

# APPRAISING BLUEBERRY RESIDUES AS A NATURAL SOURCE OF BIOACTIVE COMPOUNDS

i-Grape Laboratory

María Celeiro a\*, Aly Castillo a, Alicja Rosalowska a, Carmen Garcia-Jares a, Marta Lores b



<sup>a</sup> CRETUS, Department of Analytical Chemistry, Nutrition and Food Science, Universidade de Santiago de Compostela, E-15782, Santiago de Compostela, Spain.

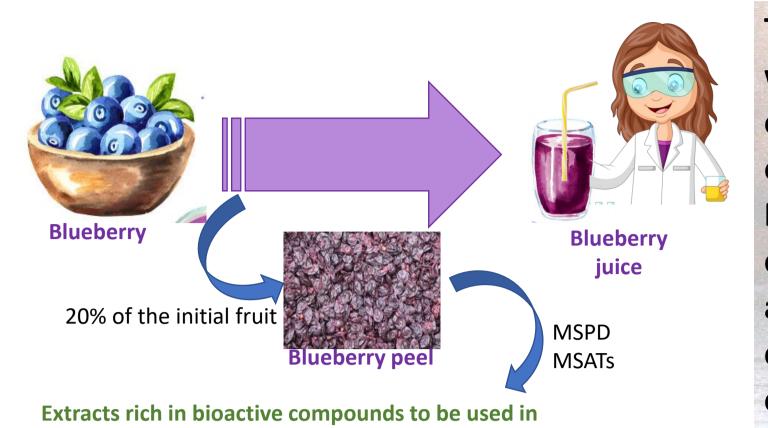
<sup>b</sup> LIDSA, Department of Analytical Chemistry, Nutrition and Food Science, Universidade de Santiago de Compostela, E-15782, Santiago de Compostela, Spain.

\*maria.celeiro.montero@usc.es



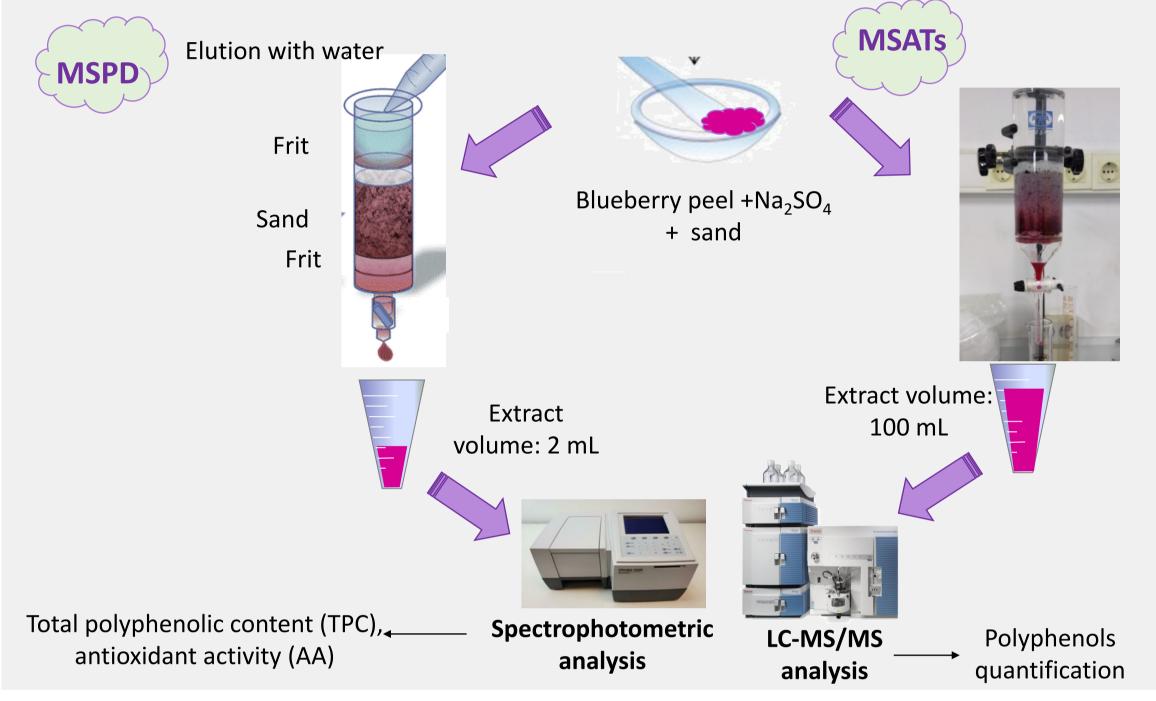
#### INTRODUCTION

other applications

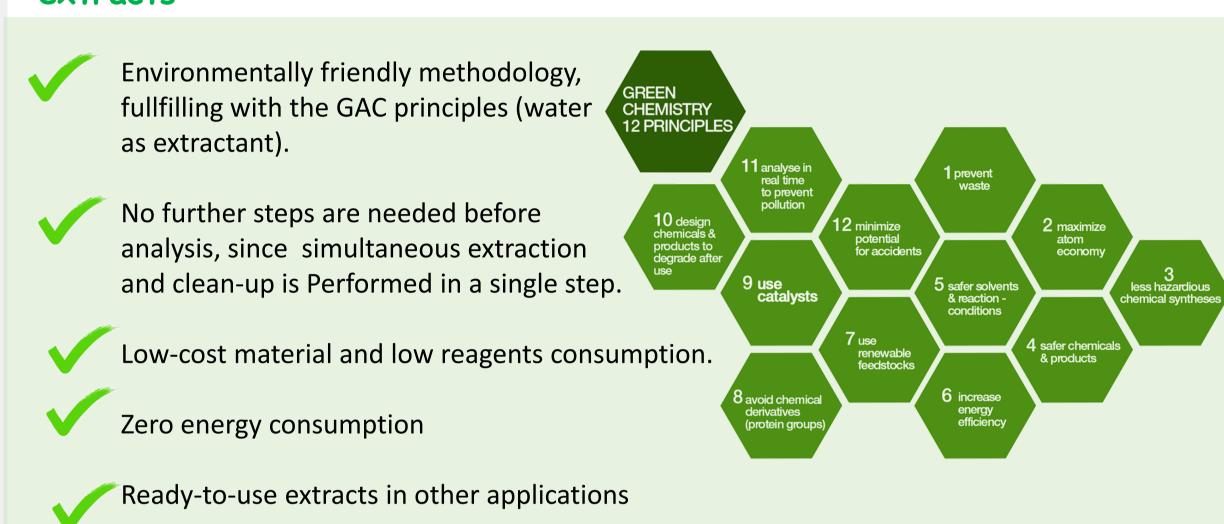


The main solid by-product of the blueberry juice processing is peel, representing up to 20% of the initial fruit weight. Its excessive accumulation causes a seasonal management and environmental problem. The main objective of this work is to perform a deep characterization of the blueberry peel residues, and their derived extracts. To obtain extracts, environmentally friendly procedures, matrix solid-phase dispersion (MSPD) and Medium-scale Ambient Temperature Systems (MSATs), have been employed. Physical, mechanical, and chemical characteristics of the raw material (blueberry), as well as the total polyphenolic content (TPC) and antioxidant capacity of the corresponding by-products and derived extracts were assessed. Liquid chromatography-tandem mass spectrometry (LC-MS/MS) was employed to quantify individual phenolic compounds.

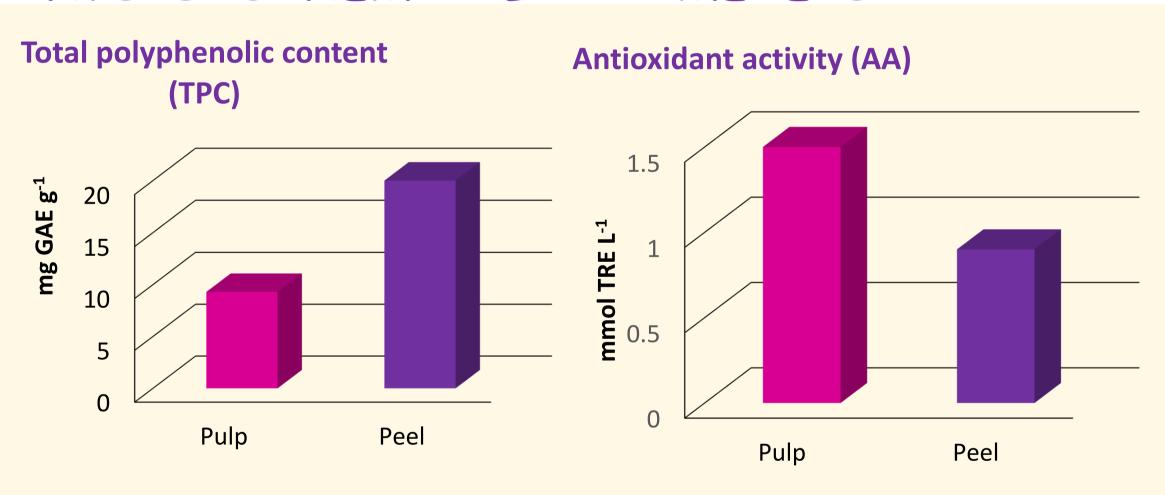
#### OBTENTION OF READY-TO-USE EXTRACTS



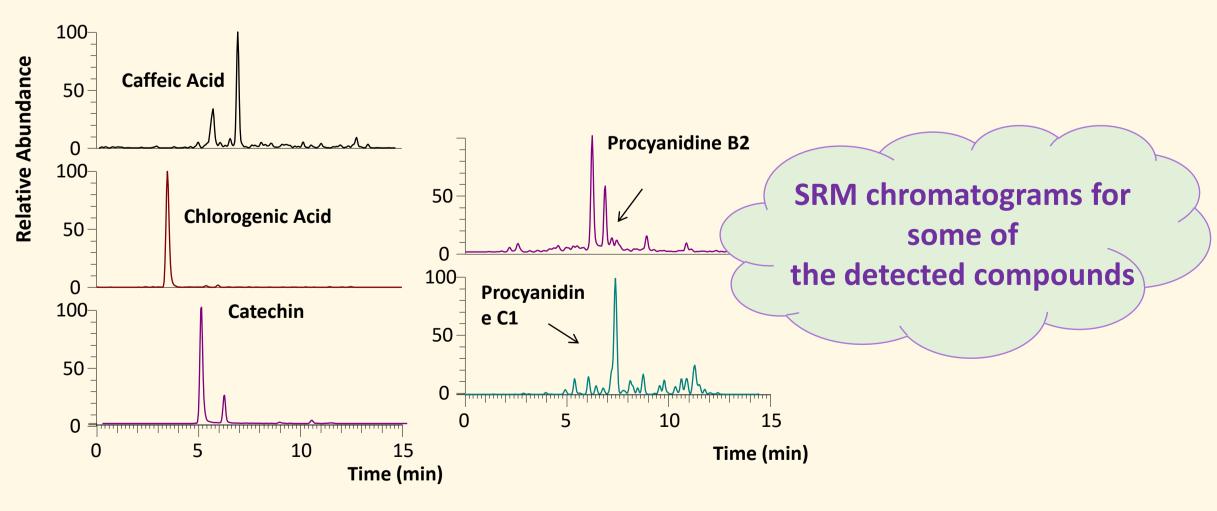
## Advantages of using MSPD and MSATs to obtain blueberry peel extracts



#### PHYSISCOCHEMICAL PARAMETERS



## TPC content was clearly higher in the peel extracts, whereas results were similar for AA in both extracts



### CHARACTERIZATION OF BIOACTIVE COMPOUNDS

Polyphenols Group	Compound name	Liquid extract*	Solid extract*
			(freeze-dryed)
		Concentration	Concentration
		(μg/g dw)	(µg/g dw liophilized)
Hydroxycinnamic acids	Caffeic Acid	$1.4 \pm 0.3$	$0.18 \pm 0.01$
	Clorogenic Acid	1225 ± 331	107 ± 15
	∑ Non-Flavanoids	1226	107
Flavan-3-ols	Catechin	37 ± 10	$1.2 \pm 0.6$
	Epicatechin	$4.8 \pm 0.2$	0.6 ± 0.2
	∑ Flavan-3-ols	42	2
Flavan-3-ols oligomeric	Procyanidine A1	$107 \pm 14$	$3.6 \pm 0.1$
_	Procyanidine B2	77 ± 29	3 ± 1
derivatives	Procyanidine C1	551 ± 251	21 ± 10
	∑ Procyanidines	735	60
Flavonols	Quercetin	$144 \pm 40$	19 ± 3
	Isoquercetin	517 ± 165	44 ± 7
	Rutin	65 ± 35	5 ± 1
	Myricetin	23 ± 7	3.3 ± 0.4
	∑ Flavonols	749	71
Anthocyanidins	Delphinidin	1327 ± 278	25 ± 8
	Cyanidin	$780 \pm 74$	27 ± 5
	Petunidine	1526 ± 411	37 ± 9
	Peonidin	85 ± 29	6 ± 1
	Malvidin	2353 ± 1125	204 ± 84
Anthocyanins	Petunidin-3-O-glucoside	318 ± 141	24 ± 11
	Cyanidin-3-glucoside	510 ± 193	48 ± 20
	∑ Anthocyani(di)ns	6899	371
	∑ Flavanoids	1526	133
Σ	<b>BIOACTIVE POLYPHENOLS</b>	9651	611
		* Res	ults showed for MSATs extraction

#### CONCLUSIONS

- Anthocyanins were the most abundant polyphenolic group of compounds in the blueberry peel extracts reaching concentrations up to 6899 μg g<sup>-1</sup>dw.
- Other compounds with interesting properties such as quercetin and its derivatives (isoquercetin, rutin), caffeic and chlorogenic acids were also found at high concentration levels (1226 μg g<sup>-1</sup>dw) in the obtained blueberry peel extracts.
- The ultimate goal of this work is investigating the utility of food processing wastes as raw materials to obtain extracts with added value to be used in new products with beneficial health properties.

#### **Acknowledgements**

