

Proceedings



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Group 14 Metallafluorenes for Lipid Structure Detection and Cellular Imaging ⁺

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Abstract: Fluorescent compounds have been shown to be useful in probing lipid dynamics, and 10 there is ongoing interest in nontoxic, photostable, and sensitive dyes. Recently we evaluated a 11 number of 2,7-disubstituted-alkynyl(aryl)-3,6-dimethoxy-9,9-diphenyl-sila- and germafluorenes for 12 their potential as cellular fluorescent probes. These compounds exhibit remarkable quantum yields 13 in hydrophobic environments and dramatic increases in emission intensity in the presence of sur-14 factants. Here we show that they exhibit significant emission enhancements in the presence of small 15 unilamellar vesicles and are nontoxic to E. coli, S. aureus, and S. cerevisiae. Further, they luminesce 16 in S. cerevisiae cells with strong photostability and colocalize with the lipid droplet stain Nile Red, 17 demonstrating their promise as lipid probes. 18

Keywords: fluorescence; lipid; metallafluorene

1. Introduction

The introductory understanding of the role of biological membranes is that they are barriers and are used to regulate transport and for energy processes. Only fairly recently has there been a developing understanding that membranes are dynamic in their lipid composition and properties, and that these local differences participate in cellular processes in a profound way and have been linked to disease states [1, 2], including cellular stress [3].

Fluorescence spectroscopy is an accessible and powerful tool for characterizing membrane composition and behavior [4]. This technique relies on the development of suitable probes. Commonly used probes such as Nile Red, dansyl and NBD [5] vary with respect to relevant properties such as excitation and extinction wavelength, extinction coefficient, working concentration (sensitivity), photostability, and quantum yield, all of which can impact their utility. When this is coupled with the rapidly expanding research area, it is no surprise that a call for more probes to meet expanding needs is prominently articulated [4].

Recently we examined the potential of a small library of sila- and germafluorenes36(metallafluorenes, or MFs) containing alkynyl(aryl) substituents at the 2,7- position37([6,7]; Fig. 1) for their potential as fluorescent probes of surfactants. These compounds38are soluble and luminescent in aqueous solution and exhibit high quantum yields and39dramatic emission enhancements in the presence of various surfactants (5-25 fold) [8].40These results suggested that MFs could have potential as probes of lipids both in vitro41and in vivo. Here we examine the sensitivity, toxicity, and photostability of MFs toward42

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Copyright: © 2021 by the authors. Submitted for possible open access publication under the terms and conditions of the Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses /by/4.0/). lipids in both settings and demonstrate the potential of these compounds as lipid probes. Indeed, they are sensitive to DOPC small unilamellar vesicles (SUVs) with significant fluorescence enhancements. These dyes show no toxicity to Gram-positive bacteria, Gram-negative bacteria, and yeast cells and demonstrate high photostability. When compared to the commercially available lipid droplet dye Nile Red, these MFs show strong colocalization with more punctate staining, demonstrating their potential as lipid probes.



Figure 1. Structures of 2,7-disubstituted sila- and germafluorenes used in this study.

2. Materials and Methods

2.1. Materials

Phospholipids were purchased from Avanti Polar Lipids (Alabaster, AL). 200 proof ethanol was purchased from Decon Labs (King of Prussia, PA). All chemicals used were of reagent grade and were used as received without further purification. DMSO and Nile Red were obtained from Millipore Sigma. Xylene was obtained from ThermoFisher. **1-4** were synthesized as previously described using an appropriate alkynyl(aryl) precursor in a palladium-catalyzed Sonagashira cross-coupling reaction [6,7] and dispensed from stocks in DMSO as previously described [8].

2.2. Preparation of Small Unilamellar Vesicles (SUVs)

At 25 °C, a stock concentration of 4.2 mM DOPC was prepared by drying under inert 20 gas and then resuspended in 10 mM Tris buffer. After 30 minutes, DOPC was sonicated 21 for 27 minutes at 25 °C until cloudy. The DOPC-SUV solution was then passed through 22 an Avanti Mini Extruder eleven times to make uniformly-sized 0.1 µm DOPC-SUVs at 25 23 °C. DOPC-SUVs were then diluted to 0.1 mM in a quartz cuvette for fluorescence meas-24 urements [9,10].

2.3. Spectroscopy

Absorbance spectra were recorded on a Shimadzu 1800 with slits (bandpass) set to 1 27 nm. Emission spectra were collected in an acid-washed quartz cuvette on a Fluorolog-3 28 (SPEX) spectrofluorimeter. The temperature was maintained at 25 °C with a thermostat-29 ted cell holder equipped with a magnetic stirrer. Emission spectra were collected with 30 the indicated excitation wavelength and slits (bandpass). MF photostability in xylene was 31 observed at the indicated emission maximum. 32

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2.4. Microbial Toxicity

Culture tubes containing LB media or YPD media were inoculated with Escherichia 2 coli (Gram-negative), Staphylococcus aureus (Gram-positive), or Saccharomyces cerevisiae, re-3 spectively. 1-4 were added such that the final DMSO concentration was 2-10% and the MF 4 at its solubility limit in the media. The tubes were incubated at either 37°C (bacteria) or 5 30°C (yeast) overnight and visually inspected for growth. 6

2.5. Confocal Laser Scanning Microscopy

Samples were prepared by smearing a small amount of cells onto a glass microscope 8 slide and heat-fixed by passing the slide through flame no more than 5 times. Then 1-4 or 9 NR was applied to heat-fixed cells at 15 μ M and incubated at room temperature for 15 10 min for MFs and 10 min for Nile Red. Slides were then rinsed with 2-3 mL of deionized 11 water, topped with coverslips and sealed with clear nail polish. Cells were imaged with a 12 Zeiss LSM 900 confocal microscope with an excitation wavelength of 405 nm. For photo-13 stability, the sample was illuminated with 1% laser power and images collected periodi-14 cally. 15

3. Results and Discussion

3.1. Spectroscopic Studies

To assess their sensitivity to a biologically relevant membrane, the emission spectra of 1-4 were compared in the absence and presence of DOPC-SUVs. As shown in Fig. 2, fold-enhancements range from 2-7 fold, with 1 and 2 showing the most dramatic changes.



Figure 2. Emission spectra of 1-4 in the absence (dashed) and presence (solid) of 0.1 mM DOPC-SUVs. Conditions: 1 µM MF, 0.1 mM DOPC, 10 mM Tris pH 8, 25 °C. The excitation wavelength was 387 nm and the slits (bandpass) set to 1.0 nm. 3 min incubation.

The photostability of these MFs was initially probed by observing the emission signal as a function of time in xylene, which is used to mimic the interior of membranes [11]. As summarized in Figure. 3, these signals are remarkably stable over two hours of 28

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3.2. Microbial Toxicity Studies

To assess their potential for use in cellular imaging, MFs were screened for toxicity against microorganisms. At the solubility limit for these compounds in media (at least 50 μ M), there was no inhibition of yeast growth. For E. coli and S. aureus no inhibition of growth was observed at the MF solubility limit in media.

3.3. Imaging of S. cerevisiae with Metallafluorenes

To determine if these MFs can be used to stain cells, **1-4** were introduced to yeast cells 13 and subsequently imaged using confocal microscopy. **Fig. 4** illustrates that in all cases, 14 MF emission intensity is visible inside fixed yeast cells. 15



10 µm



Figure 4. Confocal Imaging of MF in Yeast Cells. Conditions: 15 µM MF as indicated, 63x. The excitation wavelength was 405 nm and the scan range 400-600 nm.

To assess MF photostability in yeast cells, excitation was applied and fluorescence19observed as a function of time. As summarized in Figure. 5 for 1 and 2, fluorescence20persisted for over 2 min, with 2 showing greater photostability. See Supplemental Fig. S121for photostability studies of 3 and 4.22

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Figure. 5. Photostability of 1 and 2 in Yeast Cells. See Methods for details. S. cerevisiae were stained for 15 minutes with 15 μ M **1** or **2** and then imaged periodically during continuous excitation. Magnification is 63x.

Finally, to determine where these MF localize in yeast, we costained with Nile Red, a well known lipid droplet stain [12]. As shown in Figure. 6, 1 colocalizes with this probe and demonstrates clear specificity for *S. cereviseiae* organelles, including the vacuole and possibly lipid granules. See Supplemental Fig. S2 for a colocalization study of 2 and 4.



Figure. 6. Compound 1 Colocalizes in Yeast with Nile Red. Red, Nile Red; green, **1**; top right, transmitted light; bottom right, yellow indicates colocalization. $15 \,\mu$ M probe, 63x magnification.

4. Conclusions

We show here that these metallafluorenes have good photostability and are sensitive 13 to lipid structures in vitro, demonstrating impressive fold enhancements in the presence 14 of SUVs. Further, they are non toxic to cells and can enter cells and colocalize with Nile 15 Red but yield more punctate images. In addition, the higher extinction coefficients of MFs 16 and competitive quantum yields [8] make them more sensitive. All of these observations 17 bode well for the application of MFs as lipid probes both in vitro and in vivo. The synthetic 18 scaffolding of these MFs provides convenient tuning of desired properties by changing 19 the 2,7 substituent. This feature facilitates designs that incorporate optimal solubility, 20 emission spectra, dipole moment, and solvatochromism for specific applications. 21

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		5. Patents	1
		WO/2020/210416; PCT International Patent Application No.: PCT/US2020/027355	2
		Supplementary Materials: The following are available online at www.mdpi.com/xxx/s1, Figure S1: Photostability of 3 and 4 in Yeast Cells. Figure S2: 2 and 4 Colocalize in Yeast with Nile Red.	3 4
		Author Contributions: Conceptualization, C.D., H.S. and S.JN; Investigation, H.S. and S.JN.; Resources, C.D., J.BW, S.G. and W.O.; Data Curation, S.JN.; Writing – Original Draft Preparation, C.D. and H.S.; Writing – Review & Editing, C.D. and S.JN.; Supervision, C.D.; Project Administration, C.D.; Funding Acquisition, J.BW. and C.D. H.S. and S.JN. contributed equally to the preparation of this manuscript. All authors have read and agreed to the published version of the manuscript.	5 6 7 8 9 10
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		Data Availability Statement: The data presented in this study are available on request from the corresponding author.	12 13
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