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Biomacromolecules and the Modern World Challenges**

Structure, properties and biological activity of chitosan salts with *L*- and *D*-aspartic acid

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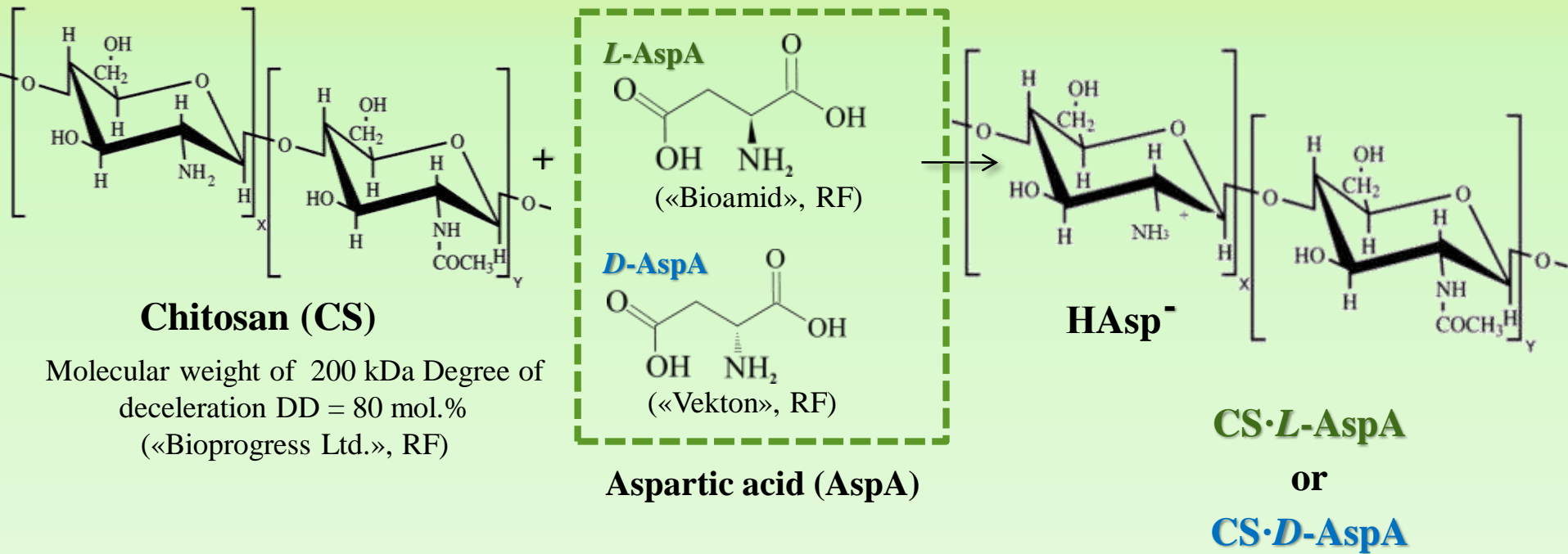
Saratov State University, Russian Federation

1–15 November 2022

Abstract

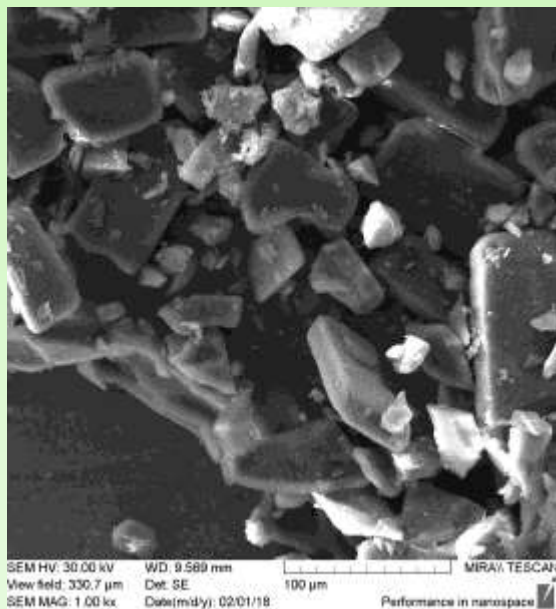
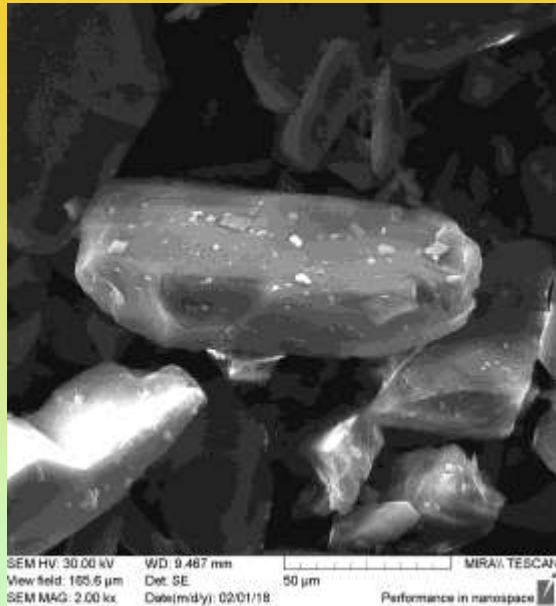
A comprehensive study of the structure, properties and biological functionality of salt chitosan complexes with *L*- and *D*-aspartic acid (AspA) was carried out. It has been established that these polymer salts differ in spatial organization, chiroptic characteristics, surface charge, and macrocoil size. In experiments *in vitro* on a wide range of biological objects (unicellular algae, planktonic crustaceans, aerobic bacterial microorganisms, cell cultures, and test plants) it was found that the chitosan salt with *D*-AspA exhibited the best biological activity. The results obtained confirm our hypothesis that the biological homochiral hierarchy principles are most consistent with chitosan (*D*-aminoglucan) derivatives with the *D*-antipode of the acid.

Preparation of chitosan aspartate

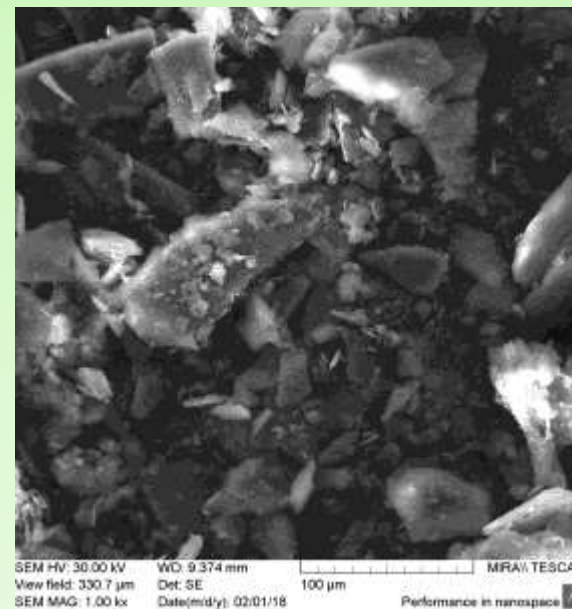
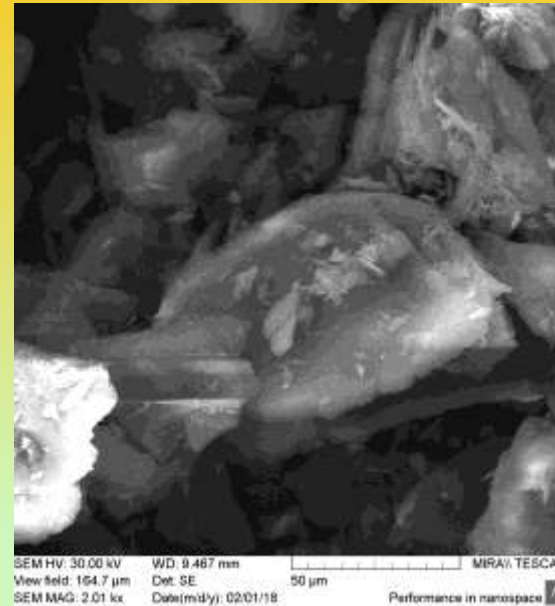


SEM photos of the powders AspA

L-AspA

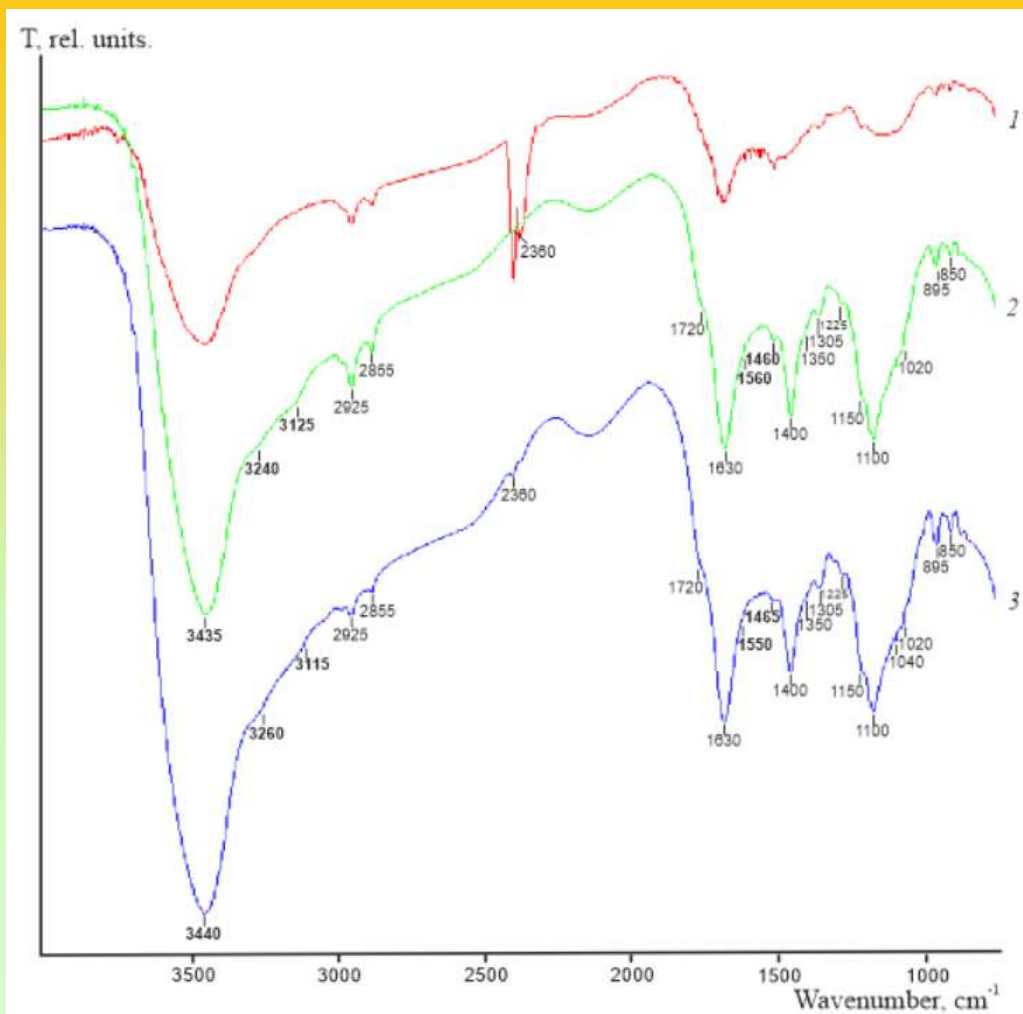


D-AspA



IF spectroscopy

Oscillation types	Wave numbers ν , cm^{-1}	
	CS·L-AspA	CS·D-AspA
$\nu_{\text{O-H}}$	3435	3440
$\nu_{\text{N-H}}$	3240	3260
$\nu_{\text{N-H}} (\text{NH}_3^+)$	3125	3115
$\nu_{\text{as}}(\text{CH})$	2925	
$\nu_{\text{s}}(\text{CH})$	2855	
$\nu_{\text{as}}(\text{C-N})$	2360	
$\nu_{\text{C=O}} (\text{COOH})$	1720	
$\delta_{\text{C=O}} (\text{Amide I})$	1630	
$\delta_{\text{N-H}} (\text{Amide II})$	1560	1550
$\nu_{\text{C-O}} (\text{COO}^-)$	1460	1465
$\delta_{\text{as}}(\text{CH})$	1400	
$\nu_{\text{C-N}} (\text{Amide III})$	1350, 1305	
$\delta(\text{OH})$	1225	
$\nu(\text{C-C}), \nu(\text{C-O}),$ $\nu(\text{C-N}), \delta(\text{C-H})$	1150	
	1020	1040, 1020
Glucopyranose ring	895	

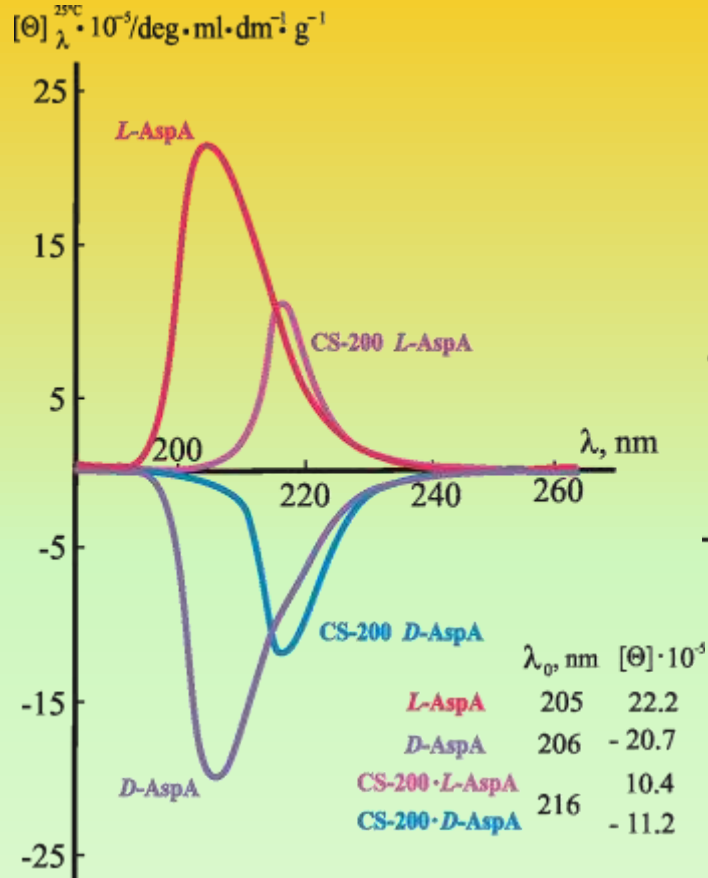


IR spectra of CS powders (1), CS·L-AspA (2) and CS·D-AspA (3)

**Vertex 70v vacuum FTIR spectrometer
(BillERICA, MA, USA)**

Chiroptic properties of solutions

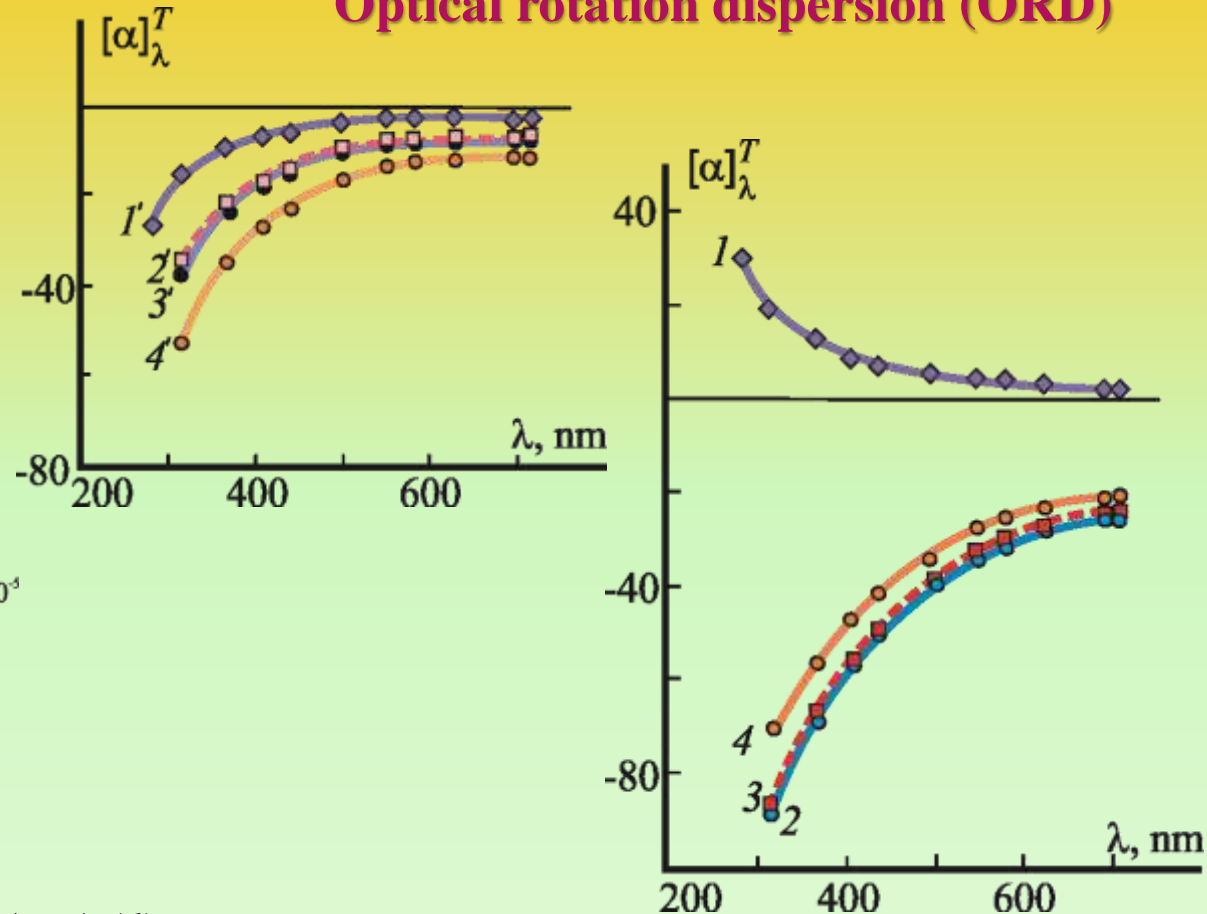
Circular dichroism (CD)



CD spectra of aqueous solutions of *L-AspA* (1), *D-AspA* (2), CS in *L-AspA* (3), and CS in *D-AspA* (4); the molar ratio $[\text{AspA}]/[\sim\text{NH}_2] = 0.85$ mol/monomol.

Chirascan™ Spectrometer (Applied
Photophysics Ltd, USA)

Optical rotation dispersion (ORD)



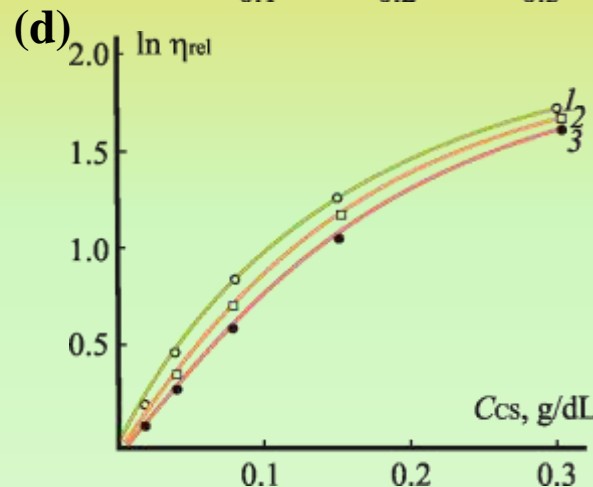
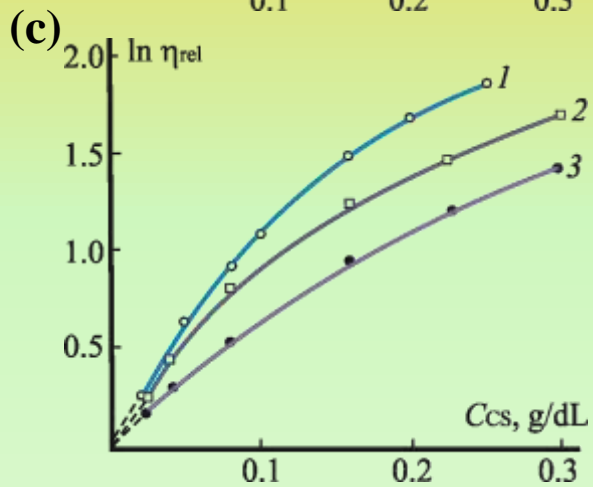
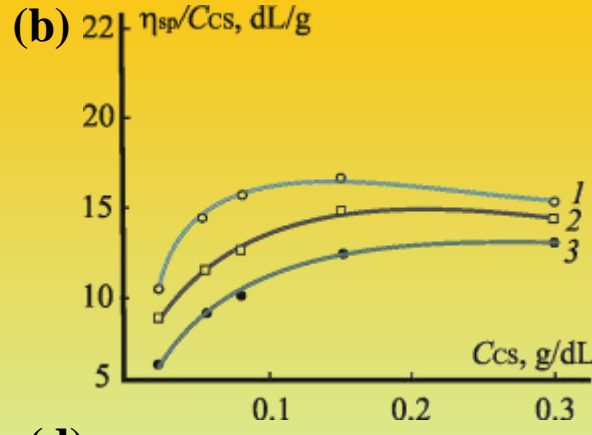
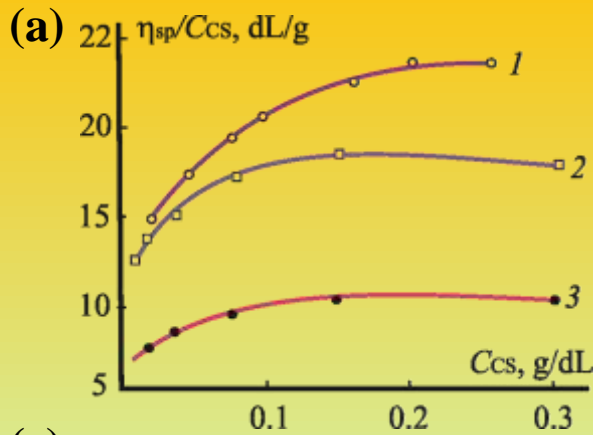
ORD curves for aqueous solutions of *L-AspA* (1), *D-AspA* (1'), CS in *L-AspA* (2–4), and CS in *D-AspA* (2'–4'); the molar ratio $[\text{AspA}]/[\sim\text{NH}_2] = 0.85$ (2, 2'), 1.0 (3, 3') and 2.0 mol/monomol (4, 4').

Automatic SPU-E spectropolarimeter (RF)

Hydrodynamic properties of solutions

Concentration dependences of the viscosity number (a, b) and the logarithm of the relative viscosity (c, d) of CS solutions in *L*-AspA (a, c) and *D*-AspA (b, d) with $C_{AspA} = 1.5 \cdot 10^{-3}$ (1), $3.0 \cdot 10^{-3}$ (2) and $6.0 \cdot 10^{-3}$ mol/dL (3).

Ubbelohde capillary viscometer (Cannon Instrument Company", USA)
 $d = 0.56$ mm



$C_{AspA} \cdot 10^3$, mol/dL	[η], dL/g		$R_{ef, HM}$	
	CS·L-AspA	CS·D-AspA	CS·L-AspA	CS·D-AspA
1.5	12.5	8.8	73	65
3.0	11.1	8.4	71	64
	3.5*	2.1*	48*	41*
6.0	7.5	6.8	62	60

* $I = const$ (0.17 M NaCl)

Limit number of viscosity

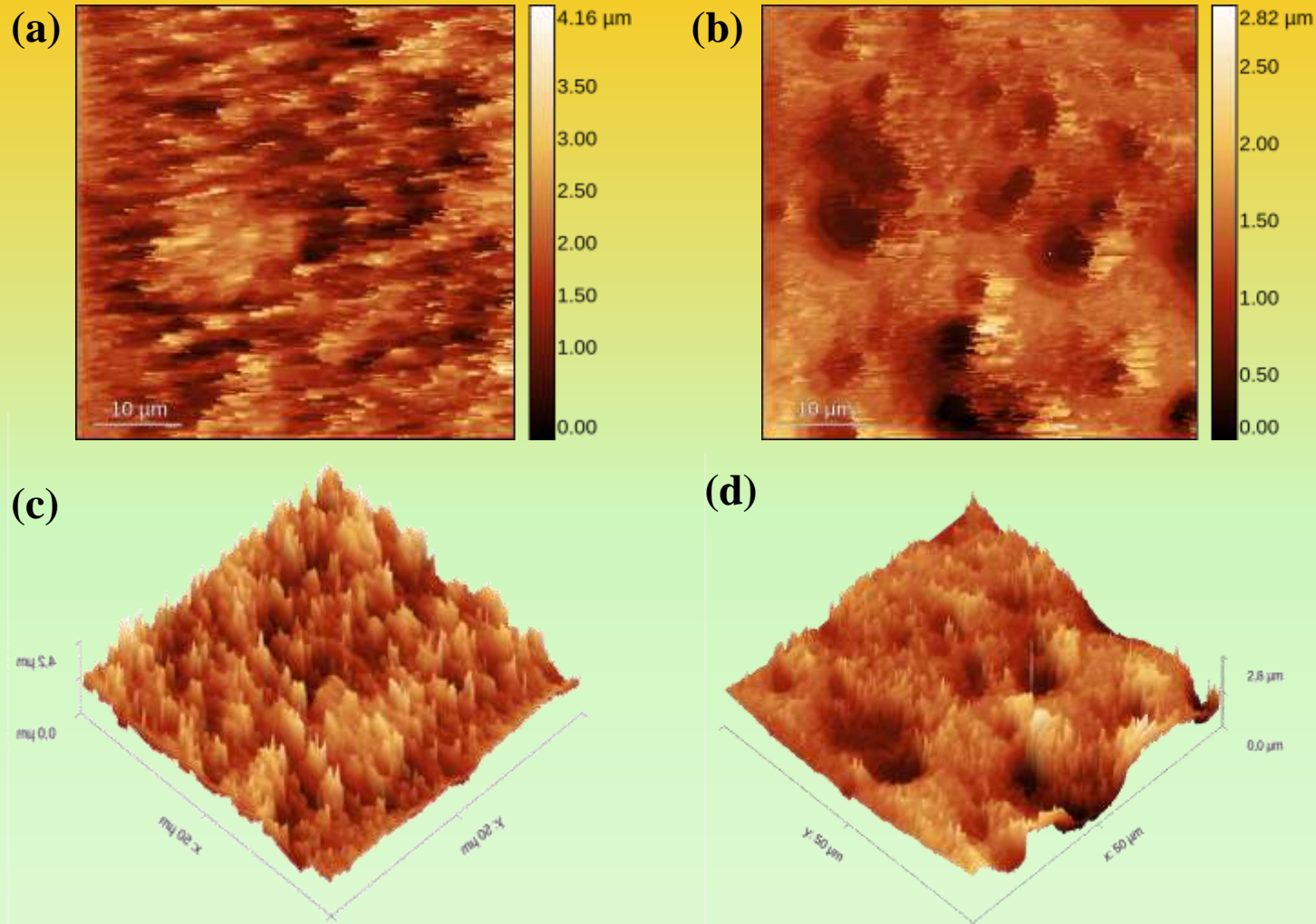
$$[\eta] = \partial \ln \eta_{rel} / \partial C_{CS}$$

[Wolf B.A. // *Macromolecular Rapid Communications*, 2007. Vol.28. No.2. P.164-170]

Effective radius of a macromolecular coil

$$R_{ef} = \sqrt[3]{\frac{3[\eta] \bar{M}_\eta}{10\pi N_a}}$$

AFM of the surface of hydrogel plates



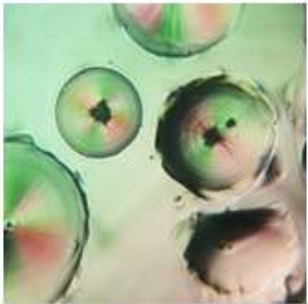
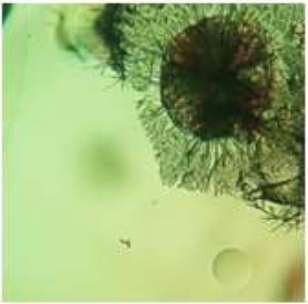
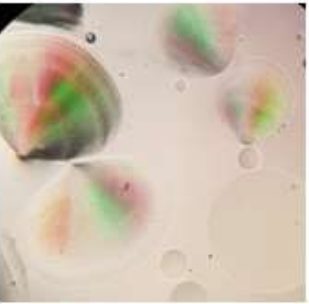
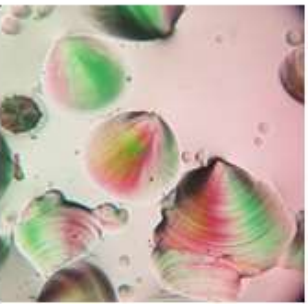
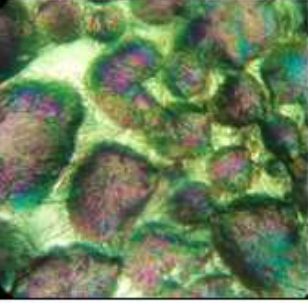
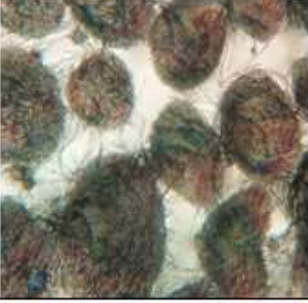
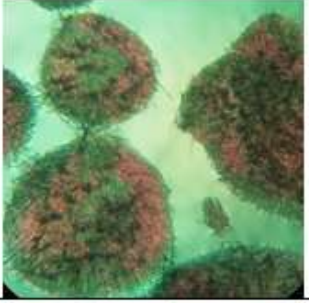


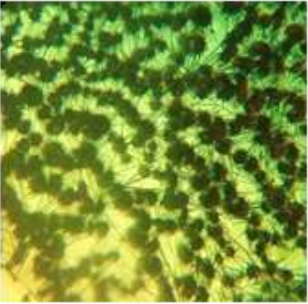
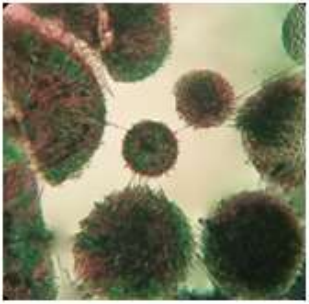

AFM images of the surface of hydrogel plates made of CS•L-AspA (a, c) and CS•D-AspA(b, d) obtained at the molar ratio of $[\text{AspA}]/[\sim\text{-NH}_2] = 0.85$ mol/monomol.

Multi-angle dynamic light scattering

Parameter	$C_{CS} \cdot 10^3$, monomol/dL	Temperature, °C					
		25			37		
		Molar ratio [AspA]/[~-NH ₂], mol/monomol					
		0.85	1.0	2.0	0.85	1.0	2.0
CS·L-AspA							
ζ-potential, mV	0.9	34.2	39.3	40.5	37.1	38.2	38.9
	2.4	39.3	39.3	35.6	33.3	37.9	35.9
	4.7	37.0	39.6	–	29.2	35.4	–
Wall ζ-potential, mV	0.9	51.0	51.8	41.5	52.2	53.6	49.5
	2.4	44.9	38.9	42.5	38.6	41.5	48.6
	4.7	37.2	29.6	–	28.6	37.3	–
Conductivity, mS/cm	0.9	0.40	0.39	0.51	0.47	0.46	0.61
	2.4	0.71	0.67	0.84	0.85	0.80	1.01
	4.7	1.30	1.24	–	1.56	1.50	–
CS·D-AspA							
ζ-potential, mV	0.9	36.5	32.3	33.7	38.0	31.0	36.2
	2.4	37.9	39.4	42.7	37.6	36.4	37.6
	4.7	42.3	33.3	–	34.8	35.5	–
Wall ζ-potential, mV	0.9	36.2	52.6	59.5	45.5	46.7	47.9
	2.4	37.3	40.2	36.0	39.0	39.0	42.9
	4.7	27.9	26.3	–	21.9	29.9	–
Conductivity, mS/cm	0.9	0.44	0.39	0.51	0.52	0.46	0.61
	2.4	0.83	0.76	0.85	0.99	0.76	1.02
	4.7	1.68	1.19	–	2.02	1.44	–

Polarizing microscopy

Kinetics of crystallization of *L*-menthol in aqueous solutions of CS·*L*-AspA and CS·*D*-AspA

$C_{CS} \cdot 10^3, \text{monomol/dL}^*$	CS· <i>L</i> -AspA		CS· <i>D</i> -AspA	
	Extractive crystallization time, min			
	5	30	5	30
1.8				
3.5				
4.7				

* $[\text{AspA}]/[\sim\text{-NH}_2] = 0.85 \text{ mol/monomol}$

Polarization microscope LaboPol-2 (RF), magnification $\times 10$

Complex formation with metal ions

System	Multiplicity of breeding, times	pH	Complex formation ***, %
<i>L</i> -AspA*	10	5.5	47.9±0.7
	100	6.4	21.4±1.1
	1000	6.7	12.4±0.9
<i>D</i> -AspA*	10	5.6	39.0±1.2
	100	6.5	10.1±0.8
	1000	6.7	6.40±0.7
CS· <i>L</i> -AspA**	10	5.9	89.6±1.4
	100	6.8	62.6±1.3
	1000	6.9	37.4±2.1
CS· <i>D</i> -AspA**	10	5.7	97.6±1.2
	100	6.6	92.2±0.9
	1000	6.8	79.8±1.7

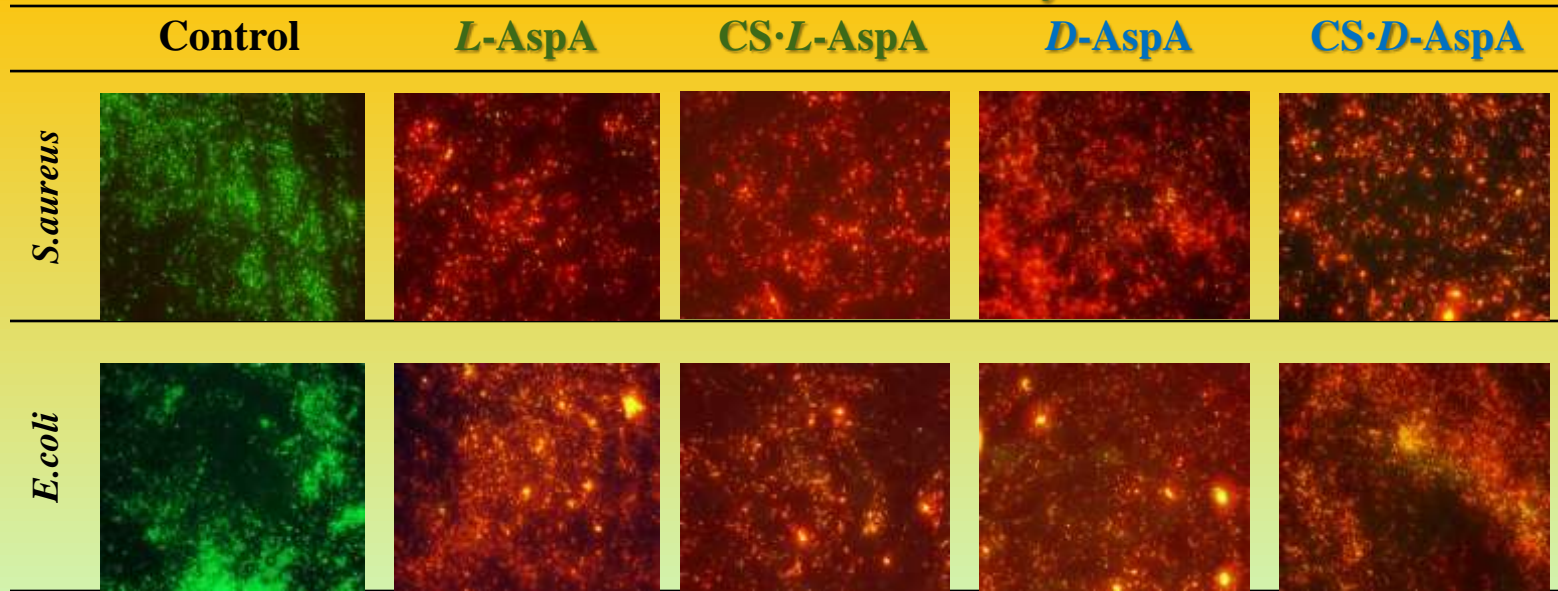
* $C_{\text{AspA}} = 0.8 \cdot 10^{-3} \text{ mol/dL}$

** $C_{\text{CS}} = 0.8 \cdot 10^{-3} \text{ monomol/dL}$, $[\text{AspA}]/[\sim\text{-NH}_2] = 1.0 \text{ mol/monomol}$

*** Complex formation was assessed by the change in the chlorophyll fluorescence intensity of *S. quadricauda* microalgae cultivated in Prat's medium, $t = 4$ days.

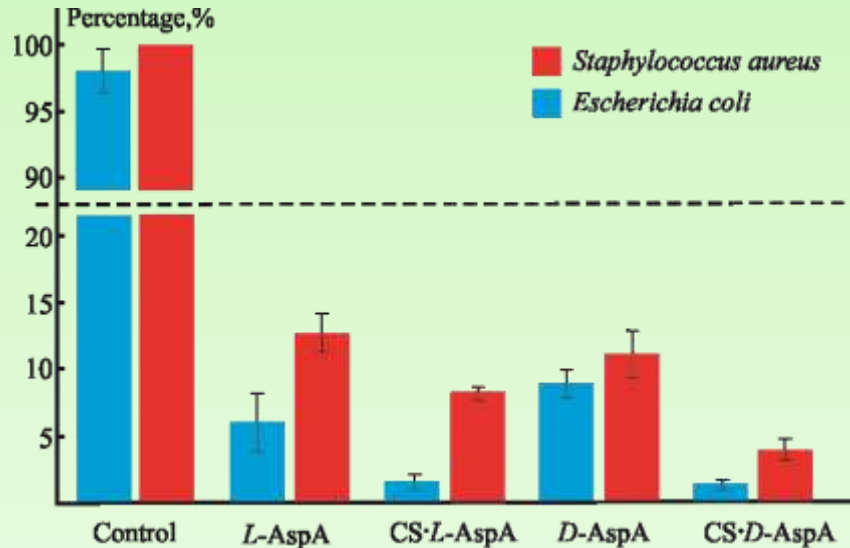
Reagent	Concentration, g/dm ³
Nutrient medium Prat	
Potassium nitrate, KNO ₃	0.1
Potassium phosphate disubstituted, K ₂ HPO ₄ ·3H ₂ O	0.01
Magnesium sulfate, Mg ₂ SO ₄ ·7H ₂ O	0.01
Iron chloride, FeCl ₃ ·6H ₂ O	0.001
Bioobject	
<i>S. quadricauda</i>	30 thousand cells/cm ³

Antibacterial activity



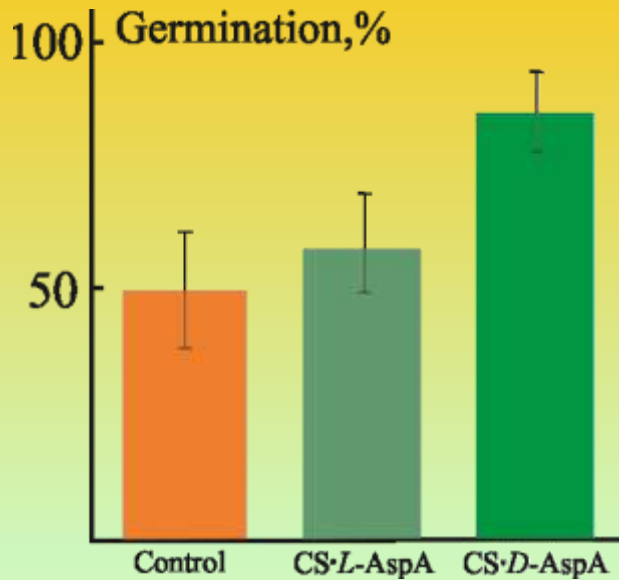
Fluorescent microscopy of cultures of *S. Aureus* 209 P and *E. Coli* 113-13 cultures in meat-peptone broth without (control) and with the addition of aqueous solutions of *L*-AspA, *D*-AspA, CS·*L*-AspA and CS·*D*-AspA; the molar ratio [AspA]/[~NH₂] = 0.85 mol/monomol; 37°C, *t* = 24 hours.

Green fluorescence - viable cells; red fluorescence - dead cells. Magnification: x 400.



Number of living cells in cultures of *S. aureus* 209 P and *E. coli* 113-13 cultures after 24 hours of growth in meat-peptone broth without (control) and with the addition of aqueous solutions of *L*-AspA, *D*-AspA, CS·*L*-AspA and CS·*D*-AspA; the molar ratio [AspA]/[~NH₂] = 0.85 mol/monomol; 37°C.

Biostimulatory activity (laboratory conditions)



Germination of test seeds of *Linum usitatissimum* in laboratory conditions on aqueous extract from the initial soil-soil (control) and from the soil-soil after biodegradation ($t = 40$ days) of hydrogel plates made from CS·L-AspA and CS·D-AspA obtained with the molar ratio $[\text{AspA}]/[\sim\text{-NH}_2] = 1.0$ mol/monomol.

Photo of germinated test-seed of *Linum usitatissimum*



CS·L-AspA

H₂O

CS·D-AspA

Biostimulatory activity (field conditions)

Sample test plants	Irrigation liquid	Time, days	Germination (field), %	Growth stimulating activity	
				Shoot height, mm	Mass of shoots, kg
Radish <i>Raphanus sativus</i>	H ₂ O	6	64±2	2.5-3.0	–
		10	76±4	3.8-4.2	–
		25	–	95-100	1.03±0.04
	CS·L-Asp +H ₂ O*	6	90±3	3.5-4.0	–
		10	93±2	6.0-8.0	–
		25	–	105-115	1.36±0.03
	CS·D-Asp+H ₂ O*	6	92±2	3.4-4.5	–
		10	96±2	6.5-8.3	–
		25	–	110-120	1.42±0.06
Mustard <i>Sinapis alba</i>	H ₂ O	6	20±5	4.5-5.0	–
		10	75±5	7.5-8.5	–
		25	–	130-140	0.61±0.03
	CS·L-Asp+H ₂ O*	6	40±4	5.0-7.0	–
		10	85±3	10.0-10.5	–
		25	–	150-170	0.82±0.02
	CS·D-Asp+H ₂ O*	6	60±3	6.0-8.0	–
		10	90±1	11.5-12.5	–
		25	–	170-190	0.85±0.01



* $C_{CS} = 0.2 \cdot 10^{-3}$ monomol/dL,
 $[AspA] / [\sim\text{-NH}_2] = 1.0$ mol/monomol

Photo of a bed of radish (*Raphanus sativus*) and mustard (*Sinapis alba*) 25 days after planting in the field conditions (open ground), watered and sprinkled with a solution of CS·L-AscA and CS·D-AscA with $C_{CS} = 0.2 \cdot 10^{-3}$ monomol/dL and the molar ratio composition $[AspA] / [\sim\text{-NH}_2] = 1.0$ mol/monomol.

Raphanus sativus

Sinapis alba

Biostimulatory activity (field conditions)

Radish *Raphanus sativus*



H₂O

CS·L-AspA

CS·D-AspA

Mustard *Sinapis alba*



H₂O

CS·L-AspA

CS·D-AspA

Photos of shoots of radish (*Raphanus sativus*) and mustard (*Sinapis alba*), cut off 25 days after planting in open ground, watered and sprinkled with a solution of CS·L-AscA and CS·D-AscA with $C_{CS} = 0.2 \cdot 10^{-3}$ monomol/dL and the molar ratio $[AspA]/[~-NH_2] = 1.0$ mol/monomol.

Thank you for your attention!

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