

The use of activated carbons to remove heavy metals and *N*-(phosphonomethyl)glycine from European wines



<u>Fragkiskos Papageorgiou¹ Theodoros Markopoulos¹, Ioannis Katsoyiannis², Athanasios C. Mitropoulos¹, George Z. Kyzas^{1,*}</u>

¹ Hephaestus Laboratory, School of Chemistry, Faculty of Science, Democritus University of Thrace, Kavala, Greece

² Department of Chemistry, Aristotle University of Thessaloniki, 54124 Thessaloniki, Greece

Email of Presenting author: frankypap1979@gmail.com

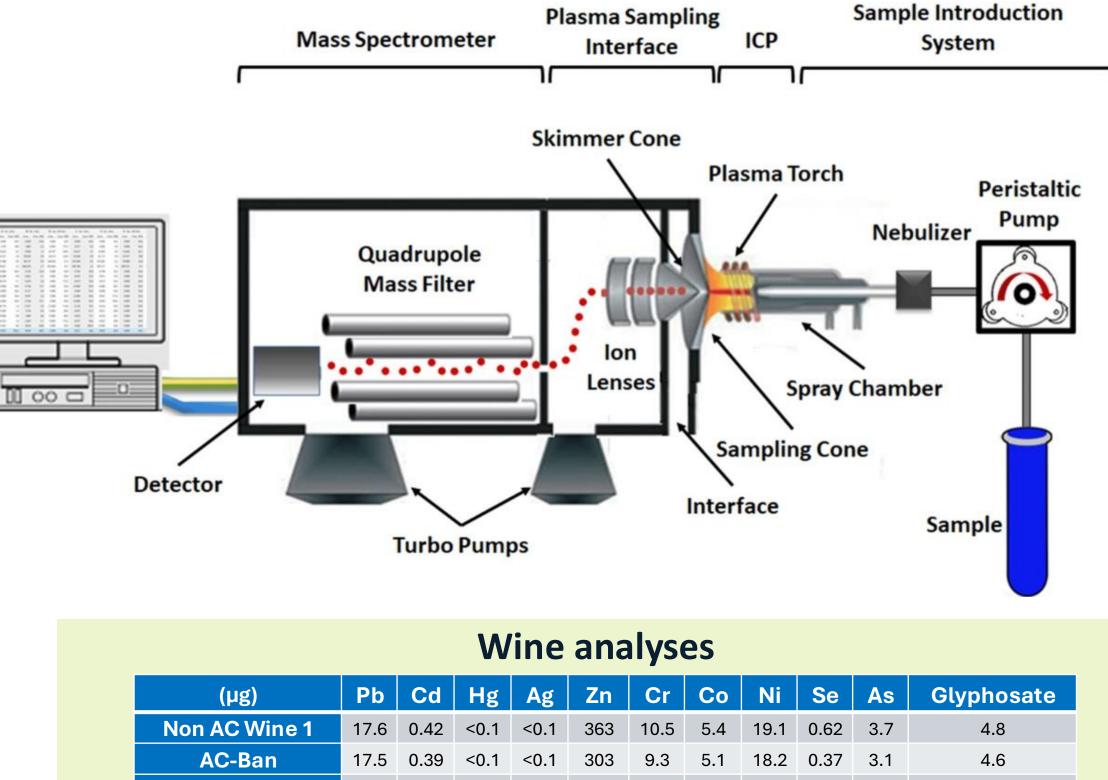
1. INTRODUCTION

➤The pollution of the planet has exceeded all limits. One aspect of concern is the environmental burden caused by heavy metals. The issue with these metals is that they tend to accumulate in the environment, leading to adverse effects.

➢Over the past century, planetary pollution has surpassed all boundaries. Climate change has manifested through extreme weather conditions, resulting in thousands of casualties. However, pollution from industrial activities and modern lifestyles has various facets, one being the environmental burden of heavy metals.

➤ In this study, we used two types of activated carbon to remove or reduce the levels of the following metals detected within permissible limits in the wines we examined, (Lead, Cadmium, Mercury, Silver, Zinc, Chromium, Cobalt, Nickel, Selenium, Arsenic) as well as the chemical substance N-(phosphonomethyl)glycine. The first type of activated carbon

2. METHODOLOGY



came from potato peels, while the second from banana peels.

➤The results were very satisfactory using 1000 grams per ton of activated carbon regarding the reduction of the content of several metals as well as the chemical substance we are examining.

3. RESULTS

Morphological characterization

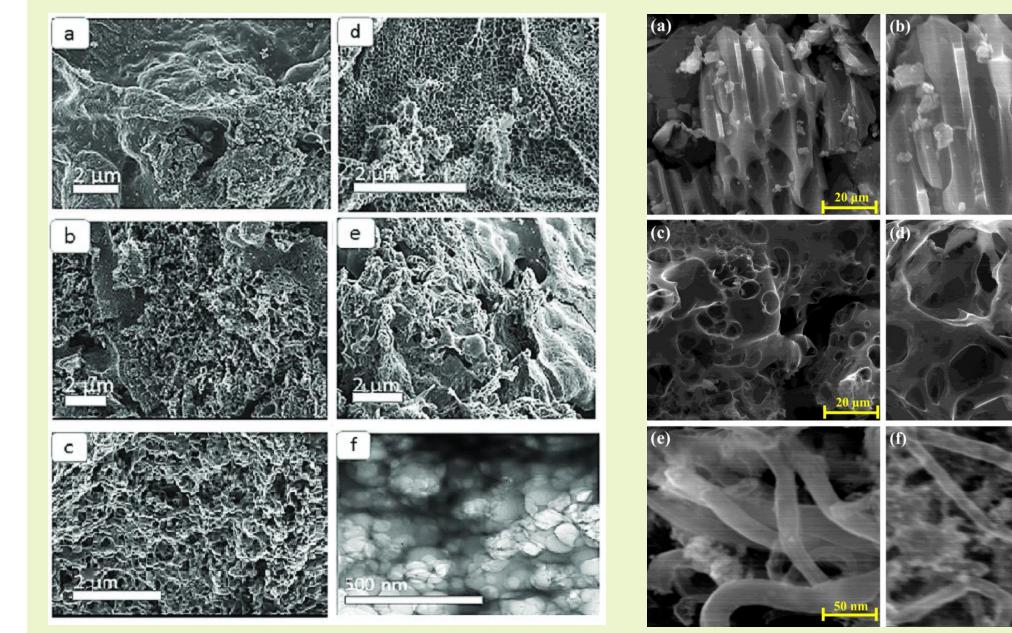


Figure 1. SEM of activated carbon derived from banana peels (left) and potato peels (right)

AC-Ban	17.5	0.39	<0.1	<0.1	303	9.3	5.1	18.2	0.37	3.1	4.6
AC-Pot	17.5	0.38	<0.1	<0.1	301	9.3	5.1	18.3	0.37	3.1	4.6
Non AC Wine 2	10.7	0.32	<0.1	<0.1	1854	35.3	5	34.1	2.1	1.2	6.3
AC-Ban	10.6	0.31	<0.1	<0.1	1695	32.1	4.8	33.6	1.8	0.8	6.1
AC-Pot	10.6	0.31	<0.1	<0.1	1693	32.1	4.8	33.5	1.8	0.8	6.1
Non AC Wine 3	14.5	0.32	<0.1	<0.1	1078	13	4	26.3	0.59	3.4	12.1
AC-Ban	14.3	0.3	<0.1	<0.1	1007	12.1	3.9	25.2	0.51	2.6	11.7
AC-Pot	14.3	0.3	<0.1	<0.1	1008	12.1	3.9	25.1	0.51	2.6	11.7
Non AC Wine 4	3.6	0.16	<0.1	<0.1	700	7.5	1.8	24.1	0.35	0.72	3.8
AC-Ban	3.4	0.11	<0.1	<0.1	589	6.4	1.8	23.4	0.27	0.62	3.6
AC-Pot	3.4	0.1	<0.1	<0.1	591	6.4	1.8	23.3	0.28	0.63	3.6
Non AC Wine 5	7.3	0.17	<0.1	<0.1	988	6.7	4.3	22.8	1.2	2.1	4.3
AC-Ban	7.1	0.15	<0.1	<0.1	975	5.8	4	21.5	1	1.8	4
AC-Pot	7.1	0.15	<0.1	<0.1	980	5.8	4.1	21.3	1	1.7	4.1
Non AC Wine 6	18	0.28	<0.1	<0.1	1285	14	4.9	29	1.4	1.8	7.3
AC-Ban	17.4	0.26	<0.1	<0.1	1057	12.8	4.7	28.5	1.1	1.5	7
AC-Pot	17.5	0.26	<0.1	<0.1	1069	12.8	4.7	28.5	1.1	1.5	7
Non AC Wine 7	7	0.2	<0.1	<0.1	533	5.3	1.9	13.3	0.21	1.2	11
AC-Ban	6.8	0.18	<0.1	<0.1	400	4.6	1.8	12.8	0.16	1	10.8
AC-Pot	6.8	0.18	<0.1	<0.1	401	4.6	1.8	12.8	0.16	1	10.8
Non AC Wine 8	18	0.19	<0.1	<0.1	354	2.7	2.3	13.5	0.31	4.3	7.1
AC-Ban	17.3	0.12	<0.1	<0.1	274	2.4	2.2	13.2	0.2	4.1	7
AC-Pot	17.4	0.12	<0.1	<0.1	271	2.4	2.2	13.1	0.21	4	7
Non AC Wine 9	1.7	0.16	<0.1	<0.1	470	2.9	1.5	10	0.31	0.59	3.4
AC-Ban	1.5	0.15	<0.1	<0.1	435	2.4	1.3	9.8	0.22	0.48	3.3
AC-Pot	1.5	0.15	<0.1	<0.1	432	2.4	1.3	9.8	0.23	0.49	3.3
Non AC Wine 10	23	0.17	<0.1	<0.1	351	5.1	2.3	26.3	0.81	0.71	6.2
AC-Ban	22.7	0.16	<0.1	<0.1	302	4.3	2.1	25.7	0.53	0.57	6
AC-Pot	22.8	0.16	<0.1	<0.1	304	4.3	2.1	25.8	0.51	0.58	6

4. CONCLUSIONS

The different origins of the activated • In carbons used to reduce the content of wheavy metals and the chemical the substance did not have significant it differences. The reduction was almost • The similar in the case of both the carbon reduced from potato and the carbon reduced from banana.

 The higher the concentration of the substance we detected, the greater the quantitative reduction we achieved with the application of activated carbon.

In metals where quantitative detection was very low, such as mercury and silver, there was no reduction or, if there was, it was undetectable.

 The relationship between the atomic radius of each element and the reduction rate is still under investigation. To draw a safe conclusion, we must also take into account the saturation of the active centers of carbon.

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