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[A0013]

Synthesis of 5-Substituted a,b-Unsaturated g-Lactams from

N-Sulfinyl Azadienes by Iron-mediated Reaction Cascades and Palladiumcatalyzed Cyclocarbonylation Strategies

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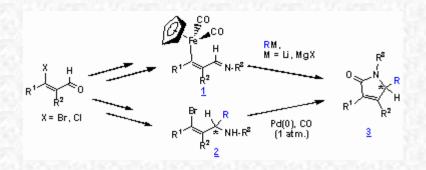
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Introduction:

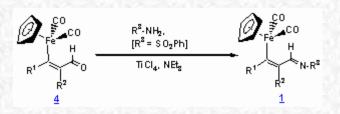
A new synthesis of 5-substituted a,b-unsaturated g-lactams is reported. $b-[Cp(CO)_2Fe]$ -Substituted *N*-sulfonyl¹ and *N*-sulfinyl azadienes **1** react with Grignard reagents or organolithium compounds furnishing 5-substituted a,b-unsaturated g-lactams **3**.^{2,3} These reaction cascades occur by initial attack of the organometallic reagent at the imine functionality followed by an intramolecular cyclocarbonylation step. In a number of cases non-*N*-protected lactams are obtained from *N*-sulfonyl azadienes besides the desired *N*-sulfonyl g -lactams **3**. Moreover, chiral b-[Cp(CO)₂Fe]-substituted *N*-sulfinyl azadienes **1** react with the organometallic reagents to furnish exclusively 5-substituted non-*N*-protected g - lactams **3**.^{2,3}

In addition, the synthesis of 5-substituted a,b-unsaturated g-lactams $\underline{3}$ by palladium-catalyzed cyclocarbonylations of chiral bromo-substituted sulfinamides $\underline{2}$, derived from the corresponding azadiene precursors and organometallic reagents, is described.⁴

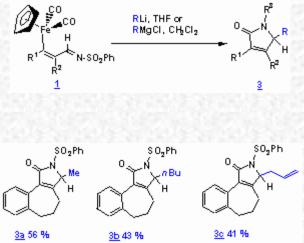


Reactions of b-[Cp(CO)₂Fe]-substituted *N*-Sulfonyl Azadienes with C-Nucleophiles:

a,b-Unsaturated imines **1** are synthesized e.g. from b-[Cp(CO)₂Fe]-substituted (*Z*)-alkenals **4** and electron poor primary amino compounds in CH₂Cl₂ in the presence of TiCl₄ and NEt₃ in dichloromethane (66-100 %).^{1,4}

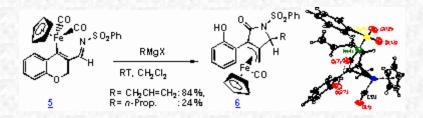


The g-lactams $\underline{3}$ are accessible by treatment of the a,b-unsaturated imines $\underline{1}$ with organolithiums in THF at -78°C or 0°C followed by prolonged stirring at room temperature (39-56%). Reactions with Grignard-reagents carried out in dichloromethane at room temperature furnished the g-lactams $\underline{3}$ in 37-62% yield.



 $\frac{32}{56} \frac{56}{10} \frac{32}{56} \frac{43}{10} \frac{32}{56} \frac{41}{10} \frac{41}{10}$ The reaction of imine 5 with Grignard reagents is found to furnish the allyliron complexes 6. The structure of the allyl-

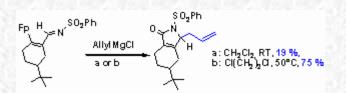
substituted product was elucidated by X-ray analysis. Therefrom an anti-orientation of the metal fragment and the allyl residue at carbon-5 is concluded.



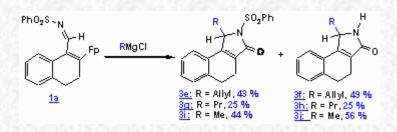
Influence of the temperature on the carbonylation step:

When the cyclohexene-derivative shown below is treated with allyl Grignard reagent at room temperature the *N*-sulfonyl g -lactam is isolated in only 19% yield. However, the yield could be increased by raising the reaction temperature after the initial 1,2-addition for the cyclocarbonylation key step. Thereby, in 1,2-dichloroethane at 50°C complete turnover to the desired product is achieved. The *N*-sulfonyl g -lactam is isolated in 75 % yield after flash chromatography.

Examples:



Unexpected Deprotection: What role does the iron play in this game?

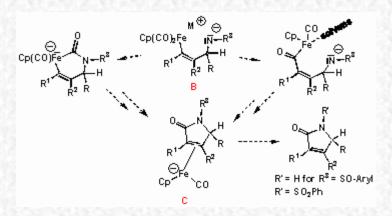


Reactions of compound <u>1a</u> with either organolithium or Grignard reagents led to the non-*N*-protected lactams <u>3f</u>, <u>3h</u>, <u>3j</u> besides the lactams <u>3e</u>, <u>3g</u>, <u>3i</u>. At the current stage the cleavage of the N-S bon is not conclusive. Iron-mediated redox processes my be involved in the formation of the non-*N*-protected g -lactams <u>3f</u>, <u>3h</u>, <u>3j</u>.

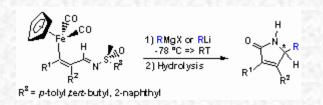
Mechanistic Considerations:

After 1,2-additon of the organometallic reagents to the *N*-sulfonyl azadienes, furnishing the metallated amides (**B**), the carbonylation steps may proceed via an acyl iron intermediate or a ferrilactam intermediate.¹⁻⁴

For the formation of the allyliron complexes presented above it seems reasonable to propose an anionic p -alkene-iron complex as intermediate (C) being formed after the reductive elimination step leading to the five-membered ring system. In an intramolecular nucleophilic substitution reaction the attack of the iron moiety at the neighboring methylene group could lead to a phenolate prior aqueous work-up yielding the allyliron complexes.



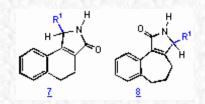
Application of Iron-substituted N-Sulfinyl Azadienes:



Chiral sulfinimines react with Grignard- or organolithium reagents to give exclusively non-*N*-protected a,b-unsaturated g-lactams. Only the *N*-tert-butyl sulfinyl imines can be treated with organolithiums and alkyl Grignard reagents since

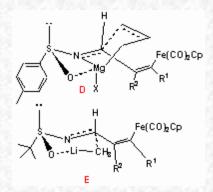
the sulfur atom is effectively shielded by the tert-butyl residue leading to exclusive attack at the imine carbon atom. For the reactions summarized in the Table, the enantioselectivities were determined by shift experiments to range from 10% *ee* up to 62% *ee*.

Iron-mediated redox processes during aqueous work-up seem to be responsible for the N-S bond cleavages observed.



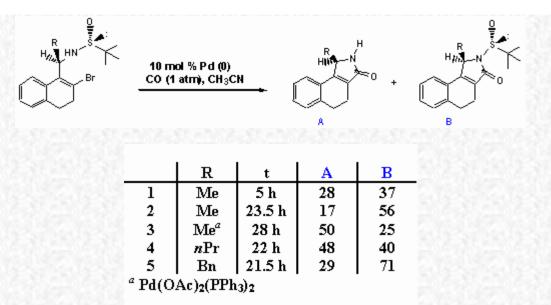
	R ³	Lactam	R ¹ -M	Yield	ee (%)
1	<i>p</i> -tolyl	8	AllylMgCl	64 %	40
2	<i>p</i> -tolyl	8	BnMgCl	27 %	62
3	2-naphthyl	8	AllylMgCl	63 %	40
4	<i>tert</i> -butyl	8	AllylMgCl	45 %	46
5	<i>p</i> -tolyl	7	AllylMgCl	66 %	52
6	<i>p</i> -tolyl	7	AllylMgBr	27 %	45
7	tert-butyl	7	AllylMgCl	55 %	38
8	tert-butyl	7	MeLi	91 %	10

For the addition of organolithiums and Grignard reagents to sulfinimines six-membered ring transition state models have been proposed by Davis and Ellman to predict the diastereofacial selectivity. ^{5,6}

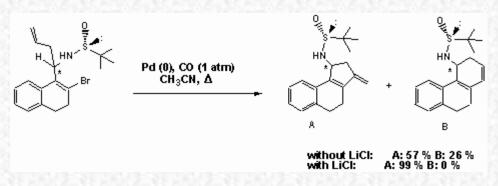


Palladium-catalyzed Cyclocarbonylations:

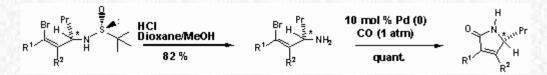
Enantiopure g-lactams 3 are accessible by palladium(0)-catalyzed cyclocarbonylation reactions of chiral sulfinamides. The chiral sulfinyl amides used are synthesized from the corresponding imines by 1,2-addition of organolithium or Grignard-reagents in 50-100 % yield. The diastereoselectivity obtained ranged from 51:49 up to 91:9. Fortunately, the diastereomers are easily separated by chromatography. The absolute configuration of an allyl-substituted and a methylsubstitued optically pure sulfinamide was elucidated by X-ray analysis. In the palladium-catalyzed cyclocarbonylation reactions surprisingly *N*-sulfinyl-substituted (B) and non-*N*-protected derivatives (A) are formed. The reactions are carried out with Pd(PPh₃)₄ and *n*-Bu₃N as base in acetonitrile at reflux for 5-28 h. The N-S bond cleavage can be attributed to attack of hydrogen bromide formed from the palladium hydride intermediate within the catalytic cycle.



The palladium-catalyzed cyclocarbonylation of allyl-substituted sulfinamides gave the desired lactam derivatives in less than 5% yield. The chiral amines shown below are found to be formed instead by Heck-reaction. In the presence of LiCl compound A is exclusively obtained.



From the optically pure sulfinamide shown below the amine derivative is prepared by treatment with HCl in dioxane/methanol for cyclocarbonylation studies, yielding the non-*N*-protected a , b -unsaturated g -lactam quantitatively.



In summary, 5-substituted a , b -unsaturated g -lactams were obtained in novel reaction cascades starting from ironsubstituted *N*-sulfonyl or *N*-sulfinyl azadienes and organometallic reagents. Enantiomerically pure derivatives were prepared from chiral sulfinamides by a palladium-catalyzed cyclocarbonylation strategy starting from b -bromosubstituted *N*-sulfinyl azadienes.

References:

- 1. K. Rück-Braun, Angew. Chem. Int. Ed. Engl. 1997, 36, 509.
- 2. P. Amrhein, D. Schollmeyer, K. Rück-Braun Organometallics 2000, in press.
- 3. K. Rück-Braun, P. Amrhein Eur. J. Org. Chem. 2000, in press.
- 4. P. Amrhein, D. Schollmeyer, K. Rück-Braun Org. Lett. 2000, submitted for publication.
- 5. F. A. Davis, W. McCoull J. Org. Chem. 1999, 64, 3396.
- 6. D. A. Cogan, G. Liu, J. A. Ellman Tetrahedron 1999, 55, 8883.

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