

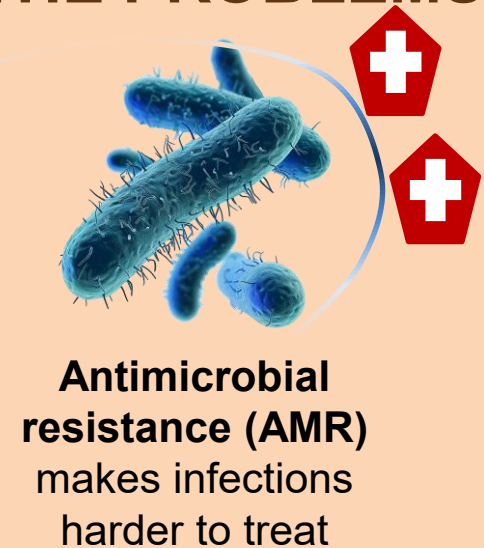
DESIGN AND CHARACTERISATION OF ZnAl-LDH-PALMITIC ACID NANOCOMPOSITES WITH PH-RESPONSIVE RELEASE AND ANTIMICROBIAL ACTIVITY

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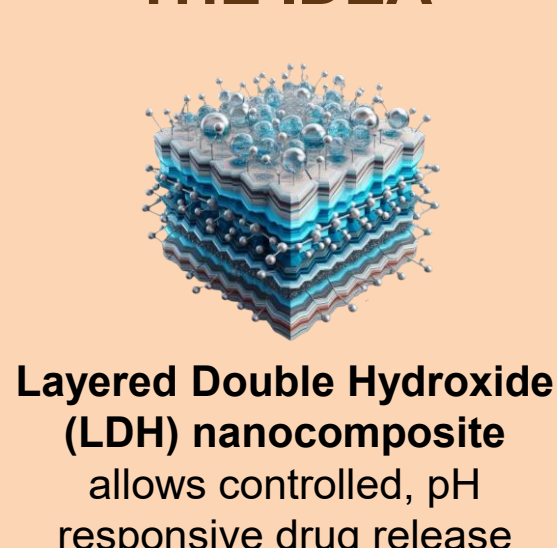
INTRODUCTION & AIM

THE PROBLEMS



Antimicrobial resistance (AMR) makes infections harder to treat

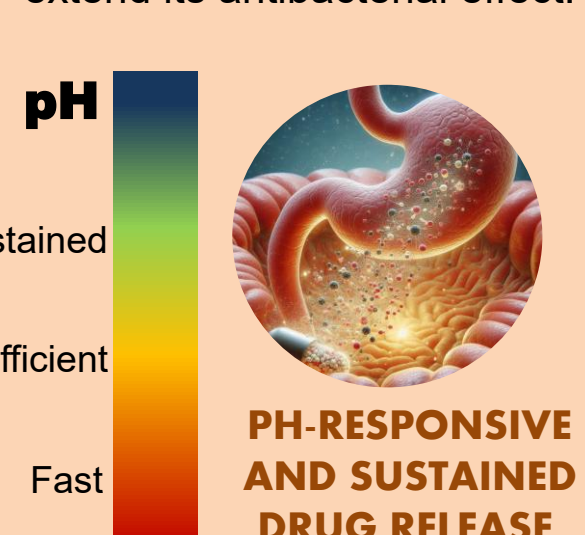
THE IDEA



Layered Double Hydroxide (LDH) nanocomposite allows controlled, pH responsive drug release

THE SOLUTION

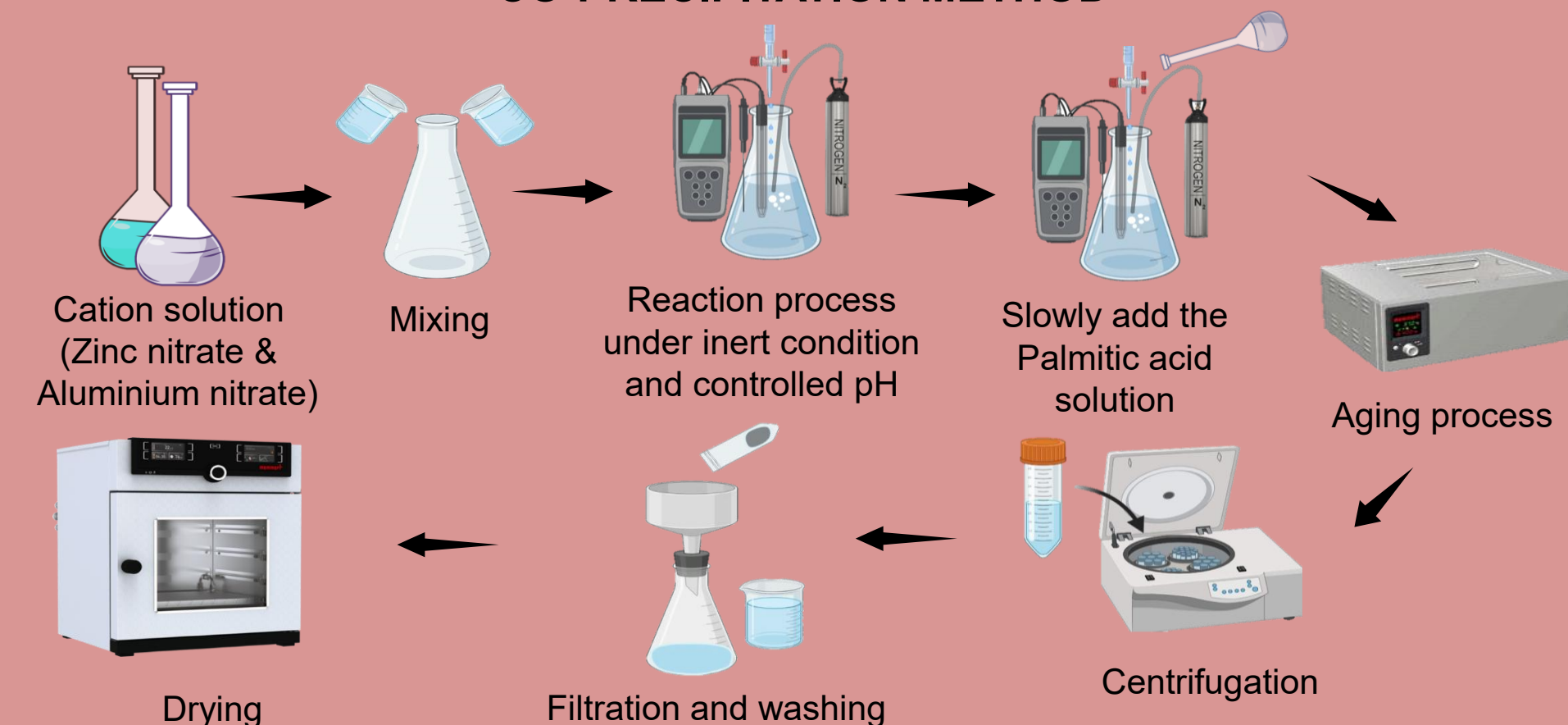
Loading PA into LDHs can improve its delivery and extend its antibacterial effect.



AIM To develop ZnAl-LDH-PA nanocomposite that protects the drug and improves bioavailability.

METHODOLOGY

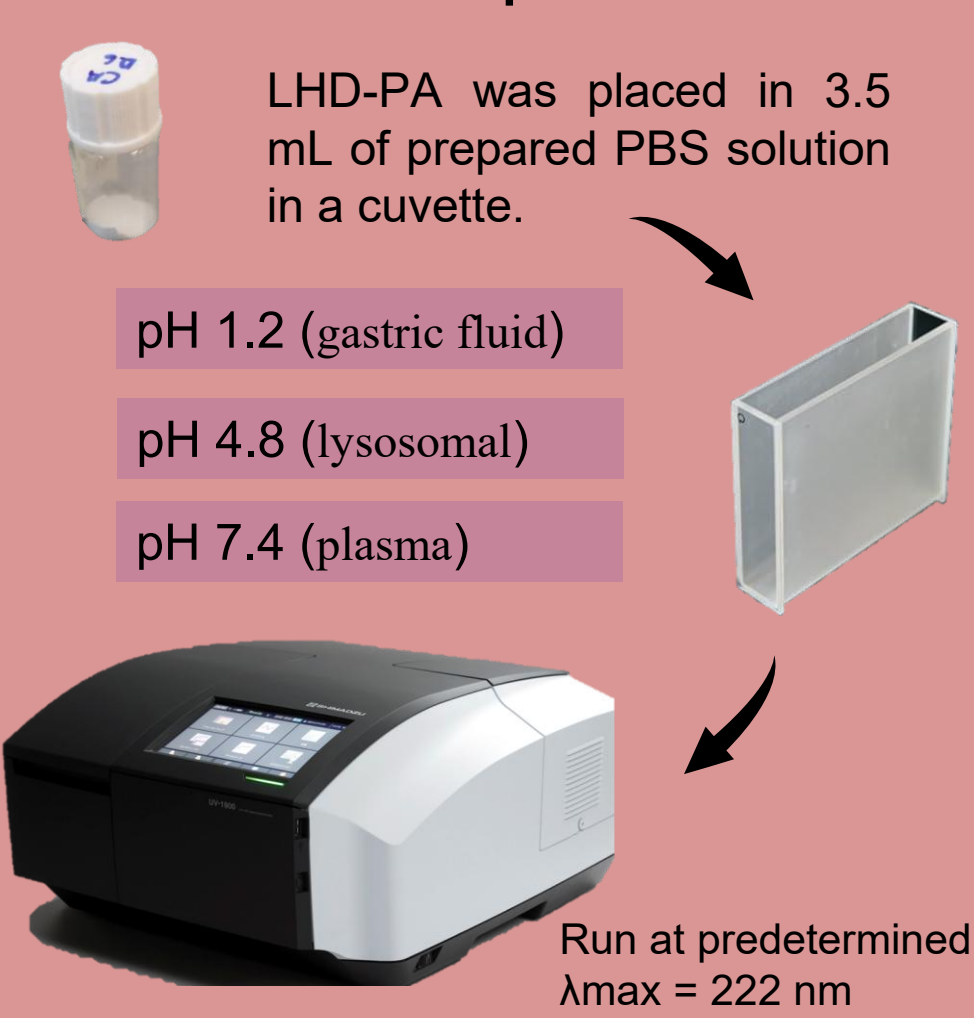
SYNTHESISING ZnAl-LDH-PA NANOCOMPOSITE VIA CO-PRECIPITATION METHOD



CHARACTERISATION

- PXRD** : crystalline structure study.
- FTIR-ATR** : Structural and chemical composition study
- FESEM-EDX** : morphological and elemental study
- TGA-DTG** : Thermal stability and decomposition study
- BET** : Surface area and porosity of study

CONTROLLED RELEASE STUDY AT DIFFERENT pH MEDIA



ANTIBACTERIAL STUDY



CONCLUSION & FUTURE WORKS

KEYFINDINGS

- ZnAl-LDH-PA nanocomposite was successfully synthesised.
- Demonstrates controlled and pH-responsive drug release behavior.
- Most efficient release: 68% at pH 4. Most sustained release: at pH 7.4
- Shows promising antibacterial activity.

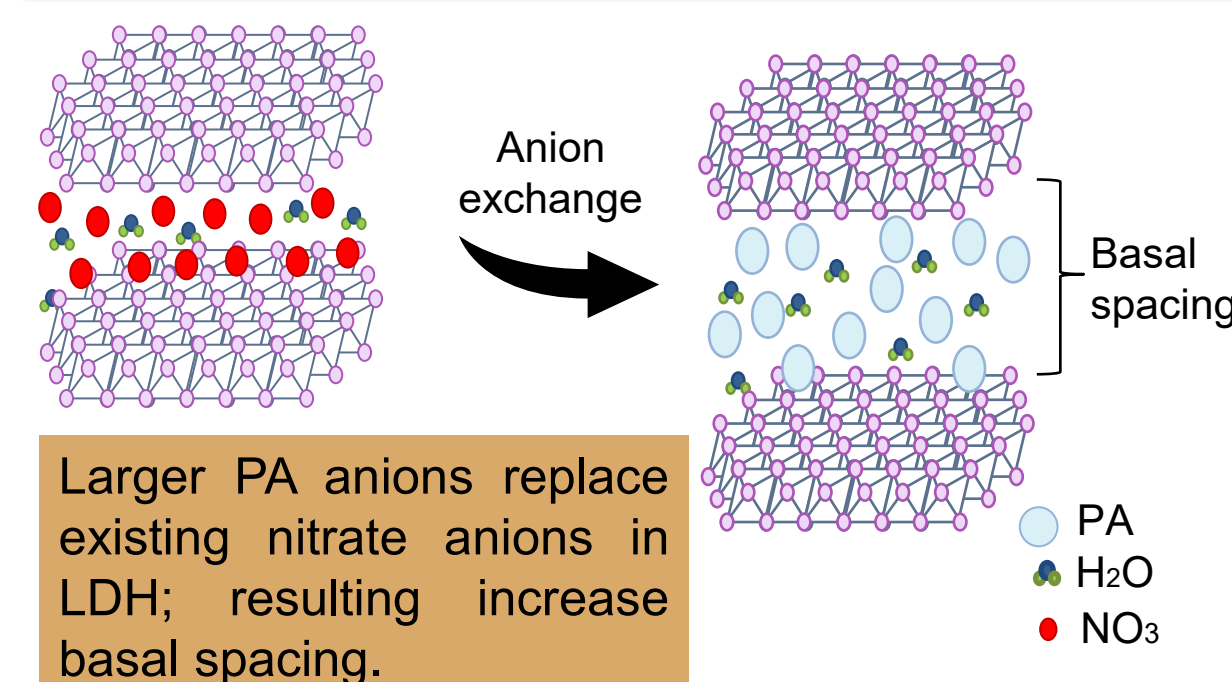
FUTURE WORKS

- Expand its antibacterial scope by testing against diverse bacterial strains.
- Validate its biocompatibility, toxicity, and release efficiency.
- Explore potential for localized antibacterial delivery in medical applications.

REFERENCE

- [1] Jadam, M. L., Syed Mohamad, S. A., Zaki, H. M., Jubri, Z., & Sarijo, S. H. (2021). Antibacterial activity and physicochemical characterization of calcium-aluminium-ciprofloxacin-layered double hydroxide. *Journal of Drug Delivery Science and Technology*, 62.

RESULTS & DISCUSSION



Successful intercalation of PA into LDH layer

- ✓ Increase in basal spacing (PXRD)
- ✓ Removal of nitrate group (FTIR)
- ✓ Addition of palmitate group (FTIR)
- ✓ Increase surface area (BET)
- ✓ Removal of nitrogen element (EXD)
- ✓ Increase thermal stability (TGA)

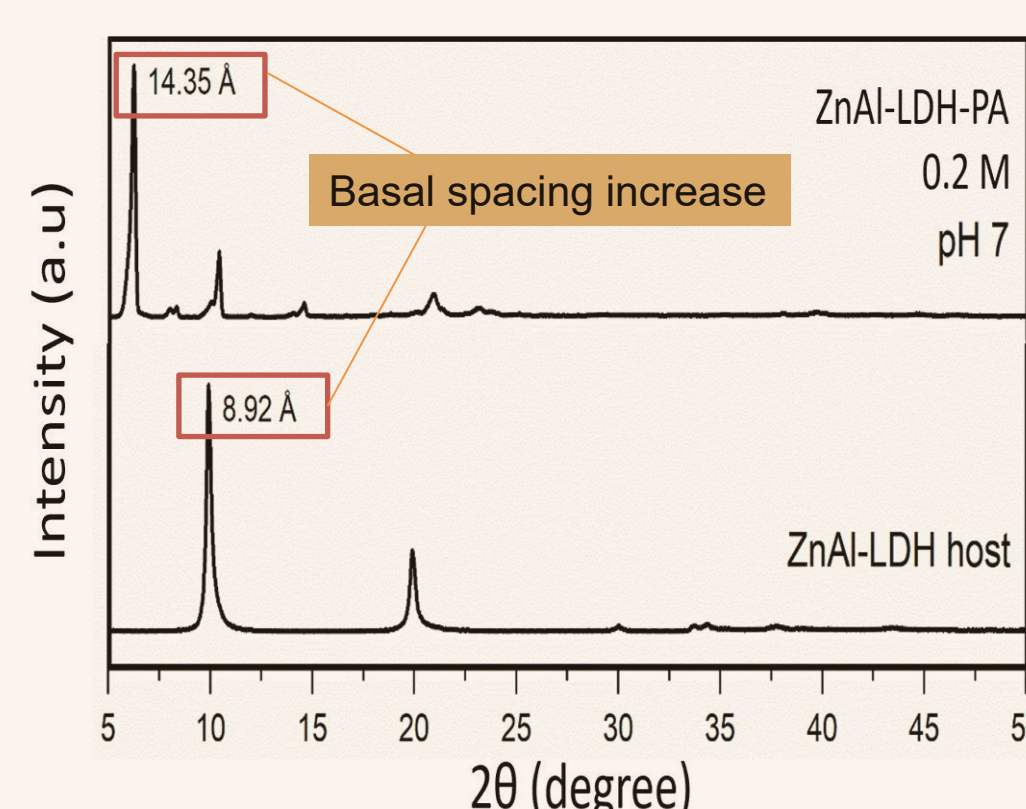


Fig. 1 PXRD diffractogram of ZnAl-LDH host and 0.2 M ZnAl-LDH-PA nanocomposite.

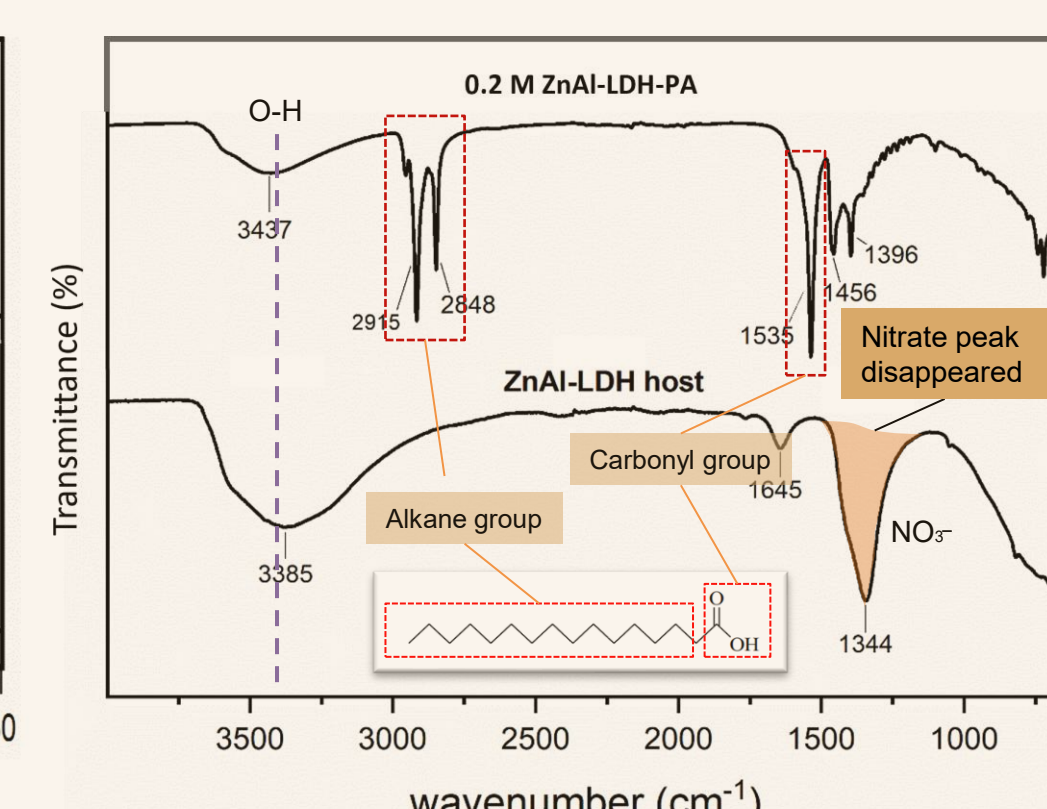


Fig. 2 FTIR spectrum of ZnAl-LDH host and 0.2 M ZnAl-LDH-PA nanocomposite.

Table 1 Surface area and pore volume

Sample	Surface area (m ² g ⁻¹)	pore volume (cm ³ g ⁻¹)
ZnAl-LDH host	4.82	4.0
ZnAl-LDH-PA	21.35	50.0
	increase	increase

H3-type: Indicative of slit-shaped pores (IUPAC)

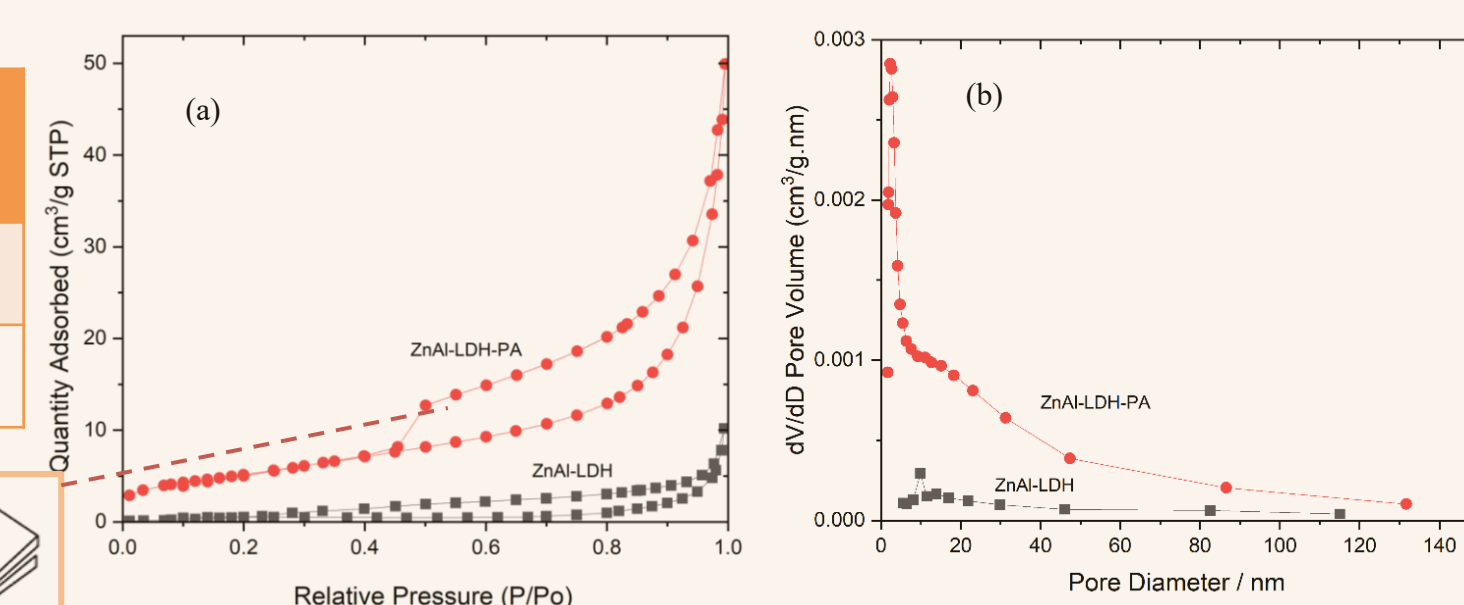


Fig. 3 (a) N₂ adsorption-desorption isotherms and (b) pore size distribution of the ZnAl-LDH host and the 0.2 M ZnAl-LDH-PA.

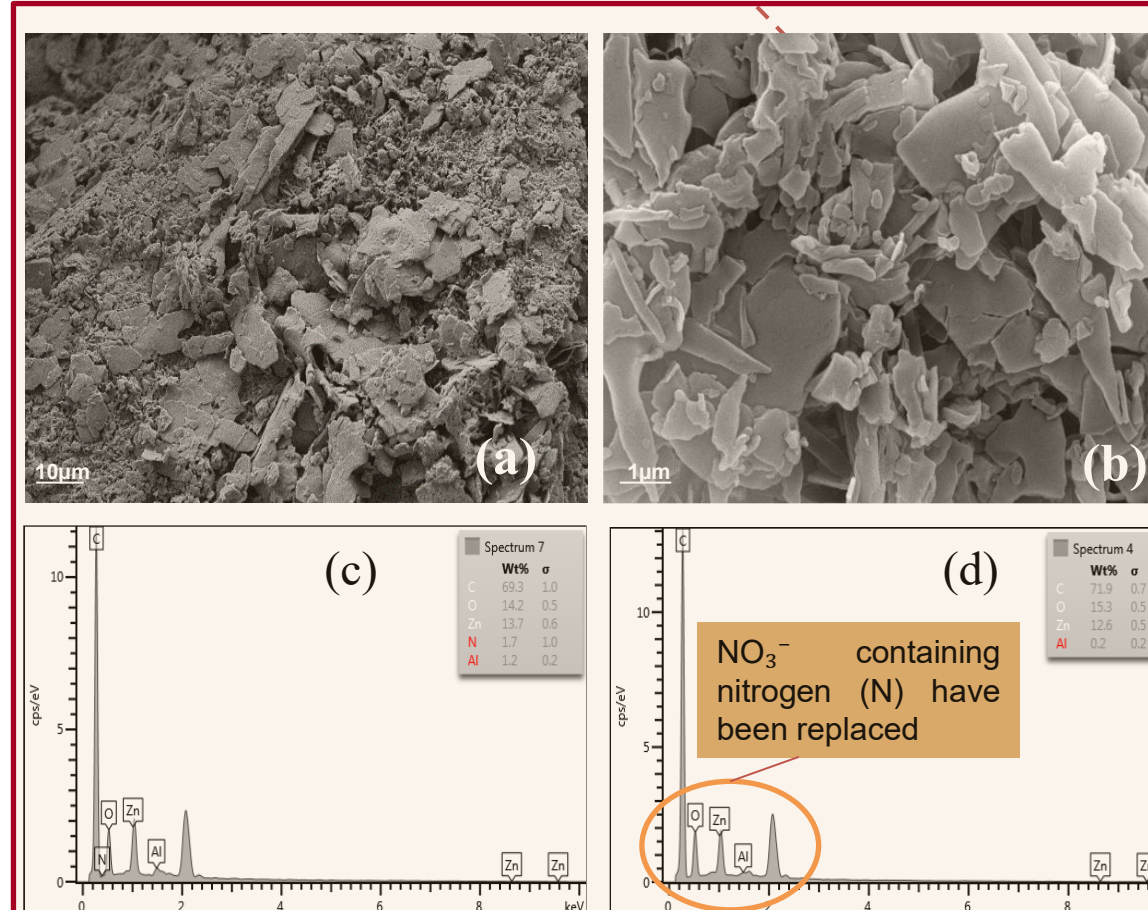


Fig. 4 (a) FESEM micrograph of ZnAl-LDH host; (b) FESEM micrograph of 0.2 M ZnAl-LDH-PA; (c) EDX spectrum of ZnAl-LDH host; (d) EDX spectrum of 0.2 M ZnAl-LDH-PA.

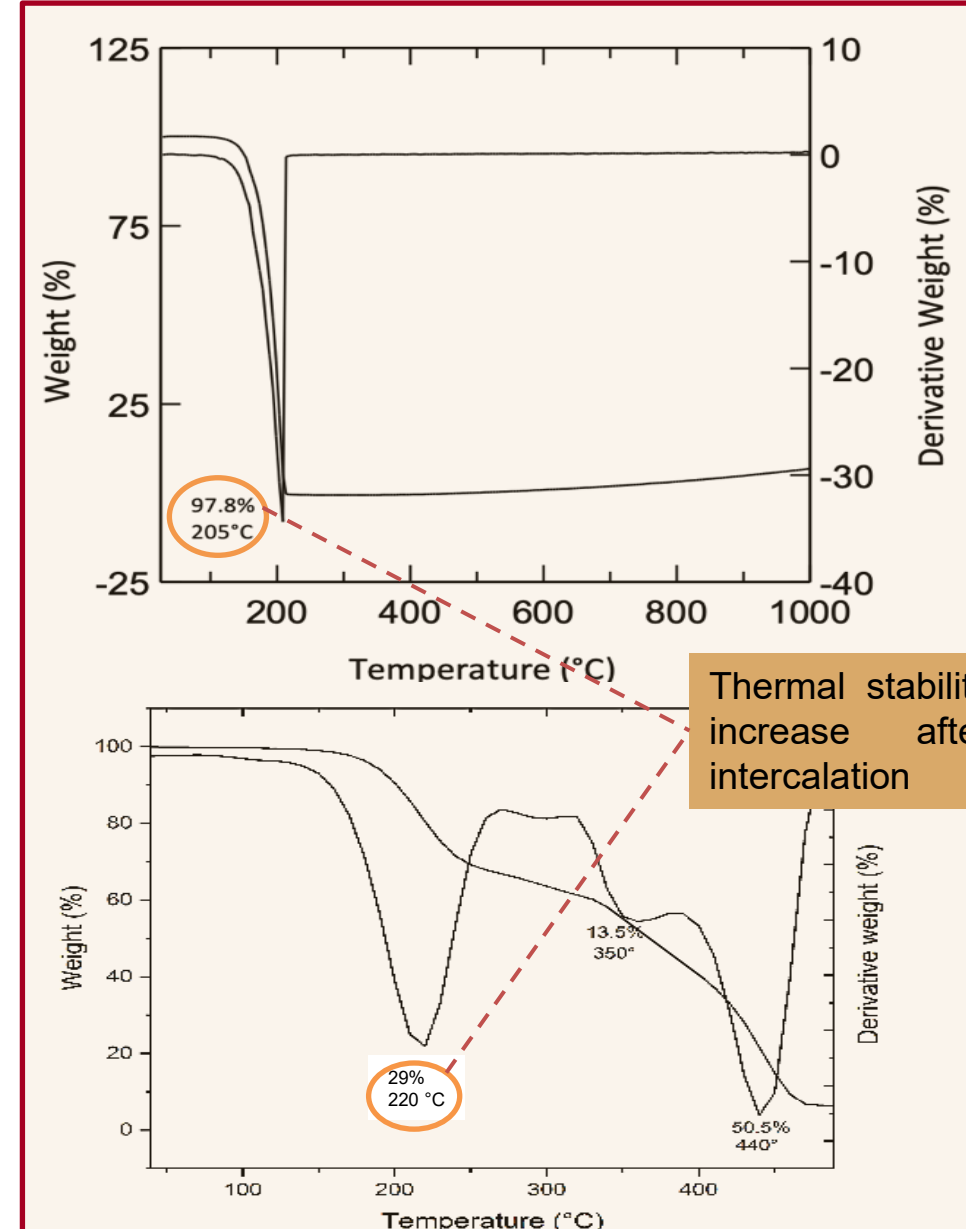


Fig. 5 Thermogram of (a) Palmitic acid and (b) 0.2 M ZnAl-LDH-PA.

Table 2 Inhibition zone diameter of PA, ZnAl-LDH host, and ZnAl-LDH-PA on *S. aureus* & *E. coli*

Bacteria	Inhibition zone (cm)	
	<i>S. aureus</i>	<i>E. coli</i>
ZnAl-LDH Host	0.6	0.6
Drug (PA)	0.6	0.6
ZnAl-LDH-PA	0.6	0.7
Positive Control	3.0	3.0
Negative Control	-	-

The intercalation of PA into ZnAl-LDH and preserves their antibacterial activity as the nanocomposite produces a similar inhibition zone to the pure drug.

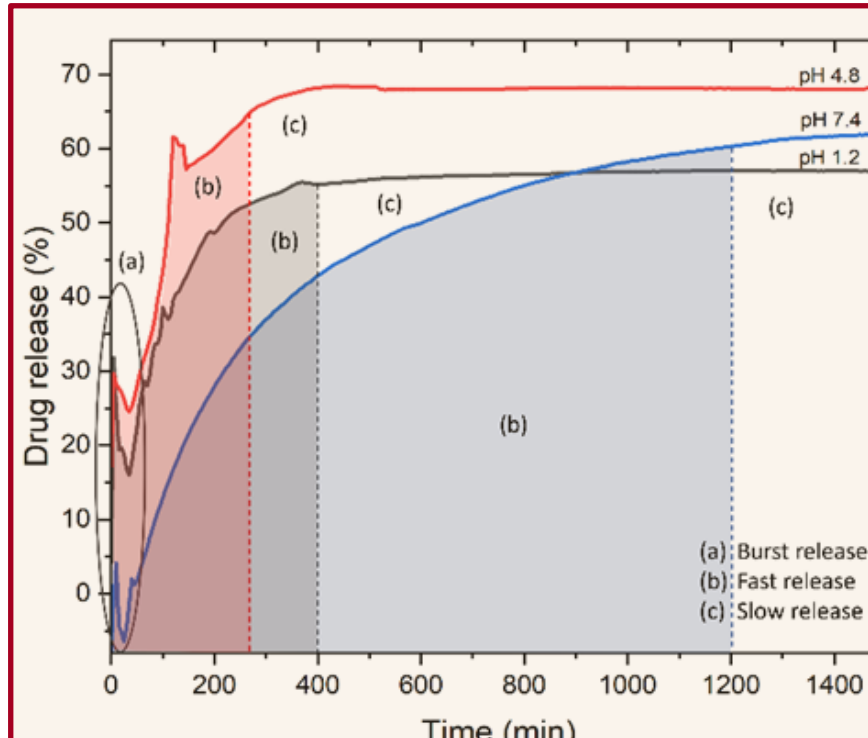


Fig. 6 The release profile of PA from ZnAl-LDH-PA in different pH levels of PBS solution

- Maximum release observed:
68% at pH 4.8 (540 min)
57% at pH 1.2 (1060min)
62% at pH 7.4 (1500 min)
- The most sustained drug release occurs at pH 7.4 where the drug is released gradually over a prolonged period.