

Synthesis, Structural, Morphological and Optical Characteristic insight of NbSe₂ Nanoparticles

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Abstract: The representative member of group VB TMDCs, NbSe₂ nanoparticles, have attracted significant attention due to their unique structural and optical properties, which make them promising candidates for various applications in nanoelectronics and optoelectronics. In this report, the author reports the synthesis of NbSe₂ nanoparticles via the sonochemical method at ambient temperature with controlled size, a well-defined crystal structure, and desirable optical properties. The investigation of the compositional and structural analysis revealed that the synthesized nanoparticles are well-defined, near stoichiometry, with a hexagonal crystal structure belonging to the space group P6₃mmc. The other morphological and optical characteristics of synthesized nanoparticles studied through Scanning Electron microscopy (SEM), Transmission Electron microscopy (TEM), Atomic Force Microscopy (AFM), UV-VIS NIR spectroscopy, etc. were discussed here.

Keywords: TMDCs Material; NbSe₂ Nanoparticles; Sonochemical Method; XRD and Optical Properties

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1. Introduction

Nanoparticles, characterized by their myriad distinctive and tunable properties at the nanoscale, have evolved into a compelling focal point of research within the discipline of materials science. Materials in the nano range, exhibit distinctive attributes that differentiate them from their macroscopic counterparts, which appealing candidates for a diverse array of technological applications. Transition metal dichalcogenides (TMDs), a diverse group of nanomaterials, have attracted a lot of interest lately.

The TMD family member NbSe₂ has become well-known because of its fascinating electrical, optical, and mechanical characteristics. Niobium (Nb) and selenium (Se) atoms are placed in a hexagonal lattice arrangement to form the layered substance [1,2]. NbSe₂, which is two-dimensional, exhibits extraordinary quantum confinement characteristics that have a wide range of uses in optoelectronics, nanoelectronics, catalysis, and energy storage [3,4]. The peculiar qualities of NbSe₂, including as its three-dimensionality, electrical, optical, and catalytic properties, contribute to it being a versatile substance with a variety of uses. The potential for the utilization of NbSe₂ nanoparticles (NPs) in advanced technologies and devices is expected to expand with ongoing research.

The creation of NbSe₂ NPs allows a special chance to investigate and modify the material's characteristics at the nanoscale. Each technique affects the size, shape, and crystallinity of the nanoparticles and has advantages and NbSe₂ NPs have been made employing

a variety of synthesis methods, such as wet chemical method, hydrothermal processes, and sol-gel procedures disadvantages of its own [5–9].

To fully grasp the promise of NbSe₂ NPs for practical applications, it is essential to comprehend their structural, morphological, and optical properties[10-12]. The crystal structure, phase purity, and crystallite size of the nanoparticles can be determined by structural characterization methods including X-ray diffraction (XRD). Researchers can see and measure the particle size, shape, and distribution utilizing morphological analysis using scanning electron microscopy (SEM) and transmission electron microscopy (TEM). To better understand the material's interaction with light and its potential in optoelectronic devices, it is also helpful to understand its optical properties, such as the bandgap and light absorption spectra. The aim of this study is to advance our knowledge of NbSe₂ at the nanoscale by offering insightful information about its synthesis, structural, morphological, and optical features.

2. Materials and Methods

2.1. Chemicals

Niobium chloride dihydrate (NbCl₅·2H₂O) [minimum assay 99%, AlfaAesar, United States], hydrochloric acid (HCl) [minimum assay 35%, HiMedia Laboratories Pvt. Ltd., Mumbai, India], tri-ethanol amine (TEA) [minimum assay 98%, Sisco Research Laboratories (SRL) Pvt. Ltd., India], sodium selenite (Na₂SeO₃) [minimum assay 98.5%, HiMedia Laboratories Pvt. Ltd., Mumbai, India] and Hydrazine Hydrate [minimum assay 80%, Sisco Research Laboratories Pvt. Ltd., New Mumbai, India] are used for synthesis.

2.2. Synthesis of NbSe₂ nanoparticles by sonochemical method

First Weight the 1 M of NbCl₅·2H₂O and add 5 mL of 50% dilute HCl to the beaker. Place the beaker in an ultrasonic wave generator and expose it to ultrasound waves for 15 minutes under continuous stirring. While continuing ultrasonic treatment and stirring, add 2.79 mL of TEA to the transparent solution obtained from step 1. Add the separately prepared solution dropwise to the sonochemically treated solution while maintaining ultrasound exposure. The solution contains Na₂SeO₃ solution by dissolving 2 M of Na₂SeO₃ in 10 mL of de-ionized water and then adding 4 mL of hydrazine hydrate to the solution. Adjust the total volume of the solution to 50 mL by adding double distilled water under continuous ultrasonic treatment. Allow the solution to undergo sonochemical treatment for an additional duration of 1 hour. The solution will exhibit a dark brown coloration attributable to the formation of nanoparticles. After completion of the sonochemical treatment, isolate the dark brown solution and allow it to cool down to room temperature. To get the NbSe₂ Nps, the dark brown precipitates that had accumulated at the bottom of the beaker were filtered using Whatman filter paper (Grade 5). Following multiple washing steps, they are left to dry for 8 hr. at atmospheric temperature, resulting in the production of dark brown NbSe₂ nanoparticles. To preserve the synthesized NbSe₂ nanoparticles within an appropriate air tight container at ambient conditions.

2.3. Characterizations

Stoichiometric elemental composition (EDS) and surface morphology of synthesized NbSe₂ NPs were characterized using Scanning electron microscope (SEM) attached with Nova Nano SEM-450, FEL, Ltd. (SICART, Vallabh Vidyanagar). The unit crystal structure is determined by XRD using Rigaku Ultima IV Powder X-ray diffractometer (SICART, Vallabh Vidyanagar) using Cu-K_α radiation. The surface topography of synthesized NbSe₂ NPs is observed using high resolution electron microscope (HRTEM) done with Thermo Scientific Talos F200i S/TEM (CSMCRI, Bhavnagar). The optical properties of synthesized NbSe₂ NPs were identified by Lambda 19 Perkin Elmer, UV-VIS NIR spectroscopy (SICART, Vallabh Vidyanagar).

3. Results and Discussion

3.1. Compositional and Structural Analysis:

3.1.1. Compositional Analysis

The synthesized NbSe₂ NPs can be qualitatively analysed using the non-destructive EDS approach. The elemental analysis helps to determine the any contaminants or phase purity of NbSe₂ NPs. High phase purity is necessary for consistent and repeatable outcomes in a variety of applications. Here, the elemental analysis of synthesized NbSe₂ NPs were analyzed by without any elemental restriction. The EDS spectrum of synthesized NbSe₂ NPs is shown in Fig. (1). The spectrum displays the intensity of X-rays emitted by the sample at various energy levels. Based on the diffraction peak profile, all of the observable peaks of Nb-Se are clearly visible and close to stoichiometric composition.

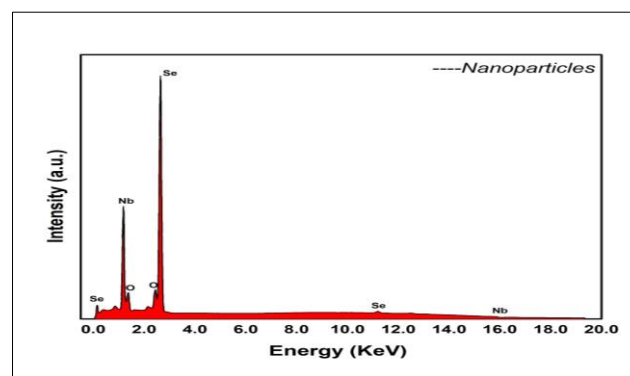


Figure 1. shows the EDS spectrum of synthesized NbSe₂ NPs synthesized via sonochemical method at room temperature.

3.1.2. Structural Analysis:

The X-ray diffraction profile of sonochemically synthesized NbSe₂ NPs X-ray diffraction in the 2θ range of 10° to 80° is depicted in Figure (2). The results showed that the hexagonal structure and high degree of crystallinity belonged to the $P6_3/mmc$ group. The determined lattice parameter was $a=b=3.446 \text{ \AA}$ and $c=12.55 \text{ \AA}$, and the diffraction peak profiles of all the peaks were well matched with the standard JCPDS No. 01-072-0864.

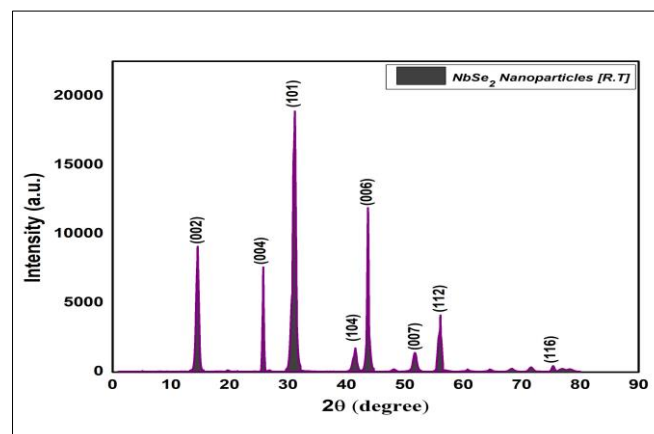


Figure 2. shows the XRD profile of synthesized NbSe₂ NPs synthesized via sonochemical method at room temperature.

This non-destructive technique reveals important details about the nanoparticle's atomic configuration and the determination of the crystal structure, lattice parameters, and phase purity. The foundation of XRD is Bragg's law, which describes how a crystal lattice bends X-rays. Crystallite size and lattice parameters are crucial structural characteristics of NbSe₂ NPs that significantly influence their properties and performance in various applications. This interference occurs only when the Bragg condition is met, which is given by [13],

$$n\lambda = 2d \sin\theta \quad (1)$$

Where n is the order of the diffraction peak, λ is the wavelength of the X-rays, d is the interplanar spacing of the crystal lattice, and θ is the diffraction angle. The Scherrer equation is commonly used to estimate the crystallite size (D) using the full-width at half-maximum (FWHM) of the diffraction peak [14],

$$D = K * \lambda / \beta * \cos\theta \quad (2)$$

Where K is the shape factor (typically taken as 0.89 for spherical nanoparticles), λ is the wavelength of X-rays, β is the FWHM of the diffraction peak, and θ is the diffraction angle. The average crystallite size from XRD analysis is found to be 15.12 nm.

The comprehensive structural characterization ensuing from the analysis of crystallite dimensions and lattice attributes has augmented our comprehension of the correlations between the synthesis parameters, structural attributes, and properties of NbSe₂ NPs.

3.2. Morphological and topographical analysis

3.2.1. Scanning Electron Microscope (SEM)

For morphological investigation and in order to comprehend more about the physical properties such as size, shape, and dispersion of these tiny objects of NbSe₂ NPs a high-resolution Scanning Electron Microscopy (SEM) were employed. Figure 3 depicts the surface morphology of synthesized NbSe₂ NPs. The image displaying the smooth surface and spherical nanoparticles. Additionally, they have a compact texture, without any pinholes or cracks and appear to be tightly packed nanoparticles have been formed

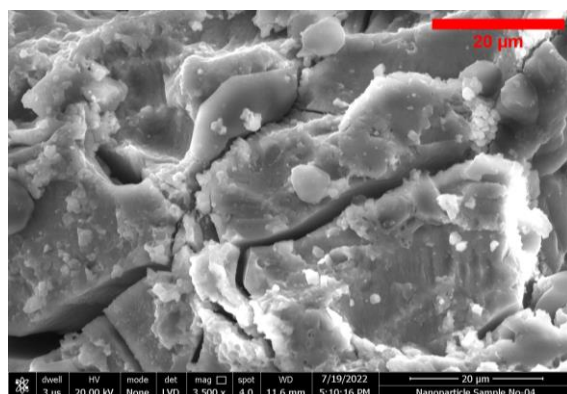


Figure 3. shows the surface morphology SEM image of synthesized NPs synthesized via sonochemical method at room temperature.

3.2.2. Transmission Electron Microscope (TEM)

In order to study about the topographical and internal makeup of synthesized NbSe₂ NPs, high resolution transmission electron microscopy (TEM) were utilized. Fig. 4(a) and 4(b) shows the topographic image and particle size distribution curve of NbSe₂ NPs respectively. The Fig. 4(a) demonstrates that nanoparticles are crystalline and have a spherical shape with an average particle size of 42.62 nm. Where the Fig. 4(b) displays the particle size distribution curve of NbSe₂ NPs, with a standard deviation of 0.271.

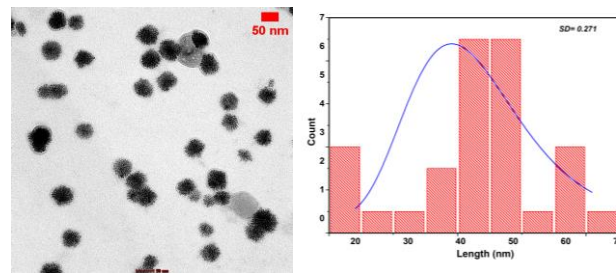


Figure 4. a) and 4(b) shows the TEM image and particle size distribution curve of NbSe₂ NPs synthesized via sonochemical method at room temperature.

3.3. Optical Analysis

UV-Vis spectroscopy is frequently used to characterize the optical properties of synthesized NbSe₂ NPs, where the intensity of the absorption is displayed as a function of wavelength. It offers intuitions into the bandgap and absorption of the nanoparticles. Here, for the absorption spectra analysis, the sample was prepared by being ultrasonically dispersed in acetone and recorded in the spectral wavelength of 800 to 1400 nm are showed in fig. 5(a).

The absorption spectra fig. 5(a) shows strong absorption in the spectral wavelength range of 850-950 nm. The absorption edges for synthesized NbSe₂ NPs are 910 nm. The bandgap of the synthesized NbSe₂ NPs has a direct bearing on this range. The intercept of the linear section of the plot with the energy axis is used to calculate the bandgap (E_g). Fig. 5(b) displays the optical energy bandgap as 1.418 eV for NbSe₂ NPs as calculated from the absorption spectrum using the near band edge absorption relation. Tauc plot analysis is one approach that is frequently used to find the bandgap. The absorption coefficient (α) as a function of the photon energy ($h\nu$) on a linear scale or $(h\nu)^2$ on a logarithmic scale is shown to create the Tauc graphic. The bandgap energy of 1.418 eV places the material in a favorable range for applications in photodetectors and solar cells. In the case of photodetectors, the material's bandgap falls within the visible spectrum, making it sensitive to a broad range of incident light. For solar cells, the 1.418 eV bandgap is well-suited for harvesting sunlight, as it corresponds to the energy range of visible light.

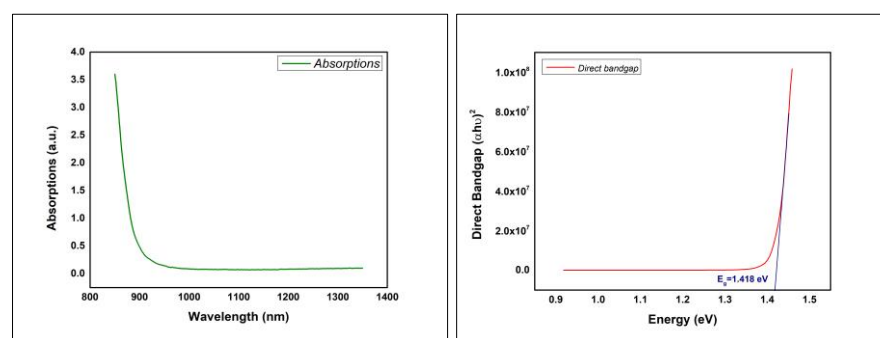


Figure 5. a) and 5(b) shows absorption and optical direct bandgap of synthesized NbSe₂ NPs synthesized via sonochemical method at room temperature.

4. Conclusion

In essence, this study advanced understanding of sonochemically synthesized NbSe₂ NPs at room temperature, providing valuable insights into their synthesis, structure, morphology, and optical properties. The analysis revealed a high phase purity of NbSe₂ NPs, laying the foundation for consistent performance in various applications. Structural investigations, including XRD analysis, elucidated the crystallographic arrangement of NbSe₂ NPs. The hexagonal structure with well-matched diffraction peak profiles underscored the nanoparticles high degree of crystallinity. Morphological intuitions from SEM

and TEM showcased the nanoparticles smooth, spherical morphology and compact arrangement, emphasizing their potential for diverse applications. The optical analysis shed light on the nanoparticles absorption capabilities within the spectral wavelength. The characterization of the material with a bandgap of 1.418 eV opens up exciting possibilities for its application in photodetectors and solar cells. The material's position within the visible spectrum range, coupled with its electronic properties, makes it a promising candidate for optoelectronic devices. Further research is warranted to fully exploit its potential in these applications.

References

1. A. Gupta, T. Sakthivel, and S. Seal. Recent development in 2D materials beyond graphene. *Prog. Mater. Sci.* 2015, vol. 73, pp. 44–126. doi: 10.1016/j.pmatsci.2015.02.002.
2. X. Zhang, Z. Lai, Q. Ma, and H. Zhang. Novel structured transition metal dichalcogenide nanosheets. *Chem. Soc. Rev.* 2018, vol. 47, no. 9, pp. 3301–3338. doi: 10.1039/c8cs00094h.
3. X. Wang, G. Sun, N. Li, and P. Chen. Quantum dots derived from two-dimensional materials and their applications for catalysis and energy. *Chem. Soc. Rev.* 2016, vol. 45, no. 8, pp. 2239–2262. doi: 10.1039/c5cs00811e.
4. S. Das, M. Kim, J. W. Lee, and W. Choi. Synthesis, properties, and applications of 2-D materials: A comprehensive review. *Crit. Rev. Solid State Mater. Sci.* 2014, vol. 39, no. 4, pp. 231–252. doi: 10.1080/10408436.2013.836075.
5. C. Suryanarayana and M. G. Norton. *X-Ray Diffraction, MA: Springer US. Boston* 1998. ISBN 978-1-4899-0150-7, doi: 10.1007/978-1-4899-0148-4.
6. R. Delhez, T. H. De Keijser, and E. J. Mittemeijer. Determination of Crystallite Size and Lattice Distortions through X-Ray Diffraction Line Profile Analysis Recipes, Methods and Comments. *Fresenius Z Anal Chem* 1982, vol. 312, pp. 1-16.
7. R. Das, S. S. Nath, and R. Bhattacharjee. Preparation of linoleic acid capped gold nanoparticles and their spectra. *Phys. E Low-Dimensional Syst. Nanostructures* 2010, vol. 43, no. 1, pp. 224–227. doi: 10.1016/j.physe.2010.07.008.
8. P. C. Dey and R. Das. Effect of silver doping on the elastic properties of CdS nanoparticles. *Indian J. Phys.* 2018, vol. 92, no. 9, pp. 1099–1108. doi: 10.1007/s12648-018-1214-4.
9. R. Yogamalar, R. Srinivasan, A. Vinu, K. Ariga, and A. C. Bose. X-ray peak broadening analysis in ZnO nanoparticles. *Solid State Commun.* 2009, vol. 149, no. 43–44, pp. 1919–1923. doi: 10.1016/j.ssc.2009.07.043.
10. Y. Liaoa, L. Wua, Y. Shana, X. Daia, H. Caia, Y. Xiang, and Dn Fana. Nonlinear optical response, all optical switching, and all optical information conversion in NbSe₂ nanosheets based on spatial self-phase modulation. *Nanoscale* 2019, vol. 11, no. 10, pp. 4515–4522. doi: 10.1039/c8nr08966c.
11. M. Dave. Optical analysis for few TMDC materials. *Bull. Mater. Sci.* 2015, vol. 38, no. 7, pp. 1791–1796. doi: 10.1007/s12034-015-0960-6.
12. S. Bharucha, M. Dave, and R. Vaidya. Electronic and optical studies of NbS₂ semiconductor material. *Mater. Today Proc.* 2022, vol. 55, no. 1, pp. 118–121. doi: 10.1016/j.matpr.2021.12.543.
13. A. X. Gray, S. Nemšák, and C. S. Fadley. Combining Hard and Soft X-ray Photoemission with Standing-Wave Excitation, Resonant Excitation, and Angular Resolution. *Synchrotron Radiat. News* 2018, vol. 31, no. 4, pp. 42–49. doi: 10.1080/08940886.2018.1483659.
14. K. He, N. Chen, C. Wang, L. Wei, and J. Chen. Method for Determining Crystal Grain Size by X-Ray Diffraction. *Cryst. Res. Technol.* 2018, vol. 53, no. 2, pp. 1–6. doi: 10.1002/crat.201700157.