



The use of the Stille cross-coupling reaction in the prenylation of naphthoquinones

Pascal Habonimana, Sven Claessens and Norbert De Kimpe

*Department of Organic Chemistry, Faculty of Bioscience Engineering,
Ghent University, Belgium*

pascalhabonim@yahoo.fr

sv.claessens@ugent.be

Abstract: Prenylation reactions are important reactions in natural product chemistry. The Stille reaction was evaluated for the prenylation of naphthoquinones and applied to the synthesis of the natural product 2,3-bis-(3-methylbut-2-enyl)-1,4-naphthoquinone.

Keywords: naphthoquinones, prenylation, Stille reaction, *Tabebuia guayacan*

Introduction

Prenylation reactions are important reactions in the chemistry of quinones. Important examples of prenylquinones are vitamin K derivatives which play an important role in the vital body as hemostatic vitamins. Other prenylquinones are coenzyme Q derivatives which are useful as therapeutic agents for ischemic heart diseases.¹ The unsubstituted prenylnaphthoquinone **1**, known as desoxylapachol **1** was isolated from *Tectona grandis*, the real teak wood. The compound causes a dermatitis allergy and gives the wood resistance to termites.^{2,3} Desoxylapachol **1** was also isolated from a New Zealand brown alga *Landsburgia quercifolia*. It was shown that this compound is active against P-388 leukemia cells (IC₅₀ 0.6 µg/ml) and fungi.⁴ The double prenylated naphthoquinone **2** was isolated from *Tabebuia guayacan*.⁵ Besides these examples the prenylquinones are the obvious precursors of important natural products, e.g. lapachol **3** and its derivatives. As a result, research was conducted to an efficient prenylation reaction.

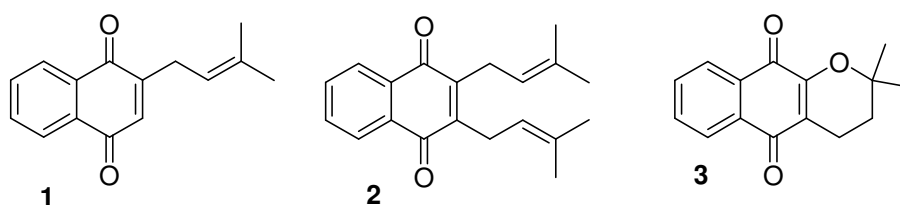
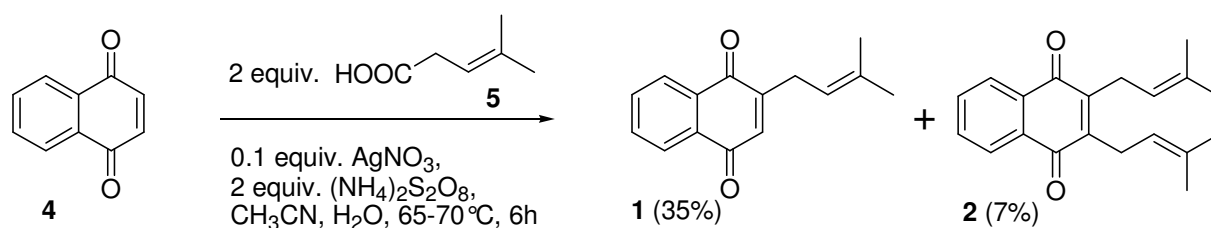


Figure 1

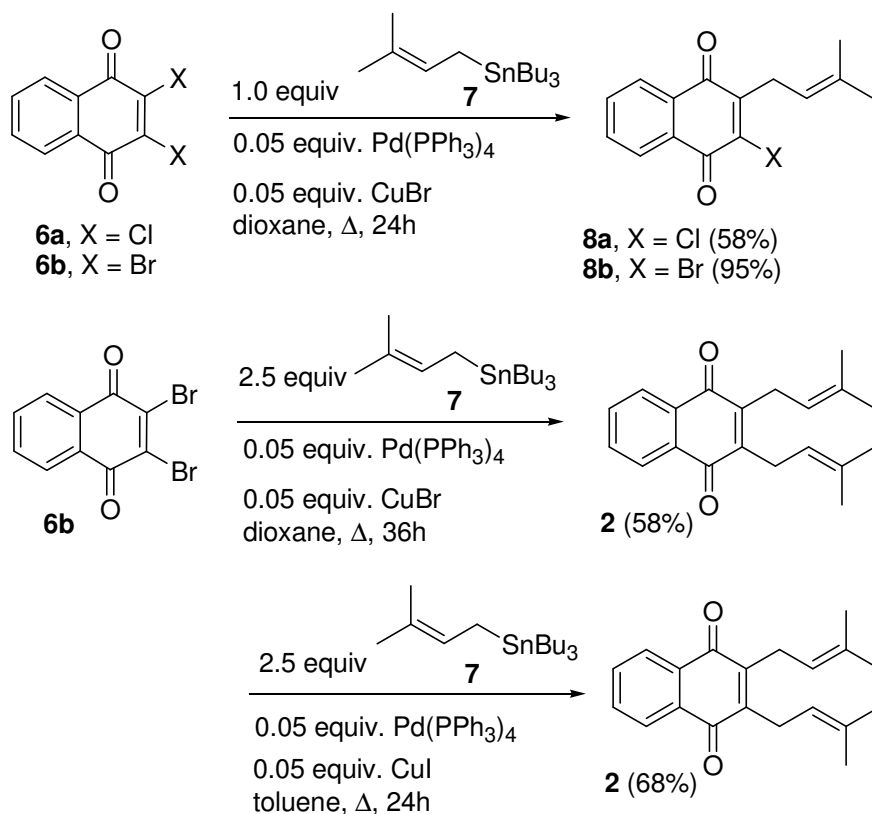
Results and discussion

The radical prenylation of 1,4-naphthoquinone **4** with 4-methyl-3-pentenoic acid **5** in the presence of silver nitrate and ammonium persulfate was reported previously to afford 2-prenyl-1,4-naphthoquinone **1** in 58% yield.⁶ However, in our hands using the published procedure, mixtures of starting material together with the monoprenylated **1** and the diprenylated 1,4-naphthoquinone **2** were found in 35% and 7% yield, respectively (Scheme 1). As our interest in the double prenylated naphthoquinone arose, a series of reactions was performed in order to improve the formation of the diprenylated compound. Increasing the amount of 4-methyl-pent-3-enoic acid gave no satisfactory improvements.



Scheme 1

The Pd-catalyzed Stille cross-coupling of a prenylstannane with an arylhalide has become popular in the recent past.⁷ In contrast to most other allylmetallic reagents the 3,3-



Scheme 2

disubstituted allylstannanes tend to react without allylic inversion or rearrangement in coupling reactions.⁸ Performing the reaction on the dibrominated compounds (Scheme 2) **6b** with only one equivalent of the organostannane **7** resulted in a 95% yield of the monoprenylated compound **8b**, without an indication of the diprenylated compound **2**. This result made clear that the diprenylation is a difficult reaction that requires more severe reaction conditions. In a first attempt, 2.5 equivalents of the organostannane **7** were used and the reaction time was prolonged to 36 hours. This resulted in a 56% yield of the diprenylated compound **2**. When changing from dioxane to toluene and using CuI instead of CuBr, the yield increased to 68%.

In conclusion, it can be stated that the Stille cross coupling reaction is a useful reaction for the prenylation of halogenated naphthoquinones and the application of this reaction resulted in the synthesis of the natural product 2,3-bis-(3-methylbut-2-enyl)-1,4-naphthoquinone **2**.

References

- ¹ Hamamura, K.; Iwama, T.; Seki, C.; Konishi, M. Eur. Pat. Appl. EP 613877 [EP 94-102979 19940228] (1994); *Chem. Abstr.* **1995**, 122, 229207.
- ² Dietrichs, H. H. *Naturwissenschaften*, **1964**, 51, 408-9.
- ³ Schmalle, H. W.; Hausen, B. M. *Naturwissenschaften*, **1984**, 71, 581-2.
- ⁴ Perry, N. B.; Blunt, J. W.; Munro, M. H. G. *J. Nat. Prod.* **1991**, 54, 978-85.
- ⁵ Manners, G. D.; Jurd, L. *Phytochemistry* **1976**, 15, 225-6.
- ⁶ Jacobsen, N.; Torsell, K. *Acta Chem. Scand.* **1973**, 27, 3211-3216.
- ⁷ (a) Jung, Y.-S.; Joe, B.-Y.; Seong, C.-M.; Park, N.-S. *Bull. Korean Chem. Soc.* **2000**, 21, 463-464. (b) Takemura, S.; Hirayama, A.; Tokunaga, J.; Kawamura, F.; Inagaki, K.; Hashimoto, K.; Nakata, M. *Tetrahedron Lett.* **1999**, 40, 7501-7505.
- ⁸ Kaiser, F.; Schmalz, H.-G. *Tetrahedron*, **2003**, 59, 7345-7355.