

Proceedings

# Impact of the Sensor Temperature on Low Acetone Concentration Detection Using AlGaN/GaN HEMTs

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**Abstract:** In this work, we report on AlGaN/GaN HEMT sensors for acetone concentration below 100 ppm and in a broad range of the sensor temperature varying from RT to 300°C. At RT, in presence of acetone, a smooth and monotonic decrease of the current is observed with a rather large response of 15 μA/ppm and with large response time (several minutes) and memory effect. At high temperature (300°C), a current decrease is first observed just after the acetone injection and then followed by an increase which saturates and stabilizes at a constant value. In order to clarify this unexpected behaviour, a detailed study of the sensors response versus the temperature and acetone injection flow has been carried out. The outcome of this investigation is that a competition between the current variations induced by both the sensor and gas flow temperature difference from one side and acetone dipolar moment from the other side can explain this transient. Our study highlights that AlGaN/GaN HEMTs-based sensors allow for very sensitive acetone detection at both room and high temperature. Nevertheless, care must be taken during the characterization and operation of such sensors especially at high operating temperature. On the other hand, high temperature operation helps to improve the sensor response suppress the memory effect.

**Keywords:** HEMT; AlGaN/GaN; acetone; temperature; drain current; sensor; sensing; transistor

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## 1. Introduction

With the recent progress in biomedicine and biosensors, in environment and industry control, acetone detection emerges as a huge challenge to be met with small, wearable and cheaper sensors. In fact acetone is one of the important volatile organic compound (VOC) used as a biomarker of several diseases [1] and is widely used in research laboratories and biochemical industries. Modern electronic devices, with their reduced sizes and fast time response can play a major role in this field of sensing.

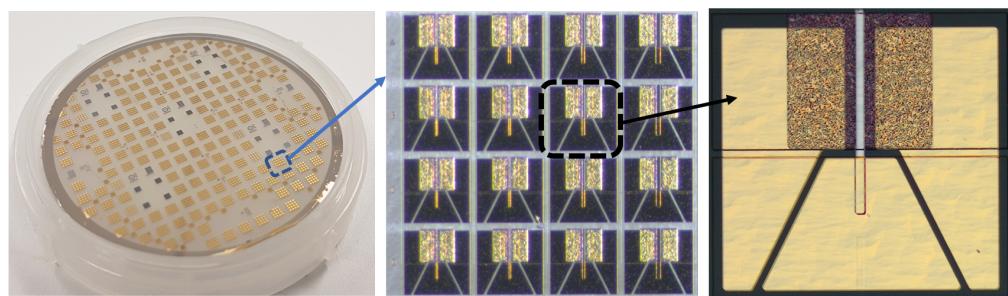
AlGaN/GaN HEMTs have been shown to be efficient sensors for a broad range of physical parameters, in either liquid or dry condition, such as pressure sensor [2,3], gas detection [4,5], pH sensor [6], and more recently used as biosensors for the rapid detection of viruses [7]. These achievements could pave the way for the use of these HEMT transistors in electronic nose development particularly useful for VOCs detection. In the acetone case, sensors with responsivity in the range below 1 ppm are desired [1].

Pioneer works on chemical and gas sensing using AlGaN/GaN HEMTs started in the beginning of 2000s [8–11]. Neuberger et al. reported on polar liquid sensing at room temperature using a gate-less HEMT. They tested both acetone, propanol and methanol in liquid phase, and showed that, whatever the polar liquid, the drain current decreased from its clean air value to a lower value in presence of acetone. The same trend has been reported by Mehandru et al. [11]. They used AlGaN/GaN HEMT to measure acetone and alcohol effects and the sensing mechanism was attributed to the interaction between the dipole moments and the 2D gas. However the magnitude of the changes did not correlate with the dipole moment of the liquids, suggesting that steric hindrance effects or mismatch orientation of the adsorbed molecules should play a role in the detection mechanism [10]. HEMT-based sensing of acetone in gaseous phase has also been tested by Sun et al. [12]. For that, they have used a  $\text{WO}_3$  gated AlGaN/GaN HEMT integrated with a micro-heater allowing the operation of the HEMT at temperature higher than 200 °C. In contrary to the previous cited studies, they observed a large increase of the drain current of the HEMT in presence of acetone.

In this work, in an attempt to elucidate the reason of discrepancies in the sensor response to acetone in gaseous phase, we report on a Pt gated AlGaN/GaN HEMT sensor for acetone concentration ranging from 30 to 150 ppm and for operating temperature varying from room temperature up to 300 °C.

## 2. Device Fabrication and Measurement Setup

The AlGaN/GaN heterostructure was grown by Metal Organic Vapor Phase Epitaxy (MOVPE) on a semi-insulating GaN template on sapphire substrate using trimethylgallium (TMGa), trimethylaluminum (TMAI) and ammonia ( $\text{NH}_3$ ) as the gallium, aluminum and nitrogen sources, respectively. The epilayers consisted of a 260 nm thick undoped GaN layer and a 19 nm thick undoped AlGaN layer with an aluminum (Al) composition of 30%, which was verified by XRD measurements. AFM measurements show a smooth surface with RMS roughness of 0.8 nm. For the sensor fabrication, source and drain contacts were defined by optical lithography. Using an electron beam evaporator, the ohmic contact is achieved by the multilayers 12/200/40/100 nm Ti/Al/Ni/Au deposition followed by a rapid thermal annealing at 870 °C for 30 s under nitrogen atmosphere. A rectangular-shaped gates were defined by optical lithography and Platinum (Pt) contact was evaporated with a thickness of 15 nm followed by lift-off to obtain 16 sensors in a single chip with a size of 3 mm<sup>2</sup> (see Figure 1).



**Figure 1.** (Left): Photograph of a 2 inch wafer processed HEMT acetone sensors. (Middle): single chip of 16 sensors and (Right): SEM image showing a single HEMT sensor.

For gas sensing, the sensors were placed in a closed chamber and electrically connected to a source measurement unit for device polarization and current measurements. The acetone flow was prepared by bubbling a controlled carrier flow of nitrogen in a bubbler containing the solvents in liquid phase and carried out to the chamber.

### 3. Sensing Mechanism

The acetone sensing mechanism using AlGaN/GaN HEMT device has been described by Rabba and Stien [13]. The drain current can be described by:

$$I_{DS} = W \times q \times v_{drift} \times N_s(m), \quad (1)$$

where  $W$  is the gate width,  $v_{drift}$  is the channel position-dependent electron drift velocity,  $N_s(m)$  is the charge density in the transistor channel, and  $m$  is the Al mole fraction in the AlGaN layer. The expression of this sheet density is obtained by solving the 1-D Poisson's equation and given by:

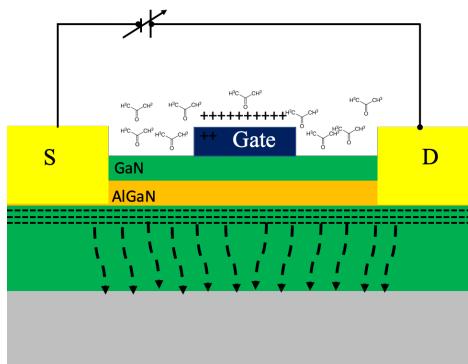
$$N_s(m) = \frac{\epsilon_0 \epsilon_r(m)}{q \times d} \times \left( V_{GS} + \Delta V - v_{th} - V_{DS} - \frac{E_F(m)}{q} \right), \quad (2)$$

where  $\epsilon_r(m) = 9.5 - 0.5m$  is the dielectric constant of the  $Al_mGa_{1-m}N$ ,  $q$  is the electronic charge,  $d$  is the thickness of the AlGaN layer,  $V_{GS}$  is the gate voltage,  $V_{DS}$  is the drain-source voltage,  $E_F(m)$  is the Fermi level energy with respect to the GaN conduction band,  $m$  is the aluminium composition and  $v_{th}$  is the threshold voltage of the HEMT, which is negative for a normally-on HEMT.  $\Delta V$  corresponds to the potential variation induced by acetone molecule dipole calculated using the Helmholtz condenser model and given by:

$$\Delta V = \frac{N_{sp} \times p \times \cos\theta}{\epsilon_0 \times \epsilon_r}, \quad (3)$$

where  $N_{sp}$  is the dipole density per unit area,  $p$  is the dipole moment of the polar molecule in Debye ( $D$ ),  $\theta$  is the angle between the dipole moment vector and the surface normal,  $\epsilon_r$  is the relative permittivity of the gas and  $\epsilon_0$  is the vacuum permittivity.

For normally-on HEMT, the 2D electron gas induces positive charges on the gate which attract and trap the acetone molecular dipoles leading to a depletion of the 2D gas and thus current decrease (see Figure 2). This corresponds to an induced negative  $\Delta V$ , in Equation (2).

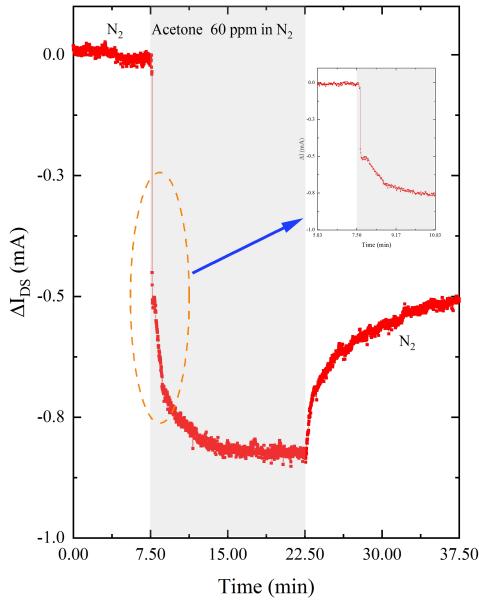


**Figure 2.** Sketch of the operation principle of a HEMT for acetone sensing.

### 4. Results

#### 4.1. Room Temperature Measurements

Figure 3 shows a typical time dependence of the induced drain current variation,  $\Delta I_{DS}$ , when the gas flow is switched from pure  $N_2$  ( $200 \text{ cm}^3$ ) to acetone diluted in  $N_2$  at a concentration of 60 ppm, and then to pure  $N_2$ . A sharp decrease of the current is observed just after the acetone switching. The current then stabilizes for a few seconds (see inset of Figure 3) and starts again to decrease with a smaller slope and finally saturates to reach a constant value. When the gas flow is switched back to pure  $N_2$ , a slow increase of the current followed by a saturation without full recovering is observed. It is to be noticed that a full recovery is obtained after a few hours of operation.



**Figure 3.** Typical time dependence of the induced drain current variation,  $\Delta I_{DS}$ , when the gas flow is switched from pure  $N_2$  ( $200 \text{ cm}^3$ ) to acetone diluted in  $N_2$  at a concentration of 60 ppm, and then to pure  $N_2$ . Measurements were carried out at room temperature and at a bias polarisation  $V_{DS}$  of 5 V.

The sensitivity ( $S = \frac{I_{DS,acetone} - I_{DS,N_2}}{\Delta C} = \frac{\Delta I_{DS}}{\Delta C}$ ), relative variation ( $\frac{\Delta I_{DS}}{I_{DS}} = \left| \frac{I_{DS,acetone} - I_{DS,N_2}}{I_{DS,N_2}} \right| \right)$  and response time (10% to 90% of the total induced current variation) of the sensor have been derived from the data of Figure 3. The sensitivity, relative sensitivity and response time were equal to  $15 \mu\text{A}/\text{ppm}$ ,  $2.4\%$ , and  $131 \text{ s}$ , respectively. Table 1 summarizes the values of this set of characteristics obtained for different acetone concentrations.

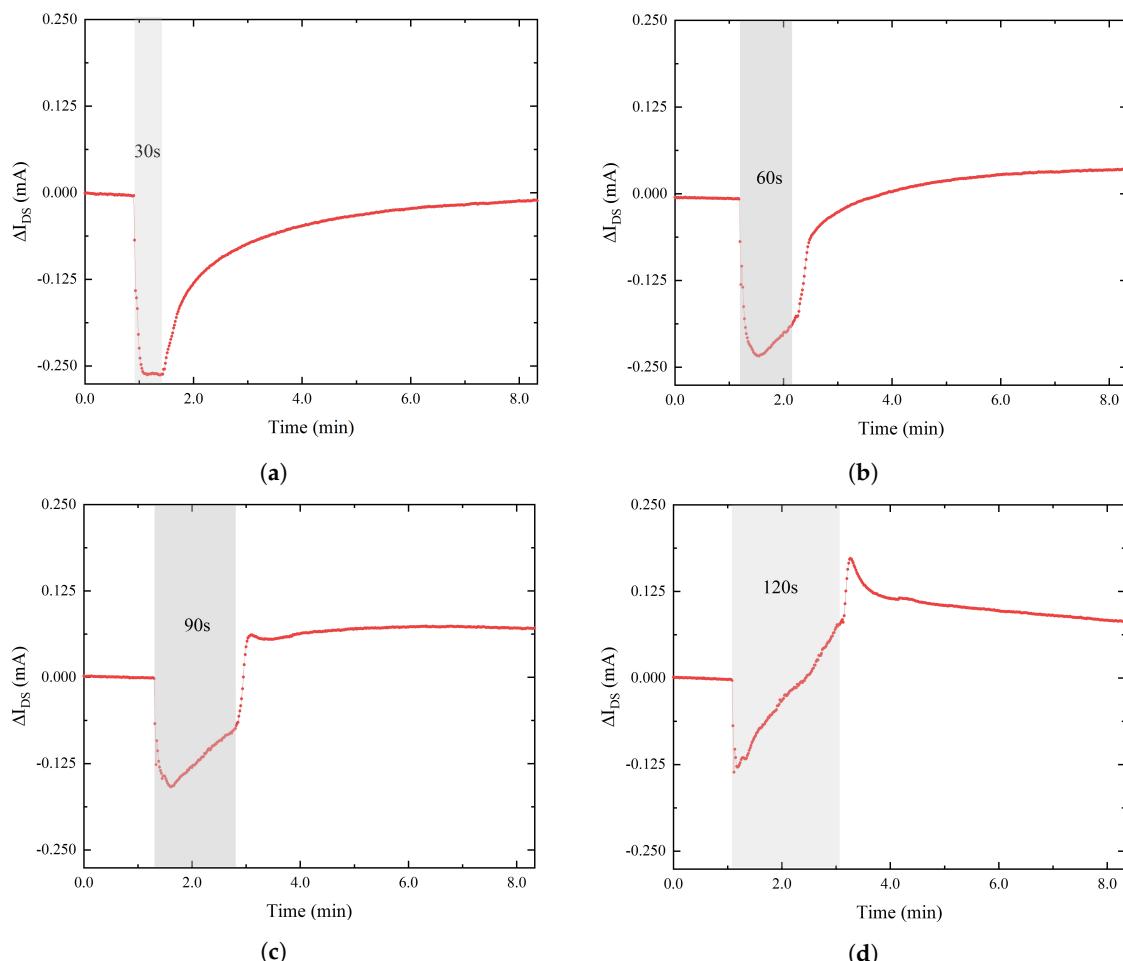
**Table 1.** Values of the sensitivity  $S = \frac{I_{DS,acetone} - I_{DS,N_2}}{\Delta C} = \frac{\Delta I_{DS}}{\Delta C}$ , relative variation  $\frac{\Delta I_{DS}}{I_{DS}} = \left| \frac{I_{DS,acetone} - I_{DS,N_2}}{I_{DS,N_2}} \right|$  and response time of the HEMT sensor at room temperature and for different concentrations.

N <sub>2</sub> Flow (cm <sup>3</sup> /s)	Acetone Concentration (ppm)	$\frac{\Delta I}{\Delta C} (\mu\text{A}/\text{ppm})$	$\frac{\Delta I}{I} (\%)$	Response Time (s)
100	30	33.3	2.7	70
200	60	15	2.4	131
300	90	7	1.7	166
400	120	6	1.9	150

A large decrease of the sensitivity is observed when the acetone concentration is increased from 30 to 120 ppm. Such behaviour cannot be explained by a saturation of the current variation due to a memory effect because each measurement started only after a full recovering of the sensor. A possible reason for this unexpected decrease of the sensor performance is a cooling of the sensor by the gas flow. When the gas flow is introduced into the chamber, the temperature of the sensor decreases and reaches a minimum before to increase when the equilibrium is reached in the chamber. This transient cooling of the sensor induces current transient which might partially compensate for the decrease of current induced by the acetone leading to the behaviour shown in the inset of Figure 3. The transient cooling of the sensor becomes larger when the acetone concentration is increased since the gas flow is increased correspondingly. This assumption is supported by data reported recently by Sun et al. [12] who reported a low sensitivity of  $0.3 \mu\text{A}/\text{ppm}$  for acetone in air at  $300^\circ\text{C}$ , conditions for which the gas flow cooling might play a large role. The increase of the response time with the increase of acetone concentration could also be explained by the transient cooling which affects mainly the saturation region of the sensor response and thus the time at which 90% of the response is reached.

#### 4.2. High Temperature Measurements

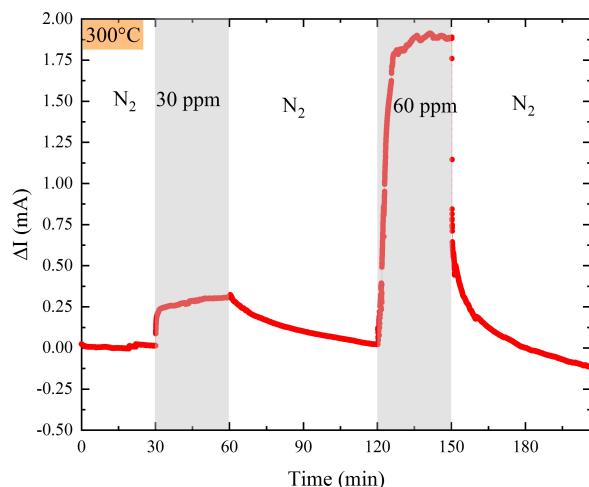
To explore the effect of the cooling transient on the sensor response, we have performed acetone sensing at higher temperature up to 300 °C, for which larger cooling transient are expected, and for different exposure time of the sensor to the acetone flow. Results obtained at 300 °C for an acetone concentration of 60 ppm and time exposure varying from 30 s to 120 s are reported in Figure 4a–d.



**Figure 4.** Typical time dependence of the induced drain current variation,  $\Delta I_{DS}$ , when the gas flow is switched from pure  $N_2$  ( $200 \text{ cm}^3$ ) to acetone diluted in  $N_2$  at a concentration of 60 ppm, and then to pure  $N_2$  for different time exposure varying from 30 s to 120 s. Measurements were carried out at 300 °C and at a bias polarisation  $V_{DS}$  of 5 V.

For the shortest exposure time, a sharp and fast decrease of the current is observed just after the acetone switching. Then, the current variation saturates and reaches a constant value. When the gas flow is switched back to pure  $N_2$ , a slow increase of the current followed by a saturation with a full recovering is observed. The sensitivity, relative variation and response time were equal to  $4.2 \mu\text{A}/\text{ppm}$ , 1.9%, and 6 s, respectively. As expected, the sensor response and the recovery time are smaller at high temperature than at room temperature. Nevertheless, it is to be noticed that the sensitivity is almost four times smaller than the one obtained at room temperature. This is likely due to the transient cooling effect which partially compensates for the acetone induced current decrease. For larger exposure time, the cooling of the sensor by the gas flow increases and the compensation effect increases meanwhile. A larger current increase due to the cooling occurs leading to an effective sensitivity decrease. This cooling induced current increase extends after the switching to pure  $N_2$  gas and then slowly decreases before reaching the initial current value (not shown here). For large exposure time there is thus a strong competition between a current decrease due to the acetone interaction with

the gate and a current increase due to the sensor cooling by the gas flow. As shown in Figure 5, for very long exposure time (30 mn) the sensor cooling effect on the current becomes fully dominant and completely hides the acetone response.



**Figure 5.** Typical time dependence of the induced drain current variation,  $\Delta I_{DS}$ , when the gas flow is switched from pure  $N_2$  ( $100 \text{ cm}^3$ ) to acetone diluted in  $N_2$  at a concentration of 30 ppm for 30 mn, then to pure  $N_2$ , then to acetone diluted in  $N_2$  at a concentration of 60 ppm for 30 mn and finally to pure  $N_2$ . Measurements were carried out at  $300^\circ\text{C}$  and at a bias polarisation  $V_{DS}$  of 5 V.

## 5. Conclusions

At room temperature, in presence of acetone, a smooth and monotonic decrease of the HEMT sensor current is observed with a rather large sensitivity but with slow response time and very long recovering time. The sensor response is in agreement with the electrostatic interaction between the 2D gas of the HEMT with the dipolar moment of the acetone molecules. At high temperature ( $300^\circ\text{C}$ ), a cooling effect due the gas flow which counter balance the acetone response is observed. This cooling effect becomes dominant for long exposure time of the sensor to acetone gas flow and can completely hide the sensor response to acetone. AlGaN/GaN HEMTs-based sensors are shown to allow for very sensitive acetone detection at both room and high temperature. Nevertheless, care must be taken during the characterization and operation of such sensors especially at high operating temperature. Increasing the latter, can help to improve response and recovery time, but requires the control or the cancellation of the current transient due to the cooling of the sensor by the gas flow.

**Author Contributions:** maintext

**Funding:** maintext

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**Conflicts of Interest:** maintext

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