

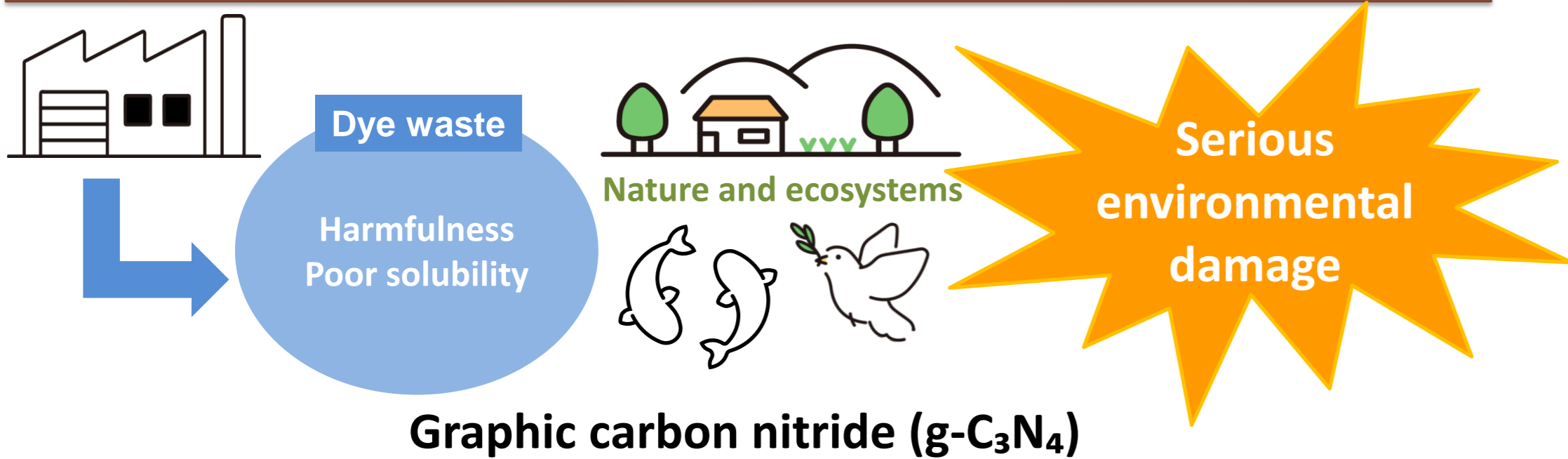
Photocatalytic dye degradation using modified $g\text{-C}_3\text{N}_4$

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INTRODUCTION & AIM



Advantages

- Inexpensive
- Easy to prepare
- Chemically and thermally stable

Disadvantages

- Narrow visible light absorption
- Quick recombination photogenerated electron-hole pair

In this research, we aim to enhance the dye degradation performance by synthesizing $g\text{-C}_3\text{N}_4$ incorporating heteroatoms.

METHOD

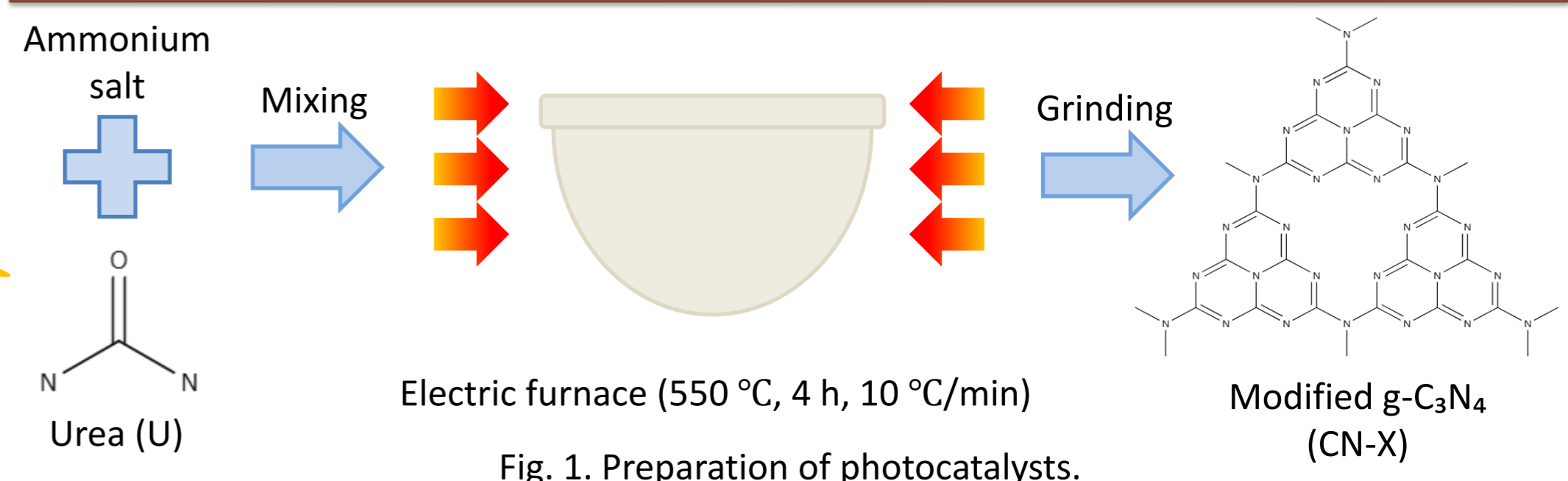
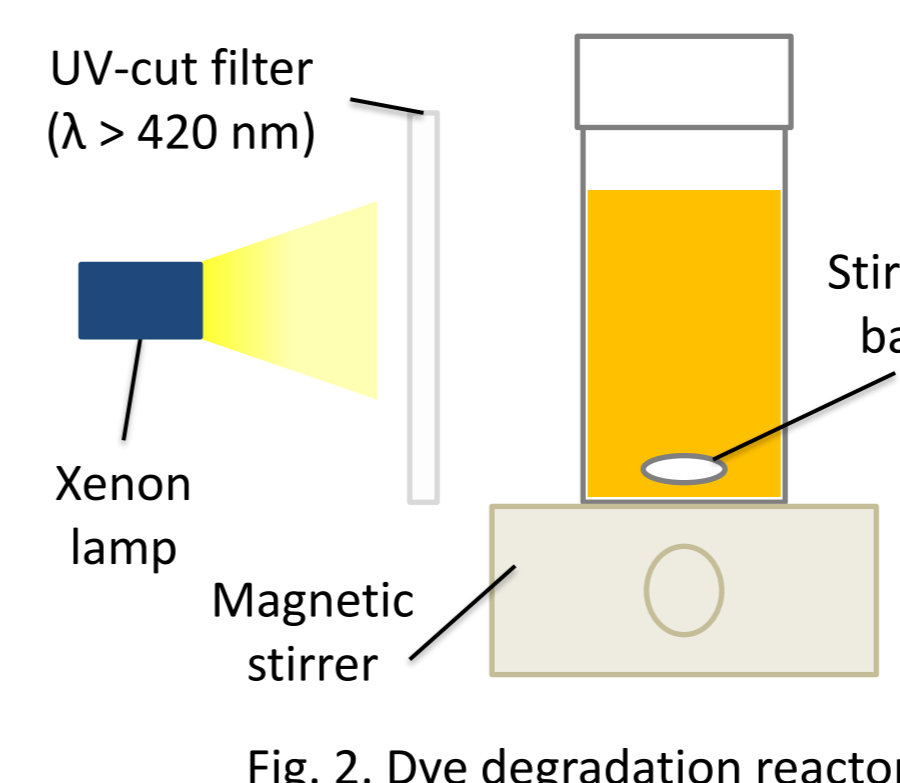


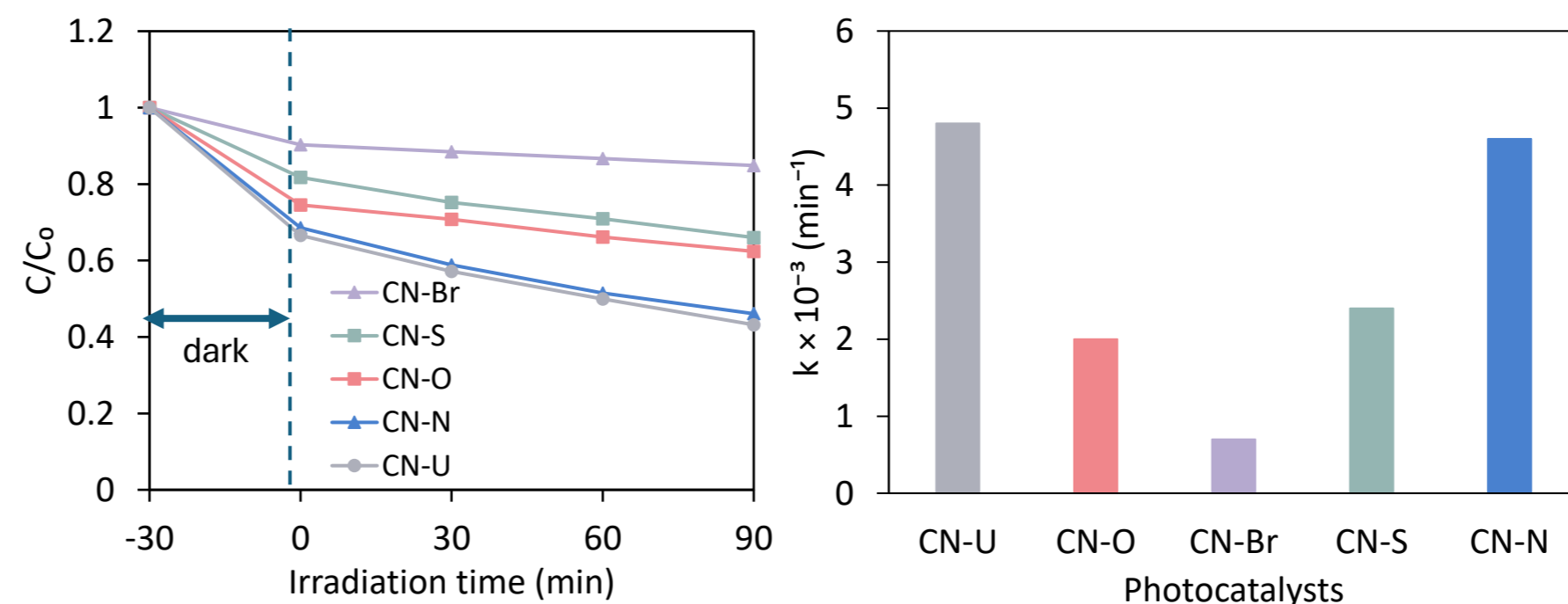
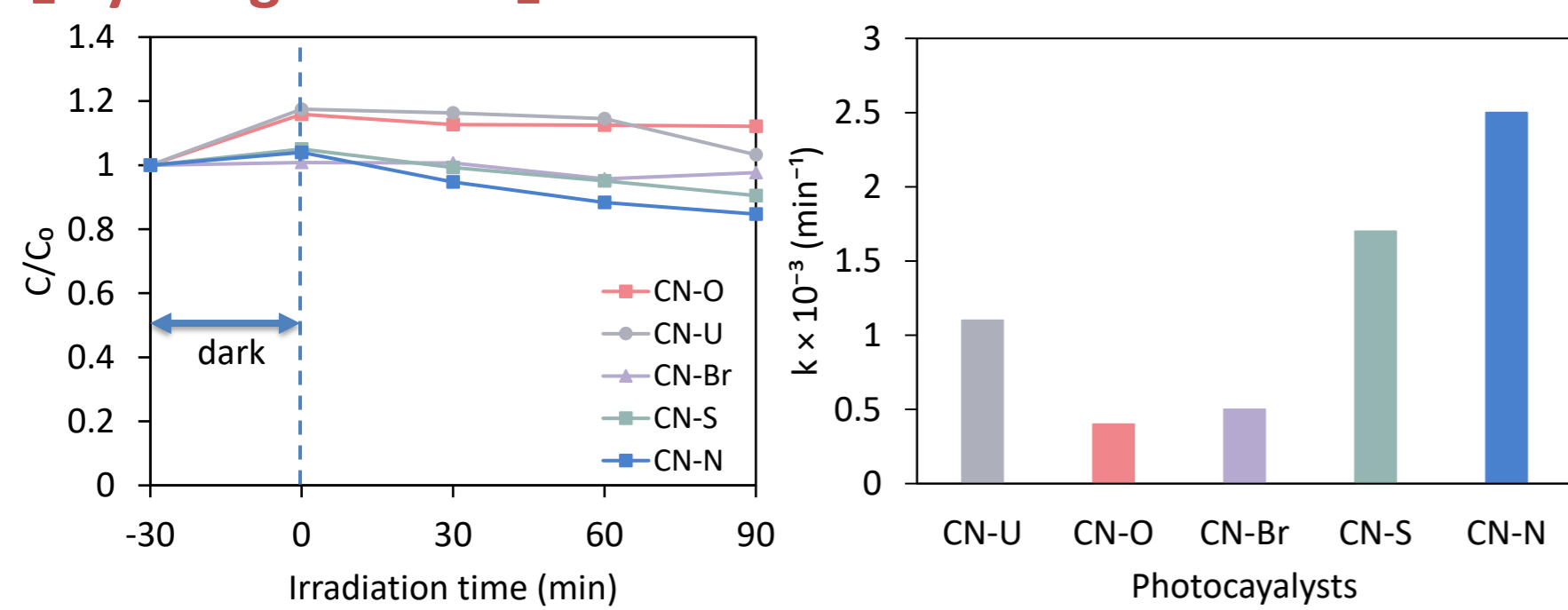
Table 1. Experimental conditions.

Sample	Methyl orange, Methylene blue (10 ppm, 35 mL)
Photocatalyst	$g\text{-C}_3\text{N}_4$ (20 mg)
Light source	Xenon lamp ($\lambda > 420$ nm)
Detection	MO : 465 nm, MB : 663 nm
Detector	UV-vis spectroscopy



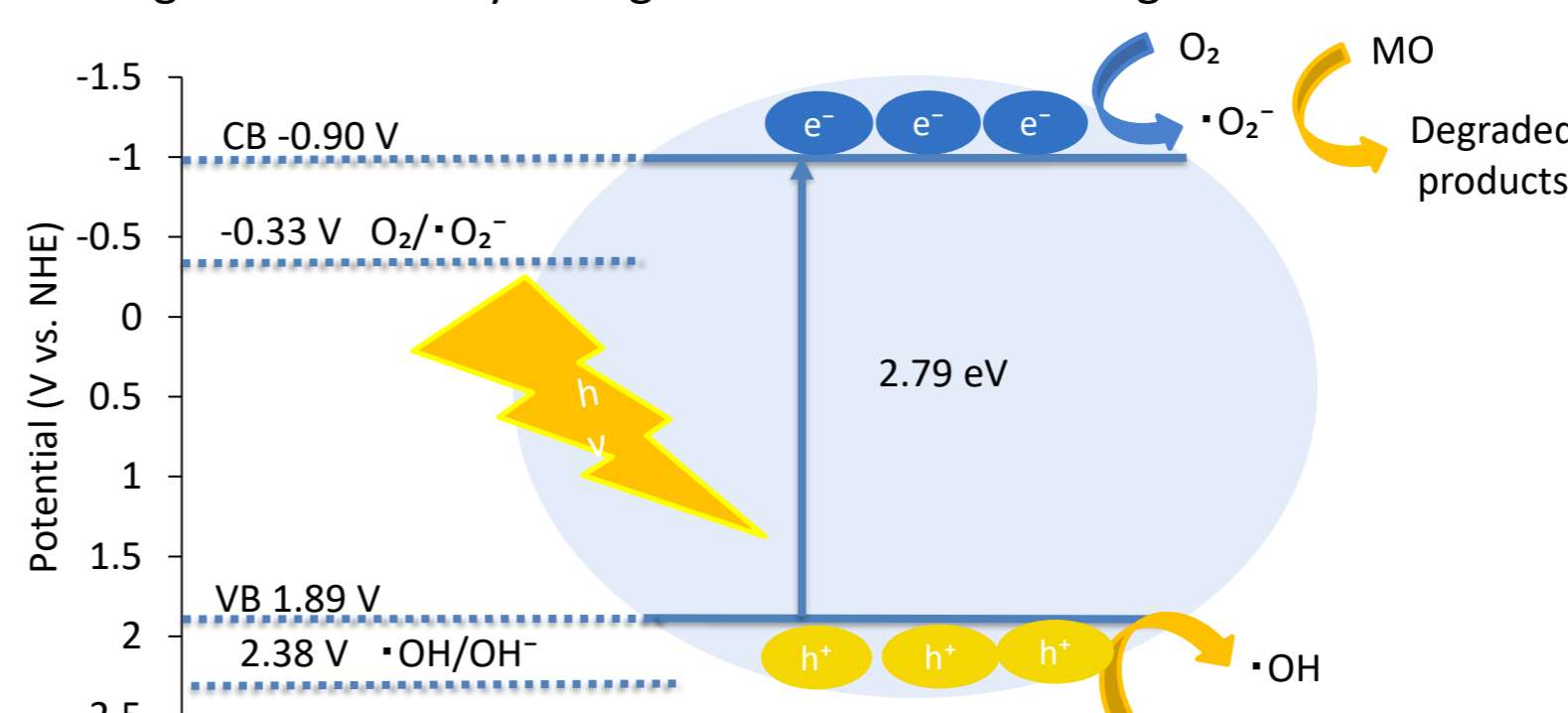
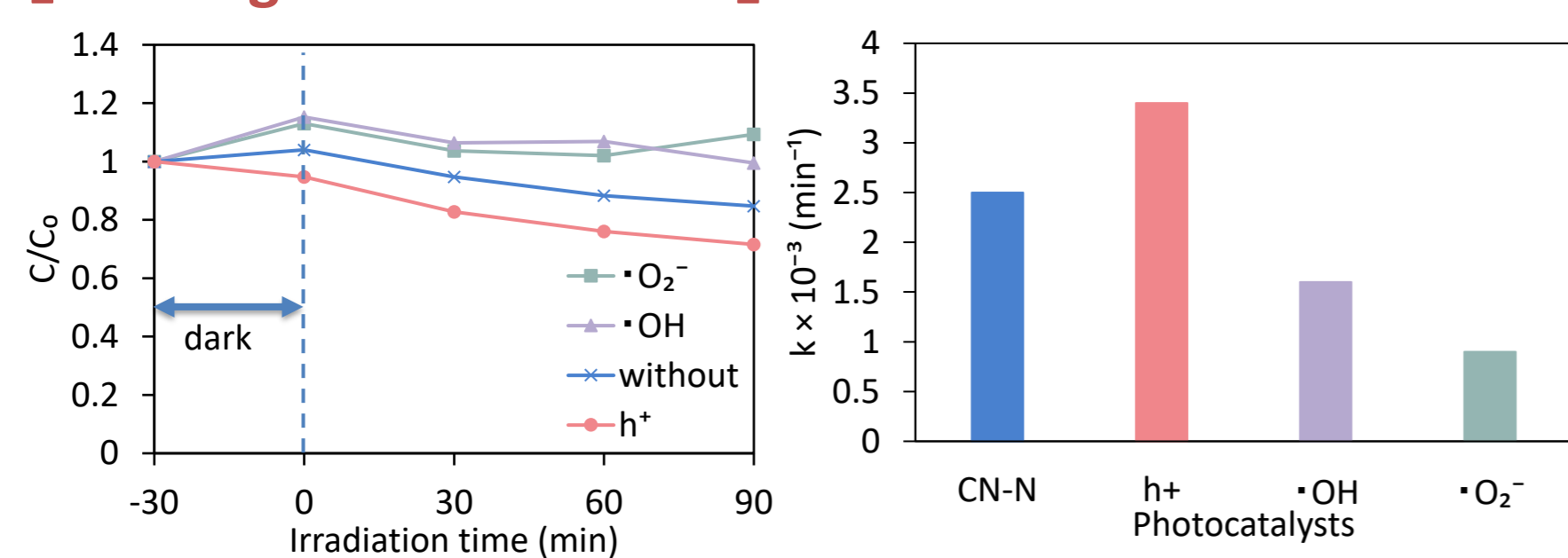
RESULTS & DISCUSSION

[Dye degradation]



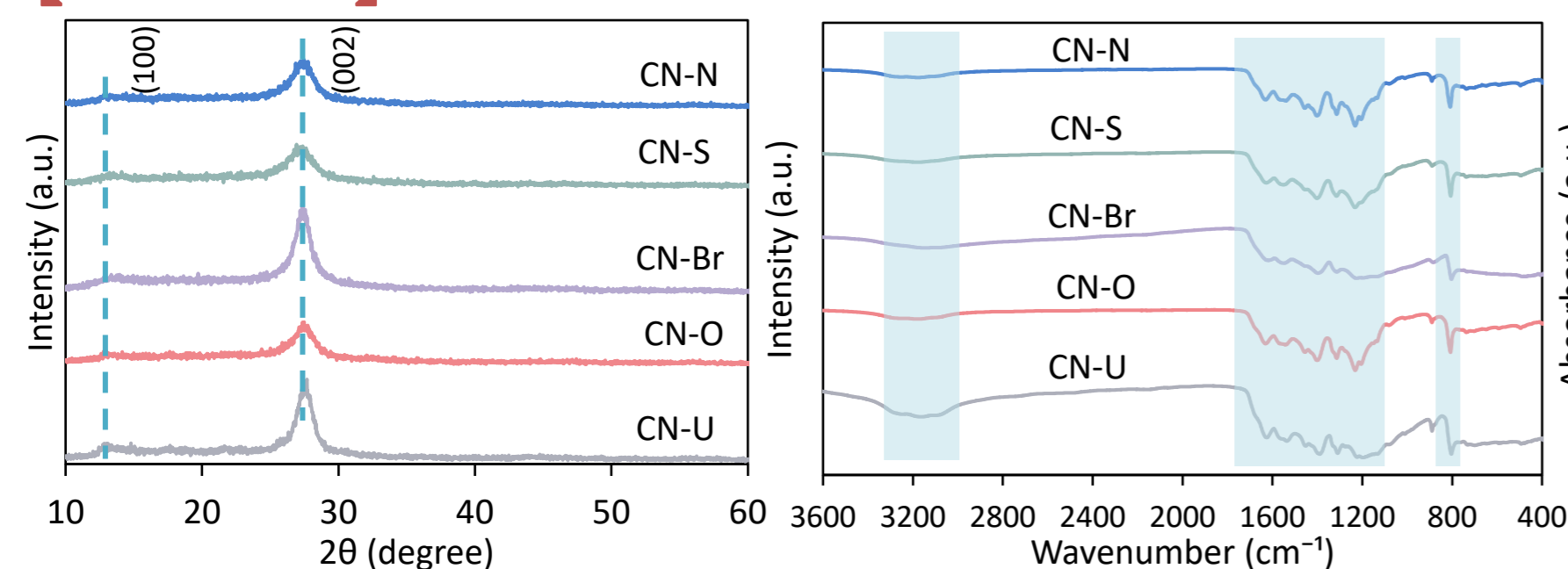
From dye degradation experiment results, for MO, CN-N degraded the most. For MB, CN-U and CN-N degraded the most.

[Scavenger & Mechanism]

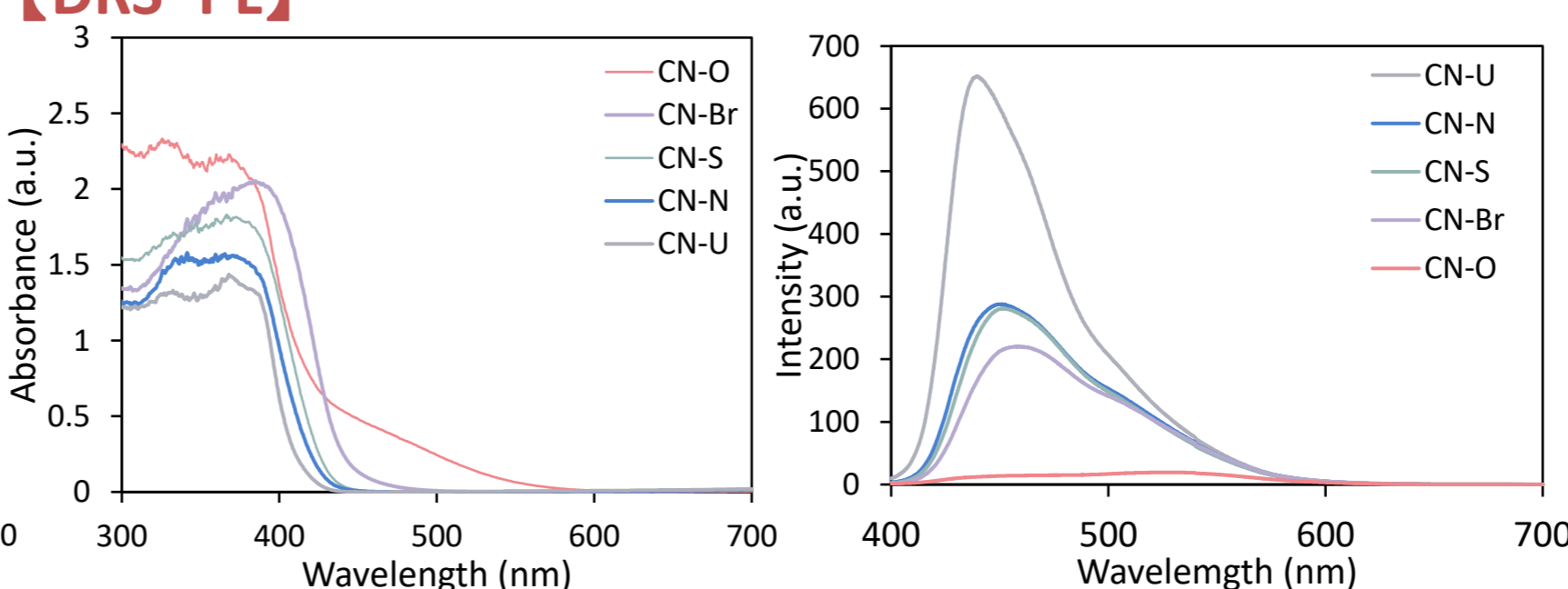


From scavenger experiment results, The main active species is $\cdot\text{O}_2^-$. The MO was decolorized by the $\cdot\text{O}_2^-$ produced by the excited electrons. The h^+ in the VB reacts with water to form $\cdot\text{OH}$.

[XRD-FT-IR]



[DRS-PL]



From XRD-FT-IR, typical structure of $g\text{-C}_3\text{N}_4$ was maintained. From DRS-PL, Optical absorption in the visible light region was increased and recombination of electron-hole pairs was suppressed.

CONCLUSION

- Successful calcined of $g\text{-C}_3\text{N}_4$ incorporating heteroatoms
- CN-N showed the best photocatalytic activity
- We think this is due to improved optical properties resulting from the incorporation of heteroatoms.

FUTURE WORK / REFERENCES

- Perform characterization
- Consider other heteroatoms (ex. Cl, Pyridine ring...)
- Consider the amount of Photocatalyst used in the experiment