

Mitigating Environmental Risks: Efficient Removal of Metronidazole from Pharmaceutical Wastewater Using Functionalized Graphene Membrane

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INTRODUCTION



Fig 2. Metronidazole different form of its administration

- ❑ Metronidazole is commonly used antibiotic
- ❑ It belonging to the nitroimidazole class of antibiotics
- ❑ It can be administered as injectable or capsules
- ❑ It is used for treating gastrointestinal infections
- ❑ And it is also treated parasitic infections like trichomoniasis and giardiasis, and amebiasis

- ❑ It adverse reactions, which include brain toxicities¹.
- ❑ It long term use of drugs like metronidazole or ornidazole can cause toxicities like infertility².
- ❑ It is difficult to treat its abuse^{2,3}.

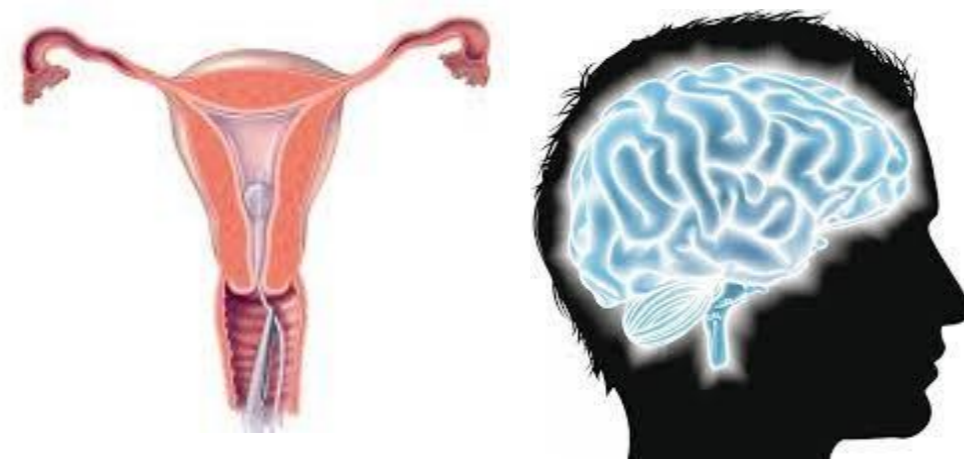


Fig 2. Human Part affected for when the drug is abused

AIM

This study explores the potential of functionalized graphene membranes for the removal of metronidazole from industrial and pharmaceutical wastewater. Employing molecular simulations and the AM1 semi-empirical calculation method, we designed and simulated functionalized membranes to enhance metronidazole removal efficiency.

METHOD

- ❑ Molecular modeling using AM1 semi-empirical method
- ❑ With evaluation of vacuum and aqueous system
- ❑ Adsorption strength calculation: $E_{ads} = E_{ax} - E_a - E_x$
- ❑ In the study, different material functionalization method like alcohol, acid, and others were explored.

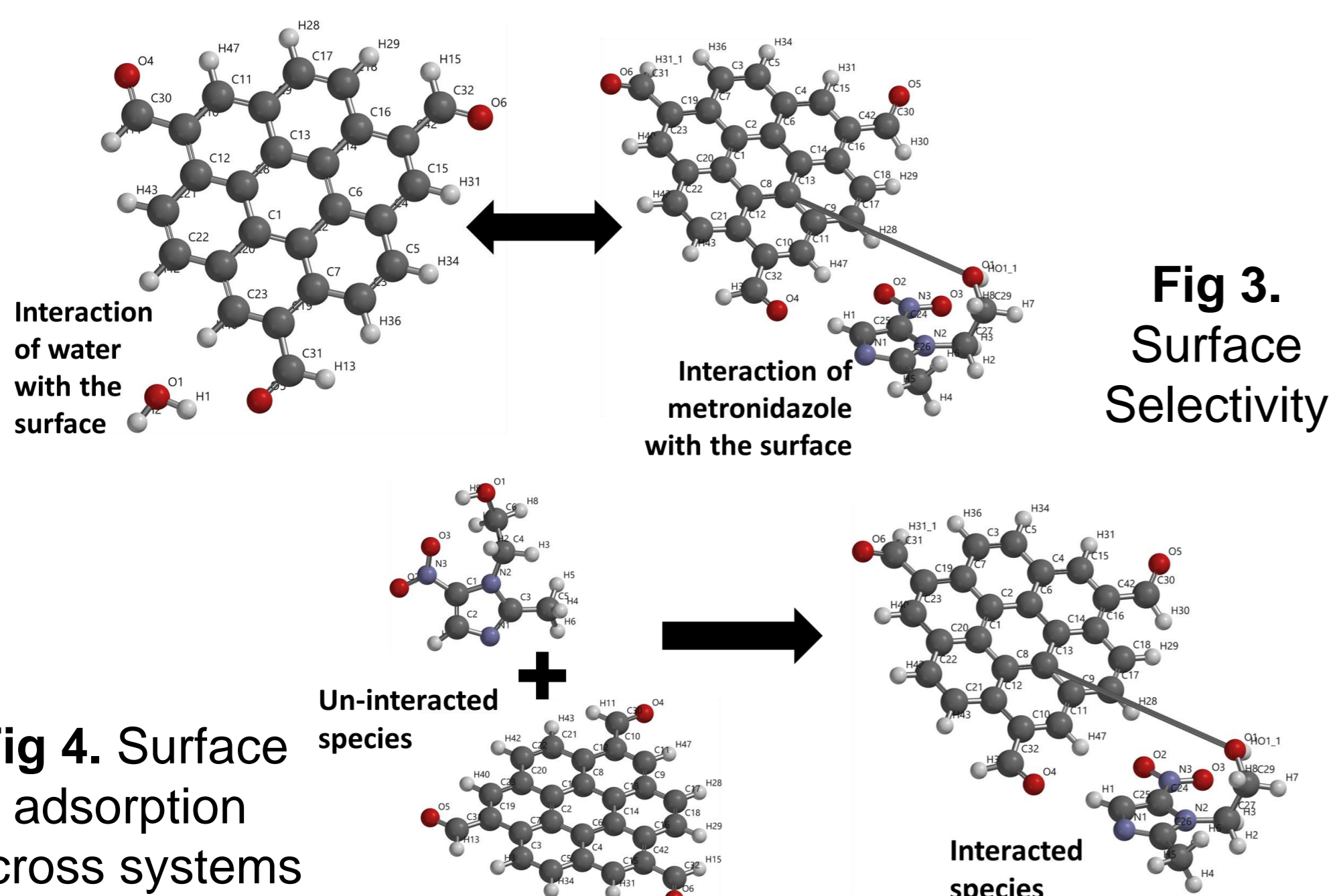


Fig 4. Surface adsorption across systems

Fig 3. Surface Selectivity

RESULTS & DISCUSSION

Table 1: The effect of water presence in metronidazole adsorption on the surfaces (all energies are in eV).

Functionalization	$E_{ads}^{met,va}$	$E_{ads}^{met,aq}$	$E_{ads}^{met,va-aq}$	$E_{ads}^{met,va-aq}$ %
Normal (g-H)	-0.06	0.06	0.12	200.00
Alcohol (g-OH)	-0.36	-0.06	0.3	83.33
Aldehyde (g-CHO)	-0.21	-0.13	0.08	38.10
Acid (g-COOH)	-0.46	-0.23	0.23	50.00
3 rd Amine (g-CN)	-0.15	-0.07	0.08	53.33
2 nd Amine (g-CHNH)	-0.32	-0.22	0.1	31.25
1 st Amine (g-CH ₂ NH ₂)	-0.19	0.02	0.21	110.53

Table 2: The selectivity of the surface for metronidazole over water across the surfaces (all energies are in eV).

Functionalization	E_{ads}^{met}	E_{ads}^{wat}	$E_{ads}^{met} > E_{ads}^{wat}$	$E_{ads}^{met} - E_{ads}^{wat}$
Normal (g-H)	0.06	-0.02	FALSE	0.08
Alcohol (g-OH)	-0.06	-0.05	TRUE	-0.01
Aldehyde (g-CHO)	-0.13	-0.03	TRUE	-0.10
Acid (g-COOH)	-0.23	-0.29	FALSE	0.06
3 rd Amine (g-CN)	-0.07	-0.03	TRUE	-0.04
2 nd Amine (g-CHNH)	-0.22	-0.17	TRUE	-0.05
1 st Amine (g-CH ₂ NH ₂)	0.02	0.07	TRUE	-0.05

Note: Met = Drug, Wat = Water, va = Vacuum system, aq = Aqueous system

CONCLUSION

- ❑ Functionalized membranes show selective adsorption for metronidazole, promising efficient wastewater treatment.
- ❑ Adsorptive properties significantly drop for most membranes, except aldehyde (-CHO) and secondary amine (-CHNH) functionalized ones.
- ❑ Aldehyde-functionalized graphene membranes exhibit superior selective adsorption of metronidazole over water.
- ❑ Functionalized graphene membranes have strong potential for mitigating pharmaceutical effluent risks.
- ❑ Improved understanding of adsorption processes can lead to better wastewater treatment technologies.

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