

Proceeding Paper

Multicomponent, Solvent-Free Synthesis of 4-Substituted Aminopyrido [2,3-*d*] Pyrimidines Derivatives [†]

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Abstract: 4-substituted aminopyrido [2,3-*d*] pyrimidine derivatives **1a-f** were synthesized via multicomponent reaction of 2-aminopyridines, triethyl orthoformate and various primary amines under solvent-free conditions. The present work creates a variety of fluorescent heterocyclic compounds in short time and good yields. The structures of all synthesized compounds were established by IR, ¹H and ¹³C NMR analysis.

Keywords: pyrido [2,3-*d*] pyrimidine; 2-aminopyridine; solvent-free conditions; multi-component reaction

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1. Introduction

Pyrido [2,3-*d*] pyrimidine ring structure is one of the most interesting heterocycles in drug design [1], these later witches containing this moiety have various pharmacological activities such as antitumor [2] antipyretic [3], antihypertensive [4], antifungal [5], antibacterial [6] and anti-inflammatory activities [7]. More specifically pyrido [2,3-*d*] pyrimidines were considered as inhibitors of dihydrofolate reductases (DHFR) [8], tyrosine kinases and adenosine kinase [9]. Moreover, the synthesis of these fused heterocyclic compounds provides an interesting challenge in medicinal chemistry [10–12].

Continuing our research in the field of new heterocyclic compounds of biological interest [13–15], we have previously reported on the synthesis of functionalized pyrido [2,3-*d*] pyrimidines [16] (Figure 1). Encouraged by these results, we decided to extend this methodology to the synthesis of new pyrido [2,3-*d*] pyrimidines via multicomponent reaction under solvent-free conditions.

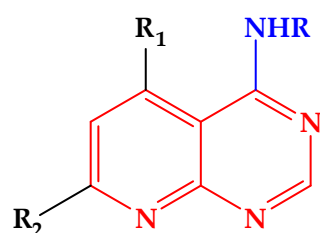


Figure 1. General structures of 4-substituted aminopyrido [2,3-*d*] pyrimidines.

2. Results and Discussion

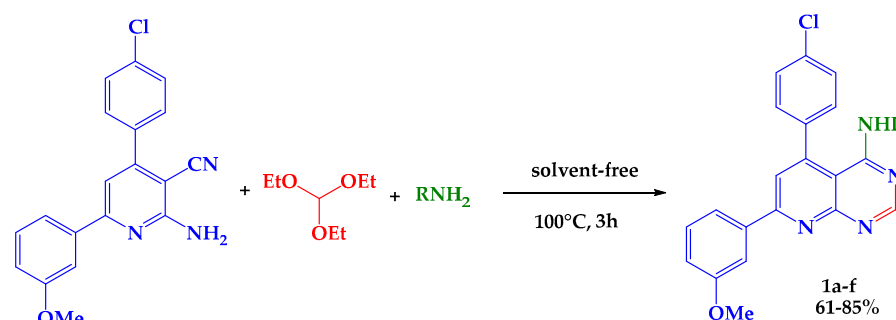
In our current studies on the synthesis of 4-substitute aminopyrido [2,3-*d*] pyrimidines, we have reported a simple and new multicomponent reaction in the eco-friendly economical and environmental conditions.

Recently, multicomponent reactions (MCR) have become an interesting approach to achieve such molecular diversity and complexity [17]. In this work, we present a new efficient method for the synthesis aminopyrido [2,3-*d*] pyrimidines derivatives from 3-cyano-2-aminopyridine under solvent-free conditions.

2.1. Synthesis of 4-Aminopyrido [2,3-*d*]pyrimidines Derivatives

The 4-aminopyrido [2,3-*d*] pyrimidines **1a-f** were easily obtained via one pot reaction of 3-cyano-2-aminopyridines, triethyl orthoformate and various primary amines. The mixture was heated during 3h without solvent to obtain compounds **1a-f** in good yields (61–85%) (Table 1).

Table 1. Synthesis of 4-aminopyrido [2,3-*d*] pyrimidines derivatives. The primary amines used are: benzylamine, butylamine, propylamine, hexylamine, phenylethylamine and tryptamine.



Product	Primary Amine	Yield (%)
1a	Benzylamine	85
1b	Butylamine	79
1c	Butylamine	71
1d	Cyclohexylamine	69
1e	Phenylethylamine	65
1f	Tryptamine	61

The Structure of the synthesized compounds **1a-f** were confirmed by spectral analysis. The IR spectra (KBr, ν_{\max} , cm^{-1}) showed the absence of NH_2 , CN and the appearance of $(\text{C}=\text{C})$ at $1542\text{--}1559\text{ cm}^{-1}$, $(\text{C}=\text{N})$ at $1669\text{--}1690\text{ cm}^{-1}$ and NH at $3462\text{--}3540\text{ cm}^{-1}$. the ^1H NMR (CDCl_3 , δ , ppm) showed the appearance of OCH_3 stretch at $\delta_{\text{H}} 3.86\text{--}3.88$ ppm and NH stretch at $\delta_{\text{H}} 5.17\text{--}5.79$ ppm, H_{pyrid} stretch at $\delta_{\text{H}} 7.09\text{--}7.36$ ppm and $\text{H}_{\text{pyrimid}}$ stretch at $\delta_{\text{H}} 7.76\text{--}8.65$ ppm.

3. The Proposed Mechanism for the Formation of 4-Aminopyrido [2,3-*d*] Pyrimidines **1a-f**

The proposed mechanism for the formation of 4-aminopyrido [2,3-*d*] pyrimidines **1a-f** is described in Figure 2.

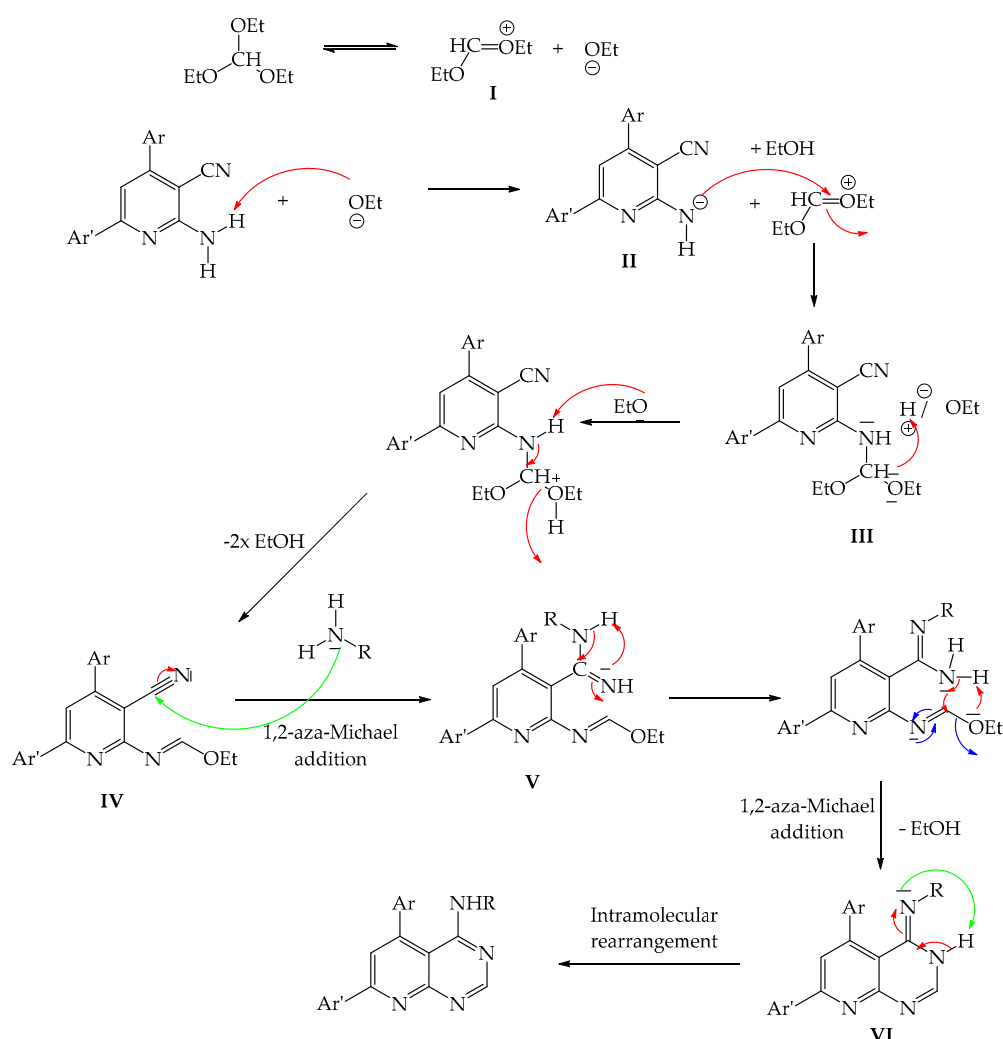


Figure 2. The proposed mechanism for the formation of 4-aminopyrido [2,3-*d*]pyrimidines **1a-f**. The reaction begins with the formation of the intermediate **I** followed by nucleophilic addition of the “NH₂” group of 2-aminopyridines on the double bond to form intermediate **III**. After rearrangement and 1,2-aza-Michael addition between the primary amine and the “CN” group of the product **IV**, the intermediate **V** is obtained. This latter undergoes a rearrangement and a 1,2-aza-Michael addition intramolecularly to form the product **VI**. Finally, an aromatization step to obtain the desired 4-aminopyrido [2,3-*d*] pyrimidines.

4. Experimental Procedure

General procedure for the synthesis of pyrido [2,3-*d*] pyrimidines **1a-f**: A mixture of 3-cyano-2-aminopyridine (10 mmol), primary amine (10 mmol) and triethyl orthoformate (10 mmol) was heated for 3 h. After the completion of the reaction (TLC). The residue was purified by column chromatography over silica gel using a mixture of nhexane- EtOAc (5:5) as the eluent. All the desired compounds were obtained as white solid [16].

5. Conclusions

In conclusion, we have successfully developed a novel route for the synthesis of new 4-substituted aminopyrido [2,3-*d*] pyrimidines derivatives via multicomponent reaction under solvent free conditions with good yields. This new MCR provided a general and efficient strategy for the construction of structurally diverse fused pyridopyrimidines skeleton.

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Conflicts of Interest: The authors declare no conflict of interest, financial or otherwise.

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