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2020The 2nd International Electronic
Conference on Mineral Science
16-30 NOVEMBER 2020 | ONLINE

Spectral response (VNIR-SWIR) associated to isomorphic substitutions in the octahedral sheet of the smectites.

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💥 minerals

MDPI

Abstract:

A mineralogical characterization of a group of bentonites samples was done by X-ray diffraction (XRD), chemical analysis by inductively coupled plasma mass spectroscopy (ICP-MS) and visible-near infrared and shortwave infrared spectroscopy (VNIR-SWIR). As tested by XRD, all samples are very pure, composed mainly by smectites with very small amounts of impurities like quartz and feldspar. The results of the chemical analysis show high contents of Al_2O_3 in all the samples except COU(V), which has high contents of Fe_2O_3 , and R4 which is a trioctahedral smectite. Within VNIR-SWIR spectra, absorption features characteristics of the smectites due to the presence of Fe are observed at lowest wavenumber, while at the SWIR region the absorptions are related to the M-OH bonds, and there are differences among the samples according to their octahedral content.

Keywords:

dioctahedral smectites; trioctahedral smectites; Fe-rich smectite; XRD; VNIR-SWIR spectroscopy.

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Materials and Methods

Materials

Samples	Localization				
APA	Arizona, USA (Clay Repository)				
COU(V)	Washington, USA (Clay Repository)				
PUT	Putifigari, Italy				
R4	Esquivias, Spain				
SAN	Arizona, USA (Clay Repository)				
SUD	Cabo de Gata, Spain				
CGA	Cabo de Gata, Spain				

Methods



X-Ray diffraction



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Chemical analysis

	SiO ₂	Al ₂ O ₃	MnO	MgO	CaO	Na ₂ O	K ₂ O	FeO	Fe ₂ O ₃
APA	51.95	14.18	0,08	5.04	2.48	0.08	0.17	0.00	1.23
COU(V)	49.90	7.41	0.02	1.41	1.61	0.02	0.02	0.50	18.17
PUT	50.52	14.02	0.07	4.11	1.17	0.50	0.18	0.00	5.06
R4	52.91	1.83	0.01	25.81	0.46	0.03	0.32	1.00	0.00
SAN	53.71	14.52	0.09	5.16	2.57	0.16	0.17	0.40	0.98
SUD	53.82	16.60	0.02	4.27	1.26	1.05	0.56	0.40	3.15
CGA	53.03	16.82	0.06	5.10	1.38	0.99	0.34	0.20	2.20



VNIR-SWIR spectroscopy



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Conclusions

- 1. The samples present clear and discriminatory absorption bands related to the octahedral content of the smectites:
 - a) Dioctahedral Al-rich smectites are characterized by the absorption at 2210 nm
 - b) Fe-rich dioctahedral smectites present the most intense absorption at 2290 nm.
 - c) The presence of dioctahedral Al-Fe³⁺-OH bonds causes an adsorption located at 2250 nm.
- 2. Trioctahedral 3Mg-OH bonds produce significant absorption bands located at ~ 2310 nm and ~ 2390 nm.
- 3. These discriminatory absorptions allow to identify the smectite type by means of the field spectroscopy.

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Acknowledgments



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