*Fourth International Electronic Conference on Synthetic Organic Chemistry (ECSOC-4), www.mdpi.org/ecsoc-4.htm, September 1-30, 2000* 

[A0048]

# X-Ray Structures of Bicyclic Organophosphates

## **Illustrating the Anomeric Effect**

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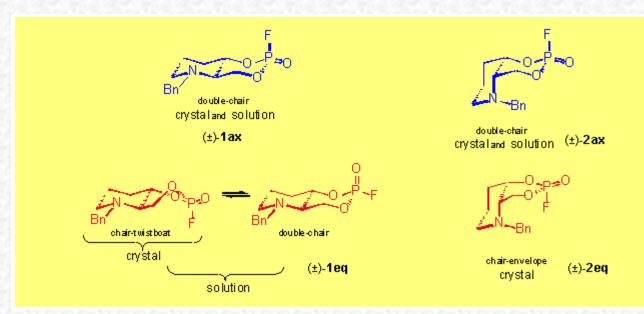
Received: 21 July 2000 / Uploaded: 30 July 2000

#### Introduction

The influence of the anomeric effect on bicyclic organophosphates could be shown for the first time by means of X-ray analysis.

In the course of our syntheses of bicyclic organophosphates as inhibitors of acetylcholinesterase we have been able to crystallize a series of four isomeric compounds:  $(\pm)$ -7-aza-7-benzyl-3-fluoro-2,4-dioxa-31<sup>5</sup>-phosphabicyclo[4.4.0]decan-3-ones  $(\pm)$ -**1ax,eq** and  $(\pm)$ -**2ax,eq**, *cis*- and *trans*-decaline-type congeners with the F-substituent in the *axial* and *equatorial* position.

The syntheses of the mentioned and of some related compounds as well as their inhibitory effects on acetylcholinesterase are described on a separate <u>Poster A0047</u>.



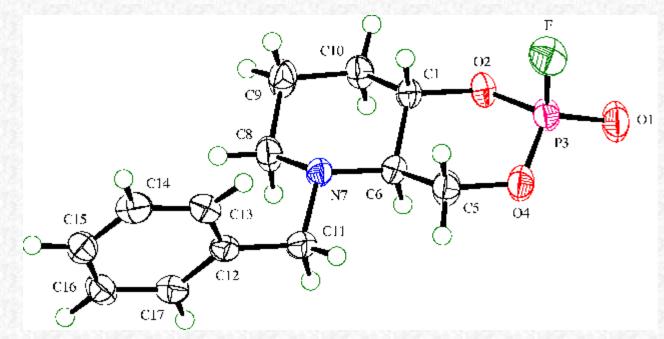
#### **X-Ray Structures**

 $(\pm)$ -1ax,eq and  $(\pm)$ -2ax,eq had been crystallized and submitted to X-ray analysis. As expected, both axially substituted

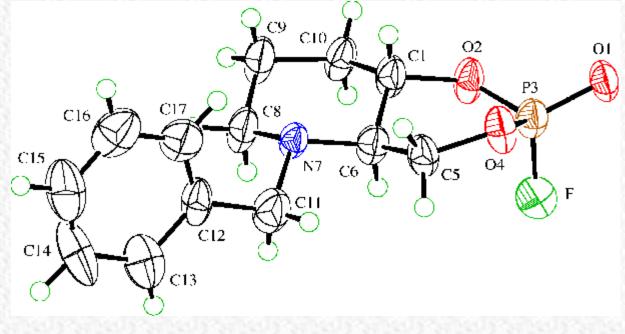
isomers adopt the sterically and stereoelectronically favoured double-chair conformation in the crystals (as well as in solution).

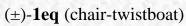
For the two equatorially substituted compounds the electronegative substituent (F) tends to occupy the sterically unfavoured *pseudo-axial* position according to the anomeric effect [1]. This results *e.g.* in a chair-twistboat conformation. Although we showed by means of NMR-spectroscopy that in solution  $(\pm)$ -**1eq** and  $(\pm)$ -**2eq** exist in a mixture of double-chair and chair-twistboat conformations ( $(\pm)$ -**1eq** in CD<sub>2</sub>Cl<sub>2</sub>: 33:67 at 293 K, 21:79 at 213 K) we obtained crystals of a chair-twistboat form ( $(\pm)$ -**1eq**) and a chair-envelope form ( $(\pm)$ -**2eq**) as they seem to have a slightly lower DG than the double-chair forms. The envelope conformation is highly surprising, it represents a middle position between the two possible twistboat conformations with five ring-atoms in plane, only C(6) laying outside.

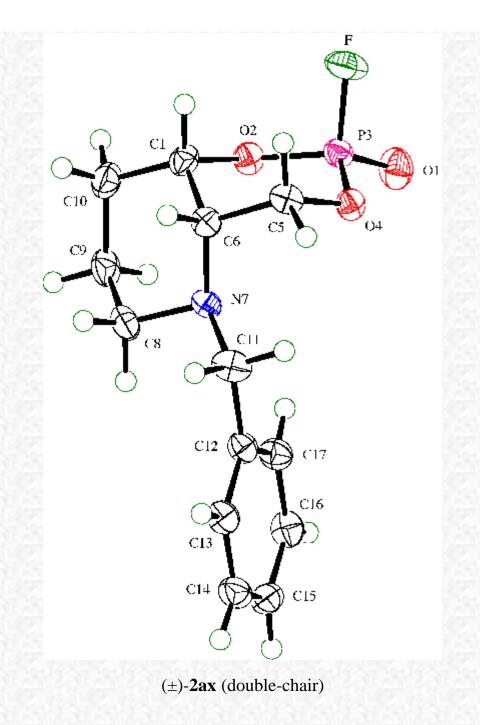
This is the first time the anomeric effect could be illustrated by X-ray analysis of such bicylic phosphates. Former attempts to crystallize related compounds lead to X-ray structures only with double-chair conformations, the electronegative substituent (*O*-phenyl, *O*-4-methoxyphenyl,  $N(CH_2CH_2Cl)_2$ , resp.) being located in the *axial* resp. *equatorial* position [2], see also poster. However, a 1,4-*bis-t*-butyl substituted monocylic organophosphate yielded an X-ray structure showing a boat conformation [3].

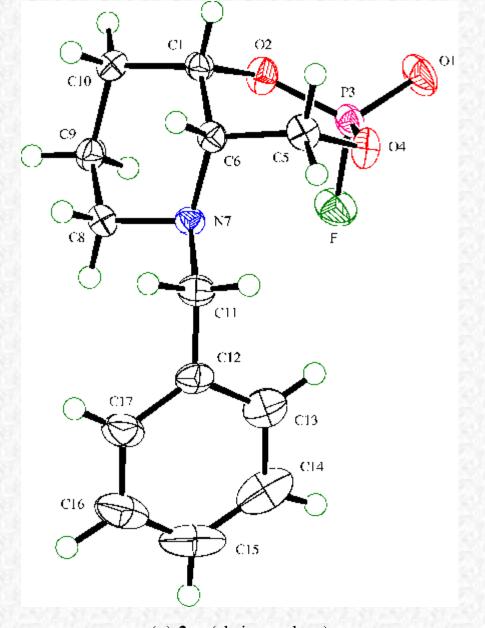


(±)-1ax (double-chair)









(±)-2eq (chair-envelope)

Conformation analysis in solution (CDCl<sub>3</sub>) was carried out by means of <sup>31</sup>P- and <sup>1</sup>H-NMR-spectroscopy using the <sup>3</sup> $J_{PH}$ -coupling constants. According to the *Karplus*-equation a torsion angle between P–O–C–H of 180° leads to a maximal <sup>3</sup> $J_{PH}$  ca. 25 Hz, while a 90° angle corresponds with zero. The characteristic pattern of the three <sup>3</sup> $J_{PH}$ -coupling constants in each compound gives information about the conformations ocurring in solution [4].

### References

[1] D.G. Gorenstein, R. Rowell, J. Findlay, J. Am. Chem. Soc. 1979, 102, 5077.

[2] R.O. Day, D.G. Gorenstein, R.R. Holmes, *Inorg. Chem.* 1983, 22, 2192; P. Van Nuffel, A.T.H. Lenstra, H.J. Geise, *Bull. Soc. Chim. Belg.* 1982, 91; J.-C. Yang, D.O. Shah, N.U.M. Rao, W.A. Freeman, G. Sosnovsky, D.G. Gorenstein, *Tetrahedron* 1988, 44, 6305.

[3] R.O. Day, W.G. Bentrude, K.C. Yee, W.N. Setzer, J.A. Deiters, R.R. Holmes, J. Am. Chem. Soc. 1984, 106, 103.

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