Passive samples (POCIS) in ponds and wetlands to evaluate pesticide degradation using compound-specific isotope analysis

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Content

• Passive samplers and compound specific isotope analysis (CSIA)

• Effect of concentrations, matrices, flow conditions and hydraulic regimes on pesticide and isotope fractination

• Field application : POCIS for pesticide CSIA in stormwater wetland

Objectives

- Develop the use of passive samples (POCIS) to facilitate CSIA analysis (preconcentration / in situ purification step) at water-sediment interface
- Collect the average isotope signature / concentration of pollutants accounting for highconcentrated short runoff events
- Compare POCIS as an alternative to grab sampling less cost/time-consuming
- Tease apart destructive and non-destructive processes in the field with pesticide CSIA



Compound Specific Isotope Analysis (CSIA)

CSIA measures the isotope ratios e.g., ¹³C/¹²C of individual contaminants





Carbon-12 6 protons 6 neutrons 98.9%

Carbon-13 6 protons 7 neutrons 1.1%

Isotopic analysis of contaminants is directly linked to degradative processes





POCIS: passive sampler



Stainless steel disks

Semi-permeable polyethersulfone (PES) membrane filled with 200 mg

Widely used as alternative to grab water samples for concentrations = > Now tested for **CSIA** (this study)

Lab experiments

Estimate sorption and isotope signatures for different compounds under varying pesticide concentrations, matrices, flow conditions and hydraulic regimes







Experiment 2

Experiment 1

POCIS: sorption experiment 1

Effect of concentrations, matrices and flow conditions tested for atrazine, S-metolachlor, terbutryn, metalaxyl, dimetomorph and tebuconazole



DI - dionized water; RF water - stormwater water

POCIS sampled after 5,10, 20 and 30 days

K_{OW Atrazine, Metalaxyl} < K_{OW Terbutryn, S-Metalochlor}

EXP 1. NO carbon isotope effects due to sorption on POCIS

RF water Slow flow 10μg/L RF water Slow flow 1μg/L

Tank 3

Tank 2

Summary 1st experiment POCIS

- 1. Flow, pesticide concentrations and compound physicochemical characteristics, but not the matrix, affected sorption (N_{POCIS}). The equilibrium was not reached for 1µg/L condition
- 2. No isotope effects due to sorption on POCIS, however, changes in isotope signatures due to matrix effects at low concentrations for some compounds
- 3. <u>Projection</u>: POCIS (after 30 days) could allow isotope fractionation-free measurements of environmentally relevant concentrations
 -Atrazine, metalaxyl, terbutryn, dimetomorph at 0,3-0,5µg/L
 -Tebuconazole, S-metolachlor at 0,1-0,2µg/L

POCIS: sorption experiment 2

Estimation of sorption for each compound under distinct hydraulic patterns

• When to collect the samplers from the field and what information it will represent?

low = low flow, low concentration *high* = high flow, high concentration (simulation of storm event)

Tank 1 : HIGH concentration / HIGH flow

Tank 2 : LOW concentration / LOW flow

Metalaxyl (Ridgold) $\delta^{13}C = -29.5\%$ Tebuconazole S-Metolachlor D-11 Atrazine Metalaxyl (Sigma) $\delta^{13}C = -31.5\%$ Tebuconazole S-Metolachlor

• Metalaxyl proxy for mixing

• S-Metolachlor D-11 and Atrazine proxy for accumulation during storm event

Compounds in both tanks

Com - commercial POCIS Self - self-made POCIS

Similar pattern for all compounds Self-made ≈ COM Self-made POCIS 1≈2=3>4 as expected Com POCIS Scenario 2>1 ≈ 3>4

Compounds in High tank

Com - commercial POCIS Self - self-made POCIS

Same pattern as before

- $\sim \Delta 1\%$ in Self-made between Series 4 and 1,2,3
- $\sim \Delta 1\%$ in COM between Series 4 and 2
- Tank1,2,3 are influenced by High flow tank
- Isotope values : generally between Ridgold and Sigma sources

No isotope effects due to sorption on POCIS as in experiment 1

S-Metolachlor and Tebuconazole - stability in all tanks

Summary 2nd experiment POCIS

- Self-made POCIS have similar sorption (except for scenario 2) as commercial
- Isotope analysis were identical and concentration were identical for all mixed regime regardless HIGH/LOW mixtures scenario
- No isotope effect caused by sorption on POCIS, similar to experiment 1

Field application stormwater wetland Rouffach

Sampling from June 2nd to September 22nd 2020

- Weekly sampling of grab water (10L), inlet/outlet (autosampler)
- Monthly sampling of water from piezometers in gravel filter (GF) (12L)
- Monthly sampling of plants, sediment, POCIS
- Concentration and isotope analysis of all samples to establish mass balance and monitor degradation

Installation of POCIS in the field

Metal pole inserted into sediment

POCIS hanged with a fishline on a metal pole

Protective cage installed

3 POCIS kept in water and replaced every 4 weeks

Preliminary results Rouffach campaign

Almost all compounds detected in water were found in POCIS

Matrix	Dimetom orph	Metalaxyl	S-met.	Cyprodynil	Tetraconazole	Pyrimethanil	Terbuthylazine	Atrazine	Myclobutanil/ Tebuconazole	Pendimethalin
Water	+	+	+	+	+	+	+	+	+/+	+
POCIS	+	+	+	+	+	+	+	+	-/+	-
Plant	+	+	-	-	-	+	-	-	-	-
Sediment	-	-	-	+	-	-	-	- + de	- tected - not	- detected

Preliminary results Rouffach campaign

Period of POCIS application

POCIS concentration

Average concentration in the pond ~100 = ~2000 μ g/L in POCIS extract ng/L

Conclusions

- POCIS : a suitable alternative to grab sampling less cost/time consuming sampling. Up to 20'000 times preconcentration of dimetomorph in the sampling campaign in stormwater catchment.
- POCIS collected the average isotope signature/ concentrations of pollutants and account for high-concentrated short runoff events.
- Sorption of tested pesticides on POCIS : no isotope effects and is linear at environmental concentrations.
- High pre-concentration factor of POCIS allows CSIA of micropollutants =>
 insight into pesticide dissipation processes in the water column/ at watersediment interface.

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