

Abstract

Electrocatalytical Chemical Sensor for Hydrogen Peroxide †

Arwa Laroussi ^{1,2,*}, Nouredine Raouafi ² and Vladimir Mirsky ¹

1. Nanobiotechnology - Institute of Biotechnology, Brandenburg Technical University, 03046 Cottbus-Senftenberg, Germany
2. Sensors and Biosensors Group, Laboratory of Analytical Chemistry and Electrochemistry (LR99ES15), Department of Chemistry, Faculty of Science, University of Tunis El Manar, Rue Béchir Salem Belkheria, Tunis El-Manar, 2092 Tunis, Tunisia

* Correspondence: larouarw@b-tu.de

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Abstract: The fast and selective determination of hydrogen peroxide (H₂O₂) is of importance not only because of strong interest to this widely applied analyte but also because of the development of enzymatic biosensors for glucose or other metabolites where the sensor for H₂O₂ can be used as the transducer. We report here electrocatalytical amperometric sensor for detection of H₂O₂. The sensor consists of a gold electrode covered by self-assembled monolayer (SAM) with immobilized p-benzoquinone. To provide highly stable immobilization of p-benzoquinone at the distance of effective electron tunneling, a new anchor compound - 1,3-dimercaptopropan-2-ol - was synthesized and used for the preparation of the SAM. Due to two thiol groups binding gold surface this compound provides a high stability of the SAM. The surface concentration of p-benzoquinone obtained from cyclic voltammetry is $2.5 \pm 0.2 \times 10^{-10}$ mol·cm⁻². Cyclic voltammetry and chronoamperometry experiments proved that the immobilized benzoquinone exhibited high electrocatalytic activity towards the decomposition of H₂O₂. Depending on the used potential range, different sensing modes can be realized. For example, one can measure electrochemical response due to the oxidation of H₂O₂ at anodic potentials or due to the reduction of oxygen formed during oxidative decomposition of H₂O₂. Also amperometric response at fixed potential of +0.4 V vs. Ag/AgCl corresponding to the oxidation of benzoquinone to hydroquinone was studied. The sensor exhibited a linear response over a concentration range of 0.1–2 mM with a low detection limit of 4.24 μM. The reproducibility of three different electrodes prepared was examined at the H₂O₂ concentration range from 0.1 till 3 mM, which resulted in a relative standard deviation below 4.2%.

Keywords: self-assembled monolayer; p-benzoquinone; electron transfer; cyclic voltammetry; chronoamperometry; hydrogen peroxide; electrocatalysis; chemical sensor