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Smart design nano-hybrid formulations by machine learning

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ABSTRACT: Nano-hybrid formulations combine organic and inorganic materials in self-assembled platforms for drug delivery. Laponite is a synthetic clay, biocompatible, and a guest of compounds. Poloxamines are amphiphilic four-armed compounds and have pH-sensitive and thermosensitive properties. The association of Laponite and Poloxamine can be used to improve attachment to drugs and to increase the solubility of β -Lapachone (β -Lap). β -Lap has antiviral, antiparasitic, antitumor, and anti-inflammatory properties. However, the low water solubility of β -Lap limits its clinical and medical applications. All samples were prepared by mixing Tetronic 1304 and LAP in a range of 1-20% (w/w) and 0-3% (w/w), respectively. The β -Lap solubility was analyzed by UV-vis spectrophotometry, and physical behavior was evaluated across a range of temperatures. The analysis of data consisted of response surface analysis (RMS), and two kinds of machine learning (ML): multilayer perceptron (MLP) and support vector machine (SVM). The ML techniques, generated from a training process based on experimental data, obtained the best correlation coefficient adjustment for drug solubility and adequate physical classifications of the systems. The SVM method presented the best fit results of β -Lap solubilization. In silico tools promoted a fine-tuning, and near experimental data show β -Lap solubility and classification of physical behavior to be an excellent strategy for use in developing new nano-hybrid platforms.

Keywords: Laponita RD; Tetronic T1304, response surface methodology; machine learning; thermoresponsive gels, drug solubilization.



INTRODUCTION

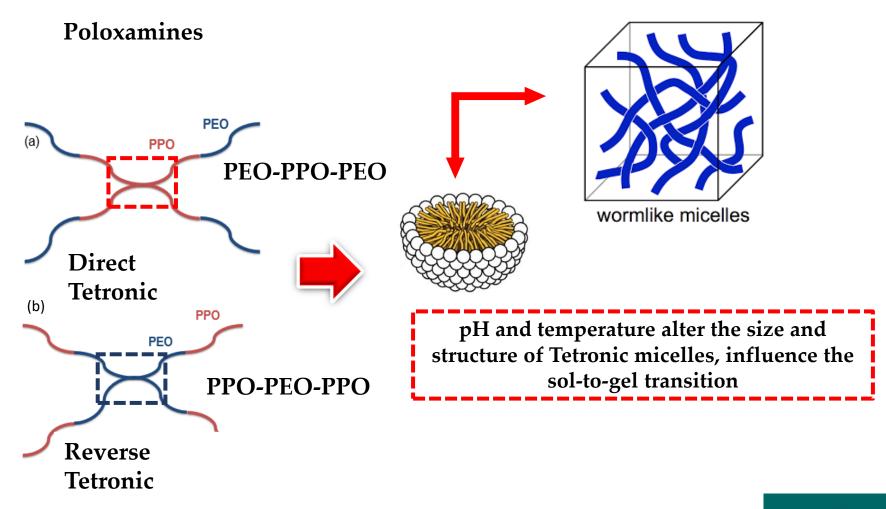


Figure 1: Schematic representation of Tetronic (a) direct and (b) reverse.



INTRODUCTION

- A swelling nano-clay, composed of layered synthetic phyllosilicates with an octahedral MgO sheet sandwiched between two tetrahedral silica sheets;
- a chemical formula given by: $Si_8Mg_{5,4}5Li_{0,4}O_{24}Na_{0,7}$;
- widely used as a mechanical crosslinker to create hydrogels with robust mechanical properties;
- can also be used as a rheological modifier due to its ability to form gels when dispersed in polar solvents such as water.

Laponita® (Lap)

(diameter) Laponite 25nm OH OH OH Nat Nat Nat Nat 0.92nm thickness Multi-step mechanism to gel form "house-of-cards" H_2O H_2O H_2O

Figure 2: Schematic representation of the multistep mechanism to gel forms from Laponite.

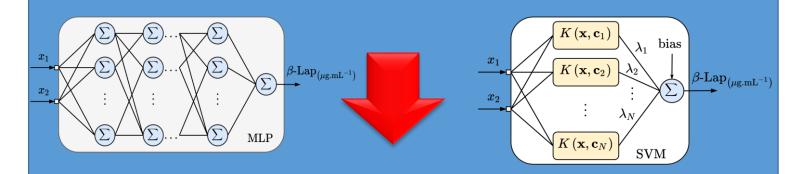
LAP interact with charged, polar, and nonpolar species. This interaction is due to its large charged surface, which enables the adsorption of ions or molecules through ion exchange, van der Waals forces, hydrogen bonding, and cation/water bridges, as well as protonation and ligand exchange at the edges of the crystal.



AIM

Develop multifunctional platforms to be used as nanotechnological systems, consisting of polymeric micelles prepared with the Tetronic® copolymer with or without the presence of LAP to deliver drugs with low water solubility.

Prepare and compared nano-hybrid formulation by physical behavior assays and their ability to increase the solubility of the β -Lap, a low solubility drug model.



Machine learning (MLP and SVM) and response surface analysis (RMS) to model the influence of the compounds used (T1304 and LAP) for the development of the nano-hybrid system.

2020

MATERIALS AND METHODS

Preparation of nano-hybrid

Tetronics (T1304) (1, 5, 10, 15 e 20 %, w/w)



Laponite (0, 1.5 e 3.0 %, w/w)

Nanocomposites were prepared by adding LAP to water under constant stirring and adding Tetronic after 20 minutes and then leaving to stir for 24 hours. All samples were made in triplicates.

Characterization of the nano-hybrids

1. Quantification of βlap solubility

UV-Vis at 257 nm; Dataset analysis by RSM, SVM and MLP.

2. Phase behavior

- Thermal bath in the range of 20-80 °C with 5 °C of step;
- Classification by SVM.

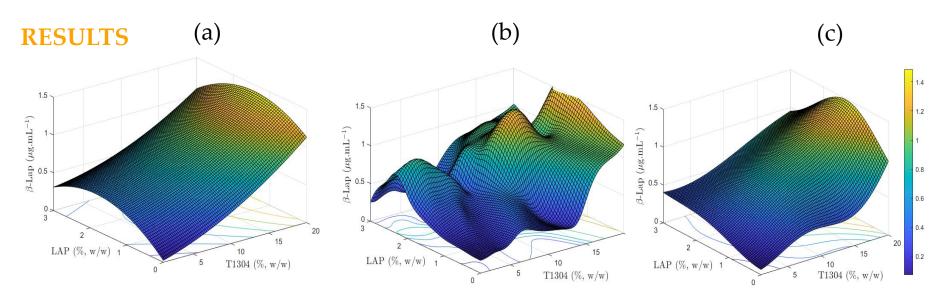


Figure 3: β-Lap solubility response surface: RSM (a), MLP (b), and SVM (c).

Table 1: The coefficients (β_{1-5} ,) obtained from the RSM model.

Parameters	values	Parameters	values
β0	-0.0005	β11	0.0015
β1	0.0262	β22	-0.1345
β2	0.5031	β12	-0.0057

Table 2: Comparison between the techniques used.

Surface	MSE		$ m R^2$	
method	Fitting	Val.	Fitting	Val.
RSM	0.0105	0.0109	0.9279	0.9368
	Training	Val.	Training	Val.
MLP	0.0106	0.0098	0.9332	0.9433
SVM	0.0030	0.0045	0.9814	0.9737



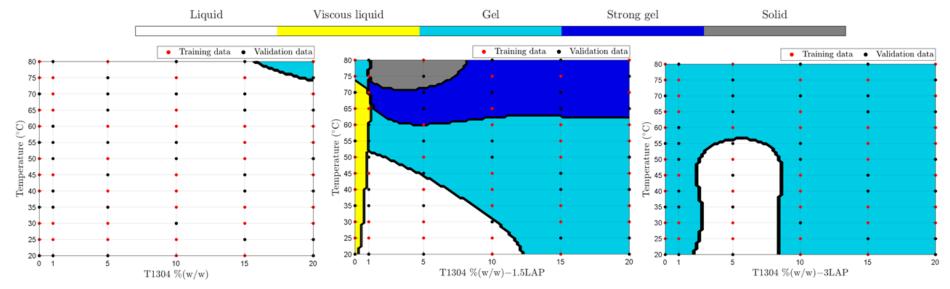


Figure 4: Phase behavior of different concentrations of T1304 and LAP from 20 to 80 °C ramp classified by SVM. The red dots are values used for ML training and the and black are values used for ML validation.

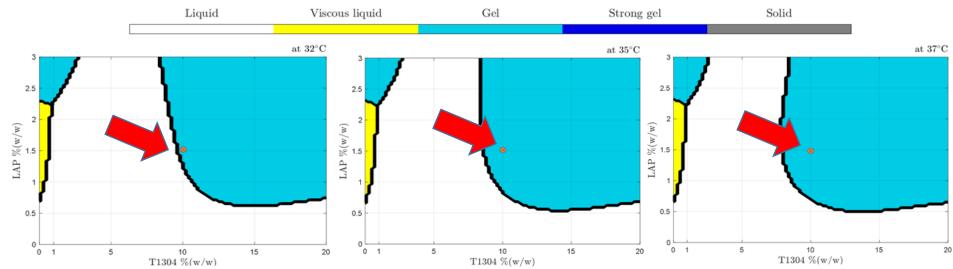


Figure 5: Phase behavior diagrams of different concentrations of T1304 and LAP classified by SVM at 32, 35 and 37°C (body temperatures in different parts such as skin, mouth and mucous membranes).



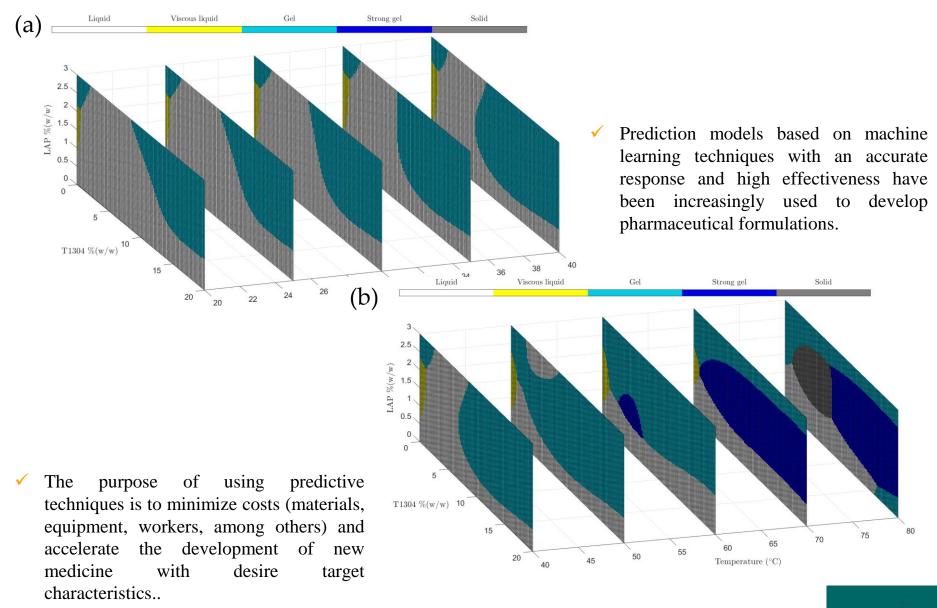


Figure 6: Phase behavior diagrams of different concentrations of T1304 and LAP classified by SVM at ramp from 20 °C to 40 °C (a) and from 40 oC to 80 °C (b).



CONCLUSIONS

- ✓ The association of Tetronic and Laponite in different concentrations allowed the formation of systems that present different phase behaviors as a function of temperature.
- ✓ LAP have a great influence in the liquid-gel transition of the systems. β-Lap solubility have expressive increase in samples T1304 (over 10%) and 1.5% LAP, or systems with only LAP (1.5%).
- ✓ *In silico* tools promoted a fine-tuning and near experimental data shown to be an excellent strategy for use in the development of news nano-hybrid platforms

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