



Halogenated 2,1,3benzoxadiazoles as potential fluorescent warheads for covalent protease inhibitors

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Traditional Covalent Inhibitors of Proteases

Derivatives of:

- Alkyl halides,
- Michael acceptors,
- Alkyl boronates,
- ➢ Nitriles,
- Sulfonyl fluorides,
- Epoxides or aziridines.



The aim is the synthesis of fluorescent S_NAr -type warheads based on halogenated 2,1,3-benzoxadiazole derivatives.

New «warheads» for Covalent Protease Inhibitors

Electrophilic Derivatives of:

- Aromatic compounds,
- Heteroaromatic compounds,
- Quinoid compounds.

SYNTHESIS SCHEME 1



1

2

CI

CI

NO₂ 3 О



Fluorescence Spectrum of Compound 4a (7 · 10⁻⁵, MeCN)



The location of the maximum emissions in a relatively narrow wavelength region can be explained by the presence of a N-H…O=N-hydrogen bond.

Fluorescence Spectrum of Compound 4a $(7 \cdot 10^{-5}, MeOH)$

SYNTHESIS SCHEME 2



To eliminate the influence of the hydrogen bond on the fluorescent properties, we prepared a model compound from a secondary amine and studied its photophysical properties.



THE SYNTHESIS SCHEME 3



of Compound 6 ($7 \cdot 10^{-5}$, MeCN) 3000 2500 max = 540 nm2000 Intensity 1500 ex 441 1000 500 0 470 520 570 620 Wavelenght (nm)

We studied the S_NAr reaction of this compound with mercaptoethanol simulating the active cysteine residues in certain proteases and also studied the impact on the fluorescent properties.

Conclusions

- Derivatives of 2,1,3-benzoxadiazoles were synthesized as potential protease inhibitors allowing a multiparameter optical read-out and their photophysical properties were investigated.
- > The emission maxima of these compounds are located at relatively short wavelengths due to the presence of an intramolecular hydrogen bond.
- > Derivatives devoid of NH-protons have longer emission wavelengths and a second S_NAr reaction with a thiolate leads to an increase in the fluorescence intensity while keeping the emission wavelengths basically unchanged.

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Thank you for your attention!