heating **2a-c** at 140 °C without solvent for 24 h did not result in the formation of triazoles **3a-c**.



Annular tautomerism is possible in the prepared triazoles (**3a-c**); they may exist in three tautomeric forms (**A**, **B** and **C**). A study of triazoles (**3a-c**) in DMSO solution using NMR spectroscopy concluded that the tautomer **A** dominated in the equilibrium. The tautomer **B** was found to be a minor, whereas the form **C** was not present to any measurable extent. The characteristic signals of ^1H NMR spectra of tautomers **A** and **B** as well as the tautomeric equilibrium constant (K_T) and the relative Gibbs free energies (ΔG_{298}) of individual tautomers.

TableTautomerism in 5(3)-amino-3(5)-(het)aryl-1,2,4-triazoles (**3a-c**)

Compound	¹ H NMR signals of tautomeric forms A and B in DMSO- <i>d</i> ₆ , ppm				K _T	ΔG ₂₉₈ , kJ mol ⁻¹
	3(5)-NH ₂		H-N(1)			
	A	B	A	B		
2a	6.12	5.39	12.09	13.24	9	-5.4
2b	6.21	5.43	12.24	13.44	20	-7.4
2c	6.28	5.52	12.40	13.69	33	-8.7

In the conclusion, the cyclization of (het)aroylaminoguanidines (**2a-c**) in the aqueous medium seems to be an effective method for preparation of 5-amino-3-(het)aryl-1,2,4-triazoles (**3a-c**) which are predominant tautomers in the equilibrium with 3-amino-5-(het)aryl-1,2,4-triazoles.

Experimental

Melting points (uncorrected) were determined on a Gallenkamp melting point apparatus. ¹H and ¹³C NMR spectra were recorded on a Bruker DPX-300 spectrometer, using DMSO-*d*₆ as a solvent and TMS as an internal reference.

(Het)aroylaminoguanidines (**2a-c**) were prepared from hydrazides of appropriate acids (**1a-c**), methyl isothioureia sulfate and sodium hydroxide in water according to the reported method [3].

Benzamidoguanidine (2a)

Yield 86%, mp 176 °C.

¹H NMR (300 MHz, DMSO-*d*₆): δ 6.97 (2H, br. s, NH₂), 7.16 (2H, br. s, NH₂), 7.29 (3H, m, W_{1/2} = 9 Hz, H-3, H-4 and H-5), 7.95 (2H, m, W_{1/2} = 12 Hz, H-2 and H-6), 11.01 (1H, br. s, NH).

¹³C NMR (75 MHz, DMSO-*d*₆): δ 126.5 (C-3 and C-5), 127.3 (C-2 and C-6), 128.0 (C-4), 138.5 (C-1), 152.9 (N=C(NH₂)₂), 160.5 (C=O).

Nicotinamidoguanidine (2b)

Yield 72%, mp 211-212 °C.

¹H NMR (300 MHz, DMSO-*d*₆): δ 7.08 (2H, s, NH₂), 7.12 (2H, s, NH₂), 7.33 (1H, dd, J = 7.5, 4.9 Hz, H-5), 8.24 (1H, dt, J = 7.9, 1.9 Hz, H-4), 8.49 (1H, dd, J = 4.5, 1.5 Hz, H-6), 9.14 (1H, dd, J = 1.1 Hz, H-2), 10.86 (1H, br. s, NH).

¹³C NMR (75 MHz, DMSO-*d*₆): δ 122.6 (C-5), 133.6 (C-3), 133.7 (C-4), 148.2 (C-6), 148.7 (C-2), 152.9 (N=C(NH₂)₂), 158.7 (C=O).

Isonicotinamidoguanidine (2c)

Yield 76%, mp 270-271 °C.

¹H NMR (300 MHz, DMSO-*d*₆): δ 6.97 (2H, s, NH₂), 7.08 (2H, s, NH₂), 7.84 (2H, dd, J = 4.5, 1.5 Hz, H-3 and H-5), 8.50 (2H, dd, J = 4.5, 1.5 Hz, H-2 and H-6), 10.66 (1H, br. s, NH).

^{13}C NMR (75 MHz, $\text{DMSO-}d_6$): δ 121.1 (C-3 and C-5), 146.0 (C-4), 149.0 (C-2 and C-6), 152.9 ($\text{N}=\text{C}(\text{NH}_2)_2$), 158.7 (C=O).

5-Amino-3-(het)aryl-1,2,4-triazoles (**3a-c**)

(Het)aroylaminoguanidines (**2a-c**) (0.02 mole) were heated under reflux in 15 ml of water for 3-5 h. After cooling, the precipitated triazoles (**3a-c**) were filtered, washed with ice-cold water and dried. The prepared compounds were sufficiently pure and could be used without further purification. After recrystallization from water mp of the compounds did not change.

5-Amino-3-phenyl-1H-1,2,4-triazole (**3a**)

Yield 97%, mp 186-187 °C (lit. mp. 186-187 °C [3]).

^1H NMR (300 MHz, $\text{DMSO-}d_6$): δ 6.12 (2H, s, NH_2), 7.28-7.52 (3H, m, H-3', H-4' and H-5'), 7.92 (2H, d, $J = 7.9$ Hz, H-2' and H-6'), 12.09 (1H, s, NH).

^{13}C NMR (75 MHz, $\text{DMSO-}d_6$): δ 126.3 (C-3' and C-5'), 128.1 (C-4'), 128.3 (C-2' and C-6'), 132.3 (C-1'), 157.3 (C5), 158.3 (C3).

5-Amino-3-pyridin-3-yl-1H-1,2,4-triazole (**3b**)

Yield 98%, mp 224-225 °C (lit. mp. 223 °C [4]).

^1H NMR (300 MHz, $\text{DMSO-}d_6$): δ 6.21 (2H, s, NH_2), 7.44 (1H, dd, $J = 7.5, 4.9$ Hz, H-5), 8.19 (1H, dt, $J = 7.9, 1.9$ Hz, H-4), 8.55 (1H, d, $J = 4.1$ Hz, H-6), 9.07 (1H, dd, $J = 1.5$ Hz, H-2), 12.24 (1H, s, NH).

^{13}C NMR (75 MHz, $\text{DMSO-}d_6$): δ 123.6 (C-5'), 127.8 (C-3'), 132.4 (C-4'), 146.5 (C-6'), 149.0 (C-2'), 156.1 (C-3), 157.5 (C-5).

5-Amino-3-pyridin-4-yl-1H-1,2,4-triazole (**3c**)

Yield 98%, mp 272-274 °C (lit. mp. 273-274 °C [4,5]).

^1H NMR (300 MHz, $\text{DMSO-}d_6$): δ 6.28 (2H, s, NH_2), 7.81 (2H, dd, $J = 4.9, 1.5$ Hz, H-3 and H-5), 8.62 (2H, dd, $J = 4.9, 1.5$ Hz, H-2 and H-6), 12.40 (1H, s, NH).

^{13}C NMR (75 MHz, $\text{DMSO-}d_6$): δ 119.5 (C-3 and C-5), 139.2 (C-4), 149.9 (C-2 and C-6), 156.4 (C-3), 157.6 (C-5).

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