

Abstract

# $\text{Li}_{1+y}\text{Ti}_{2-x-y}\text{Ge}_x\text{Al}_y(\text{PO}_4)_3$ NASICON-Type Electrolytes with Enhanced Conductivity for Solid State Lithium-Ion Batteries <sup>†</sup>

Ekaterina Kurzina \* and Irina Stenina

Institute of general and inorganic chemistry of Russian Academy of Science; irina\_stenina@mail.ru

\* Correspondence: katya.kurzina@gmail.com

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**Abstract:** The use of lithium-ion batteries allows to reliable and efficient storage of electricity. Commercial batteries use flammable liquid organic electrolytes, which have low thermal and electrochemical stability. Replacing liquid electrolytes with solid ones will solve these problems. NASICON structured electrolytes, in particular LATP ( $\text{Li}_{1+y}\text{Ti}_{2-y}\text{Al}_y(\text{PO}_4)_3$ ) and LAGP ( $\text{Li}_{1+y}\text{Ge}_{2-3y}\text{Al}_y(\text{PO}_4)_3$ ), are among the most promising electrolytes for all-solid-state batteries. Partial replacement of titanium ions by germanium ions can lead to materials that combine the high lithium-ion conductivity of LATP with the high chemical stability of LAGP. The aim of this work was to synthesize and study the ionic mobility of  $\text{Li}_{1+y}\text{Ti}_{2-x-y}\text{Ge}_x\text{Al}_y(\text{PO}_4)_3$  ( $x = 0-2$ ,  $y = 0-0.3$ ) with the NASICON structure.  $\text{Li}_{1+y}\text{Ti}_{2-x-y}\text{Ge}_x\text{Al}_y(\text{PO}_4)_3$  ( $x = 0-2$ ,  $y = 0-0.3$ ) electrolytes were synthesized by the solid-state method and investigated using X-ray diffraction and scanning electron microscopy, impedance spectroscopy, and NMR spectroscopy. The processes occurring during the solid-state synthesis of  $\text{Li}_{1+y}\text{Ti}_{2-x-y}\text{Ge}_x\text{Al}_y(\text{PO}_4)_3$  have been studied. An increase in conductivity from  $10^{-7}$  S/cm to  $4.6 \cdot 10^{-6}$  S/cm at 25 °C was found when 10% titanium ions were replaced by germanium. Additional introduction of aluminum results in increase in lithium conductivity up to  $1.4 \cdot 10^{-4}$  S/cm (25 °C). Since grain boundaries are of decisive importance for the overall ionic conductivity of the NASICON-structured phosphates, the influence of precursor mechanical treatment on the microstructure and ionic conductivity of the prepared materials was studied. The use of mechanical treatment leads to a significant increase in grain size (reducing the grain boundaries and its resistance) and an increase in ionic conductivity (up to  $6.4 \cdot 10^{-4}$  S/cm at 25 °C). The obtained materials can be considered as promising solid electrolytes for all-solid-state lithium batteries with high safety and stability.

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