



Proceeding Paper

Functionalization of Graphene Oxide for Label-Free Electro-Chemical Detection of Hepatic Cancer Cells[†]

Samar Damiati 1,*, Shakil A. Awan 2, Martin Peacock 3 and Bernhard Schuster 4

- ¹ Department of Chemistry, College of Sciences, University of Sharjah, Sharjah, UAE
- ² Wolfson Nanomaterials and Devices Laboratory, School of Engineering, Computing and Mathematics, Faculty of Science and Engineering, University of Plymouth, Plymouth, UK
- ³ Zimmer and Peacock Ltd., Royston SG8 9JL, UK
- ⁴Institute for Synthetic Bioarchitecture, Department of Bionanosciences, University of Natural Resources and Life Sciences, 1190 Vienna, Austria
- * Correspondence: sdamiati@sharjah.ac.ae
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Abstract: In this study, a graphene oxide modified screen-printed carbon electrode was functionalized with 1-pyrenebutyric acid-N-hydroxy-succinimide ester and conjugated with antibodies for the label-free detection of human hepatoma HepG2 cells. Using a user-friendly reservoir chamber, the functionalized film was exposed continuously to the cancer cells. The continuous flow is intended to enhance capture of the target cells onto the sensing platform. The response of the biosensor was evaluated using cyclic voltammetry. The preliminary data showed good sensitivity for detection of hepatic cancer cells. The developed biosensor could detect the HepG2 cells from 1×10^3 to 3×10^5 cells/mL range. Thus, it provides a simple tool for electrochemical detection of cancer cells and offers low-cost and disposable sensing platform.

Keywords: graphene oxide; electrochemical biosensors; cancer cells; PANHS; label-free detection

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1. Introduction

Electrochemical biosensors are popular tools in the field of analytical chemistry due to their high sensitivity and selectivity to detect analytes of interest. These biosensors are highly effective for different applications such as medical, pharmaceutical, environmental, and food sectors because they use the principles of electrochemistry to convert biological interactions into quantifiable signals [1]. Cyclic voltammetry (CV) is one of the most used techniques for electrochemical biosensors. CV measures the current and voltage by reduction and oxidation reactions by applying a voltage sweep to a working electrode surface and monitoring the current response as a function of potential. The resulting signal provides information about the electrochemical behavior of the sensor, including the redox potential, electron transfer kinetics, and the concentration of different analytes such as small molecules, proteins, and nucleic acids [1,2]. Cyclic voltammetry is highly versatile and can be adapted to a variety of electrode materials, including metals, carbon, and conducting polymers. The sensitivity and selectivity of fabricated biosensor can be enhanced using electrode modifications, such as the addition of nanoparticles, linkers, antibodies, enzymes or other biomolecules such as surface-layer proteins [3–5].

Graphene has attracted increasing attention in recent years. It is a 2D sheet of carbon atoms that are arranged hexagonally and exposed on its surface [2,6]. Graphene, graphene oxide (GO) and reduced graphene oxide (rGO) are biocompatible carbon-based materials that lately used in bioelectrochemical systems. These materials have been employed in many biosensing platforms due to their chemical stability, high electronic conductivity,

large specific area, and high carrier mobility [7]. GO is synthesized by chemically oxidizing graphite and then can be immobilized on solid support and functionalized with capturing molecules such as antibodies and nucleic acid probe [8]. To improve the selectivity and sensitivity of fabricated biosensors the immobilization of antibodies in the correct orientation on the electrode surface is critical step [5,8]. Therefore, linker molecules can be used as a scaffolding molecule that offer an appropriate base for immobilizing antibodies [9]. In this study, 1-pyrenebutyric acid-N-hydroxy-succinimide ester (PANHS) was immobilized on GO electrode through π -stacking, while succinimidyl ester group binds to the amine groups of an antibody. In the proposed label-free electrochemical-based biosensor, screen-printed carbon electrodes modified with GO was functionalized with the scaffolding molecule, PANHS, and capturing molecule, anti- OV-6 antibody to detect intact hepatic cancer cells (HepG2) (Figure 1). Further, a reservoir chamber module was used to enable continuous exposure of cancer cells to the functionalized surface and thus increase capture of target cells in the specimen. By optimizing the detection conditions and assessing the generated signals from CV measurements, promising biosensing platforms are developed.

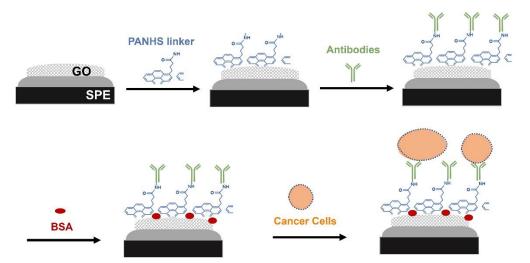


Figure 1. Schematic representation of the developed electrochemical biosensor for detection of cancer cells in the HepG2 cell line.

2. Materials and Methods

2.1. Fabrication of the Sensing Platform

A three-electrode system (Zimmer & Peacock Ltd., UK) consists of a counter electrode (CE, carbon), reference electrode (RE, Ag/AgCl), and working electrode (WE, diameter of carbon electrodes is 4 mm, and functionalized with GO) was used to perform the electrochemical measurements. Initially, 2 mM PANHS (Sigma Aldrich, UK) in methanol were applied to cover the surface of the GO electrode. After incubation for at least 4 hours at 4 °C, 50 μ g/mL of anti- OV-6 antibody (R&D Systems, UK) in PBS was dropped onto the modified surface and incubated at 4 °C overnight, followed by blocking with 1% BSA in PBS for 1 hour to minimize unspecific adsorption on the activated surface. A rinsing step with PBS was performed after each step.

2.2. Electrochemical Measurements

CV measurements were recorded for each functionalized layer of the fabricated sensor and after cell injection in a solution of 5 mM $[Fe(CN_6)]^{3-}$ and 5 mM $[Fe(CN_6)]^{4-}$ (1:1), containing 100 mM KCl and cyclic voltammograms were recorded from -0.6 to 0.6 V at scan rate of 0.05 V/s. Indeed, the developed electrodes were embedded into a reservoir

chamber module that allowed HepG2 cell injection at different numbers and continuous exposure to the developed sensor.

2.3. Cell Lines and Cell Culture

Human liver hepatocellular carcinoma cell line (HepG2) (American Type Culture Collection) was cultured in DMEM supplemented with 10% fetal bovine serum (FBS), and a 1% antibiotic/antimycotic solution at 37 $^{\circ}$ C in 5% CO₂ and 95% air humidified atmosphere. After 48 h, the cells were detached using Trypsin and separated from the medium via centrifugation.

3. Results and Discussion

This study aims to design a simple approach to detect intact cells by using electrochemistry technique. The fabricated electrochemical sensor was characterized using CV measurements and all measurements were conducted in 5 mM ([Fe(CN)₆]^{3-/4-}) redox couple and 100 mM KCl. CV is a powerful electrochemical method that is commonly used to monitor the changes of modified sensor surfaces. To confirm the functionalization of the developed sensor on the GO electrode, CV measurements were measured after each electrode modification step as follow: bare GO, GO/PANHS, GO/PANHS/Ab, and GO/PANHS/Ab/BSA electrodes. The current response decreased noticeably after each addition step onto the electrode surface (Figure 2).

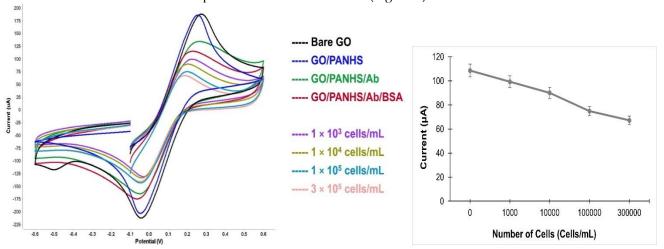


Figure 2. Cyclic voltammetry spectra of the developed GO/PANHS/Ab/BSA sensor with different number of hepatic cancer cells (Left). Peak current values as a function of number of cells that added onto the developed sensor. (Right).

The bare GO surface showed the highest peak current due to a small resistance in the electron transfer. The noticeable change in the peak current for GO/PANHS compared to that of the bare GO electrode shows the formation of an additional layer. The further reduction in the CV current peaks for GO/PANHS/Ab and GO/PANHS/Ab/BSA indicate the immobilization of the antibodies and the steric hindrance of BSA molecules in electron transfer, respectively, which attributed to the decrease in electron transfer rate reflects successful fabrication of the sensing platform. Subsequently, a suspension of HepG2 cells was added to the fabricated sensor at different cell densities, and CV measurements were conducted in presence of cancer cells. The ability of the developed biosensor to capture cancer cells due to the specific antigen-antibody interaction was verified by the CV spectra. When cell suspension was added to the functionalized electrode, the peak current decreased relative to the initial value measured in the absence of cells. The cyclic voltammograms of blank test (no cells) and those with cells were presented in Figure 2. The peak current consistently decreased as the number of captured cells increased. HepG2 cancer cell line

can be identified by several cell surface markers such as OV6. Further, using the chamber reservoir offers continuous flow system that improves the capturing efficiency by allowing continuous exposure of the cells to the functionalized sensor. Cell injection resulted in generation of well-defined cyclic voltammograms at different cell numbers. The voltammograms show peak shifts corresponding to increasing cell numbers because captured cells blocked direct access of the ions to the electrode surface.

4. Conclusions

This work offers simple platform for diagnostic purposes that enables the use of disposable screen-printed electrodes and chamber reservoir for continuous exposure of the cancer cells to the fabricated electrode. The developed GO-modified electrochemical sensor was efficient in capturing hepatic cells via recognition of the OV6 marker that highly expressed on the cancer cell surface. The obtained results present a potential low-cost, effective, and reliable biosensor for cell detection. This electrochemical platform could be further modified with other biomarkers to detect other types of cancer and may be useful for point-of-care diagnostics.

Conflicts of Interest: The authors declare no conflict of interest.

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