

VALIDATION OF A METHOD FOR THE CONTROL OF **ERGOT ALKALOIDS IN OAT-BASED FUNCTIONAL FOODS**





Laura Carbonell-Rozas^{*}, Francisco J. Lara, Laura Gámiz-Gracia, Ana M. García-Campaña **Department of Analytical Chemistry, Faculty of Sciences, University of Granada** Avda. Fuente Nueva s/n, 18071, Granada <u>*rozas@ugr.es</u>

Research group: "Quality in Food, Environmental and Clinical Analytical Chemistry" (FQM-302)

INTRODUCTION

Ergot alkaloids (EAs) are mycotoxins produced mainly by fungi of the *Claviceps* genus, as *Claviceps purpurea*. The fungus infects the seed heads of living plants, specially cereals, at the time of flowering replacing the developing grain or seed with specialized fungal structures known as sclerotium (or ergot body), which contains alkaloid substances. Although the sclerotia can be mechanically removed during the harvesting process, EAs can be found in cereal-based food, and their ingestion might cause adverse health effects in humans. The European Comission has established a maximum content of 0.5 g/kg of ergot sclerotia in most unprocessed cereals; however, the maximum content for EAs in food is still under study.



EFSA has stated that collection of analytical data are required in order to define the variability of EAs in food and feed commodities, paying special attention to processed foods



European Food Safety Authority





As functional foods containing cereals as ingredients have been scarcely explored so far, in this work, we propose the extraction and quantification of the main 6 EAs; Ergometrine (Em), Ergosine (Es), Ergocornine (Eco), Ergorkriptine (Ekr) and Ergocristine (Ecr), and their corresponding epimers; Ergometrinine (Emn), Ergosinine (Esn), Ergotaminine (Etn), Ergocorninine (Econ), Ergokriptinine (Ekrn), Ergocristinine (Ecrn) in different oat-based products using QuEChERS combined with UHPLC-MS/MS.

		ANALYTIC		HOD				
Chromatographic conditions				QuEChERS procedure				
Column	Zorbax Eclipse Plus RRHD C18 (50×2.1 mm, 1.8 µm)	Grain	1 g of	4 mL aceton	itrile:			3 mL of supernatant
Organic solvent (B)	MeOH + 0.3% formic acid	grinding	powder sample	ammonium carbonate		Vortex 30 s	9000 rpm 5 min	Ŧ
Aqueous solvent (A)	Ultrapure water + 0.3% formic acid			5 mM (85:15),V/V)			
Gradient	0-6 min 30-60% B; 6-9 min 60% B; 9-10 min 60-30% B; 10-12 min 30% B				Supernatan collection			
Column temperature	35 °C			Drying and		9000 rpm 5 min	Vortex 30 s	\$
Flow rate	0.4 mL/min		Filtering	reconstitution	-			
Injection volume	5 µL	UHPLC-MS/MS						
Ν	S conditions							150 mg C18:Z-Sep+ (1:1)
Ionization mode and polarity ESI +								
Scan type	MRM			Some impr	ovements	were achieved		
Target scan tim	e Is	Lowei	r consump	tion organic s	olvents by	reducing the	volume of	the extraction

The monitored ions were the protonated molecules [M+H]⁺ for all of them, except for Esn, Etn, Econ, Ecrn and Ekrn, where the signal at m/zcorresponding to [M–H₂O+H]⁺ was higher than that of the protonated molecules.



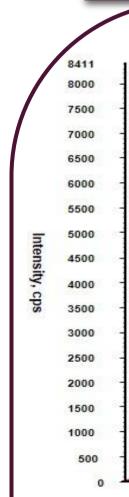


- mixture to 4 mL, which leads to an increase in sensitivity.
- Different dispersive sorbents (C18, PSA and Z-Sep+) were evaluated. The best clean-up of the matrix without affecting the recoveries was achieved using a mixture of C18:Z-Sep+ (1:1).

RESULTS

Method Characterization

- ✓ Procedural calibration curves were established with LODs and LOQs between 0.1-1.0 μg/kg and 0.2-3.20 μg/kg, respectively. Good linearity was obtained (R>0.994).
- \checkmark The precision, evaluated in terms of repeatability and intermediate precision, was lower than 15% RSD in all cases.
- \checkmark Recovery experiments were carried out on oat samples at two concentration levels (5) and 50 µg/kg). The recoveries ranged between 90 and 106%. Matrix effects were lower than 20% in most cases.



Chromatogram of a blank oat flake sample spiked with EAs at 10 µg/kg

Ekr

