[A0038]

NUCLEOPHILIC ADDITIONS TO 10-ISOBORNYLSULFINIMINES

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Received: 25 August 1997 / Uploaded: 25 August 1997

Keywords: sulfinimines, sulfinamides, nucleophilic addition, asymmetric synthesis, Diels-Alder reaction.

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1. INTRODUCTION

Optically active sulfinimines proved to be versatile chiral auxiliares in the synthesis of amino-compounds. In this presentation new camphor derived sulfinimine is described, as well as its reactions with some nucleophiles. Lewis acid catalyzed additions of ketene silyl acetal to sulfinimine is presented and hetero Diels-Alder reaction with Brassard's diene. As opposed to majority of sulfur containing chiral auxiliares, title compound may be recycled without loss of optical purity.

2. SYNTHESIS OF (Ss)-10-ISOBORNYLSULFINIMINE.

The reaction of 10-isobornylsulfinate (Kawecki R., Urbanczyk- Lipkowska Z. *Synthesis* **1996**, 603) with LHMDS provides sulfinamide in good yield. Subsequent condensation with aromatic aldehyde gives sulfinimine which after nucleophilic addition can be converted to initial sulfinate by acid treatment (scheme 1. lower arrow).

Scheme.1

3. ADDITIONS OF ORGANOMETALLIC REAGENTS.

Scheme 2.

RM	Yield, %	Ratio	
EtM gBr	95	1:1	_
n-BuLi	88	1:1	
MgBr	85	99 : 1	
⇒CoLi OEt	no reaction	-	
N—t Bu	no reaction	-	

Addition of organolithium or Grignard reagents to C=N bond is not stereoselective. One exception is allylmagnesium bromide, which undergoes reaction through six-membered ring transition state (scheme 3), affording exclusively one diastereoisomer.

Scheme 3.

4. LEWIS ACID CATALYZED CYCLOADDITION TO 10-ISOBORNYLSULFINIMINE.

Scheme 4.

Lewis Acid	Temp., ^O C	Yield, %	E.ə., %
BF3Et20	20	48	42
BF3Et20	0	47	43
EtAICI ₂	20	36	58

5. LEWIS ACID CATALYZED ADDITION OF KETENE SILYL ACETAL TO SULFINIMINES.

Scheme 5.

RS(0)	Lewis Aci	d (eq.)	Solvent	Temp., °C	Unreacted sulfinimine, %	Yield of product, %	D.e., %
(±)-p-ToIS(O)	BF ₃ ·Et ₂ 0	(1.3)	CH ₂ Cl ₂	20	27	53	61
	BF ₃ ·Et ₂ O	(1.1)	CH ₂ Cl ₂	-70	74	9	34
	BF ₃ ·Et ₂ O	(1.1)	CH3CN	20	47	35	42
	Et ₂ AICI	(1.1)	CH ₂ Cl ₂	20	46	28	47
	EtAICI ₂	(1.1)	CH ₂ Cl ₂	20	38	39	60
	SnCl ₄	(1.1)	CH ₂ Cl ₂	20	45	13	47
	Yb(OTf)3	(0.5)	CH ₂ Cl ₂	20	59	19	61
	Yb(OTf)3	(0.5)	CH3CN	20	61	17	26*
	TrSbCl ₆	(0.14)	CH ₂ Cl ₂	-70	67	14	40*
(S _S) X	BF ₃ ·Et ₂ 0	(1.1)	CH ₂ Cl ₂	20	42	0	-
	s EtAICI ₂	(1.1)	CH ₂ Cl ₂	20	78	0	-

^{*} D.e. of the second diastereoisomer

6. CONCLUSIONS.

Above results show that 10-isobornylsulfinimine is rather poor electrophile in reactions with enolates and azaenolates. Similar situation occurs in the case of Mukaiyama type condensation, despite simple ptolylsulfinimine reacts smoothly with medium stereoselectivity. The use of 10-isobornylsulfinimine as dienophiles in hetero Diels-Alder reaction seems to be the most promising.

Acknowledgement: This work is part of the Institute of Organic Chemistry research program sponsored by State Committee for Scientific Research.

Comments

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