

New Solid Solutions Based on Barium Borates Incorporating Bismuth and Rare Earth Elements: Thermal Expansion, Crystal Structures, and Photoluminescence

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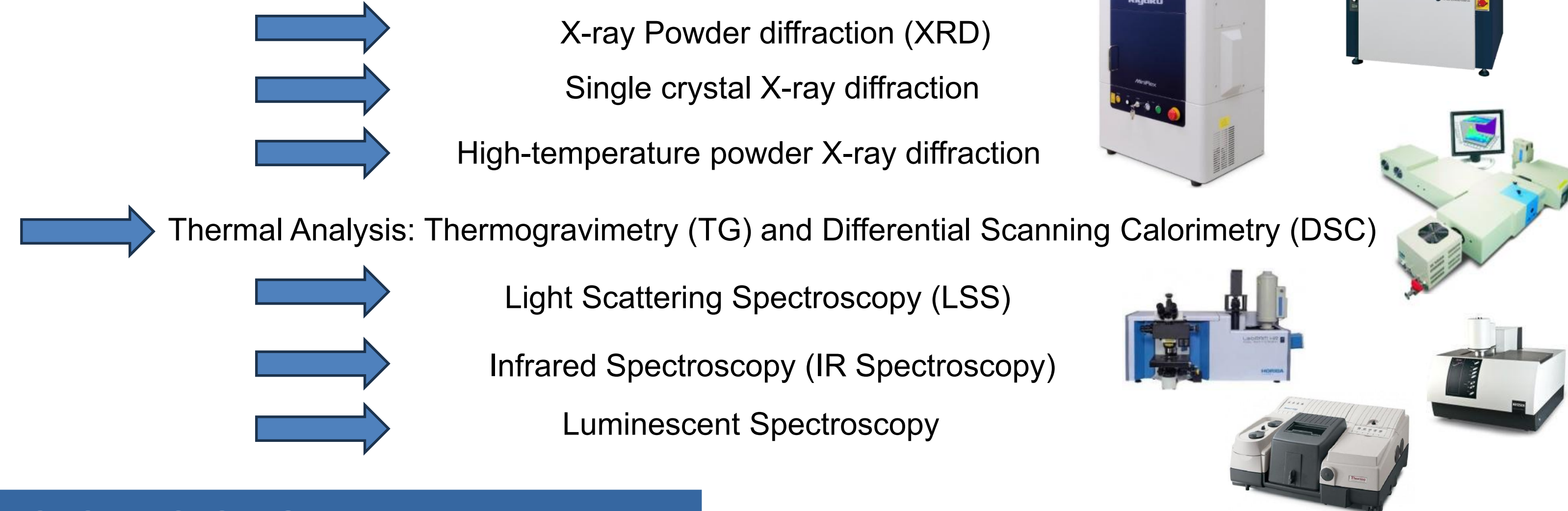
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INTRODUCTION & AIM

Seven new series of solid solutions (42 compositions total), doped with REE^{3+} ions, were synthesized based on two barium borate matrices: $BaBi_2B_2O_7$ and $Ba_3REE_2(BO_3)_4$ ($REE = Y, Eu$). Nine crystal structures were refined using single-crystal X-ray diffraction. Additional characterization included Raman spectroscopy, IR spectroscopy, thermal analysis (DSC + TG), and high-temperature powder X-ray diffraction. Photoluminescence spectra were recorded for all series, and the temperature dependence of photoluminescence intensity was examined.

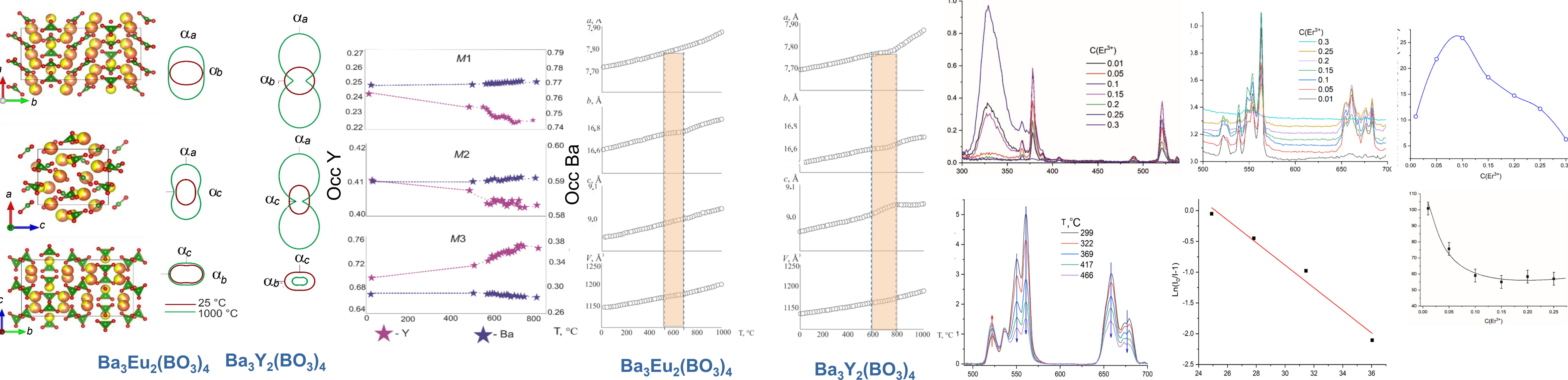
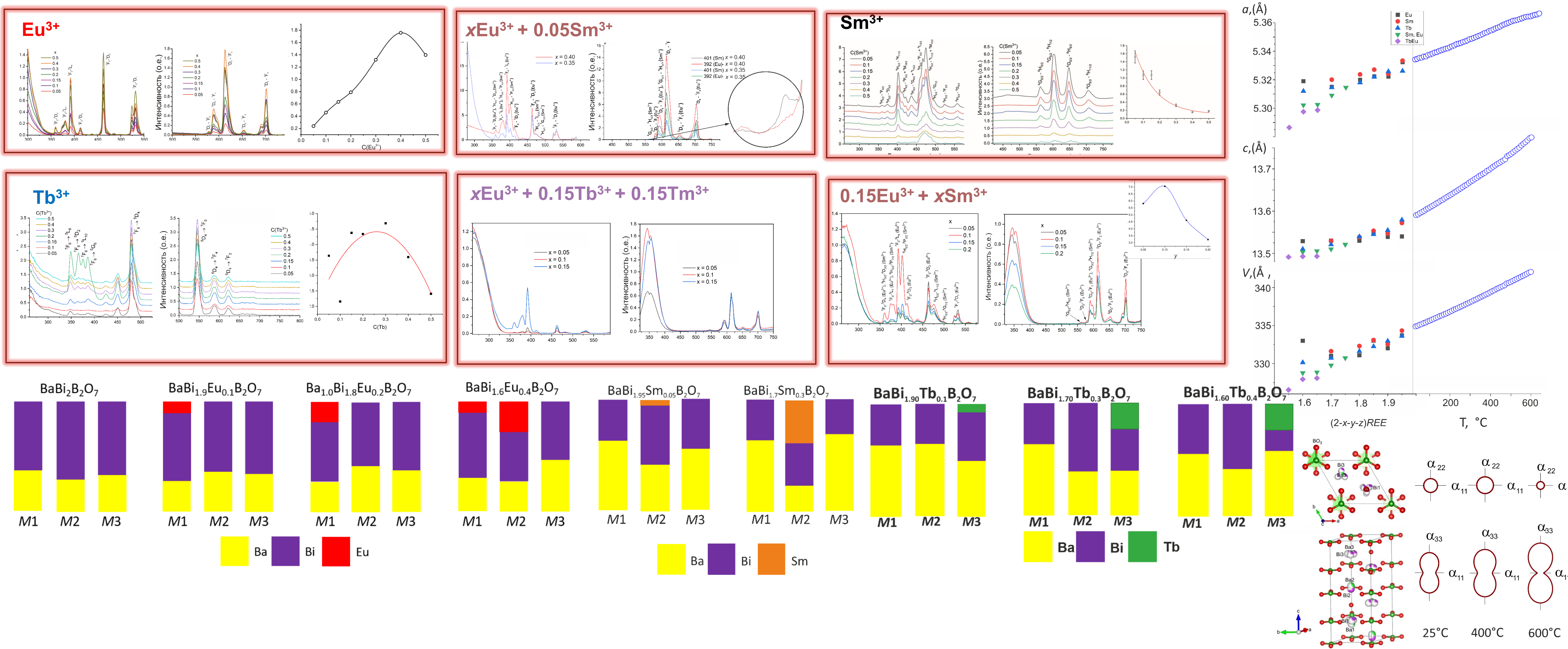
For the $BaBi_2B_2O_7$ matrix, solid solutions doped with REE^{3+} were obtained. The compositional ranges of continuous solid solutions were determined.

METHOD



RESULTS & DISCUSSION

Co-doping of the $BaBi_2B_2O_7$ lattice was found to expand the miscibility regions of the solid solutions. Eight crystal structures were refined from single-crystal data for $BaBi_{2-x}Eu_xB_2O_7$ ($x = 0.1, 0.2, 0.4$), $BaBi_{2-x}Sm_xB_2O_7$ ($x = 0.05, 0.3$), $BaBi_{2-x}Tb_xB_2O_7$ ($x = 0.1, 0.3, 0.4$). All compounds crystallize in the $BaBi_2B_2O_7$ structure type (hexagonal system, space group $P6_3$). The structure contains three independent crystallographic sites for large cations, each split into Ba and Bi subsites. REE^{3+} ions occupy the Bi subsites. Upon REE^{3+} activation, the larger Sm and Eu ions preferentially occupy the M1 and M2 sites (largest coordination polyhedra volumes), whereas Tb^{3+} , with the smallest ionic radius, occupies the M3 site (smallest polyhedral volume)



CONCLUSIONS

A new solid-solution series $Ba_3Y_{2-x}Er_x(BO_3)_4$ ($x = 0.01–0.3$), along with end-member borates $Ba_3Y_2(BO_3)_4$, $Ba_3Eu_2(BO_3)_4$, was synthesized via melt crystallization. Thermal expansion was studied for $Ba_3Eu_2(BO_3)_4$ and $Ba_3Y_2(BO_3)_4$, and the crystal structure of $Ba_3Y_2(BO_3)_4$ was refined across a wide temperature range (40 data points). Photoluminescence spectra and their temperature dependence were investigated for the $Ba_3Y_{2-x}Er_x(BO_3)_4$ series. The optimal activator concentration was determined to be $x = 0.1$

FUTURE WORK/ REFERENCES/ACKNOWLEDGMENT

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