



UNIVERSITI PUTRA MALAYSIA

RICULTURE • INNOVATION • LIFE

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**Tapered Optical Fiber for Hydrogen
Sensing Application Based on Molybdenum
trioxide (MoO_3)**

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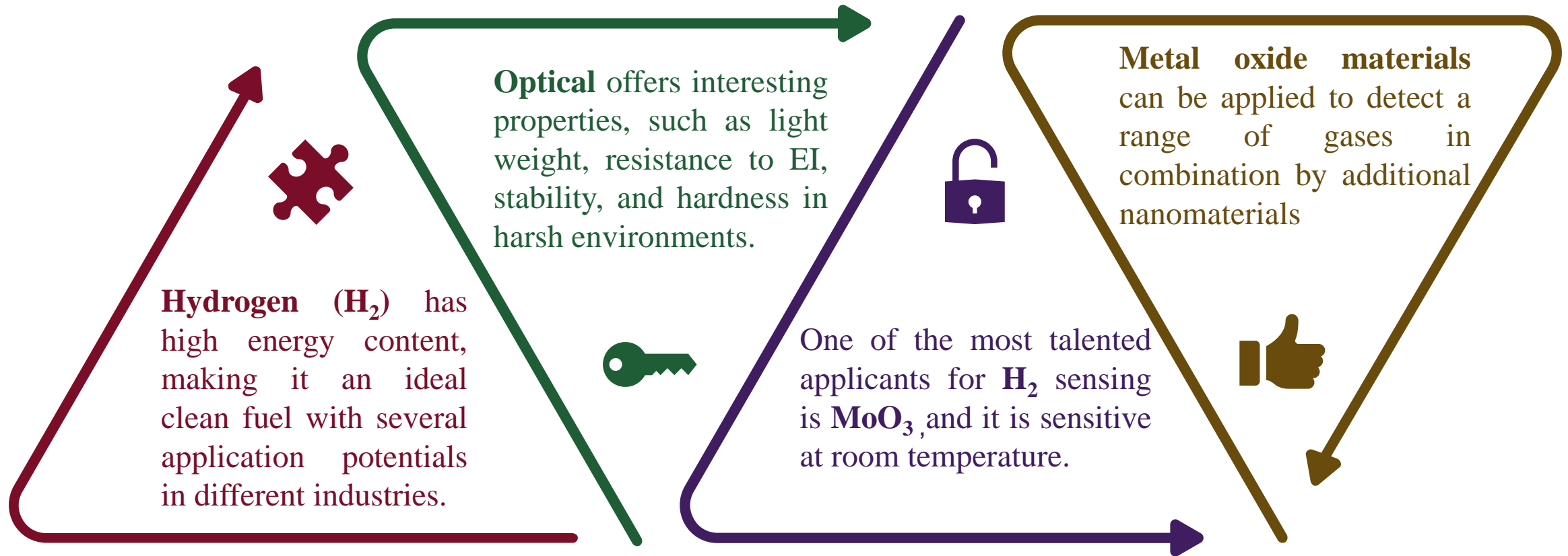
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CONTENT



INTRODUCTION



PROBLEM STATEMENT

- Hydrogen is flammable at concentrations > 4 vol% in the air and can explode at a wider range of 15–59 vol% at standard pressure.
- Currently, there are several types of H₂ sensors available.
 - Electrical sensors (i.e. chemiresistor or microelectronic) are susceptible to electromagnetic interference (EMI) which can affect their response to signals.



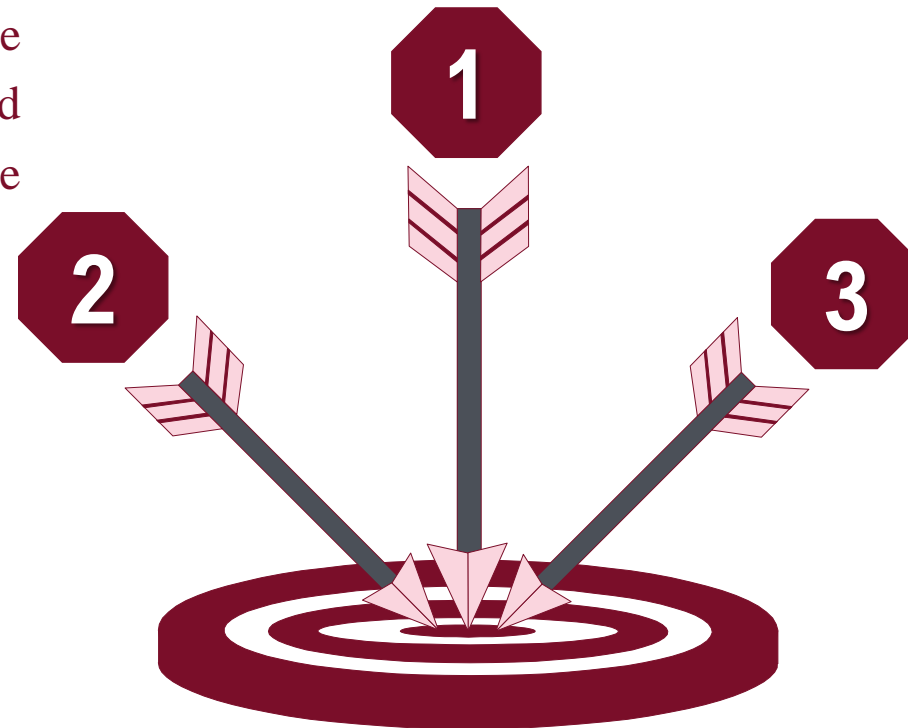
- On the other hand, optical fiber that offers other advantages, such as lightweight, small size, resistance to EMI, non-inductiveness, and ruggedness in harsh environments.
 - These properties make optical fiber an ideal candidate for H₂ detection in rugged environment.

OBJECTIVES

To design and develop hydrogen gas sensors based on PANI coated on tapered optical fiber via drop-casting technique.

To evaluate the optical fiber sensor performance (sensitivity, response and recovery time, repeatability, and selectivity) based on absorbance measurement.

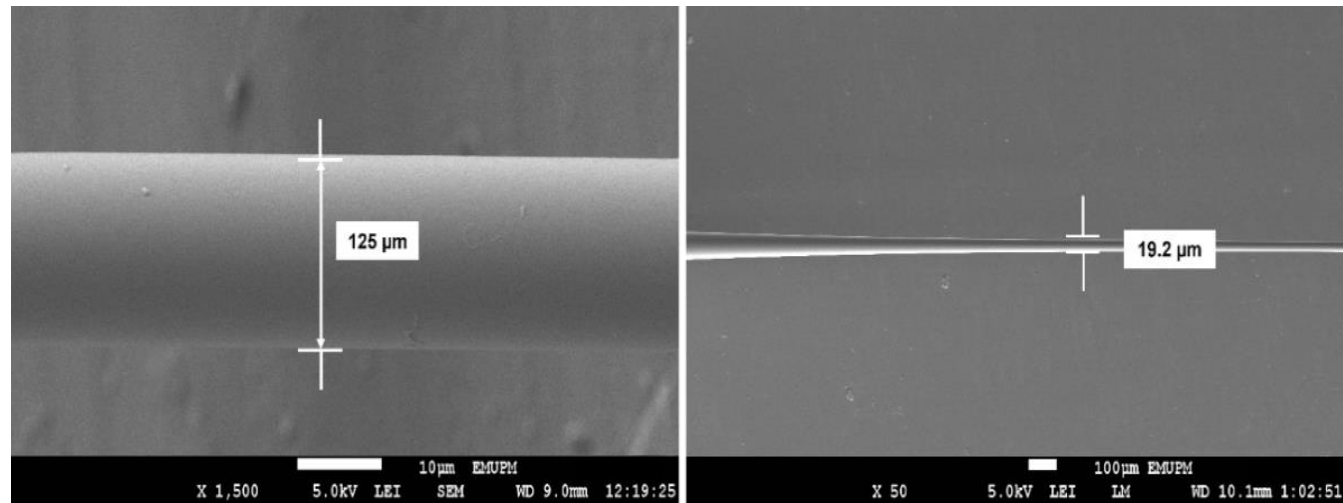
To discuss the sensing mechanism of gas molecules-sensing layer interaction of tapered optical fiber sensor.



METHODOLOGY

Fabrication of Tapered Optical Fiber

- Multimode Optical fiber (MMF) was fabricated with cladding and core diameters of 125 μm and 62.5 μm respectively, as a transducing platform.
- The MMF was tapered from cladding diameter of 125 μm to waist diameter of 20 μm , waist-length of 10 mm, and down taper and up of 5 mm.
- The tapering was done using the Vytran glass processing machine (Vytran GPX-3400).
- The machine works based on a heating and pulling process, using a graphite filament as a heater to achieve the desired geometry of the tapered profile.

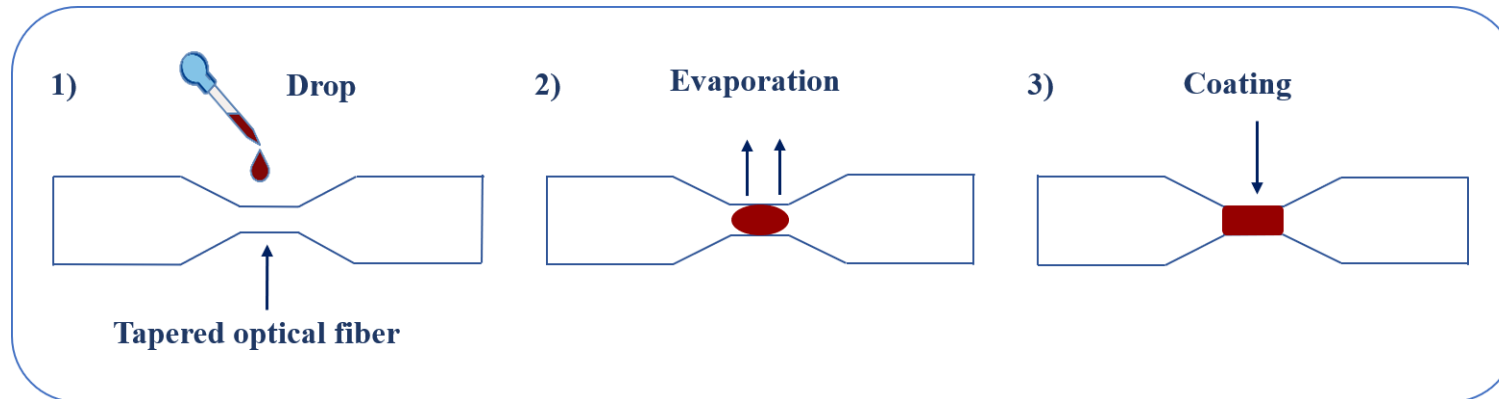


(SEM) micrograph of the transition region of the prepared tapered multimode optical fiber (MMF)

METHODOLOGY

MoO₃ Functionalization of the Tapered Optical Fiber

- Molybdenum trioxide (MoO₃) powder was synthesized by simple solid decomposition method.
- 2.50 g of ammonium heptamolybdate tetrahydrate.
- Annealed in an alumina crucible at 500 °C for three hours in the air.
- A 10 mL of deionized water.
- A milk-white suspension after ultrasonic treatment for 30 minutes.
- The coating of the tapered optical fiber was done using the drop-casting technique.
- A drop of the mixture (approx. 10 μL) was dropped into the base of the tapered optical fiber.
- Heating the sample at 80 °C for 15 minutes in the oven to ensure complete evaporation of the aqueous medium.

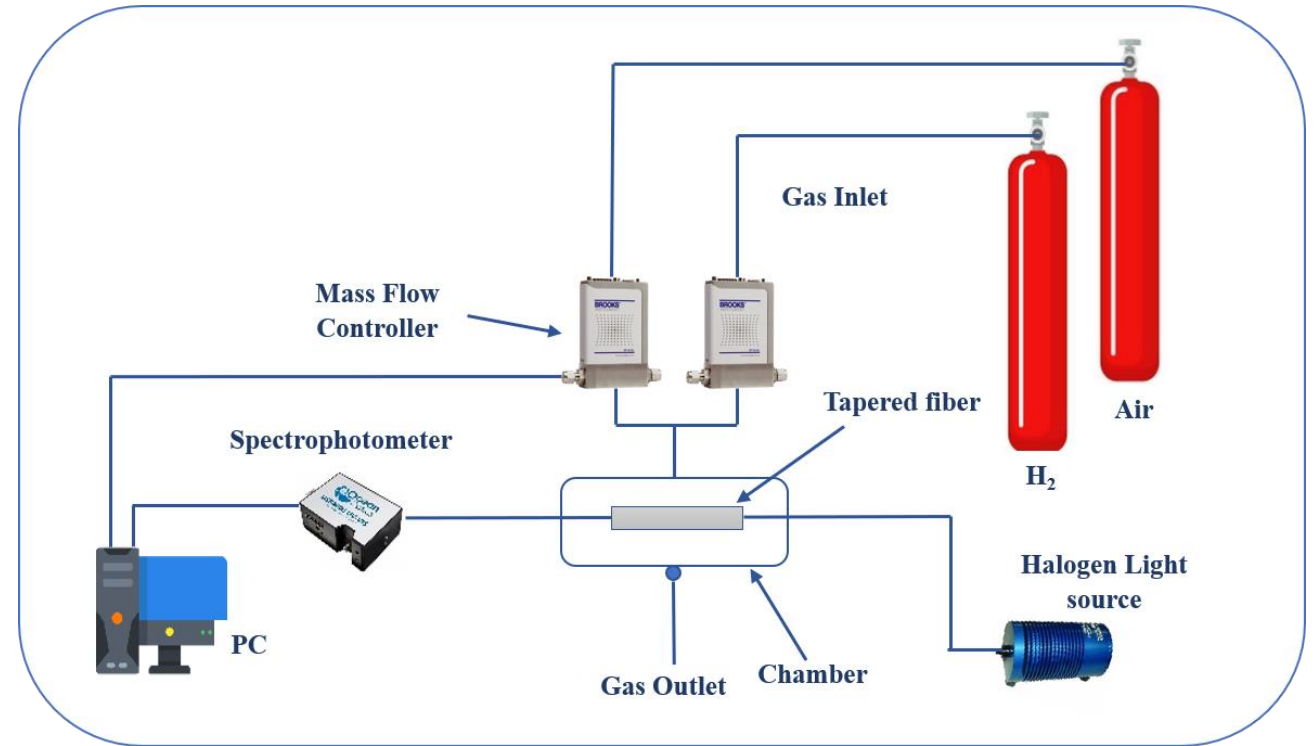


Steps of the drop-casting technique.

METHODOLOGY

The experimental setup

- The gas optical sensing system consists of a light source (Tungsten Halogen, HL-2000, Ocean Optics USA) with coverage wavelength of 360 to 2500 nm.
- A spectrophotometer (USB 4000, Ocean Optics USA) with a detection range of 200-1100 for monitoring the optical absorption spectrum.
- A dedicated gas chamber.
- The MoO₃ coated sensor was placed in a closed gas unit and purged with the centrifuge from a computer-regulated mass flow controller at a gas flow rate of 200 sccm.

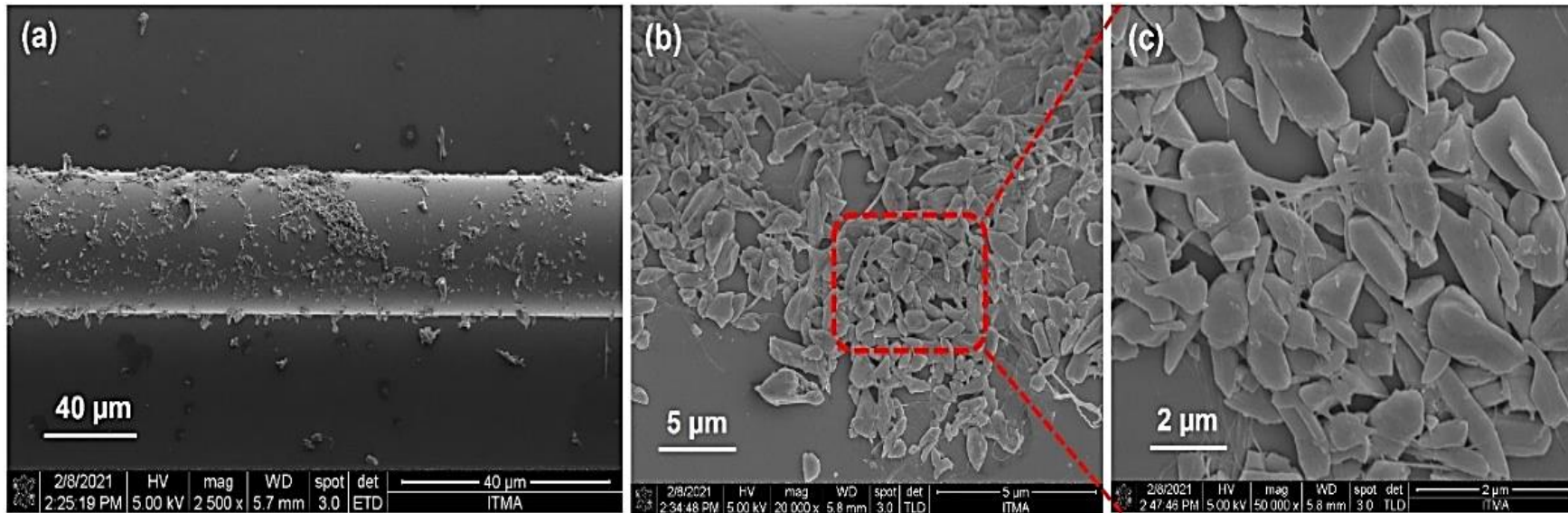


The experimental setup of the H₂ sensor.

METHODOLOGY

Material Characterization

- The films' morphology was observed using Field Emission Scanning Electron Microscope (FESEM) (JSM-7600F).
- The FESEM images of MoO_3 shows that it is The particles agglomerated in nature after attempting to form thin plate or flake-like formations. The gas analyte will interact more effectively with this plate-like structure.

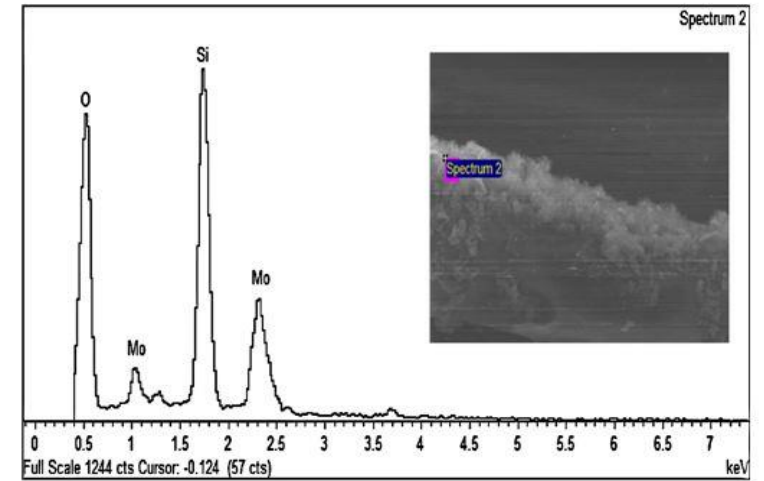


FESEM micrograph of MoO_3 .

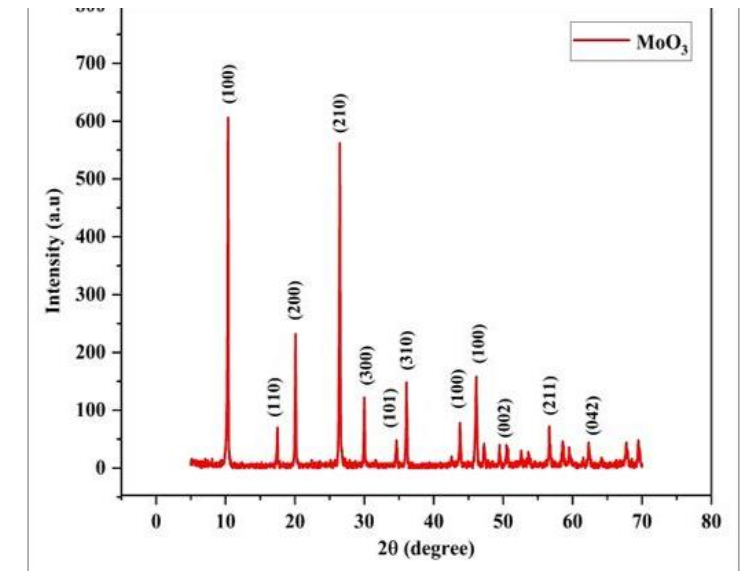
METHODOLOGY

Material Characterization

- The elemental composition of MoO_3 was determined through an Energy Dispersive X-Ray (EDX) analysis as shown in Figure (b).
- The EDX pattern of MoO_3 showed that the important elements in MoO_3 films are Mo, O and Si, as evidenced by their respective peaks.
- Material identification and phase transition of MoO_3 was observed by an X-Ray Diffraction (XRD) analysis (APD 2000) as shown in Figure (a).
- XRD patterns of the coated sensor recorded in range 2θ , from 5° to 70° .
- All the XRD peaks can be distinguished for the sample, and single-phase can be as-signed to crystal structures. The prominent diffraction peak corresponding to planes (100) and (210) placed at the highest density of MoO_3 .



(a)

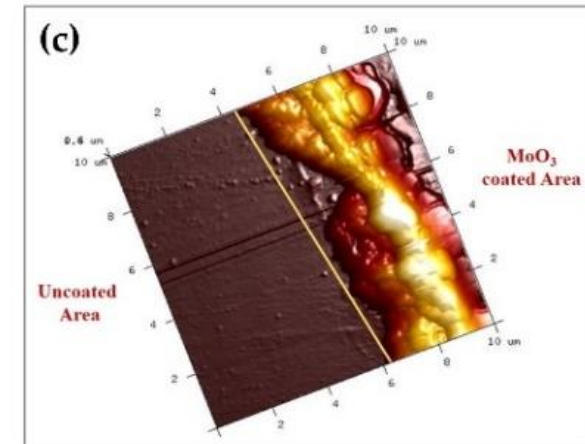
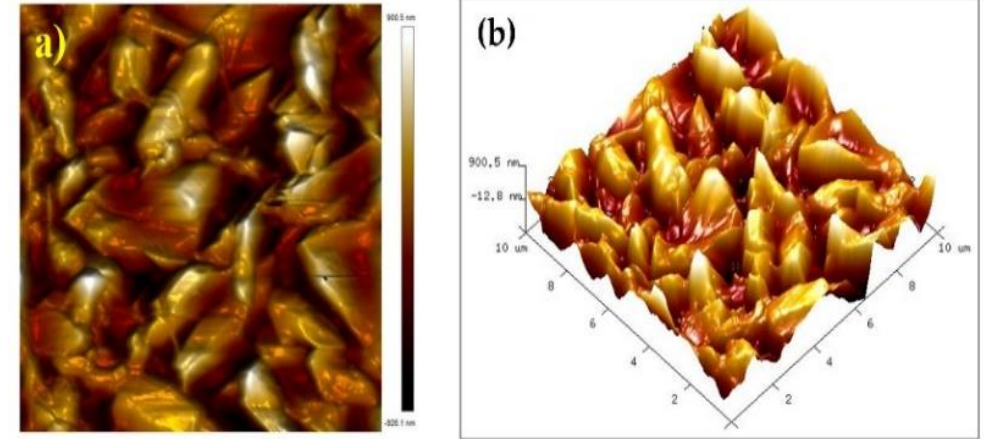


(b)

METHODOLOGY

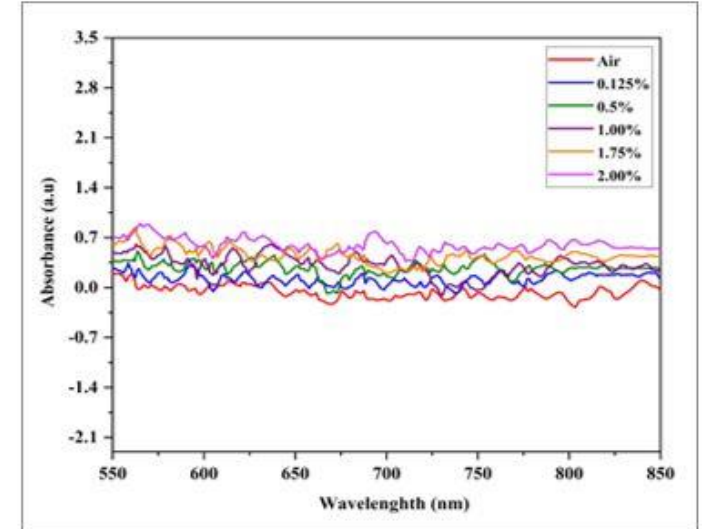
Material Characterization

- The atomic force microscope (AFM) can verify the average surface roughness and thicknesses of MoO_3
 - A $10 \times 10 \mu\text{m}$ section of the boundary area was scanned for the AFM analysis.
 - The average surface roughness values of the MoO_3 were $\approx 44.98 \text{ nm}$, as shown in (a and b).
- As part of this study, the thicknesses of the MoO_3 coatings were measured.
 - As part of this study, the thicknesses of the MoO_3 coatings were measured. As shown in Figure 5c, The average thickness of the MoO_3 coatings was 181 nm .

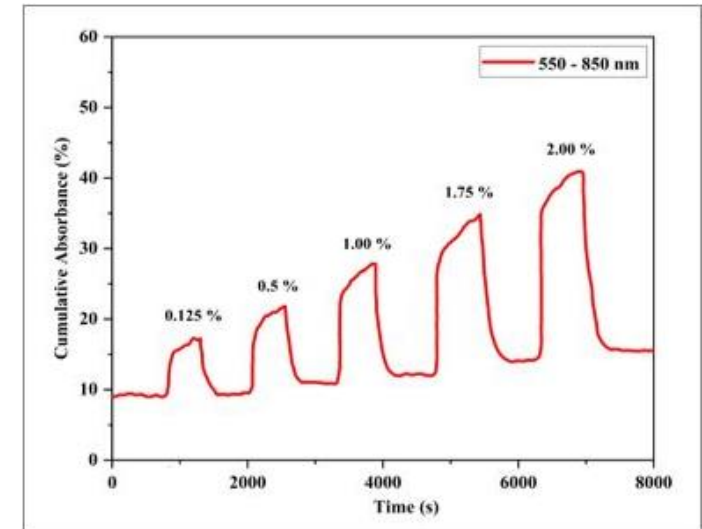


RESULTS AND DISCUSSION

- The absorption spectra of the sensor coated with MoO_3 to synthetic air at room temperature with different concentration 0.125% to 2.00% H_2 .
- The MoO_3 sensor demonstrated notable changes in absorbance, especially in the wavelength range of 550-850 nm as shown in Figure (a).
- The response time and recovery time of the MoO_3 coated sensor was 200 s and 220 s respectively. Changes in absorption at 0.125% H_2 are about 8% and 25% higher at 2.00% H_2 as shown in Figure (b).
- The MoO_3 coated sensor showed stronger absorbance and recovery of H_2 at higher absorption changes.



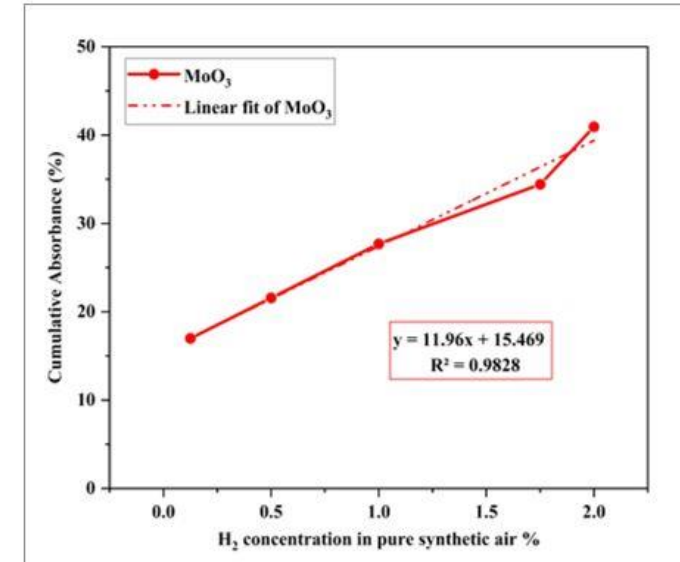
(a)



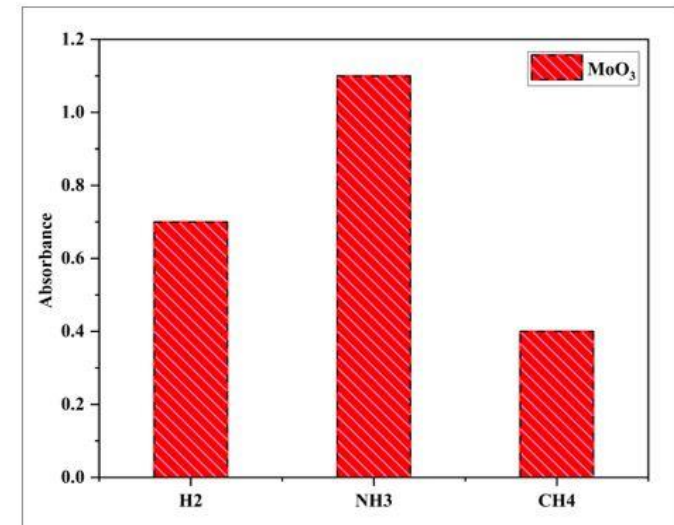
(b)

RESULTS AND DISCUSSION

- Sensor repeatability was confirmed by exposure of the sensor to 3 cycles of 2.00% H₂. Overall, the MoO₃ coated sensor showed a high level of good repeatability of H₂.
- A test for selectivity was done for MoO₃ coated sensor toward NH₃ and CH₄ gas at 1.00%.
- The MoO₃ coated based sensor had a very high NH₃ absorption response but a substantially lower response for the other gases.

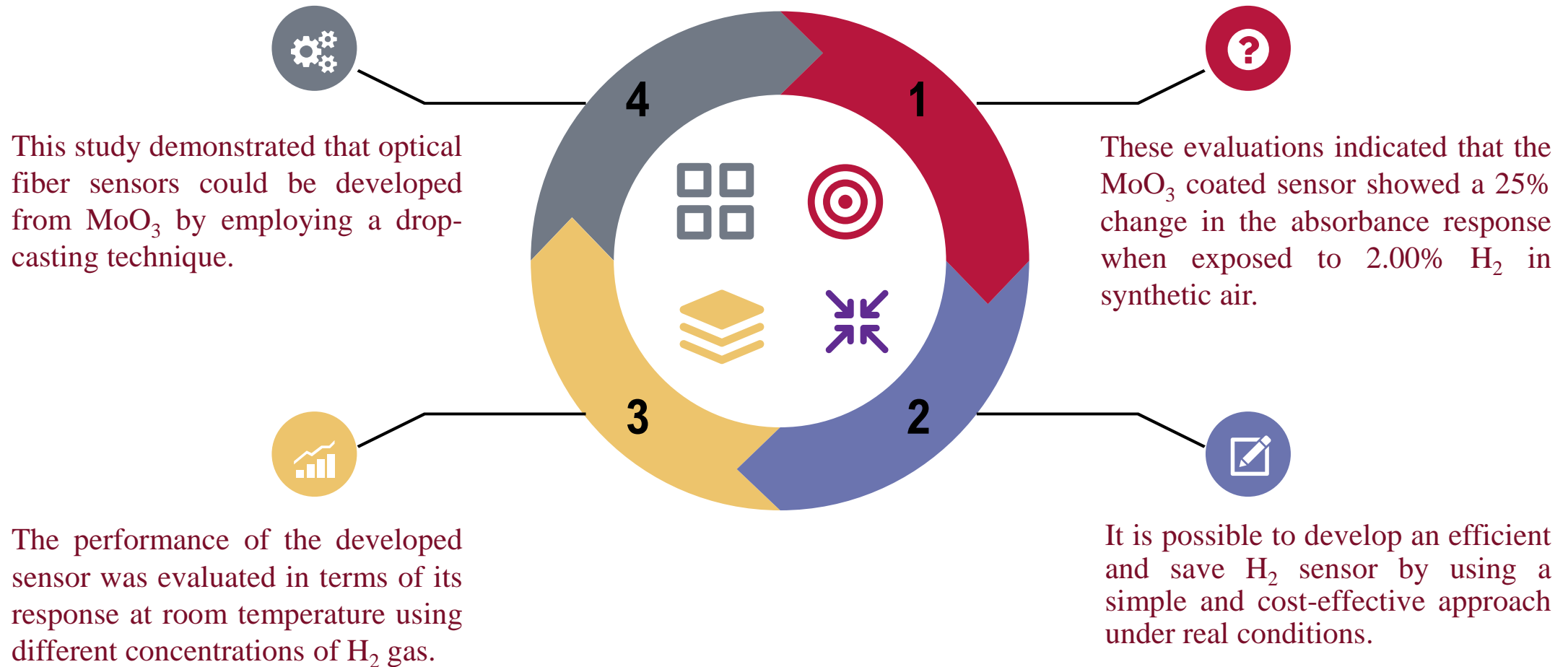


(a)



(b)

CONCLUSION





**THANK
YOU**