# THE SIBI GROUP

## Introduction

All-carbon quaternary stereocenters

- All-carbon quaternary stereocenters are carbon centers that consist of four carbon substituents.
- They are widely present in many natural products, bioactive molecules and drugs.
- Developing a catalytic method to construct all-carbon quaternary stereocenters in enantioselective fashion is a challenging goal:
- due to strong steric repulsion between the carbon substituents. • Installing all-carbon quaternary chiral centers enantioselectively in acyclic system is especially difficult due to conformational flexibility.

### Enantioselectivity and cyclopropanation

- Enantioselectivity is a preference for the formation of one enantiomer over the other in a chemical reaction.
- Cyclopropane moieties are widely found in therapeutic agents and biologically active molecules.
- Because most drugs and natural products that contains the cyclopropane ring skeleton are chiral, it is important to develop catalytic methods that produce highly enantioenriched cyclopropanes.
- Michael-Initiated Ring closure (MIRC) reaction with chiral Lewis acid catalysts enables the asymmetric formation of cyclopropanes that contain quaternary and tertiary chiral centers.

### **Previous Studies**

Chiral Lewis acid catalyzed Michael-Initiated Ring Closure (MIRC) reaction to synthesize contiguous tertiary chiral centers

		(—)			
Entry	Lewis acid (L.A.)	Ligand (L)	Time (h)	Diastereomeric ratio (d.r.)	Yield(%
			O R	$ \begin{array}{c}                                     $	a), Bn ( <b>b</b> )
BnQ N N N	O I Ph + E	$EtO_2C$ $H$ $Br$ $Br$	CO <sub>2</sub> Et Et	A. (30 mol%) 33 mol%) 33 mol%) 3N, 4Å MS CM, r.t. 16-18h E	O H, Ph + tO <sub>2</sub> C CO <sub>2</sub> Et

		(12)			
1	$Co(ClO_4)_2$	a	16	99:01	95
2	$Ni(ClO_4)_2$	a	16	99:01	95
3	$Ni(ClO_4)_2$	b	16	99:01	80
4	$Ni(OTf)_2$	a	18	99:01	95
5	$Ni(BF_4)_2$	a	18	99:01	95

Ramkumar Moorthy, Novel asymmetric approaches for the construction of small molecules by cyclization and cycloaddition, **2015**, NDSU

- obtained
- centers.
- quaternary chiral centers



Entry	Lewis acid (L.A.)	Diastereon ratio (d.	
1	$Mg(OTf)_2$ (30 mol%)	N.D.	
2	$Mg(NTf_2)_2$ (30 mol%)	N.D.	
3	$Mg(ClO_4)_2$ anh. (30 mol%)	N.D.	
4	$Cu(OTf)_2$ (30 mol%)	N.D.	
5	$Ni(BF_4)_2 \bullet 6H_2O (30 \text{ mol}\%)$	80:8:12	
6	$Ni(ClO_4)_2 \bullet 6H_2O (30 \text{ mol}\%)$	77:11:1	
7	NiBr <sub>2</sub> (30 mol%)	67:5:28	
8	NiBr <sub>2</sub> (20 mol%)	63:6:3	
9	NiBr <sub>2</sub> (10 mol%)	76:3:2	
10	$Ni(BF_4)_2 \bullet 6H_2O (10 mol\%)$	83:6:1	



- **6)** Enantiomeric excess (ee%) 26 97 13 98 98

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