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# Essential oil composition and glandular trichome structure of the weather prophet *Dimorphoteca pluvialis*

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Proceedings

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Abstract: Dimorphoteca pluvialis (L.) Moench, usually known as weather prophet, African daisy, or 12 Cape marigold, is an Asteraceae commonly found in gardens due to its appealing white to yellowish 13 flowers. Recently, its use as a non-food oilseed crop has been investigated due to the high amounts 14 of dimorphecolic acid (Δ9-hydroxy,10t,12t-octadecadienoic acid), a highly reactive C18 fatty acid 15 with value for the manufacture of paints, inks, lubricants, plastic and nylon. However, information 16 on the essential oil (EO) composition of its plant tissues is scarce. The present work focused on char-17 acterizing the glandular trichomes, the main site for secretion of natural products, of shoots and 18 sepals and analysing the EO composition of shoots and flowers of *D. pluvialis*, extracted by hydro-19 distillation for 15, 30 or 60 minutes. Shoot surface displayed sharp and elongated non-glandular 20 protection trichomes, while the sepals additionally showed shorter and wider non-glandular tri-21 chomes. A capitate trichome with a biseriate peduncle and a multiseriate head was the only type of 22 glandular trichome identified. A histochemical analysis of the glandular head revealed the presence 23 of acid lipids, terpenic and phenolic compounds. The extracted EOs showed high amounts of trans-24 2-hexenal, a C6 aldehyde that protects plants against harmful substances, but is considered toxic for 25 humans. This study described, for the first time, the composition of EOs of D. pluvialis plants. 26

Keywords: Dimorphoteca pluvialis; essential oil; non-food crops; trans-2-hexenal; trichomes

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**Copyright:** © 2022 by the authors. Submitted for possible open access publication under the terms and conditions of the Creative Commons Attribution (CC BY) license (https://creativecommons.org/licenses/by/4.0/). 1. Introduction

Dimorphoteca pluvialis Moench, usually known as weather prophet, African daisy, or 30 Cape marigold, is an Asteraceae believed to be native to South Africa and Namibia. It 31 owes its name to the fact that its flowers close at night and on cloudy days before raining. 32 The aerial parts of this annual species form a bushy plant (up to 30 cm), with shoots 33 branching from the base and holding oblanceolate leaves (ca. 7 cm long), lobed to toothed, 34 that are numerous at the base of the stems and fewer and smaller near the top. The plant 35 is covered with large flower heads that blossom at the same level, with a white appearance 36 except near the base, where they have a dark purple or violet section. The flower heads 37 are composed of fertile female and sterile male ray florets, and hermaphrodite disk florets. 38 The seeds (achenes) developing from disk florets have flattened margins (wings), while 39 those produced by the ray florets are unwinged [1,2]. In the past decades, non-food 40 oilseed crops have garnered interest for industrial use due to the extracted oils containing 41 compounds with functional groups that makes them potential substitutes of the mineral 42 oils used to produce e.g., lubricants, surfactants, coatings or polymers; with the added 43 advantage that these can be supplied at a constant and more economical rate. This is the 44 case for D. pluvialis, whose seed oil can be composed by more than 60 % of dimorphecolic 45

acid (9-hydroxy-*trans*, *trans*-10, 12-octadecadienoic acid), a valuable C<sub>18</sub> fatty acid that contains a C9 hydroxyl group, two conjugated double bonds relative to the  $\alpha$ -carbon of the hydroxy group ( $\Delta$ 10, $\Delta$ 12) and a trans- $\Delta$ 12 unsaturation, setting this compound apart from other plant hydroxy fatty acids and granting it the potential for a wide range of new applications [3,4].

In the present work, *D. pluvialis* flowers and vegetative shoots were analyzed for the structural and chemical characterization of their glandular trichomes and the chemical profiling of their essential oils.

# 2. Material and Methods

#### 2.1. Plant Material

Aerial parts of D. pluvialis in the flowering stage were collected from the vicinity of11Campo Grande, Lisbon, in the spring. The flowers were isolated from the shoot tissues12and immediately processed to be used for structural analysis and essential oil extraction.13A voucher specimen is kept in the Herbarium of the Botanical Garden of Lisbon University, Lisbon, Portugal.14

## 2.2. Structural and Chemical Characterization of Glandular Trichomes

Longitudinal and cross sections were obtained from shoots and sepals. Structural 17 characterization of glandular trichomes was performed through scanning electron micros-18 copy (SEM) and light microscopy (LM). For SEM, samples were fixed with 1.5% (v/v) glu-19 taraldehyde in 0.05 M sodium cacodylate buffer, pH 7.0 for 45 min at room temperature. 20 After 1-2 min under vacuum (26 mm Hg, 3.46 kPA), the fixative was substituted with 3% 21 glutaraldehyde in 0.1 M sodium cacodylate buffer, pH 7.0 for 2 h at room temperature. 22 The material was rinsed thoroughly in the same buffer, post-fixed with a 2% osmium te-23 troxide ( $OsO_4$ ) aqueous solution for 2 h at room temperature, dehydrated in a graded ac-24 etone series and critical point dried in a Polaron E 3500. Dried specimens were mounted 25 on stubs, coated with gold in a Polaron E 5350. Observations were carried out on a JEOL 26 T220 SEM (JEOL Ltd., Tokyo, Japan) at 15 kV. 27

For the chemical characterization of the glandular trichomes, longitudinal and cross sec-28 tions of the aerial parts were stained with Sudan black B, Sudan IV, and Nile blue A for 29 total lipids, Nadi reagent for terpenoids, periodic acid-Schiff (PAS) reagent for polysac-30 charides with vicinal glycol groups, iron (III) trichloride (FeCl<sub>3</sub>) and potassium dichro-31 mate for phenolic compounds and Ruthenium red for pectins ([5] and references therein). 32 Observations were made under a Leica DM-2500 microscope (Leica Microsystems CMS 33 GmbH, Wetzlar. Germany). The images were digitally obtained using a Leica DFC-420 34 camera (Leica Microsystems Ltd., Heerbrugg, Switzerland) and the Leica Application 35 Suite software (version 2.8.1). 36

## 2.3. Essential oil extraction and analysis

Essential oils (EOs) were obtained by hydrodistillation of shoots or flowers of the aerial parts of *D. pluvialis*, in a Clevenger-type apparatus according to the European Pharmacopoeia [6] for 15, 30 or 60 min at a distillation rate of 3 mL/min. When EO yield was below 0.05 %, distilled *n*-pentane was used to collect the volatiles. Samples were stored in glass vials at -20 °C until analysis.

Samples were analyzed by gas chromatography (GC), for component quantification, 43 and gas chromatography coupled to mass spectrometry (GC-MS) for component identifi-44 cation. Gas chromatographic analyses were performed using a Perkin Elmer Autosystem 45 XL gas chromatograph (Perkin Elmer, Shelton, CT, USA) equipped with two flame ioni-46 zation detectors (FIDs), a data handling system, and a vaporizing injector port into which 47 two columns of different polarities were installed: a DB 1 fused-silica column (30 m × 0.25 48 mm i.d., film thickness 0.25 µm; J & W Scientific Inc., Rancho Cordova, CA, USA) and a 49 DB-17HT fused-silica column (30 m × 0.25 mm i.d., film thickness 0.15 µm; J & W Scientific 50

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Inc., Rancho Cordova, CA, USA). Oven temperature was programmed to increase from 1 45 to 175°C, at 3°C / min increments, then up to 300°C at 15°C / min increments, and finally 2 held isothermal for 10 min. Gas chromatographic settings were as follows: injector and 3 detectors temperatures, 280°C and 300°C, respectively; carrier gas, hydrogen, adjusted to 4 a linear velocity of 30 cm/s. The samples were injected using a split sampling technique, 5 ratio 1:50. The volume of injection was 0.1  $\mu$ L of a pentane-EO solution. The percentage 6 composition of the oils was computed by the normalization method from the GC peak 7 areas, calculated as a mean value of two injections from each volatile oil, without response 8 factors. 9

The GC-MS unit consisted of a Perkin Elmer Autosystem XL gas chromatograph, 10 equipped with DB 1 fused-silica column (30 m  $\times$  0.25 mm i.d., film thickness 0.25  $\mu\text{m};$  J & 11 W Scientific, Inc., Rancho Cordova, CA, USA) interfaced with Perkin-Elmer Turbomass 12 mass spectrometer (software version 4.1, Perkin Elmer). GC-MS settings were as follows: 13 injector and oven temperatures were as above; transfer line temperature, 280°C; ion source 14 temperature, 220°C; carrier gas, helium, adjusted to a linear velocity of 30 cm/s; split ratio, 15 1:40; ionization energy, 70eV; scan range, 40-300 u; scan time, 1 s. The identity of the com-16 ponents was assigned by comparison of their retention indices relative to C8-C25 *n*-alkane 17 indices, and GC-MS spectra from a laboratory made library based upon the analyses of 18 reference EOs, laboratory-synthesized components, and commercially available stand-19 ards. 20

# 2.4. Data Treatment and Statistical Analysis

Statistical analysis was performed with SPSS version 29 statistics software. Statistical22significance was determined with one-way ANOVA, and individual means were compared using the Tukey's Post-Hoc test with p < 0.05. Results were presented as mean  $\pm$ 23standard error (SE) of 6 samples.25

#### 3. Results and Discussion

In *D. pluvialis*, the indumentum showed mostly a uniform distribution of glandular 27 and non-glandular trichomes in the floral and vegetative parts (including the abaxial and 28 adaxial leaf surfaces). Non-glandular trichomes were singular multicellular structures 29 with a pointed tip, devoid of any pigmentation. Morphometric differences were found 30 between non-glandular trichomes of the sepals and the shoot stems. In the sepals, non-31 glandular trichomes were long and uniseriate (with 2 to 7 stacked cells), slightly pointing 32 towards the sepal tip, with an average height of  $160.3\pm12.0 \,\mu\text{m}$  and width of  $14.5\pm1.0 \,\mu\text{m}$ , 33 preferentially distributed in the margins of the sepal. In the stem, trichomes were multi-34 seriate with three columns of 2 to 6 stacked cells and showed an average 78.3±2.6 µm in 35 height and 57.6±1.2 µm in width. The glandular trichomes observed in sepals were capi-36 tate-type multicellular structures composed of a biseriate stalk and a multiseriate globoid 37 glandular head capping the products of secretion under the subcuticular space (Fig. 1 and 38 2). These secretory structures were  $141.9\pm8.7 \mu m$  in height (4 to 7 stacked cells) and 39 43.4 $\pm$ 1.4  $\mu$ m in width (Fig. 1a and b). In the stems, the same type of capitate glandular 40 trichome was observed, however, with lower height (101.2 $\pm$ 3.2 µm) and width (37.1 $\pm$ 1.6 41 μm) (Fig. 1c and d). The glandular head was made up of 2 to 3 cell layers, where the second 42 and third cell layers showed a high chloroplast content, in contrast to the apical cells (Fig. 43 2a), whose function was, probably, to help provide the carbon and energy needs for spe-44 cialized metabolite production, since trichomes are believed to harbor specific Rubisco 45 isoforms uniquely adapted to the physiology of secretory cells [7]. The produced second-46 ary metabolites are, most likely, accumulated in the apical cells (first cell layer) from where 47 they can be exuded into the subcuticular space and then volatilized through micropores 48in the cuticle surface or released after cuticle rupture. Glandular trichomes are sites of 49 substantial production in secondary metabolites whose functions can span from the 50

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regulation of plant growth to the defense of the plant against pathogens or even other 1 plants [8].

Figure 1. Scanning electron micrographs depicting the distribution of glandular trichomes in sepals4(a), detail of glandular capitate-type trichomes of sepals (b) and vegetative shoots (c and d) of Di-5morphotheca pluvialis. Bar = 15 µm.6

In the present study, the chemical nature of the exudate was assessed through histo-7 chemical assays. Lipidic compounds were detected in the glandular head after a positive 8 result for the Sudan IV (Fig. 2b) or Sudan Black B (Fig. 2c) staining, with higher incidence 9 in the first cell layer. Through the Nile blue A staining, a strong blue colour in the cells of 10 the glandular head indicate that some of the identified lipids can be considered acidic in 11 nature, while the faint pink color in the trichome stem revealed the presence of neutral 12 lipids (Fig. 2d). The use of Nadi reagent allowed the identification of terpenic compounds 13 in the exudate of the subcuticular space (Fig. 2e). The presence of phenolic compounds 14 was confirmed with the iron (III) trichloride (FeCl3) or potassium dichromate staining. The 15 first stained intensely the cells of the glandular head (Fig. 2f) while the second stained 16 mainly the apical cells where the metabolites are accumulated before secretion to the sub-17 cuticular space (Fig. 2g). The Periodic Acid-Schiff reagent (PAS), used to detect polysac-18 charides, stained of the trichome cells (stem and glandular head) but not the subcuticular 19 space. Additionally, the Ruthenium Red, differential staining for pectins, stained the glan-20 dular head, suggesting a mucilaginous nature for the compounds accumulated in the 21 glandular head. 22

The EOs extracted from *D. pluvialis* shoots or flowers showed very low yields (≤0.05 23 %, v/f.w.), generally not reaching the lowest measure scale of the Clevenger apparatus. 24

The non-terpenic fraction was dominant in all extracted EOs, varying from 65.5±1.4 (30 1 min) to 77.5±3.3 % (60 min), for the shoots, and from 82.0±3.8 (60 min) to 91.5±1.5 % (15 2 min), for the flowers (Table 1). This fraction contained the main EO compounds ( $\geq 10 \%$ ), 3 namely, the aldehyde 2-trans-hexenal, that varied from 23.0±0.1 (15 min) to 28.9±0.8 % (60 4 min), in the shoots, and from 22.3±3.7 (60 min) to 56.3±4.1 (15 min), in the flowers; the 5 alcohol cis-3-hexen-1-ol, that varied from 16.4±1.0 (30 min) to 26.7±0.3 % (60 min), in the 6 shoots, and 3.5±1.7 (15 min) to 18.6±2.4 (60 min), in the flowers; and the alcohol hexanol, 7 that varied from 16.2±1.6 (60 min) to 21.6±0.2 % (15 min), in the shoots, and from 11.4±2.0 8 (30 min) to 18.4±2.9 (60 min) in the flowers. Overall, the increase in distillation time led to 9 a relative increase in the proportions of 2-trans-hexenal and cis-3-hexen-1-ol and a de-10 crease in hexanol in the shoots, but in the flowers, 2-trans-hexenal steeply decreased while 11 *cis*-3-hexen-1-ol increased. 2-*trans*-Hexenal is a medium chain aldehyde (C<sub>6</sub>H<sub>10</sub>O) known 12 as a potent odorant, generally the product of enzymatic oxidation of unsaturated fatty 13 acids. In plants, its protective effect is linked to the induction of biological defense re-14 sponses, however, it appears to be toxic to humans [9,10]. 15



Figure 2. Light microghaphs of glandular capitate-type trichomes (a) stained with Sudan IV (b) and17Sudan Black B (c), for the detection of total lipids; Nile blue A (d), for acidic lipids; Nadi reagent (e),18for the identification of terpenes; iron (III) trichloride (f) and potassium dichromate (g), for phenolic19compounds; Periodic Acid-Schiff reagent (h), for the detection of polysaccharides; and Ruthenium20Red (i) for pectins. Bar = 25 μm.21

Together with *cis*-3-hexen-1-ol and hexenol, 2-*trans*-hexenal belongs to the green leaf22volatiles (GLV), six-carbon long aldehydes, esters, and alcohols that are released by plants23upon attack and function on the activation of the biochemical mechanisms of biological24defense and resistance [9]. The terpene fraction occurred in low relative amounts, that25varied from  $8.1\pm0.3$  (60 min) to  $10.6\pm1.0$  % (15 min), in the shoots, and  $8.9\pm1.3$  (15 min) to26 $19.4\pm2.2$  (30 min), in the flowers. The monoterpene hydrocarbon fraction showed higher27

proportions (from 4.8±0.1, at 30 min, to 5.7±0.2 %, at 15 min, in the shoots; and from 6.4±0.9, 1 at 15 min, to 12.5±1.9 %, at 60 min, in the flowers) than the oxygen-containing monoter-2 penes (from 0.1±0.1, at 30 min, to 0.2±0.0 %, at 60 min, in the shoots; and from 0.3±0.1, at 3 15 min, to 0.6±0.1 %, 60 min, in the flowers). The dominant monoterpene hydrocarbons 4 were sabinene with proportions that varied from 2.1±0.3 (30 min) to 2.4±0.4 % (60 min), in 5 the shoots, and  $1.5\pm0.4$  (30 min) to  $4.1\pm0.2$  (60 min), in the flowers; and  $\beta$ -pinene with pro-6 portions that varied from 0.9±0.4 (30 min) to 2.3±0.0 % (15 min), in the shoots, and 1.3±0.3 7 (30 min) to 1.6±0.3 (60 min), in the flowers. For the sesquiterpenes, hydrocarbon propor-8 tions had lower relative amounts (from 0.4±0.2, at 60 min, to 0.6±0.1 %, at 15 min, in the 9 shoots; and from 0.9±0.1, at 60 min, to 3.1±0.7 %, 30 min, in the flowers) than oxygen con-10 taining molecules (from 2.7±0.1, at 60 min, to 6.0±1.9 %, at 30 min, in the shoots; and from 11 0.5±0.1, at 15 min, to 8.9±1.3 %, 30 min, in the flowers). The dominant oxygen-containing 12 sesquiterpenes were elemol with proportions that varied from 0.9±0.2 (60 min) to 1.3±0.5 13 % (15 min), in the shoots, and 0.1±0.1 (15 min) to 1.7±0.5 (30 min), in the flower; and  $\alpha$ -14 eudesmol with proportions that varied from 1.1±0.2 (60 min) to 2.3±0.7 % (30 min), in the 15 shoots, and 0.0±0.0 (15 min) to 2.0±0.7 (30 min), in the flowers. 16

**Table 1.** Composition of the essential oils extracted from the shoots or flowers of *Dimorphoteca plu-*17*vialis* through hydrodistillation with the duration of 15, 30 or 60 minutes. For each compound at18each parameter, values are presented as mean  $\pm$  standard error of 6 samples and the different letters19indicate statistically significant differences (p < 0.05).</td>20

Components	RI		Shoots			Flowers	
Time (min)		15	30	60	15	30	60
Octene	799	3.9±0.1a	1.7±0.1b	1.0±0.4b	1.6±0.2a	1.7±0.4a	1.5±0.4a
Hexanal	800	0.4±0.1b	1.7±0.2a	1.7±0.1a	1.3±0.1a	0.9±0.1a	1.2±0.2a
Octane	800	0.7±0.1b	1.3±0.0a	1.7±0.1a	1.2±0.2a	1.7±0.6a	1.1±0.2a
2-trans-Hexenal	866	23.0±0.1a	25.3±3.3a	28.9±0.8a	56.3±4.1a	34.2±3.5b	22.3±3.7b
cis-3-Hexen-1-ol	868	24.0±1.6a	16.4±1.0b	26.7±0.3a	3.5±1.7b	16.8±3.5a	18.6±2.4a
Hexanol	882	21.6±0.2a	18.8±1.1b	16.2±1.6b	18.4±0.9a	11.4±2.0a	18.4±2.9a
α-Thujene	924	0.2±0.1a	$0.0 \pm 0.0 b^1$	0.0±0.0b	0.3±0.1a	1.0±0.5a	1.5±0.7a
α-Pinene	930	0.2±0.0a	0.2±0.0a	0.2±0.0a	0.1±0.1a	0.5±0.1a	1.1±0.5a
Camphene	938	0.4±0.0a	0.2±0.0b	0.2±0.0b	0.1±0.0a	0.1±0.0a	1.7±1.2a
Sabinene	958	2.3±0.0a	2.1±0.3a	2.4±0.4a	3.9±0.3a	1.5±0.4b	4.1±0.2a
β-Pinene	963	2.3±0.0a	0.9±0.4b	1.0±0.4b	1.6±0.3a	1.3±0.3a	1.4±0.3a
Dehydro-1.8-cineole	973	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.1±0.0a	0.1±0.1a
2-Pentyl furan	973	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.1±0.0b	0.1±0.0b	0.7±0.3a
n-Octanal	973	0.0±0.0a	0.1±0.1a	1.1±0.6a	0.0±0.0a	0.2±0.1a	0.3±0.1a
β-Myrcene	975	0.0±0.0a	0.2±0.1a	0.1±0.1a	0.1±0.0a	0.6±0.2a	0.5±0.2a
$\alpha$ -Phellandrene	995	0.1±0.1b	0.5±0.1a	0.0±0.0b	0.0±0.0a	0.4±0.2a	0.5±0.1a
Benzene acetaldehyde	1002	0.1±0.0a	0.0±0.0a	0.1±0.1a	0.1±0.0a	0.1±0.1a	0.1±0.1a
α-Terpinene	1002	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a
<i>p</i> -Cymene	1003	0.1±0.0a	0.1±0.1a	0.1±0.1a	0.0±0.0b	0.2±0.1a	0.3±0.1a
β-Phellandrene	1005	0.0±0.0c	0.2±0.0a	0.1±0.0b	0.1±0.0b	0.2±0.1a	0.5±0.1a
Limonene	1009	0.1±0.1b	0.4±0.0a	0.5±0.0a	0.0±0.0b	0.5±0.2a	0.5±0.1a
γ-Terpinene	1035	0.1±0.0b	0.1±0.0b	0.2±0.0a	0.2±0.0a	0.2±0.1a	0.3±0.0a
Terpinolene	1064	0.0±0.0b	0.0±0.0b	0.2±0.1a	0.1±0.1a	0.3±0.1a	0.2±0.1a
<i>n</i> -Nonanal	1073	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0b	0.3±0.1a	0.2±0.1a
Terpinen-4-ol	1148	0.2±0.0a	0.2±0.1a	0.1±0.1a	0.3±0.1a	0.3±0.1a	0.5±0.1a
α-Terpineol	1159	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a
<i>n</i> -Decanal	1180	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.4±0.1a	0.4±0.1a
β-Damascenone	1356	0.0±0.0a	0.3±0.2a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.1±0.0a
n-Dodecanal	1397	0.0±0.0a	0.2±0.1a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a
β-Caryophyllene	1414	0.0±0.0a	0.0±0.0a	0.4±0.2a	0.8±0.2a	1.0±0.3a	0.6±0.2a

α-Humulene	1447	0.6±0.1a	0.5±0.3a	0.0±0.0a	0.4±0.3a	0.2±0.1a	0.1±0.0a
Bicyclogermacrene	1487	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.2±0.1a	1.9±0.9a	0.2±0.0a
Elemol	1530	1.3±0.5a	1.3±0.3a	0.9±0.2a	0.1±0.1b	1.7±0.5a	0.8±0.1a
Spathulenol	1551	0.2±0.1a	0.3±0.2a	0.0±0.0a	0.3±0.2b	0.9±0.2a	0.4±0.1b
β-Caryophyllene oxide	1561	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.1±0.0a
γ-Eudesmol	1609	0.7±0.2a	0.8±0.5a	0.3±0.1a	0.1±0.1b	1.0±0.2a	0.5±0.1a
β-Eudesmol	1620	0.7±0.1a	1.3±0.3a	0.5±0.0b	0.0±0.0b	1.4±0.2a	0.5±0.1a
α-Eudesmol	1634	1.3±0.4a	2.3±0.7a	1.1±0.2a	0.0±0.0b	2.0±0.7a	0.8±0.2b
% Identification		84.1±1.1a	77.3±3.5a	85.6±2.9a	91.5±1.5a	87.2±0.6ab	82.0±3.8b
Grouped compounds							
Monoterpene hydrocarbons		5.7±0.2a	4.8±0.1b	5.0±0.1b	6.4±0.9b	7.0±1.3b	12.5±1.9a
Oxygen-containing monoterpenes		0.2±0.0a	0.2±0.1a	0.1±0.0a	0.3±0.1a	0.4±0.1a	0.6±0.1a
Sesquiterpene hydrocarbons		0.6±0.1a	0.5±0.3a	0.4±0.2a	1.5±0.4ab	3.1±0.7a	0.9±0.1b
Oxygen-containing sesquiterpenes		4.2±1.1a	6.0±1.9a	2.7±0.1a	0.5±0.1b	8.9±1.3a	3.1±0.5b
C13 Norisoprenoid		0±0.0a	0.3±0.2a	0.0±0.0a	0.0±0.0a	0.0±0.0a	0.1±0.0a
Others		73.5±2.0b	65.5±1.4b	77.5±3.3a	82.5±2.4a	67.8±2.1b	64.9±5.5b
Yield (%, V / fresh weight)		< 0.05	< 0.05	< 0.05	< 0.05	< 0.05	< 0.05

<sup>1</sup> Values below quantification limits were considered as  $0.0\pm0.0$  for statistical treatment and can be referred to as compounds in trace amounts ( $\leq 0.01\%$ ).

# 4. Conclusion

The shoots of *D. pluvialis* are covered with glandular trichomes that produce and secrete mainly acid lipidic substances, terpenes and phenylpropanoids. The extracted essential oils did not show qualitative differences between shoots and flowers, however, the main compounds, 2-*trans*-hexenal, *cis*-3-hexen-1-ol and hexanol, varied with the duration of the hydrodistillation. The presence of dimorphecolic acid was not detected, suggesting it is restricted to the seeds or non-extractable by hydrodistillation. 9

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