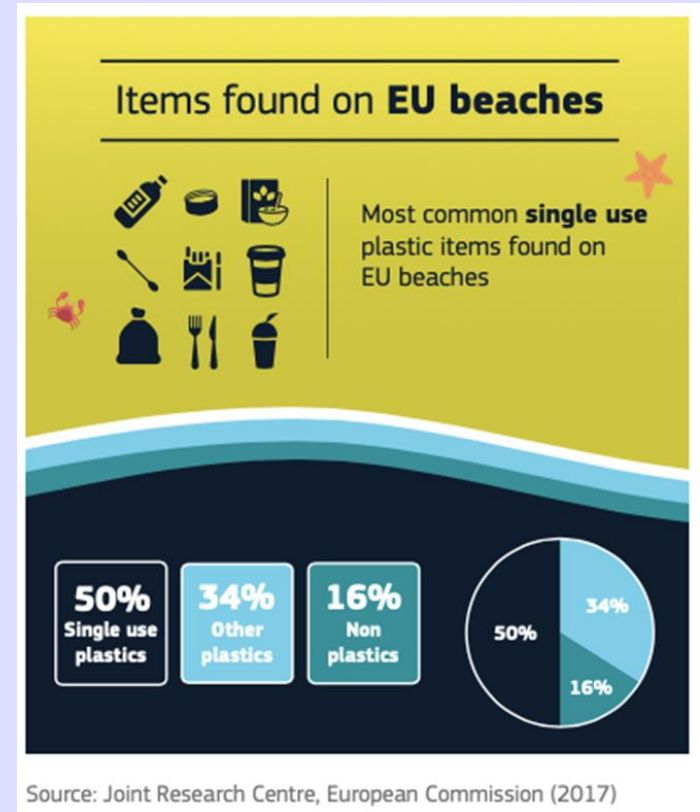
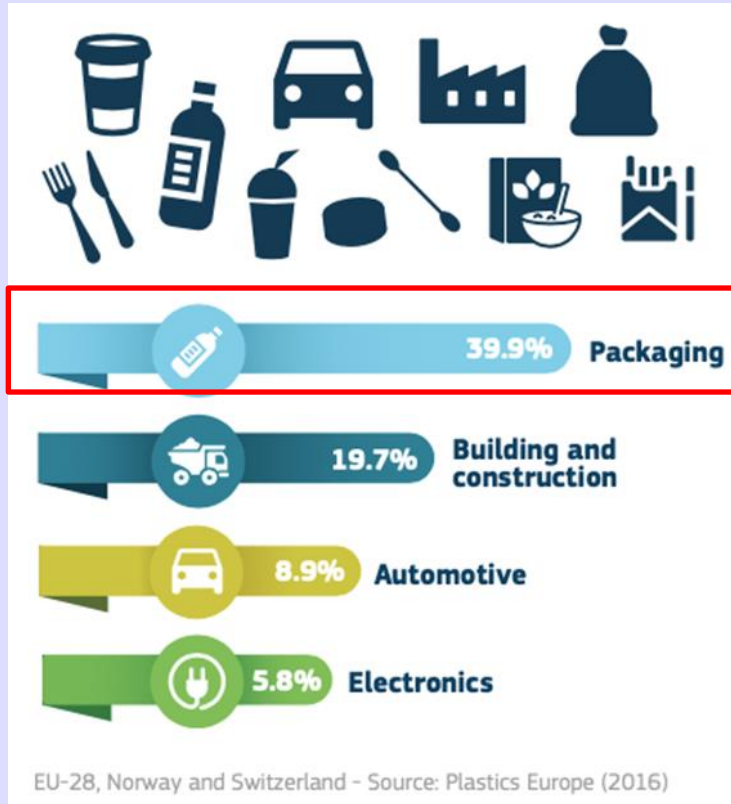




**Targeted and non-targeted gas
chromatography-mass spectrometry analysis
of volatile compounds released from
recycled and virgin polyethylene
terephthalate polymer: Authentication study**

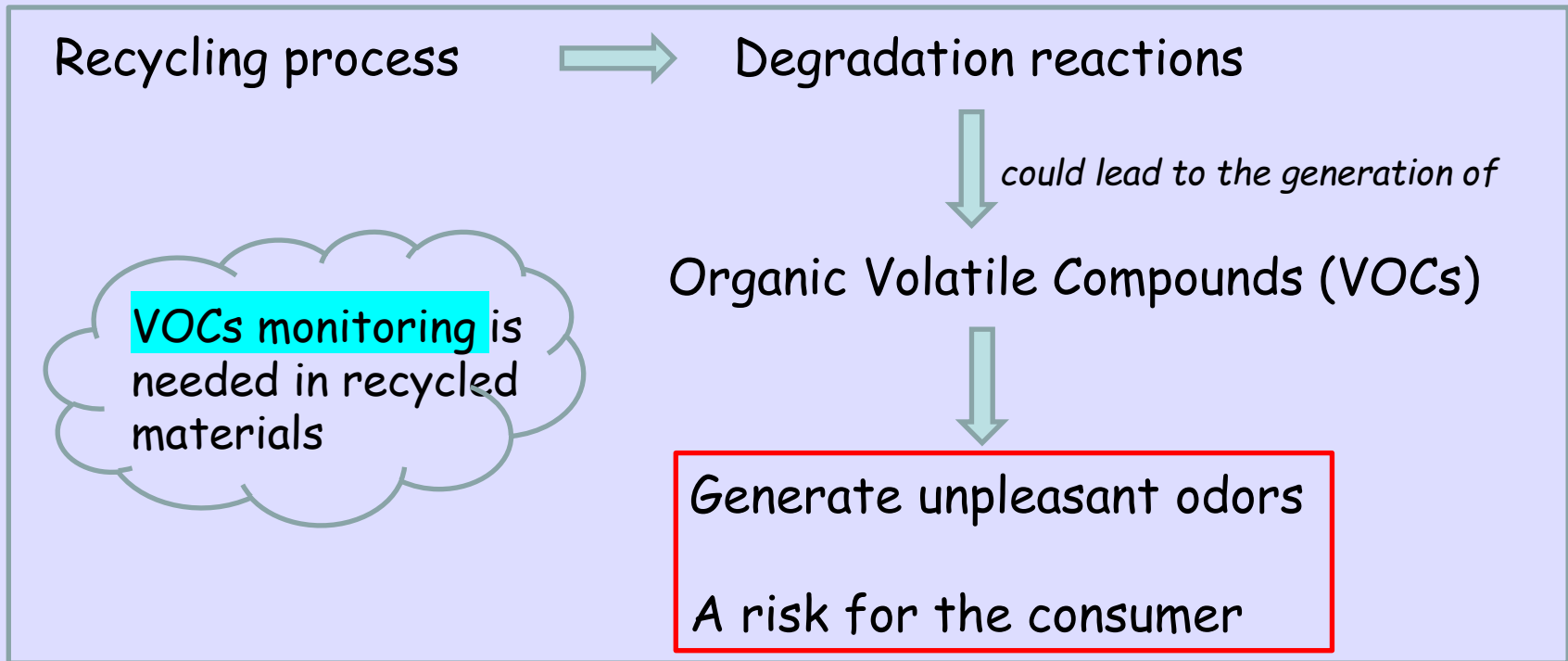
Dra. Rosa María Peñalver Soler,
University of Murcia, Spain

Plastic production has highly increased



Sustainable plastic packaging solution → Recycling

Recycling plastic packaging is needed to reduce plastic residue, BUT there is a need to control its quality



Authentication of recycled packaging is required to avoid frauds.

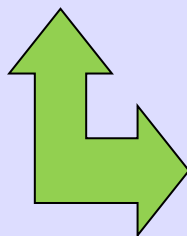
Analytical strategy for VOCs determination and authentication of recycled plastic content



HS-GC-MS



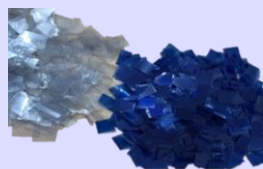
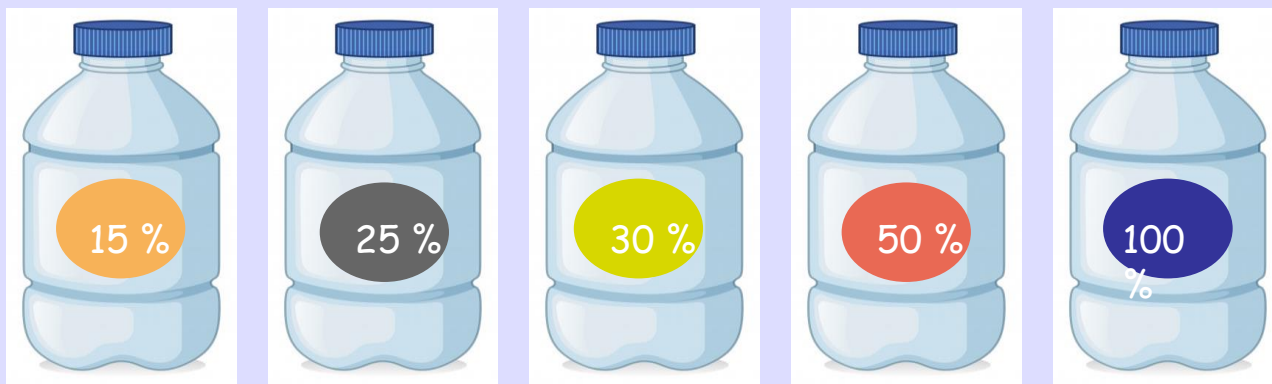
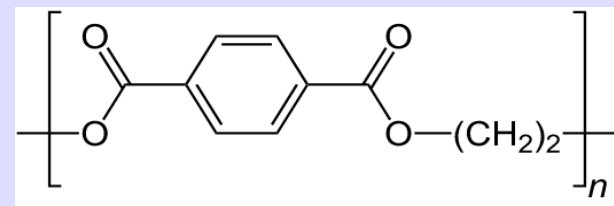
VOCs fingerprint



Chemometric (Principal Component Analysis & Partial Least Square correlation)



Polyethylene Terephthalate mineral water bottles: virgin & different recycled content



0.3 x 0.4 cm

HS-GC-MS

Volatile fingerprint

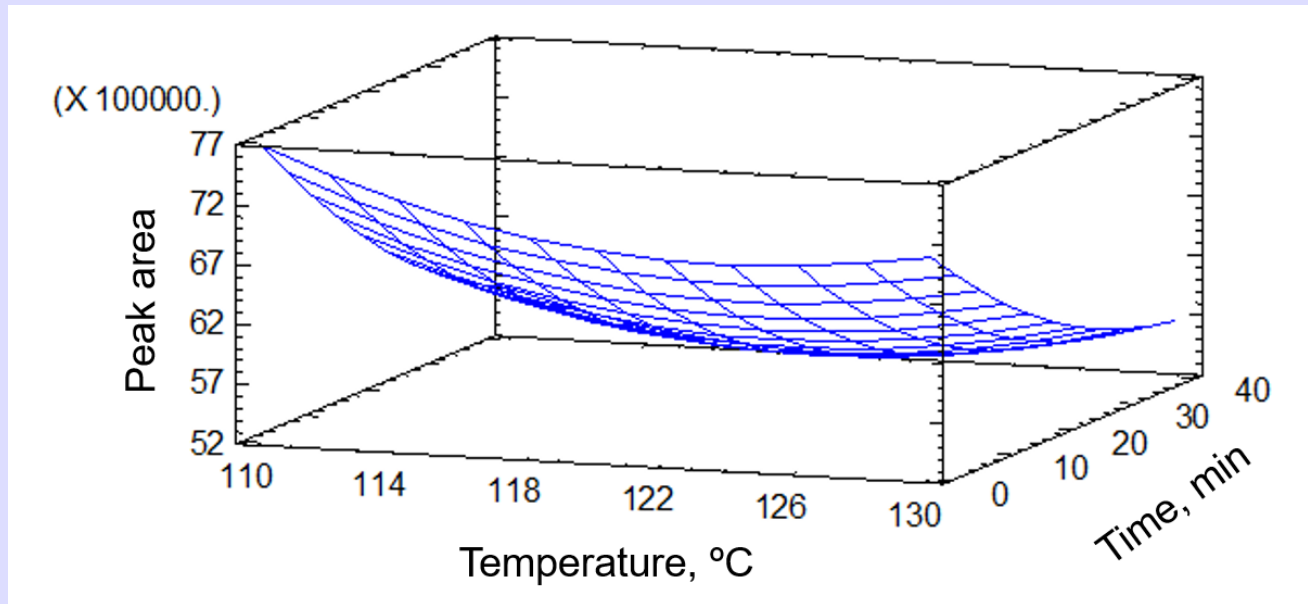
Quantification of some VOCs

Chemometric (PLS) Authentication study

2 g of recycled PET sample
+ 2-chloroanisole (Internal Standard, IS)

Selection of the HS incubation temperature and time

Surface Response Design method (face-centered):




Higher analytical signals were obtained with 110 °C and 5 min.

NON-TARGETED APPROACH FOR VOCs

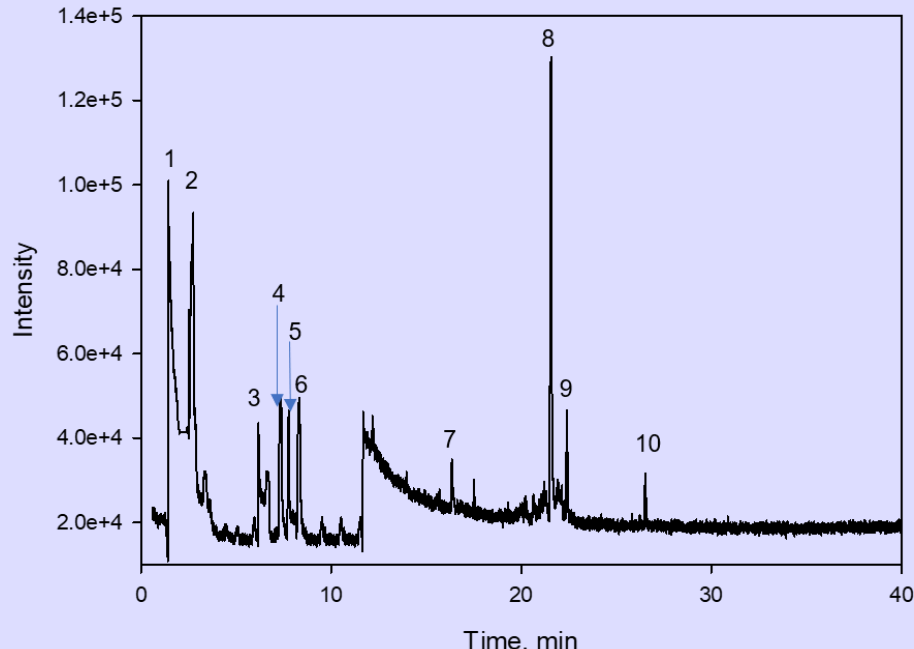
Homemade database including VOCs from the literature

Compounds	ions (m/z)
2,4-Diterbutyl phenol (DBP)	57, 191, 205
2-Ethyl-1,3-dimethylbenzene	91, 106, 107
10-Methylnonadecane	57, 71, 282
5-Methyl undecane	43, 57, 156
7-Methyl hexadecane	57, 71, 240
Tritetracontane	57, 71, 605
Tetratetracontane	57, 71, 619
1-Chlorooctadecane	43, 57, 289
Heptacosane	43, 57, 380
Decane	43, 57, 142
Hexadecane	43, 57, 226
Nonadecane	43, 57, 268
Dodecane	43, 57, 170
Octadecane	43, 57, 254
Tetradecane	43, 57, 198
Propyl octyl ether	43, 57, 71
10-Methyltricosane	43, 57, 296
3,8-Dimethyldecane	43, 57, 170
2,6,10-Trimethylidodecane	57, 71, 212
1,1'-Oxybis dodecane	43, 57, 354
1,1'-Oxybis decane	43, 57, 298
Tetratriacontane	57, 71, 479
Benzophenone	77, 105, 182
Dibutyl phthalate (DBP)	149, 150, 278
Diethyl phthalate (DEP)	149, 177, 222
Heptanal	41, 70, 114
Dodecanal	43, 57, 184
Decanal	41, 43, 156
1,2-dichlorobenzene	146, 148, 150
Toluene	91, 92, 93
1-octene	43, 55, 112
2-undecanone	43, 58, 170
2-nonanone	43, 58, 142
2-heptadecanone	43, 58, 254
2-heptenal	27, 41, 112
2-octenal	41, 55, 126
2-decenal	41, 43, 154
n-Acetic acid	43, 45, 60
n-Hexanoic acid	60, 73, 87
Octadecanoic acid	43, 73, 284
Acetone	42, 43, 58
2,6-Dimethyl-nonane	43, 57, 156
3-Methyl-5-propyl-nonane	57, 71, 184
4-Methyl-decane	43, 71, 156
2-Methyl-decane	43, 57, 156
3-Methyl-decane	57, 71, 156
3,7-Dimethyl-decane	43, 57, 170
3,6-Dimethyl-undecane	43, 57, 184
2-Methyl-dodecane	43, 57, 184
2-butanamine	41, 44, 86
n-Propyl acetate	43, 61, 73
Methyl benzene	91, 92, 93

HS-GC-MS



Chromatogram of an unspiked real sample to which the IS was added (50 ng g⁻¹)

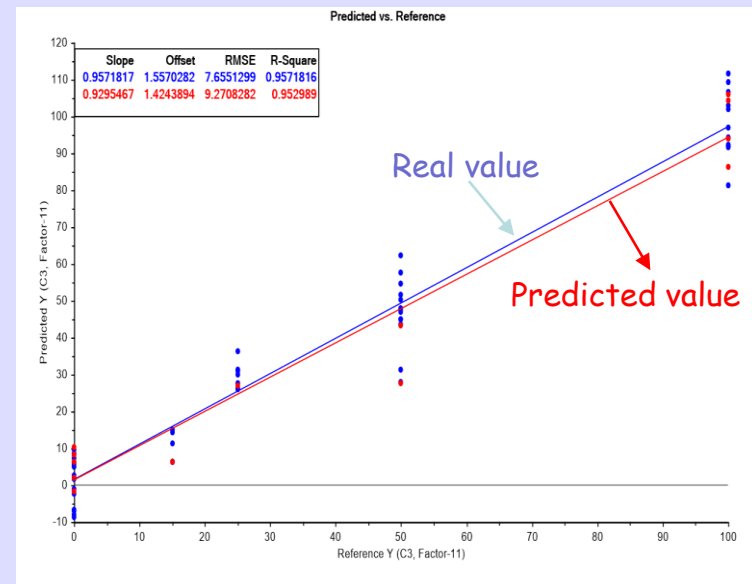
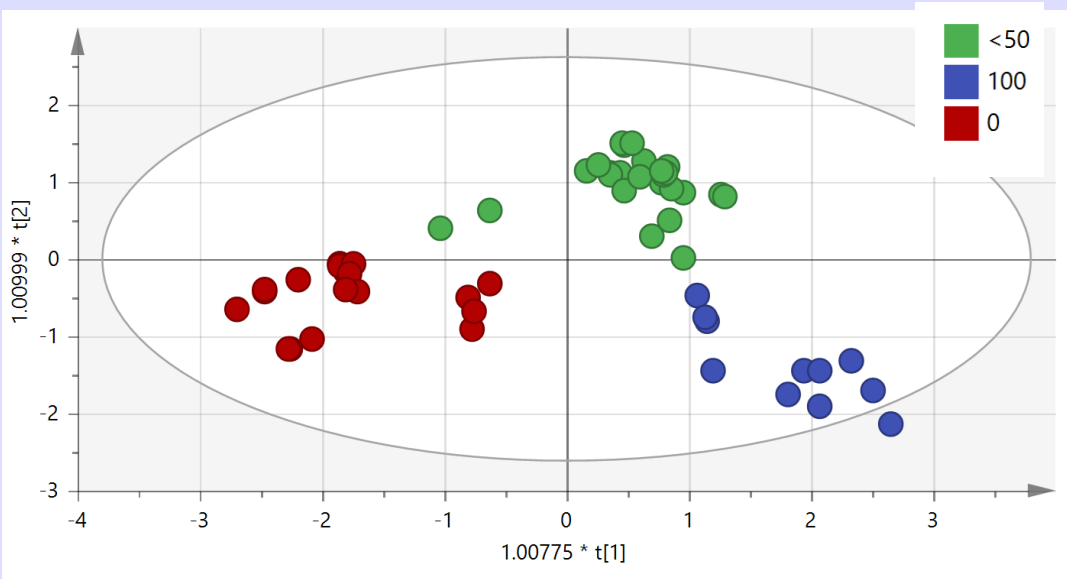


- (1) Ethyl acetate, (2) 2-Methyl-1,3-dioxolane, (3) Propyl acetate, (4) Pentanal, (5) Hexanal, (6) Heptanal, (7) Nonanal, (8) Decanal, (9) IS and (10) 5-Methyl undecane

CHEMOMETRIC ANALYSIS OF THE VOLATILE FINGERPRINT

Classificatory purposes: OPLS-DA (Orthogonal partial least squares model):
UV scaling without logarithmic transformation

Quantitative approach:
PLS regression model



100% recycled, ≤ 50 % recycled and virgin PET samples showed some differences in their VOCs and formed three different regions in the PCA score plot related to their VOCs fingerprint

Good fit between reference (real) and predicted value :
Correlation 97.7%

Analytical characteristics of the developed HS-GC-MS method for VOCs quantification

Analytical characteristics of the method

Compound	Linearity range (ng g ⁻¹)	QL ^a (ng g ⁻¹)	RSD, %	RSD, %
Ethyl acetate	10-1000	10.4	3.6	4.2
2-Methyl-1,3-dioxolane	1.25-100	1.3	2.4	2.0
Pentanal	12.5-1000	12.5	1.6	1.9
Hexanal	1.25-100	1.3	7.2	8.0
o-xylene	2-100	2.0	2.3	1.9
Ethyl benzene	10-1000	10.4	6.0	6.8
Styrene	2-100	2.0	3.7	4.9
p-xylene	2-100	2.0	4.6	5.7
Cumene	25-100	2.0	3.1	4.1
Alpha-pinene	12.5-1000	12.5	3.9	4.9
Benzaldehyde	12.5-1000	12.5	6.9	7.7
Octanal	10-1000	10.4	5.4	4.8
Alpha-terpinene	10-1000	10.4	2.4	3.2
Limonene	25-1000	25	8.9	8.1
Linalool	12.5-1000	12.5	4.5	4.7
Nonanal	1.25-100	1.3	8.7	8.1
Naphthalene	12.5-1000	12.5	3.2	5.1
Butylhydroxytoluene	12.5-1000	12.5	9.3	11.5

Virgin PET sample fortified at 50 ng g⁻¹ with all analytes including the IS (n=10)

100% recycled PET sample fortified at 50 ng g⁻¹ with all analytes including the IS (n=10)

RECOVERY STUDIES:

- Two 100% recycled PET samples
- Spiked at two levels (25 and 100 ng g⁻¹)
- Recovery values: 90-110%

^a Calculated for Signal/Noise=10 → from 1.3 to 25 ng g⁻¹

Relative standard deviation → from 1.6 and 11.5 %, corresponding to pentanal and butylhydroxytoluene, respectively.

Quantification of VOCs in the PET bottle samples

Analyte content^a (ng g⁻¹) of PET mineral bottle samples

Samples	Ethyl acetate	2- methyl-1,3-dioxolane	Pentanal	Hexanal	Styrene	p-xylene	Benzaldehyde	Octanal	Nonanal
1A	ND	885 ± 78	ND	ND	ND	ND	ND	ND	144 ± 16
2A	ND	964 ± 101	ND	21 ± 1	ND	ND	ND	ND	87 ± 8
3A	ND	650 ± 61	ND	272 ± 36	ND	ND	ND	120 ± 3	874 ± 44
5A	ND	563 ± 12	ND	32 ± 5	ND	ND	ND	65 ± 6	91 ± 22
6A	ND	513 ± 53	ND	17 ± 4	ND	ND	ND	31 ± 1	149 ± 11
7A	72 ± 1	995 ± 180	70 ± 7	63 ± 11	ND	ND	ND	44 ± 20	235 ± 16
8A	ND	777 ± 31	ND	ND	ND	ND	ND	57 ± 1	206 ± 4
9A	ND	996 ± 99	ND	ND	ND	ND	ND	ND	42 ± 3
12A	ND	977 ± 141	ND	ND	ND	ND	ND	ND	91 ± 4
10B	ND	213 ± 17	222 ± 13	11 ± 3	ND	ND	ND	ND	56 ± 13
13B	ND	563 ± 80	ND	ND	ND	ND	ND	ND	27 ± 2
4C	ND	258 ± 9	ND	29 ± 2	ND	ND	ND	ND	160 ± 9
7C	267 ± 45	248 ± 53	182 ± 13	123 ± 3	ND	ND	ND	40 ± 11	25 ± 5
2D	ND	467 ± 48	ND	63 ± 27	ND	ND	ND	ND	187 ± 28
4D	ND	369 ± 40	ND	216 ± 13	ND	ND	99 ± 5	53 ± 1	545 ± 35
6D	ND	387 ± 17	ND	28 ± 3	35 ± 1	ND	ND	62 ± 9	235 ± 25
7D	150 ± 1	381 ± 9	75 ± 6	164 ± 25	ND	ND	ND	48 ± 3	297 ± 66
11D	ND	154 ± 10	189 ± 7	88 ± 5	ND	ND	ND	22 ± 1	112 ± 7
2E	ND	94 ± 2	94 ± 2	139 ± 11	ND	ND	ND	ND	23 ± 1
5E	ND	85 ± 10	310 ± 68	ND	ND	ND	ND	ND	33 ± 13
6E	ND	142 ± 25	104 ± 3	212 ± 42	14 ± 0.7	ND	ND	ND	171 ± 19
7E	915 ± 90	146 ± 23	100 ± 8	54 ± 5	ND	ND	ND	35 ± 2	14 ± 0
11E	ND	108 ± 6	114 ± 9	117 ± 4	ND	10 ± 0.3	ND	46 ± 2	272 ± 7
14E	ND	58 ± 1	ND	79 ± 5	ND	ND	ND	ND	38 ± 2

^a Mean value ± standard deviation (n=3)

ND: No detected

- Samples named as:

A: 0% recycled PET

B: 15% recycled PET

C: 25% recycled PET

D: 50% recycled PET

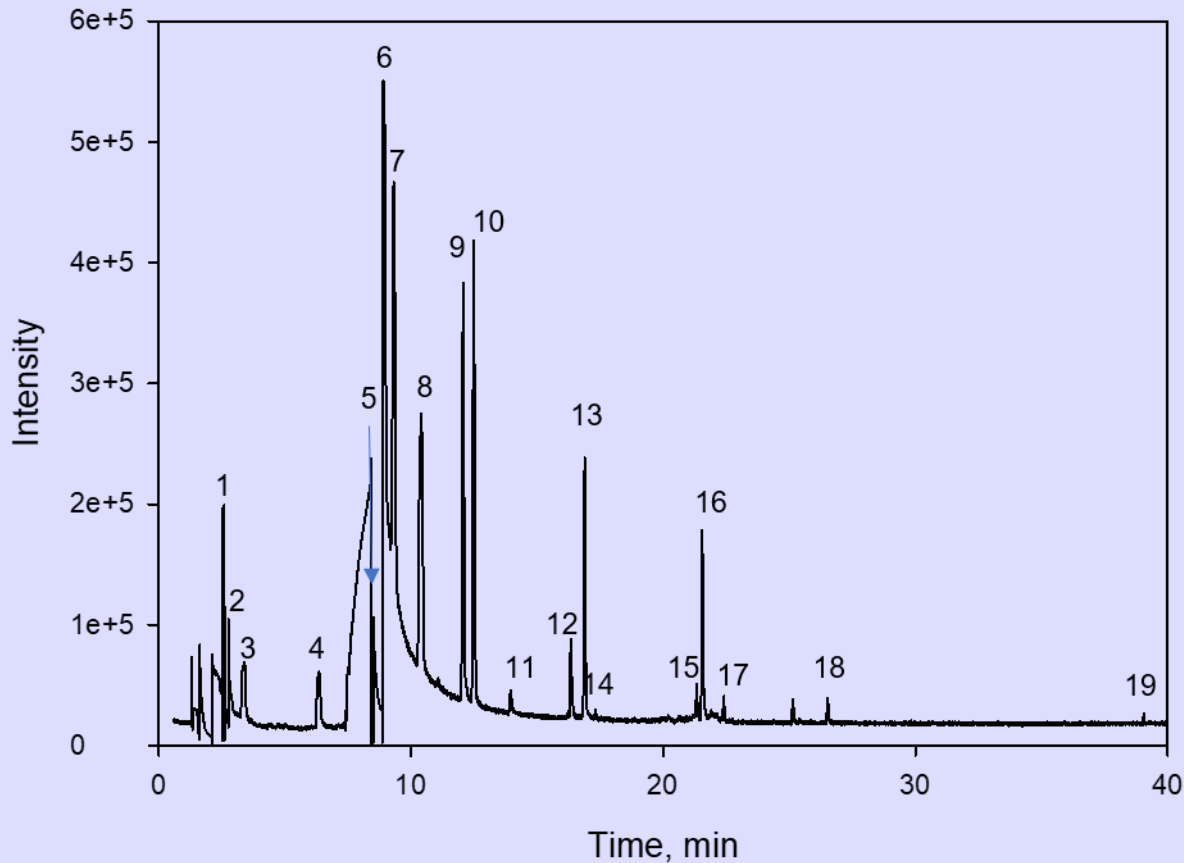
E: 100 % recycled PET

- Different numbers correspond to different PET mineral water bottles suppliers.

- Most of the compounds present in the samples were aliphatic aldehydes.

- Benzene derivatives such as benzaldehyde, p-xylene and styrene were found in very few recycled samples

HS-GC-MS chromatogram of a spiked PET bottle sample (50 ng g⁻¹)



- 1) Ethyl acetate
- 2) 2-Methyl-1,3-dioxolane
- 3) Pentanal
- 4) Hexanal
- 5) o-Xylene,
- 6) Ethyl benzene
- 7) Styrene
- 8) p-Xylene
- 9) Cumene
- 10) Alpha-pinene
- 11) Benzaldehyde
- 12) Octanal
- 13) Alpha-terpinene
- 14) Limonene
- 15) Linalool
- 16) Nonanal
- 17) 2-Chloroanisole (IS)
- 18) Naphthalene
- 19) Butylhydroxytoluene.

CONCLUSIONS

1. A novel non-targeted methodology based on HS-GC-MS and implying minimal sample preparation was developed to obtain the volatile fingerprint of virgin and recycled PET plastic bottles (containing different percentages of recycled plastic) from different suppliers.
2. The combination of chemometrics (OPLS-DA and PLS) with the data set obtained by HS-GC-MS methodology for PET samples has demonstrated to be a very useful tool to predict the percentage of recycled material in the PET bottle samples: AVOID FRAUDS.
3. Main contributors to the classification between virgin and recycled PET materials are 2-methyl-1,3-dioxolane which is present in higher amount in virgin PET bottles, and aliphatic aldehydes which are linked to secondary reactions occurring during recycling processes being, therefore, these compounds characteristics of recycled PET samples.
4. Seventeen VOCs were detected in the samples. Nine of them were quantified in the studied samples: ethyl acetate, 2-methyl-1,3-dioxolane, pentanal, hexanal, styrene, p-xylene, benzaldehyde, octanal and nonanal, being their concentrations ranging from 21 to 996 ng g⁻¹ which corresponded to hexanal and 2-methyl-1,3-dioxolane, respectively.

Thank you for your attention