

Unexpected S(O)₂-O Bond Scission upon Anionic Cyclarylation of Cyclic Sulfamidates. A New Route to N-Functionalized Benzosultams.

Magali Lorion, Vangelis Agouridas, Axel Couture,* Eric Deniau, Pierre Grandclaudon

Université Lille Nord de France, F-59000 Lille, France

USTL, Laboratoire de Chimie Organique Physique, EA CMF 4478, Bâtiment C3(2), F-59655 Villeneuve d'Ascq, France

CNRS, UMR 8181 'UCCS', F-59655 Villeneuve d'Ascq, France

[*Axel.Couture@univ-lille1.fr](mailto:Axel.Couture@univ-lille1.fr)

Abstract: 1,2-Cyclic sulfamidates undergo novel, efficient and regioselective intramolecular nucleophilic cleavage with aryllithiated species to provide an entry to poly, diversely and enantiopure N-substituted benzosultams.

Keywords: carbanion, anionic cyclization, sulfamidates, benzosultams.

Introduction

Five-membered cyclic sulfamidates **1** (Figure 1), which are readily available in enantiomerically pure forms represent a versatile set of aminoalcohol derived electrophiles that undergo nucleophilic cleavage through a SN₂ pathway [1].

The nucleophilic attack with a variety of (carbo) and (hetero)anionic species occurs selectively at the O-bearing carbon center [2] (Figure 2) and this regioselective ring opening usually results in the formation of a N-sulfate which can be hydrolyzed to chiral amines equipped with heteroatomic functionalities [3].

It can also be involved in annulation processes to provide a flexible entry to enantiomerically pure heterocyclic scaffolds, including benzofused systems [4].

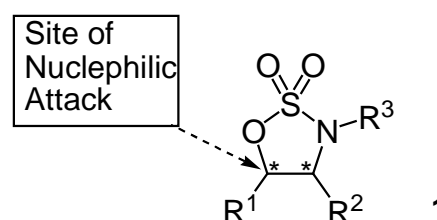


Figure 1.

Bearing these facts in mind our interest was piqued by the outcome of a carbanionic process involving an aryl-based nucleophile, e.g. an aryllithiated species, carrying an adjacent benzylic sulfamidate moiety (Figure 2). We anticipated that *intramolecular* N–S(O)₂–O bridge scission (Figure 2, **a**) could be forced and favored over *intermolecular* nucleophilic displacement at the carbon center vicinal to the oxygen atom (Figure 2, **b**).

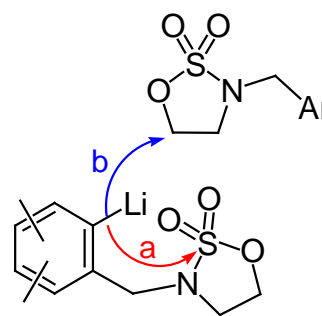


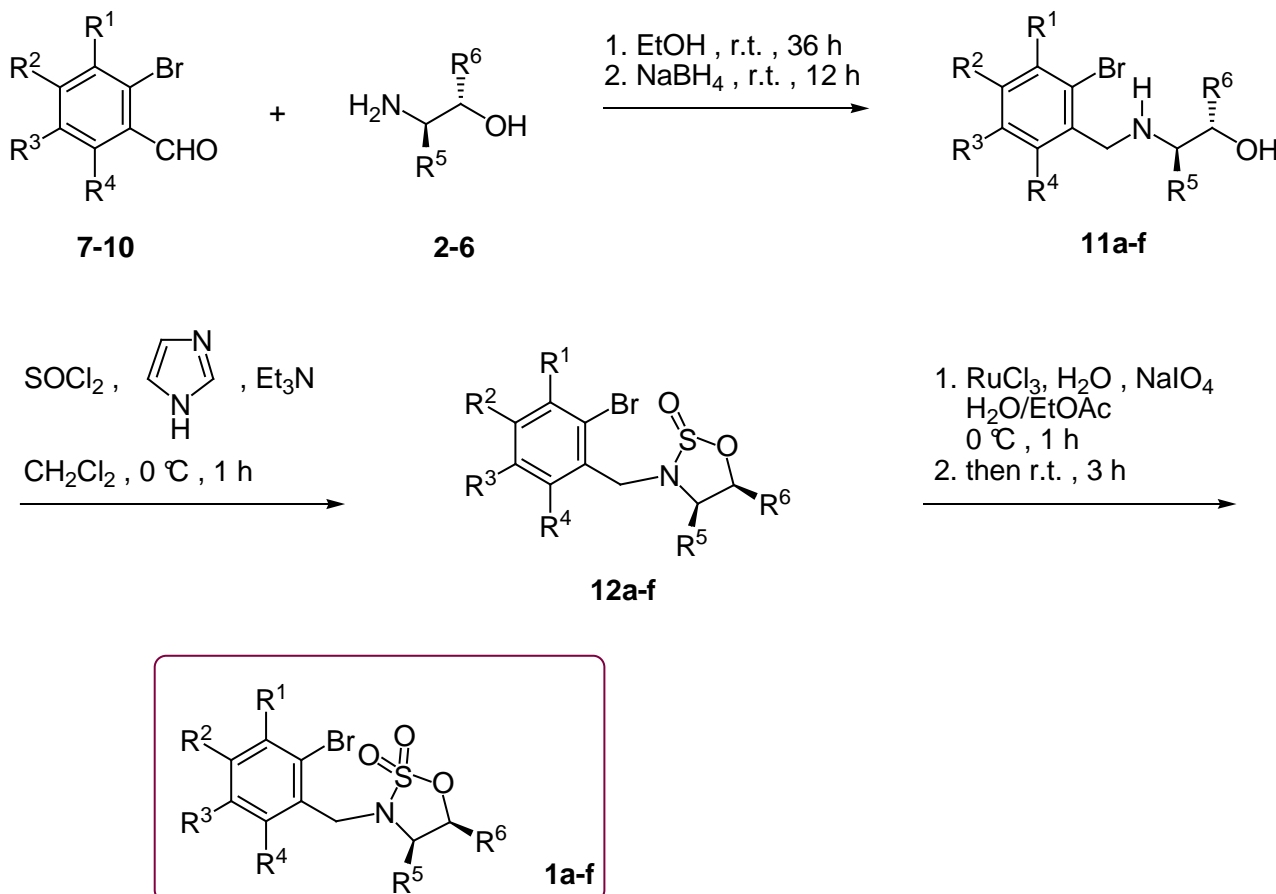
Figure 2.

We surmised that such a lithiated species could be easily generated through halogen-lithium interconversion applied to appropriate bromobenzylated precursors.

Results and Discussion

1. Synthesis of the parent bromobenzylsulfamidates **1a-f**.

A representative set of (enantiopure) sulfamidates **1a-f** was readily assembled by preliminary reductive amination reaction between a variety of bromobenzaldehydes **7-10** and an array of enantiopure aminoalcohols **2-6** (Scheme 1). Subsequent treatment of the resulting bromobenzylated aminoalcohols **11a-f** with thionyl chloride (SOCl₂) allowed for the elaboration of the five-membered cyclic sulfamidites **12a-f** (Table 1) and ultimate oxidation using catalytic RuCl₃ with NaIO₄ as reoxidant in biphasic media delivered satisfactory yields of the constitutionally diverse models **1a-f**, candidates for the planned carbanionic cycliarylation process (Table 1).



Scheme 1.

Table 1. Sulfamidites **12a-f** and Sulfamidates **1a-f** Synthesized.

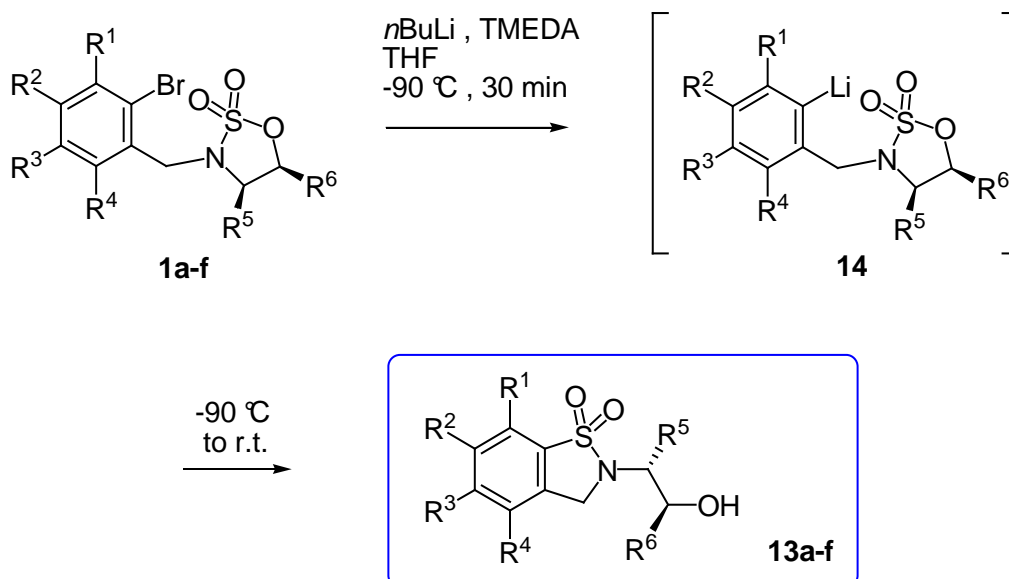
	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Starting Materials 7-10 + 2-6	Amino-alcohol 11a-f yield (%)	Sulfimidite 12a-f yield (%)	Sulfamidate 1a-f yield (%)
a	H	OMe	OMe	H	H	H	7 2	87	89	87
b	OMe	OMe	OMe	H	Ph	H	8 3	90	94	87
c	OMe	OMe	OMe	H	H	Me	8 4	91	93	90
d	H	H	OMe	OMe	H	Me	9 4	93	91	90
e	H		OCH ₂ O	H	<i>i</i> Pr	H	10 5	92	91	84
f	H		OCH ₂ O	H	Ph	Ph	10 6	47	92	87

Interestingly these operations spared the stereochemistry of the carbon centers embedded in the sulfamidate framework.

2. The anionic cyclization process. A new route to poly, diversely and enantiopure *N*-functionalized benzosultams.

To ensure the optimal formation of the mandatory lithiated species, that is **14**, variation of the ethereal solvent (THF or Et₂O), base (*n*BuLi or *t*BuLi), temperature profile (-90 °C, -78 °C, 0 °C, rt), course of the addition process (normal or reverse) and inclusion of anion modifiers (TMEDA, crown-ether) were all screened in order to facilitate halogen/metal conversion while sparing the sulfamidate moiety.

After considerable experimentation we found that adding *n*BuLi (1.1 equiv) to a degassed solution of **1a-f** (1 equiv) and TMEDA (1.1 equiv) in THF at -90 °C for 30 min (normal addition) led to the complete consumption of the starting material.

**Scheme 2.**

The intramolecular ring closure was highly efficient as demonstrated by isolation solely of the hydroxyalkyl chain tethered benzosultams **13a-f**. A representative series of annulated compounds which have been prepared by this method are presented in Table 2 where it can be seen that this simple procedure affords excellent yields of these new highly functionalized models.

Table 2. Benzosultams **13a-f** Synthesized.

	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Benzosultam 13a-f yield (%)
a	H	OMe	OMe	H	H	H	95
b	OMe	OMe	OMe	H	Ph	H	76
c	OMe	OMe	OMe	H	H	Me	82
d	H	H	OMe	OMe	H	Me	81
e	H		OCH ₂ O	H	<i>i</i> Pr	H	73
f	H		OCH ₂ O	H	Ph	Ph	90

From a mechanistic point of view one can reasonably assume that the nucleophilic attack by the preliminary formed aryllithiated species **14** triggers scission of the N-S-O bridge and ring opening is accompanied with a simultaneous new ring forming reaction to generate the sultam unit.

Benzosultams have emerged as privileged structures in drug discovery [5] and have served as key functional groups in the development of non-steroidal anti-inflammatory agents and as agonists of 5HT_{1A} receptors [6]. They have been also reported to exhibit broad inhibitory properties against a variety of enzymes, including COX-2 [7], HIV integrase [8], lipoxygenase [9], Calpain I [10] and MMP-2 [11].

Interestingly the anionic cyclization process applied to **1a-f** provoke the concomitant creation of a tailor made hydroxyalkyl appendage, possibly equipped with stereocontrolled carbon centers, which may serve as a handle for further synthetic manipulations, in particular installation of tethered functionalities liable to act on the biological profile of targeted models. The method is tolerant to a multifarious of structurally different sulfamidates and provide a flexible entry to a collection of poly and diversely alkoxyated models, one of the most challenging tasks in the elaboration of these benzofused sultams [12].

This new methodology also represents an important advance that significantly underpins the utility of cyclic 1,2-sulfamidates as effective, synthetically useful electrophiles.

Conclusion

In summary we have demonstrated that aryllithiated species are effective nucleophiles towards a set of structurally representative cyclic 1,2-sulfamidates and undergo regioselective intramolecular S(O)₂-O bond cleavage to provide the basis of a new, unusual and efficient entry to substituted benzosultams. These readily available building blocks provide the functionalized title compounds with the capacity to incorporate additional substituents/functionalities at a later stage. High yields, one-pot ring opening/ring closing reaction, procedural simplicity and easy isolation of the products are the key feature of this new methodology.

References

- [1] For a recent review, see: Bower, J. F.; Rujirawanich, J.; Gallagher, T. *Org. Biomol. Chem.* **2010**, *8*, 1505.
- [2] (a) Zubovics, A.; Toldy, L.; Varro, A.; Rabloczky, G.; Kürthy, M.; Dvortsak, P.; Jerkovich, G.; Tomori, E. *Eur. J. Med. Chem.* **1986**, *21*, 370. (b) Lyle, T. A.; Magill, C. A.; Pitzenberger, S. M. *J. Am. Chem. Soc.* **1987**, *109*, 7890. (c) Okuda, M.; Tomioka, K. *Tetrahedron Lett.* **1994**, *35*, 4585. (d) Stiasny, H. C. *Synthesis* **1996**, 259. (e) Aguilera, B.; Fernandez-Mayoralas, A.; Jaramillo, C. *Tetrahedron* **1997**, *53*, 5863. (f) Lohray, B. B.; Bhushan, V. in *Advances in Heterocyclic Chemistry*; Katritzky, A. R., Ed.; Academic Press: San Diego, 1997; vol. 68, pp. 89-180. (g) Posakony, J. J.; Tewson, T. J. *Synthesis* **2002**, 766. (h) Jiménez-Osés, G.; Avenoza, A.; Busto, J. H.; Rodríguez, F.; Peregrina, J. M. *Tetrahedron: Asymmetry* **2008**, *19*, 443 and references cited therein.
- [3] (a) Bower, J. F.; Szeto, P.; Gallagher, T. *Chem. Commun.* **2005**, 5793. (b) Bower, J. F.; Chakthong, S.; Svenda, J.; Williams, A. J.; Lawrence, R. M.; Szeto, P.; Gallagher, T. *Org. Biomol. Chem.* **2006**, *4*, 1868. (c) Bower, J. F.; Riis-Johannessen, T.; Szeto, P.; Whitehead, A. J.; Gallagher, T. *Chem. Commun.* **2007**, 728. (d) Bower, J. F.; Szeto, P.; Gallagher, T. *Org. Biomol. Chem.* **2007**, *5*, 143. (e) Bower, J. F.; Szeto, P.; Gallagher, T. *Org. Lett.* **2007**, *9*, 4909. (f) Bower, J. F.; Williams, A. J.; Woodward, H. L.; Szeto, P.; Lawrence, R. M.; Gallagher, T. *Org. Biomol. Chem.* **2007**, *5*, 2636. (g) Nguyen, H. N.; Wang, Z. J. *Tetrahedron Lett.* **2007**, *48*, 7460. (h) Xiong, Z.; Gao, D. A.; Cogan, D. A.; Goldberg, D. R.; Hao, M.-H.; Moss, N.; Pack, E.; Pargellis, C.; Skow, D.; Trieselmann, T.; Werneburg, B.; White, A. *Bioorg. Med. Chem. Lett.* **2008**, *18*, 1994. (i) Bandini, M.; Eichholzer, A.; Tragni, M.; Umani-Ronchi, A. *Angew. Chem., Int. Ed.* **2008**, *47*, 3238. (j) So, S. M.; Yeom, C.-E.; Cho, S. M.; Choi, S. Y.; Chung, Y. K.; Kim, B. M. *Synlett* **2008**, 702.
- [4] (a) Bower, J. F.; Szeto, P.; Gallagher, T. *Org. Lett.* **2007**, *9*, 3283. (b) Rujirawanich, J.; Gallagher, T. *Org. Lett.* **2009**, *11*, 5494.
- [5] Levy, I. *Drugs Future* **1992**, *17*, 451.
- [6] Dauban, P.; Dodd, R. H. *Org. Lett.* **2000**, *2*, 2327 and references cited therein.
- [7] (a) Rabasseda, X.; Hopkins, S. J. *Drugs Today* **1994**, *30*, 557. (b) Inagaki, M.; Tsurii, T.; Jyoyama, H.; Ono, T.; Yamada, K.; Kobayashi, M.; Hori, Y.; Arimura, A.; Yasui, K.; Ohno, K.; Kakudo, S.; Koizumi, K.; Suzuki, R.; Kato, Kawai, S.; M.; Matsumoto, S. *J. Med. Chem.* **2000**, *43*, 2040.
- [8] Brzozowski, Z.; Saczewski, F.; Neamati, N. *Bioorg. Med. Chem. Lett.* **2006**, *16*, 5298.
- [9] Misu, Y.; Togo, H. *Org. Biomol. Chem.* **2003**, *1*, 1342.
- [10] Wells, G. J.; Tao, M.; Josef, K. A.; Bihovski, R. *J. Med. Chem.* **2001**, *44*, 3488.
- [11] Cherney, R. J.; Mo, R.; Meyer, D. T.; Hardman, K. D.; Liu, R.-Q.; Covington, M. B.; Qian, M.; Wasserman, Z. R.; Christ, D. D.; Trzaskos, J. M.; Newton, R. C.; Decicco, C. P. *J. Med. Chem.* **2004**, *47*, 2981.
- [12] (a) Liu, Z.; Takeuchi, Y. *Heterocycles* **2009**, *78*, 1387-1412. (b) Rayabarapu, D. K.; Zhou, A.; Jeon, K. O.; Samarakoon, T.; Rolfe, A.; Siddiqui, H.; Hanson P. R. *Tetrahedron* **2009**, *65*, 3180-3188.