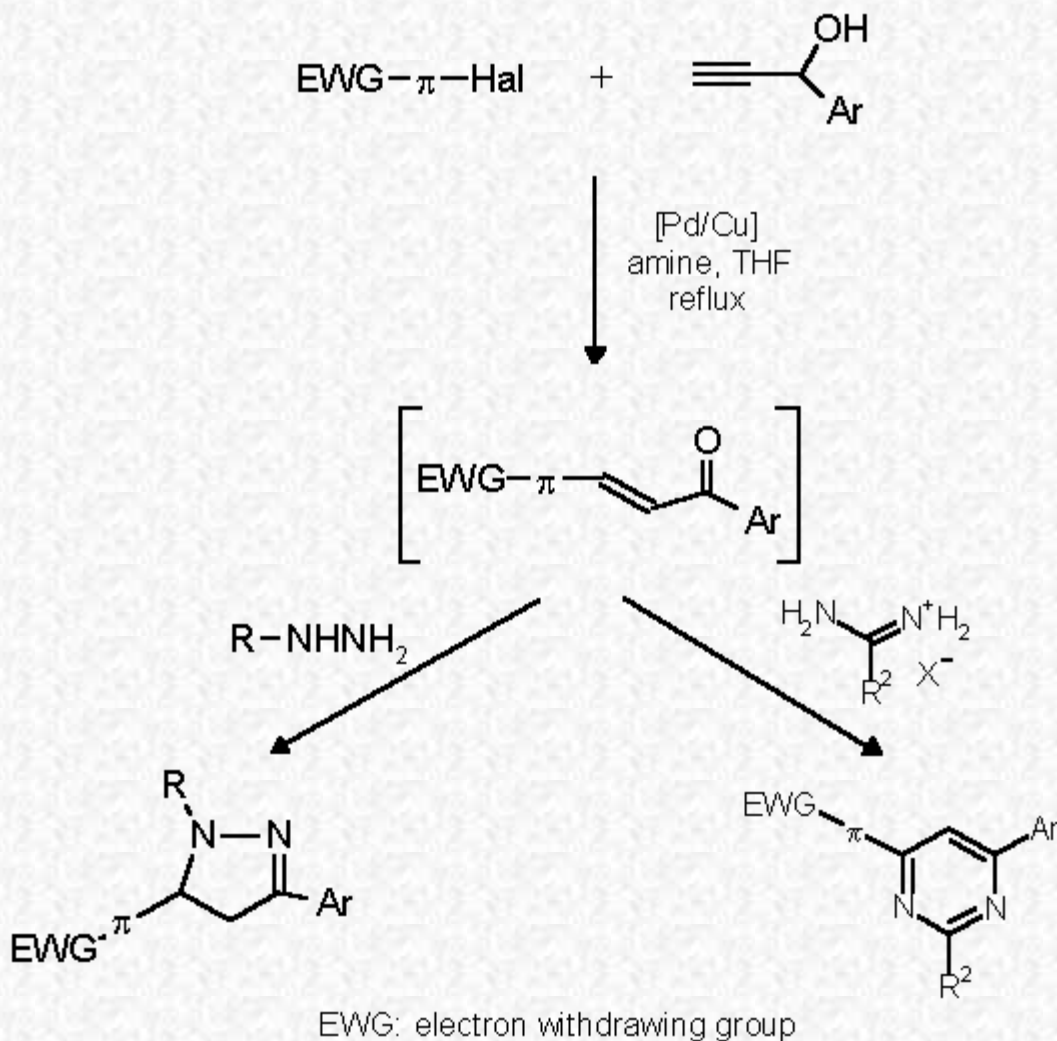




Chart 1.

Retrosynthetically, 1,5-benzoheteroazepines are synthesized by cyclocondensation of the corresponding 2-substituted anilines with suitable enones or 1,3-dicarbonyl compounds (and synthetic equivalents) [3-6,7]. However, the enones and, in particular, chalcones (i.e. 1,3-diaryl enones) are usually prepared by an aldol condensation and have to be isolated and purified prior to the cyclization step. Therefore, we set out to develop a novel synthesis of 1,5-benzoheteroazepines, preferentially in a straightforward highly convergent manner, that also can be conducted in the sense of a one-pot process. Here, we wish to communicate a facile one-pot synthesis of 2,4-di(hetero)aryl substituted 2,3-dihydro 1,5-benzodiazepines, -oxazepines, and thiazepines (**3**,  $R^1 = (\text{het})\text{aryl}$ ,  $R^2 = \text{aryl}$ ) based upon a coupling-isomerization sequence with a subsequent cyclocondensation with 2-amino, 2-hydroxy, or 2-mercapto anilines.

**Scheme 1.** One-pot pyrazoline and pyrimidine synthesis based upon a coupling-isomerization sequence.

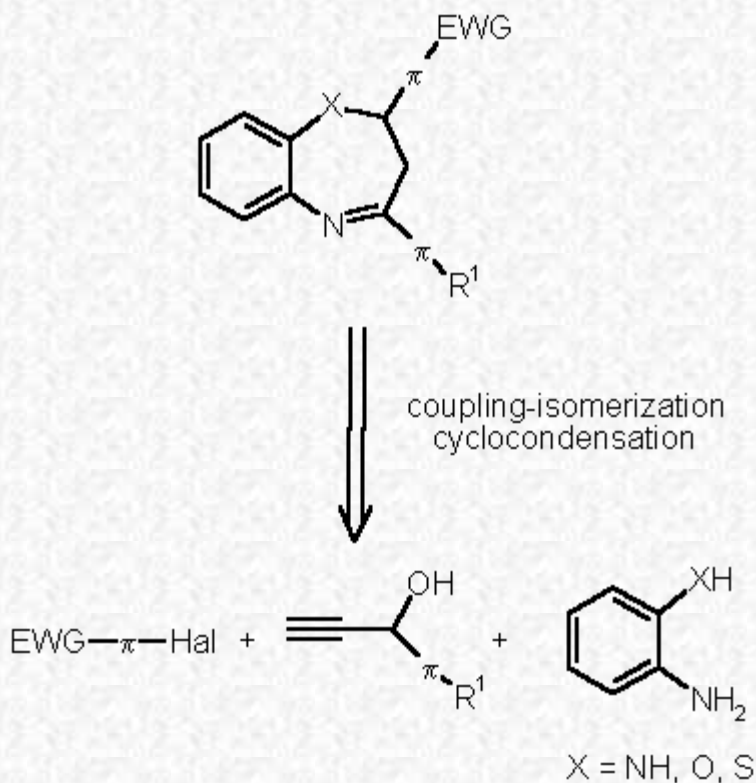


Recently, we found that palladium/copper catalyzed cross-coupling reactions of electron poor halogen substituted p-systems and 1-aryl prop-2-yn-1-ols do not furnish the expected propargyl alcohols but the isomeric enone components [8]. Mechanistically, this isomerization occurring after the cross-coupling reaction is purely base catalyzed and opens a new access to electron deficient propenones. With this powerful tool for the construction of chalcones (1,3-diaryl propenones) in hand and considering the mild reaction conditions for the Sonogashira coupling reaction we have developed novel one-pot pyrazoline [8] and pyrimidine [9] syntheses (Scheme 1).

Since cyclocondensations of 2-heteroatom substituted anilines with chalcones (1,3-diaryl propenones) give 1,5-benzoheteroazepines [2-6], retrosynthetically, an extension of the coupling-isomerization-based methodology to a one-pot synthesis of 1,5-benzoheteroazepines can be easily envisioned. Upon cyclocondensing *ortho*-phenylene diamine, 2-amino phenol, or 2-amino thiophenol as suitable 1,4-dinucleophilic components with the initially formed enone

functionality the benzoannealed seven-membered heterocycles are to be readily formed (Scheme 2). In particular, the mild reaction conditions of Sonogashira couplings [10] not only allow the presence of sensitive functional groups without tedious protection and deprotection steps but are also advantageous for base-mediated processes such as cyclocondensations. In addition, this strategy could also be extended to a combinatorial approach to 1,5-benzoheteroazepines (**3**).

**Scheme 2.** Retrosynthetic concept for a three component 1,5-benzoheteroazepine synthesis.

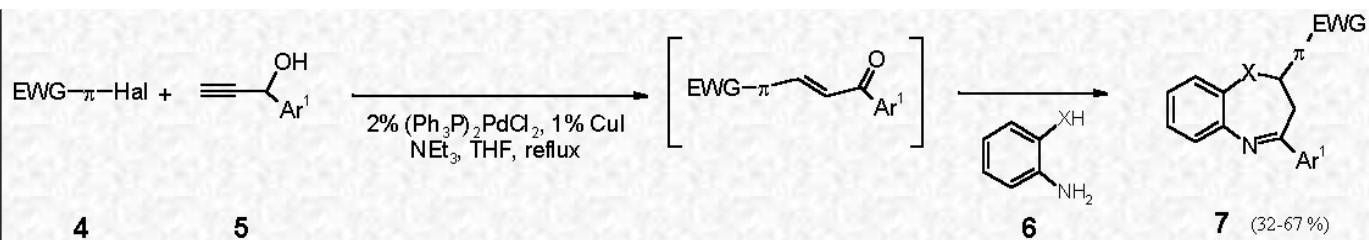


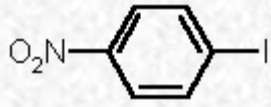
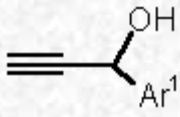
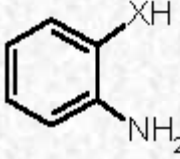
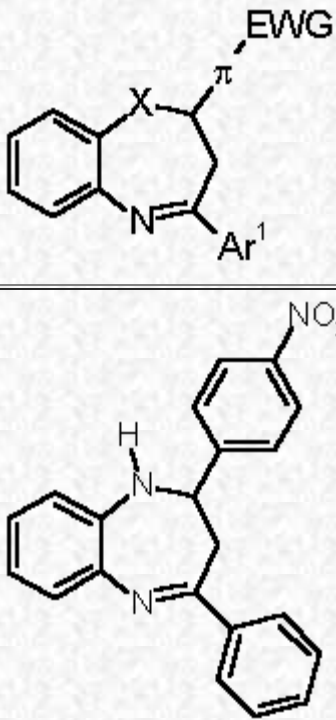
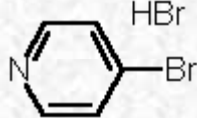
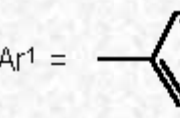
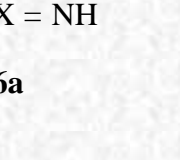
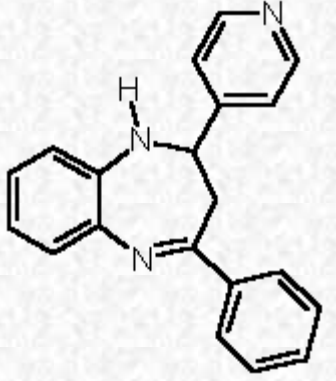
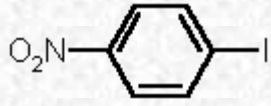
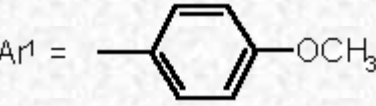
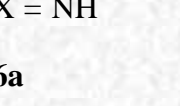
Thus, we have submitted *p*-iodo nitrobenzene (**4a**), 4-bromo pyridine (**4b**), or *p*-bromo benzonitrile (**4c**), aryl propynols **5** [11] and 2-heteroatom substituted anilines **6** to the reaction conditions of the Sonogashira coupling in boiling mixture of triethylamine and THF [12]. In all cases the isolated products were the beige to yellow 1,5-benzoheteroazepines **7** in 32-67 % yield (Table 1) [13]. As already shown for the one-pot synthesis of pyrazolines and pyrimidines the electron withdrawing nature of the (hetero)aryl halide **3** is crucial for the successful coupling-isomerization step [8,9].

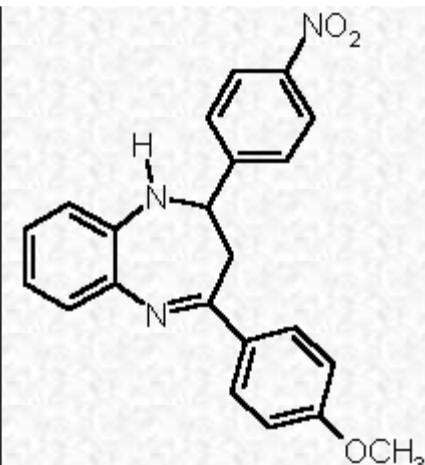
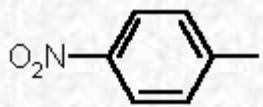
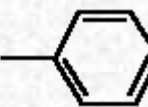
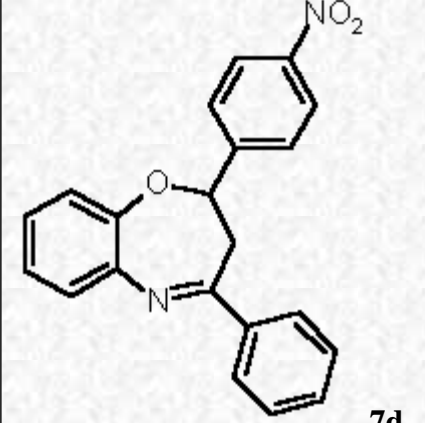
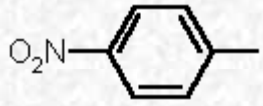
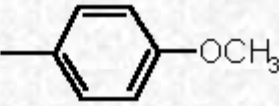
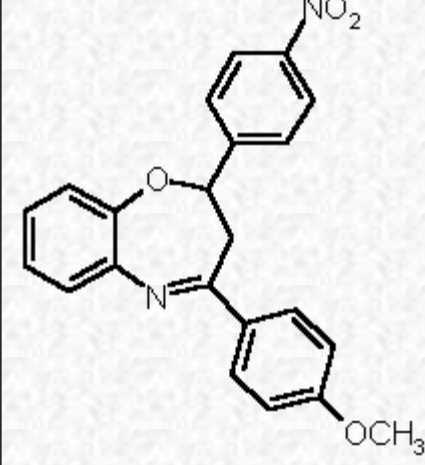
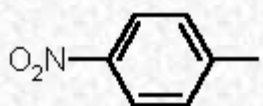
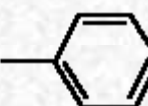
The proton and carbon NMR spectroscopic data support the formation of the 1,5-benzoheteroazepine, in particular, in the  $^1\text{H}$  NMR spectra of **7** by the indicative appearance of the ABM-spin system with the characteristic geminal and vicinal coupling constants for the methylene group resonances ( $^2J = 13.5$  Hz,  $^3J = 4.4$  Hz,  $^3J = 7.1$  Hz) and the vicinal coupling constants for the methine resonances ( $^3J = 4.4$  Hz,  $^3J = 7.0$  Hz). Furthermore, the structure of **7** was unambiguously supported by an X-ray crystal structure analysis (Figure 1) of compound **7a** (Table 1, Entry 1).

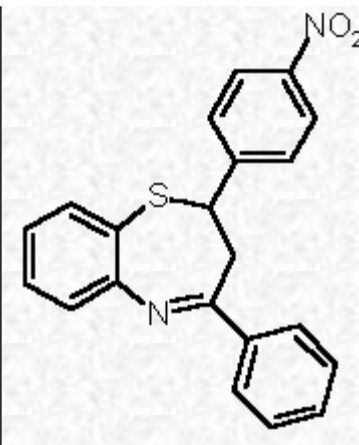
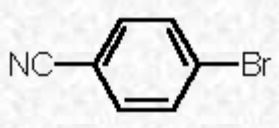
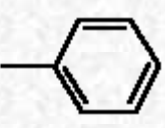
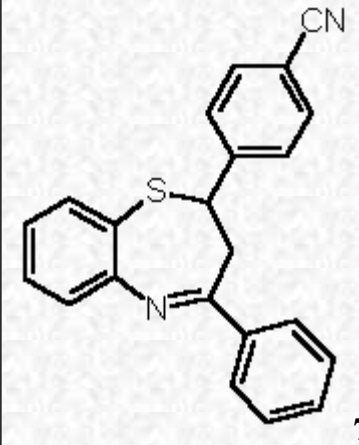
**Table 1.** Three-component 1,5-Benzoheteroazepine Synthesis Based upon a Coupling-Isomerization-Cyclocondensation Sequence<sup>a</sup>.

Entry	Heteroaryl Halide ( <b>3</b> )	Propynol ( <b>5</b> )	2-Heteroatom Substituted Aniline ( <b>6</b> )	Product ( <b>7</b> )	Yield (%)
1	<i>p</i> -Iodo nitrobenzene ( <b>4a</b> )	4-bromo propynol	2-aminoaniline	1,5-benzoheteroazepine <b>7a</b>	32
2	4-bromo pyridine ( <b>4b</b> )	4-bromo propynol	2-aminoaniline	1,5-benzoheteroazepine <b>7b</b>	67
3	<i>p</i> -bromo benzonitrile ( <b>4c</b> )	4-bromo propynol	2-aminoaniline	1,5-benzoheteroazepine <b>7c</b>	45



Entry	EWG-p-Hal 3	Propargyl alcohol 5	2-Hetero aniline 6	1,5 Benzoheteroazepine 7 (Yield %) <sup>b</sup>
1	 <b>4a</b>	 <b>5a</b>	X = NH  <b>6a</b>	 <b>7a</b> (52 %)
2	 <b>4b</b>	 <b>5a</b>	X = NH  <b>6a</b>	 <b>7b</b> (53 %)
3	 <b>4a</b>	 <b>5b</b>	X = NH  <b>6a</b>	

				 <p><b>7c</b> (42 %)</p>
4	 <p><b>4a</b></p>	<p>Ar<sup>1</sup> = </p> <p><b>5a</b></p>	<p>X = O</p> <p><b>6b</b></p>	 <p><b>7d</b></p> <p>(35 %)</p>
5	 <p><b>4a</b></p>	<p>Ar<sup>1</sup> = </p> <p><b>5b</b></p>	<p>X = O</p> <p><b>6b</b></p>	 <p><b>7e</b> (32 %)</p>
6	 <p><b>4a</b></p>	<p>Ar<sup>1</sup> = </p> <p><b>5a</b></p>	<p>X = S</p> <p><b>6c</b></p>	

			 <p><b>7f</b></p> <p>(67 %)</p>
7	 <p><b>4c</b></p>	<p>Ar<sup>1</sup> =</p>  <p><b>5a</b></p>	<p>X = S</p> <p><b>6c</b></p>  <p><b>7g</b></p> <p>(67 %)</p>

<sup>a</sup>Reaction conditions: 1.0 equiv of the (hetero)aryl halide **4**, 1.05 equiv of the propargyl alcohol **5**, 0.02 equiv of (Ph<sub>3</sub>P)<sub>2</sub>PdCl<sub>2</sub>, 0.01 equiv of CuI, 1.1 equiv of the 2-hetero aniline **6**, THF/NEt<sub>3</sub> 2:1 (10mL/mmol halide). <sup>b</sup>Yields refer to isolated yields of compounds **7** after recrystallization estimated to be > 95% pure as determined by NMR spectroscopy and elemental analysis.

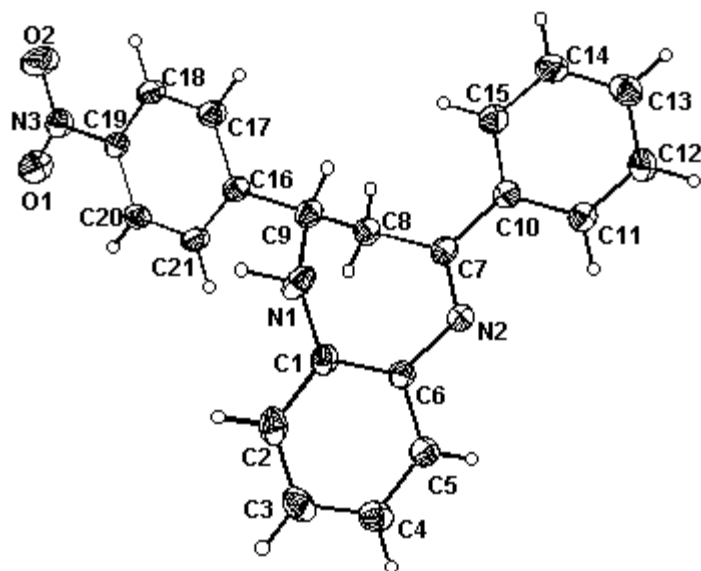


Figure 1

In conclusion, we could show that the mild reaction conditions of the coupling-isomerization sequence of electron poor (hetero)aryl halides with 1-aryl propargyl alcohols giving rise to chalcones can be extended to a one-pot three component synthesis of 2,4-di(hetero)aryl substituted 2,3-dihydro 1,5-benzodiazepines, -oxazepines, and thiazepines. Since 2-heteroatom substituted anilines can act as chelating ligands as well transition metal templated syntheses of 14-membered rings [14] for novel ligands in catalysis can be easily envisioned. Further studies directed to extend these one-pot heterocycle syntheses are currently underway.

**Acknowledgments.** The financial support of the Fonds der Chemischen Industrie, Deutsche Forschungs-gemeinschaft and the Dr.-Otto-Rohm Gedachtnisstiftung is gratefully acknowledged. The authors wishes to express their appreciation to Dr. K. Polborn for performing the X-ray structure analysis and to Prof. H. Mayr for his generous support.

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- [11] The propynols **4** were synthesized according to Krause, N.; Seebach, D. *Chem. Ber.* **1987**, *120*, 1845.
- [12] **Typical procedure (7g, entry 7):** To a magnetically stirred solution of 0.25 g (1.00 mmol) 4-bromo benzonitrile (**4c**), 22 mg (0.02 mmol) of Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub>, and 2 mg (0.01 mmol) of CuI in degassed mixture of 6 mL of THF and 3.5 mL of triethylamine under nitrogen a solution of 139 mg (1.05 mmol) of 1-phenyl propyn-1-ol (**5a**) in 6 mL of THF was added dropwise at room temperature over a period of 30 min. The mixture was heated to reflux temperature for 16 h. After cooling to room temperature 138 mg (1.10 mmol) of 2-amino thiophenol (**6c**) was added and the reaction mixture was heated to reflux temp. for 8 h. After cooling the solvents were removed *in vacuo* and the residue dissolved in dichloromethane and was filtered through a short pad of silica gel. The solvents were evaporated *in vacuo* and the residue was recrystallized from dichloromethane/pentane to give 228 mg (67 %) of analytically pure **7g** as yellow needles. Mp. 180-181 °C. <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 300 MHz): δ 3.03 (t, *J* = 12.8, 1 H), 3.30 (dd, *J* = 4.9 Hz, *J* = 12.9 Hz, 1 H), 4.97 (dd, *J* = 7.7 Hz, *J* = 12.6 Hz, 1 H), 7.16 (dt, *J* = 7.5 Hz, *J* = 1.4 Hz, 1 H), 7.32 (dd, *J* = 7.8, *J* = 1.9, 1 H), 7.41 (d, *J* = 8.3 Hz, 2 H), 7.61-7.46 (m, 7 H), 8.04 (dd, *J* = 7.6, *J* = 1.4, 2 H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>, 300 MHz): δ 36.7 (CH<sub>2</sub>), 59.4 (CH), 111.3 (C<sub>quat.</sub>), 118.3 (C<sub>quat.</sub>), 121.6 (C<sub>quat.</sub>), 125.2 (CH), 125.3 (CH), 126.6 (CH), 127.1 (CH), 128.6 (CH), 130.0 (CH), 131.0 (CH), 132.4 (CH), 134.7 (CH), 137.1 (C<sub>quat.</sub>), 148.6 (C<sub>quat.</sub>), 152.2 (C<sub>quat.</sub>), 168.1 (C<sub>quat.</sub>). MS (70 eV, *m/z* (%)): 340 (M<sup>+</sup>, 8), 211 (M<sup>+</sup> - NCC<sub>6</sub>H<sub>4</sub>CH=CH<sub>2</sub>, 100). IR (KBr):  $\tilde{\nu}$  2229 cm<sup>-1</sup>, 1609, 1574, 1452. UV/Vis (CHCl<sub>3</sub>):  $\lambda_{max}$  (ε) 244 nm (28700). Anal. Calcd. for C<sub>22</sub>H<sub>16</sub>N<sub>2</sub>S (340.45): C, 77.62; H, 4.74; N, 8.23; S, 9.42. Found: C, 77.48; H, 4.76; N, 8.26; S, 9.52.
- [13] All compounds have been characterized spectroscopically and by correct elemental analysis.
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