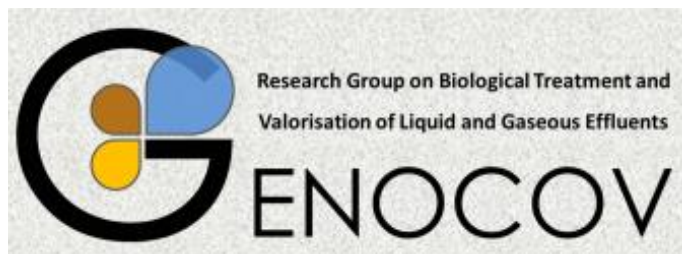


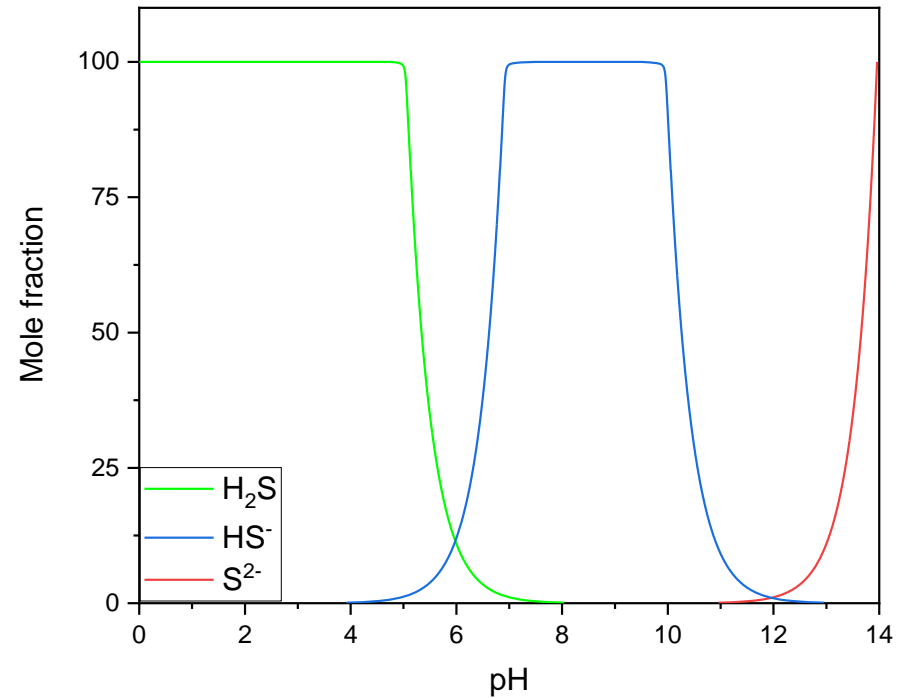
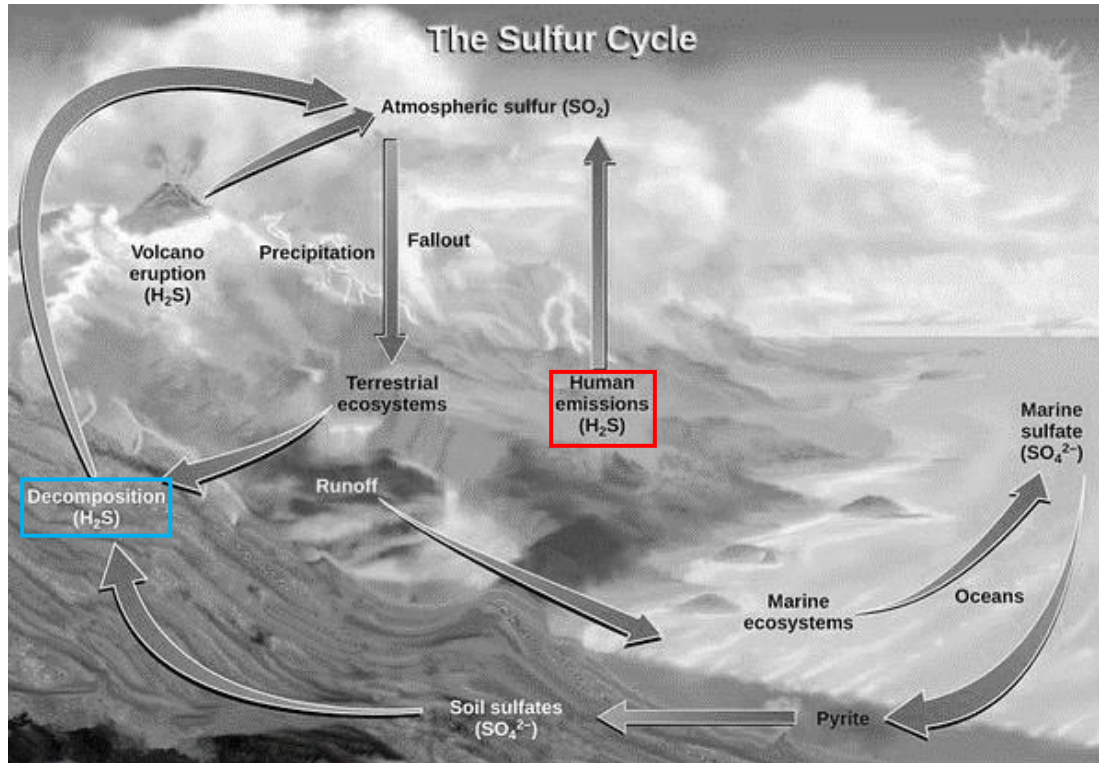
An Inkjet-printed amperometric H₂S sensor for environmental applications

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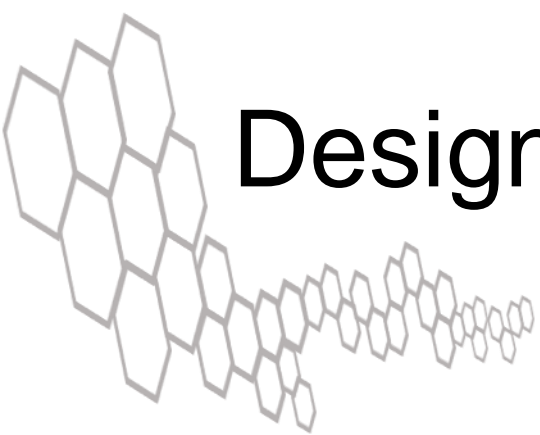


Sulfur cycle



Sulfhydic acid is a polyprotic compound with labile protons. Given that, it coexists as different species with different proportions depending on the medium's pH.

Design of the electrode



SWCNTs/SWCNTs-PVA



SU8



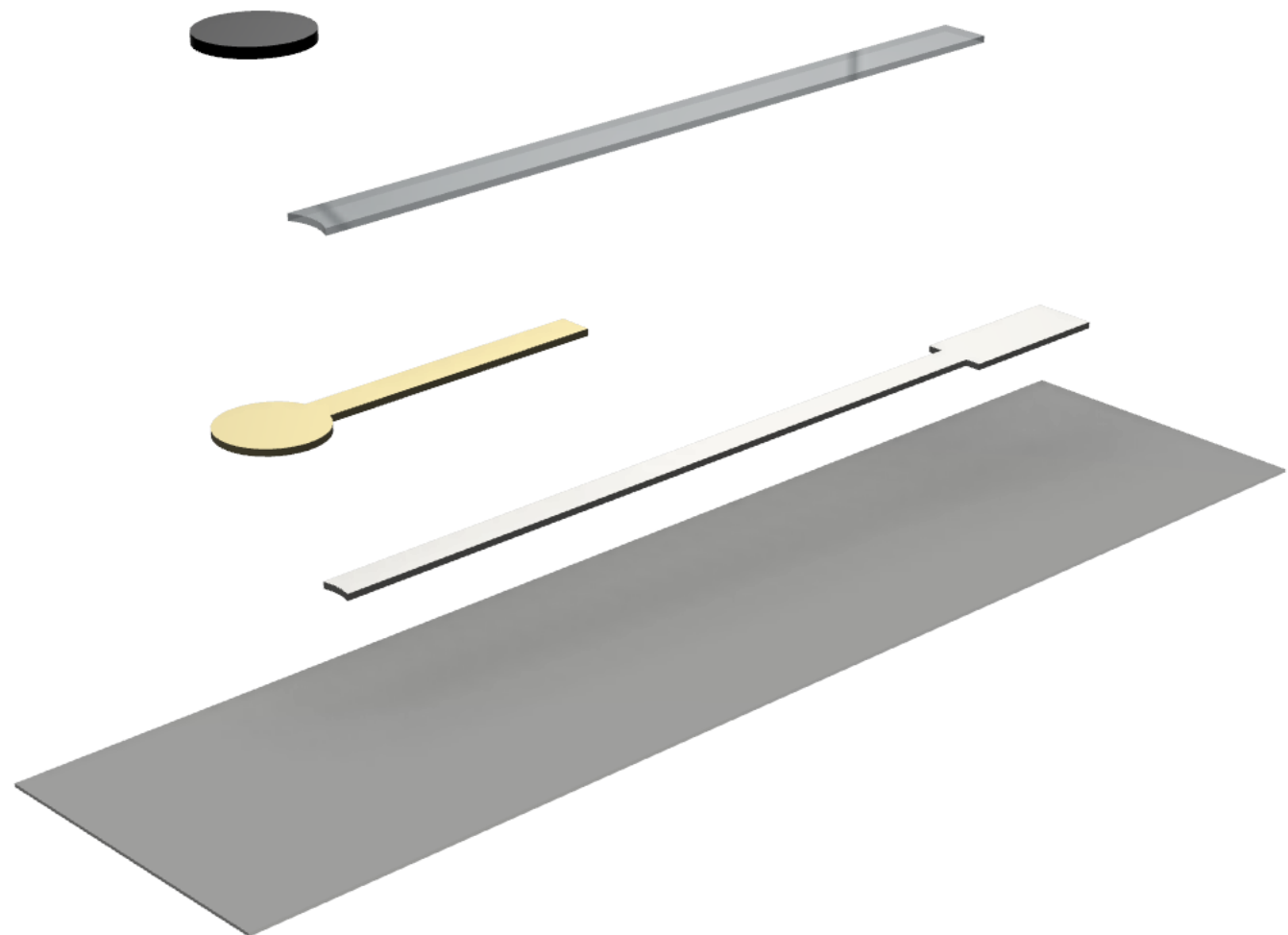
Au electrode



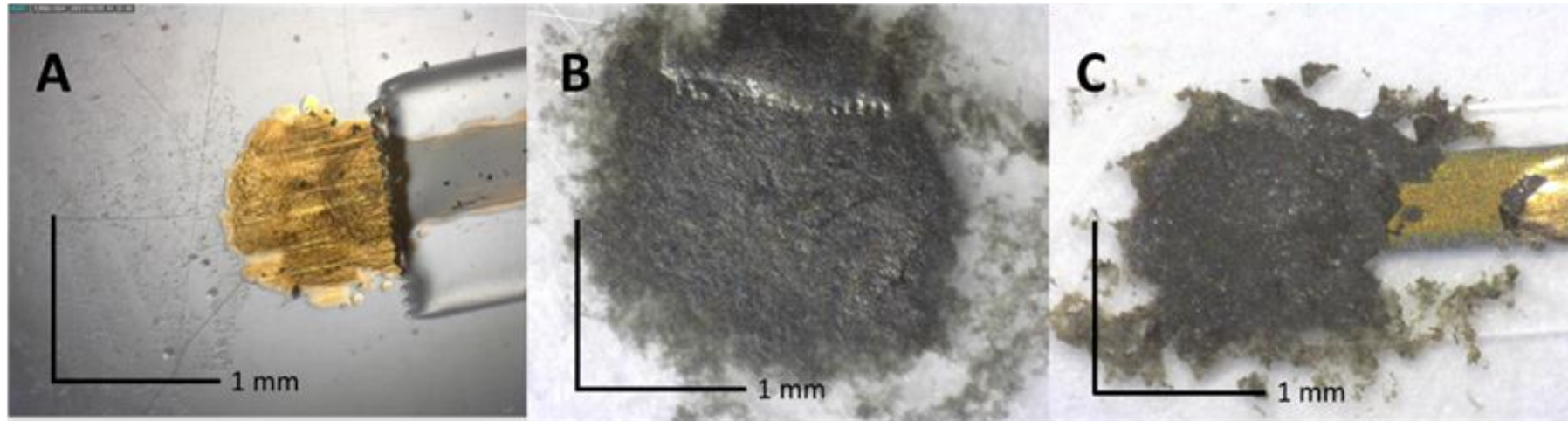
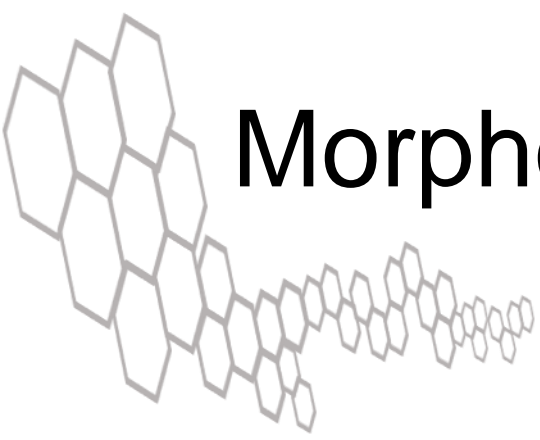
Ag tracks



PET substrate

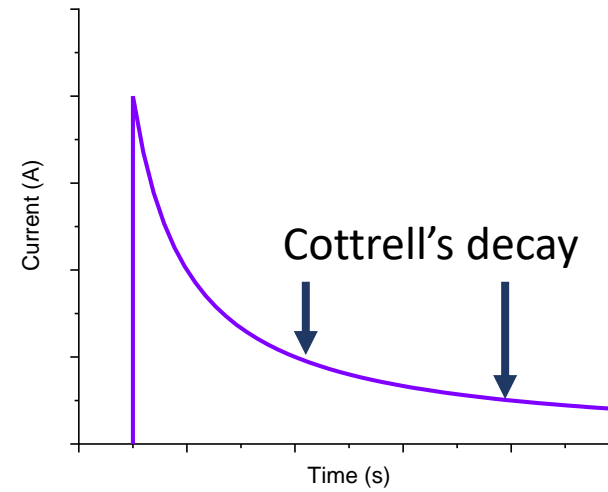
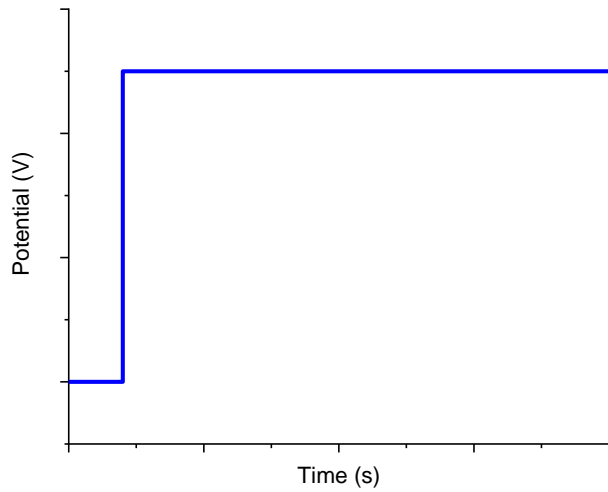
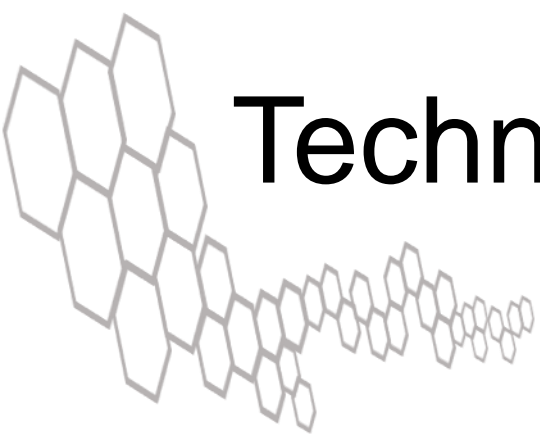


Morphological characterization



- A) Inkjet-printed Au electrode
- B) SWCNTs drop-casted over Au electrode
- C) Drop-casted SWCNTs-PVA over Au electrode.

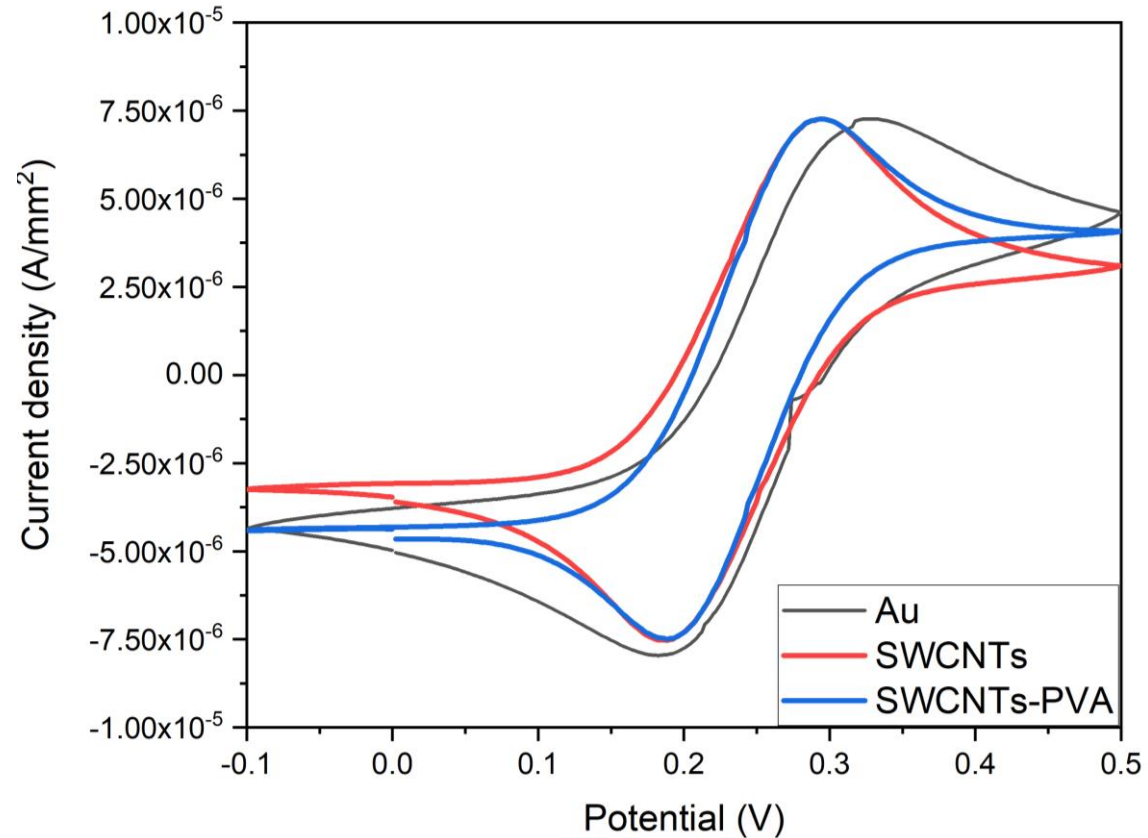
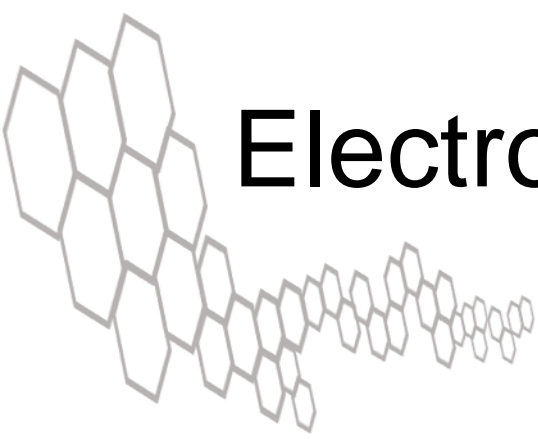
Technique basics



$$i = \frac{nFAc_j^0 \sqrt{D_j}}{\sqrt{\pi t}}$$

Measured at the same time, current has a linear correlation to analyte concentration.

Electrochemical characterization

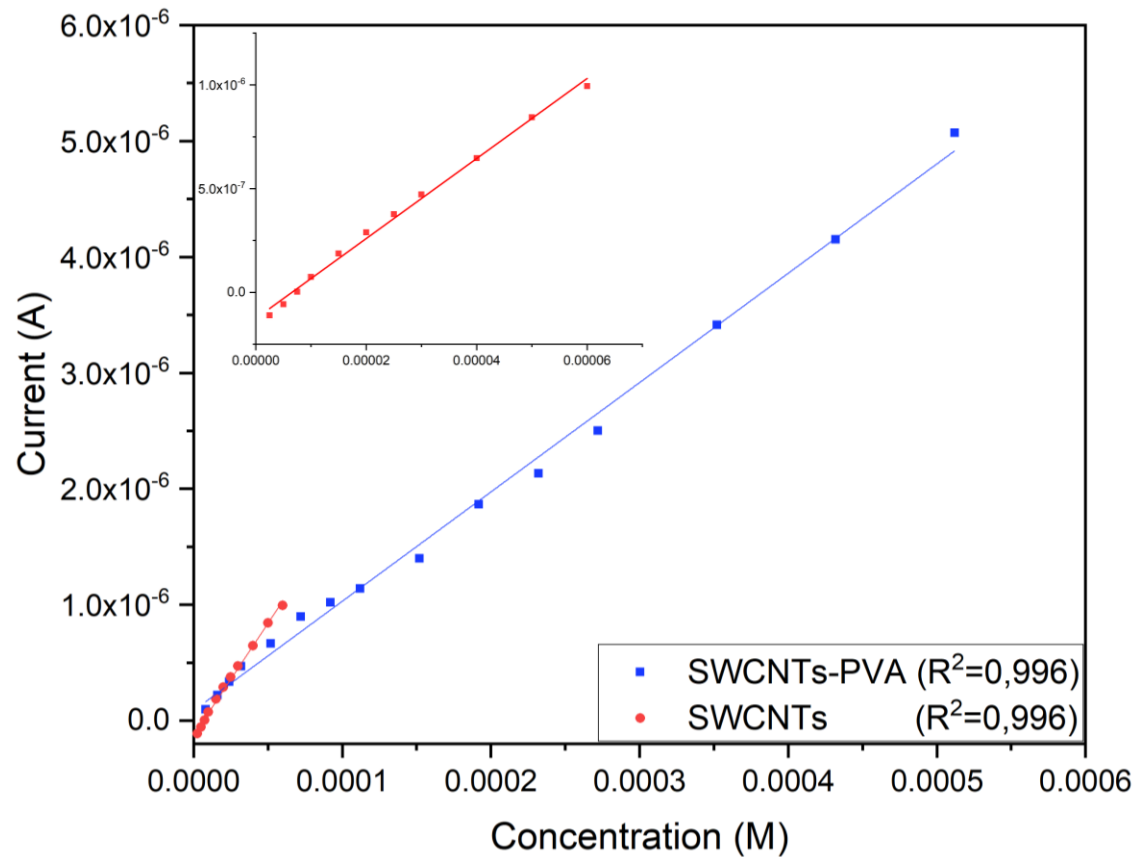
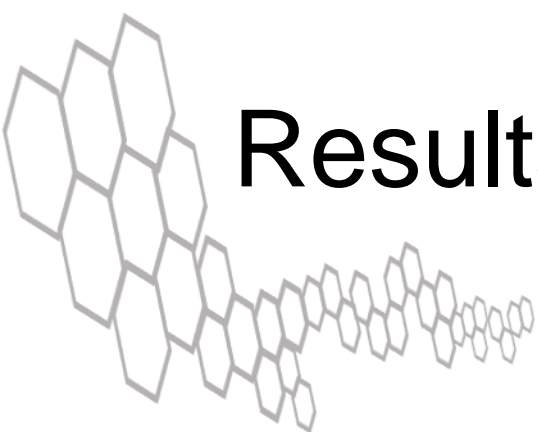


Modified electrodes presented similar current density and a smaller potential gap than Au bare electrodes

SWCNTs and SWCNTs-PVA are both favorable for an H₂S sensor due to their capacity for lower oxidation potentials

Lowered rates of sulfur deposition on their surfaces due to the less favorable interaction S-C against S-Au

Results: H₂S sensor



SWCNTs: (19.3 ± 0.4) mA/M
SWCNTs-PVA: (9.4 ± 0.2) mA/M

SWCNTs can measure H₂S concentrations from 8 μ M to 60 μ M, with a LD of 4.3 μ M

SWCNTs-PVA is capable of measuring H₂S from 52 μ M to 512 μ M, with a LD of 34 μ M

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Thank you for your attention

