

Article



Preliminary Studies on HVOF Sprayed Coatings on the Magnesium Alloys

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Abstract: In the field of modern techniques development, which improve and/or regenerate the component's surface properties, HVOF spraying of carbides or metals and its alloys is a good alternative method to other conventional surface engineering ones, including magnesium foundry alloys. Coatings manufactured by thermal spraying are used to improve the durability and life time of machine parts, both, the new and regenerated ones, by changing the surface layer properties. In this work the results of the HVOF sprayed coatings deposited onto AZ31 magnesium alloy substrate. The feeding material was composite powder Cr₃C₂ - NiCr. The coatings were investigated in terms of their microstructure and selected mechanical properties. For structure examinations the microscopy studies (light and scanning ones) were used as well as phase composition analysis. In case of mechanical properties, the wear resistance was determined also microhardness have been measured.

Keywords: HVOF spraying; AZ31 substrate; microstructure; microhardness; wear resistance

1. Introduction

The development of technology forces the demand for functional and technological surfaces that are increasingly more perfect. Knowledge of techniques for the production of surface layers and coatings allows their proper selection for specific operating conditions. One of the many methods of extending the life of machine parts, both new and regenerated, and as a result of increasing their reliability and operational durability is the application of coatings by means of thermal spraying technology, which consists in producing metal, carbide, ceramic and composite coatings on a properly prepared substrate with almost any chemical and phase composition. Spraying is mainly used to make coatings with high resistance to abrasion, protection against high temperature and corrosion [1–4].

One of the most commonly used methods is supersonic thermal spraying HVOF (High Velocity Oxygen Fuel), which allows the production of coatings with special properties, among others: low oxidation, very low porosity and high adhesion [5,6]. These coatings are free of oxides, which is often required due to their operational properties, and is difficult to obtain in other thermal spraying methods. The main materials used in thermal spraying HVOF are chromium carbide or tungsten carbide particles in a metal alloy matrix consisting of various combinations of Cr, Ni or Co. The one of the most common coating is Cr₃C₂/NiCr because chromium carbide coating with a ductile nickel binder is very often used in industrial environment to protect against abrasive wear and corrosion.

So, the main goal to achieve by coating with cermet is to increase wear resistance of parts and resistance to oxidation. It could also work as an alternative to hard chrome plating [7–11].

For several years, there has been a visible interest in non - ferrous alloys, including magnesium alloys, which are the subject of research in many research and development centers, as well as in major manufacturers of machine - building, chemical, energy, textile, electronic, office, aeronautical, in particular the automotive, shipbuilding, aviation and sports industries. The growing trends in the production of magnesium alloys indicate an increased need for their use in the global construction industry, and thus these alloys will become one of the most commonly used construction materials of our century, so it is extremely important to maintain a high pace of research on the problems of light alloys [12,13].

2. Materials and Methods

2.1. Coatings Deposition

Investigations were carried out on samples from the AZ31 magnesium alloy substrate (Institute of Non - Ferrous Metals in Gliwice, Skawina, Poland), with 5 mm in the thickness. The chemical composition of the alloy is shown in Table 1.

Table 1. The chemical composition of the AZ31 magnesium alloy [14,15].

Elements, in wt. %	Mn	Zn	Al	Ca	Cu	Mg
AZ31	0.17	1	3	0.04	0.05	balance

The feedstock material for the coatings manufacturing was $Cr_3C_2 - 25$ wt% NiCr. The powder particles size distribution was in the range - 30 + 5 μ m. It was a commercially available powder (Amperit 588.059 - Höganäs). The morphology of initial powder is presented in Figure 1.

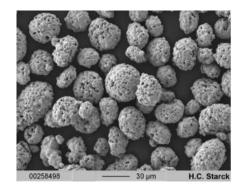


Figure 1. SEM (secondary electrons) micrographs of the Cr₃C₂ – NiCr powder [16].

The chemical composition of the feedstock powder, which was used in coatings manufacturing process is given in Table 2.

Table 2. The chemical composition of the chromium carbide powder Cr₃C₂ – NiCr [16].

Element	in wt. %	
Chromium	66 – 73	
Carbide	9 – 11	
Fe	< 0.5	
Nickiel	15 – 22	
Oxygen	< 0.6	

Before spraying all samples were sand blasted with corundum to get a surface roughness in the range of Ra equal to $17 \mu m$. Then the substrates were coated by the coating material using a C - CJS

spray system Thermico. The thermal spraying was made by CERTECH Company, Wilamowice, Poland. Kerosene and oxygen were used as the fuel gases with flow rates of 40 l/min and 350 l/min, respectively, whereas nitrogen was used as the carrier gas with a flow rate of 10 l/min. The powder feed rate was fixed at 25 g/min. The spray distance was equal to 280 mm.

2.2. Coatings Characterization

Samples for metallographic observations were prepared by standard grinding and polishing techniques. Microstructure observations were made using scanning electron microscope, SEM (Supra 35, Zeiss, Oberkochen, Germany) using secondary electron and backscattered detectors. The chemical composition was analyzed by EDS. The light microscopy (Axio Obsever, Zeiss, Oberkochen, Germany) was used to determine coatings thickness measured on random places. The average from 10 measurements was statistically calculated. In order to determine the phase composition of the Cr₃C₂ - NiCr coating, X-ray diffraction, XRD, tests were carried out. XRD investigations of sprayed coatings were done by X - ray diffractometer X'Pert Pro MPD by Panalytical (Almelo, the Netherlands) apparatus with a copper anode lamp ($\lambda K \alpha = 0.154$ nm) (Panalytical, Almelo, the Netherlands) as well as a PIXcel 3D detector (Panalytical, Almelo, the Netherlands) on the diffracted beam axis. The diffraction lines were recorded in the Bragg -Brentano geometry in the angular scope of $15 - 90^\circ$, with the step of 0.03° and the step time of 0.8 s. The analysis of the obtained diffraction patterns was made in the Panalytical High Score Plus software (Version 3.0e), containing a dedicated flat - file base of PAN - ICSD phase identification. The testing of mechanical properties was started by measurements of coating adhesion. Microhardness distribution of coated specimen from the surface to substrate was estimated using the Vickers hardness test method with a force equal to 2.94 N (HV0.3). The tests were performed along lines perpendicular to specimen surfaces, along the run face axis. The wear resistance of the manufactured coating was analyzed using the "pin - on -disc" tribological test. As a counter body, a 6 mm diameter ball of zirconium oxide, ZrO₂ was used. During the test, the friction coefficient was recorded. The test was performed at room temperature using the testing conditions given in Table 3. The wear track dimensions after tests were measured by a Sutronic 25 - Taylor Hobson profilometer (Taylor Hobson Ltd., Leicester, England), and the topography was analyzed using an SEM to determine mechanism of the wear.

Parameter	Value
Load, N	5
Linear speed, cm/s	20
Distance, m	500
Ball diameter, mm	6

Table 3. Testing conditions of the "pin – on – disc" method.

3. Results

3.1. Coatings Microstructure

The cross section micrographs show, that obtained coatings are characterized by homogeneous and dense structure. It is reveal that coating is well bonded to the substrate, there is no discontinuity at the interface. The porosity is low and the average pore size is also small. It could be observed, that the Cr₃C₂ particles are surrounded by the Ni - Cr matrix, which well adhered to the carbide particles. *Figure* 2 shows the cross - section of the manufactured coating.

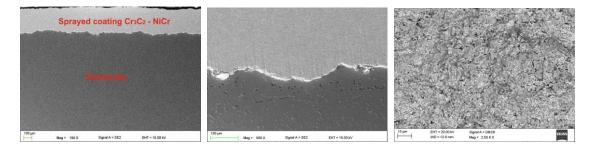


Figure 2. Cross - section of manufactured coating: (a) mag. 150×, (b) mag. 500× and (c), mag. 2500×.

The phase composition of manufactured coating is given in Figure 3. As it could be seen, there are two phases, namely Cr₃C₂ and CrNi₃. There are also some indications of C. For qualitative and quantitative analysis the RIR method [17] was used. There is about 65 % of the Cr₃C₂ phase and around 35 % of the CrNi₃ phase. Similar phase composition could be found e.g., in [18,19].

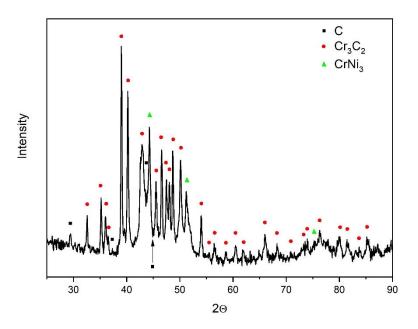


Figure 3. XRD pattern of the manufactured coating.

3.2. Microhardness

Based on the results of the investigations, it was found that the microhardness of the coating increases, and the maximum value is obtained as 790 HV0.3, while the minimum hardness is about 180 HV0.3, which means that the produced coating has increased in hardness compared to the substrate (AZ31) by about 338% (Figure 4.). An increase in temperature during the HVOF spraying process causes oxidation and the formation of a protective layer on the surface of the coating which does not affect its hardness. On the basis of XRD (Figure 3) analysis it was found that the existence of Cr₃C₂ phase provides an increase in hardness of the Cr₃C₂ - NiCr coating. It was reported in [20] that due to very high speed, powder particles can melt and dissolve carbides to varying degrees. Besides, the particles are not exactly the same size, so small particles melt completely and larger particles do not melt completely. The wide size range of powder particles causes some particles to overheat significantly while others are insufficiently melted, which may be the cause of increased porosity and reduced coating hardness. Also, an increase in the size of the carbide causes an increase in the porosity of the coating [20].

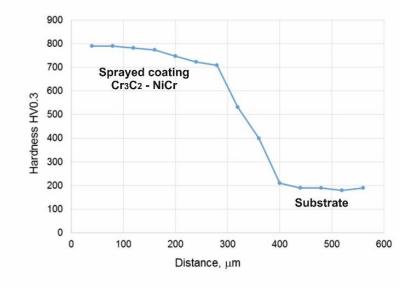


Figure 4. Hardness distribution of coated sample from the surface to substrate.

3.3. Wear Resistance

The results of dry sliding wear test show, that manufactured coating exhibit good resistance for wear. The volumetric wear of investigated coatings was equal to $13.88 \pm 3.11 \cdot 10^{-6} \text{ mm}^3/(\text{N} - \text{m})$. The friction coefficient was equal to 0.6 ± 0.03 . Similar values could be found in the literature [21,22] The typical wear trace is given in Figure 5.

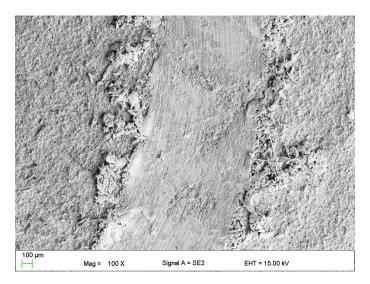


Figure 5. Wear tracks of Cr₃C₂ - NiCr coatings, mag. 100x.

During microscopic observation of wear tracks it was found, that the dominant mechanism of wear is a classic adhesive one. On the worn surface it could be observed such phenomena as decohesion areas, oxidation wear (because of increasing temperature during tests), plastic and fatigue wear and areas of smearing of the debris from counter - body (ZrO₂ ball). Similar wear mechanism could be found e.g., in [23]. Some of these phenomena are presented in *Figure* 6.

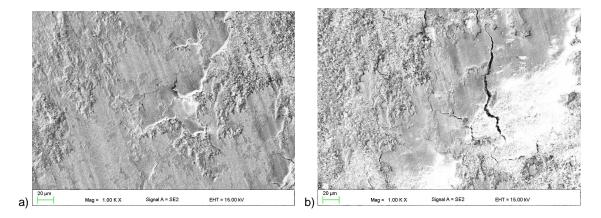


Figure 6. Different wear mechanism of Cr₃C₂ - NiCr coating: (**a**) low cycle fatigue, and (**b**) spallation and cracks.

4. Conclusions

In this premilinary studies, the Cr_3C_2 - NiCr coatings have been manufactured by HVOF method on the magnesium alloy AZ31. Based on the carried out investigations, the following conclusions can be drawn:

- The obtained coating is characterized by homogeneous and dense structure, also no cracks or discontinuities were found on the surface of the produced Cr₃C₂ NiCr coating.
- The thickness of the manufactured coating was about 300±15 μm.
- Measurements of the microhardness distribution in the sprayed coating indicate an increase up to 790 HV 0.3 (substrate material was equal to 180 HV0.3).
- In the sprayed coatings are two main phases, namely Cr₃C₂ and CrNi₃.
- The dominant mechanism of wear is a classic adhesive one.

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Conflicts of Interest: The authors declare no conflicts of interest.

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