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Application of Ring Closing Metathesis towards functionalized 1,4-dihydro-9,10-anthraquinones and anthraquinones using Grubbs' Catalyst[#]

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Abstract

A reme a and s a f o W a d syn t es s of an t a nomes W as de re o red, n W c d a y a on of ,4 na t o nomes, f o o red by R n os n Me a t t res s (R M) of re res n d a y na t o nomes W t e bbs' ca a ys t and s bse en den yd o rena t on s n d/ af o ded re des red an t a nomes t re o con t o of s bs en s and n ood y re ds.

An t a nomes a re W des read n na t re as ys o o ca y ac t re cons t t t of an t c oo an s s, n sec s and a me an a s.[3] re a ac t re b o o ca ac t es of ese an t a nomes, f o rexa re t re an t t t o a ens a re an ome,[4] oxan ome[4] and re na t a ad a yc me,[5] ex a n t re a n t es n re nome o y as a a re n o an c syn t es s. Many syn t c s a re es o ads re syn t es s of an t a nomes a re a ready been de re o red, s c as re de a s reac ons,[6] re s A de reac ons,[7] t a de an on s a re es,[2,8] and a yme ann a on reac ons. W o re e, re o red re ods s a re so re d a bac s s c as a s reac on cond t ons, o y re ds o a d ff c re ocon o . o s reason, re de re o ren t of a me and reme a re t od f o re syn t es s of an t a nomes s of sc ren f c n re es t.

n re as decade, R n os n Me a t t res s (R M) of re re oa o f nc ona zed d remes as been resen ed as a W o re f re t od f o re cons c on of an re b oaden n an re of ca bocyc es and re t ocyc es.[10] e y ecen y, a me o oco f o benzoann a on n o n do b re a sen rea an ren and R n os n Me a t t res s (R M) as been resen ed t W o a ds 2,3 d re t y , 0 an t a nome. n con n a on of o n t es n re syn t es s of na t a y occ n co o nds,[12] s co n ca on d sc oses a o re reme a re t odo o y f o re n cons c on of nome sys re s z n R n os n Me a t t res s W o a ds and/o 3 s bs t t t ed d yd oan t a nomes and an t a nomes W t re o con t o of s bs en s.

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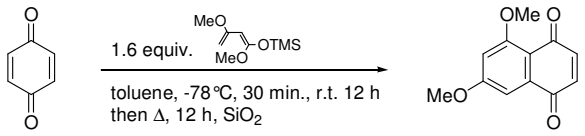
* e : +003 2 2 45 5 , f ax: +003 2 2 4 4 243

,4 Na o nones are e e co e c a y a a abe o eas y access be co o nds, W co a es e a ac e s a n a e as n na a od c syn es s. 5, e oxy ,4 na o none **2a** as e a ed o ,4 benzo none **1** s n a e no n e s A de e odo o y ³W ,3 d e oxy e y s yoxy ,3 b ad eme ⁴ (Sc e e). S nce on y e na o nones **2b** and **2d** a e co e c a y a a abe, e e y a ed ana o e **2c** o na o none **2b** ad o be e a ed z n 4e a en s o odo e ame and 3e a en s o A 2 n o x n ²2 2o 2o s (Sc e e 2).

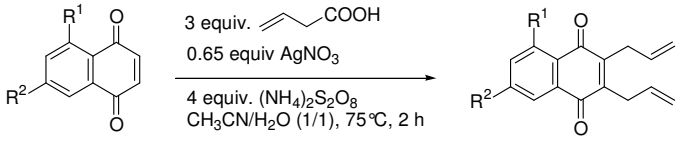
Sc e e



1 **2a** (59%)

2,3 a y na o nones **3**W e e a ed n ood y e ds (2% %) o na o nones **2** on eac on W 3e a en s o 3 b eno c ac d n ace on e n e esence o 4e a en s o d a on e s a e and a ca a y c a o n o s e n a e. A do be ad ca M c a e add on on na o nones **2** a o ded e des ed a y a ed co o nd **3** (Sc e e 2). o e y, da y a on o ,4 na o nones W ace c ac d nde ese cond ons as ed o 2,3 d e y ,4 na o none ad ca add ons.[5] A so a y a on o na o nones as been en omed n e a e.[6]

Sc e e 2



2a R^1 = OMe, R^2 = OMe
2b R^1 = OH, R^2 = H
2c R^1 = OMe, R^2 = H
2d R^1 = R^2 = H

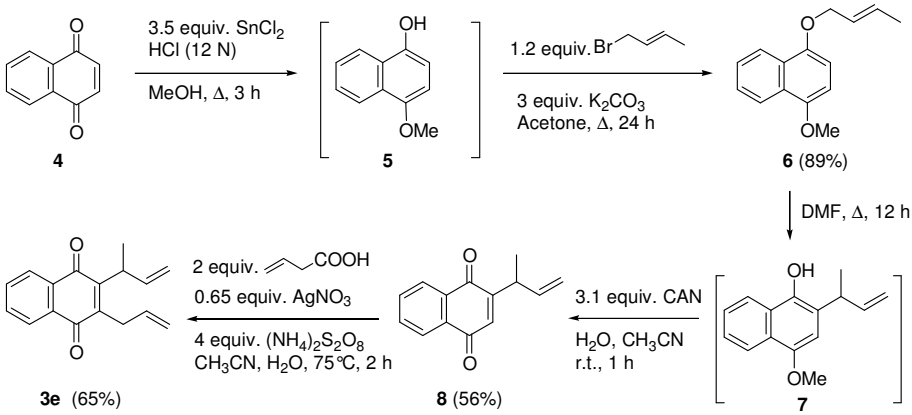
4 equiv. MeI, 3 equiv. Ag_2O, CH_2Cl_2, Δ, 12 h, 93%

3a (62%)
3b (65%)
3c (64%)
3d (67%)

n o de o n s s bs ed an a nones, a a sen ea an en a oac W as a ed o a ds e syn es s o 2 (y 2 o eny) ,4 na o none **8** a e ono *O* a y a on o e a o a e s bs a e (Sc e e 3). Na o none **4**W as ans o ed n o yd oxy 4 e oxyna o a eme **5** n e ano n e esence o 3.5e a en s o Sn 2 and 2 M yd o o c ac d. S bse en y, e e yd oxy o o co o nd **5**W as a y a ed W .2 e a en s o c o y b o de n ace ome nde bas c cond ons (3

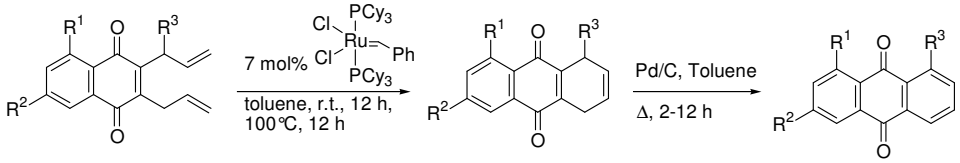
re aren s of 2 3). The s ob amed (*E*) (2' b enoxy) 4 e oxyna t arme **6** as sed as a s bs ar fo a a sen ea an e en f n s n yd oxyna t arme **7**, w as eas y ox d zed t 3. e aren s of ce a on n ar (A) n a eo s ace on e (/) es n n na t o nome **8**.[8] na y, a y a on 3 b eno c ac d n e esence of d a on e s f ar and A 2 es ed n e des ed y s bs t ed na t o nome **3e** n 5% y e d (Sc e 3).

Sc e e 3



e ey s e n e d sc osed syn es s of an a nomes co ses a R n os n Me a es s of 2,3 d a y na o nomes **3** s n e bbs' f s e me a on ca a ys. n s w ay, d yd oan a nomes **9** e s syn es zed n y e ds and y z n o % of benzy deme b s(cyc o exy os me)d o o en n o eme on ea n (Sc e e 4). n a f na s e , e eas y ob amed d yd oan a nomes **9** e e a o a zed by eans of a ad on ac e ca bon n o eme nde of x, f n s n e a ed an a nomes **10** n exce en y e ds (88% %).

Sc e e 4



3a (R^1 = R^2 = OMe, R^3 = H)	**9a** (91%)	**10a** (91%)
3b (R^1 = OH, R^2 = R^3 = H)	**9b** (93%)	**10b** (96%)
3c (R^1 = OMe, R^2 = R^3 = H)	**9c** (94%)	**10c** (92%)
3d (R^1 = R^2 = R^3 = H)	**9d** (88%)	**10d** (94%)
3e (R^1 = R^2 = H, R^3 = Me)	**9e** (89%)	**10e** (88%)

In conclusion, a theoretical and statistical foundation for synthetic design of novel organic compounds has been developed. Application of Hammett's equation, originally developed for benzene derivatives, as extended to other aromatic systems and to aliphatic compounds has been reviewed. The theoretical basis for the correlations obtained and the limitations of the method have been discussed.

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