

Short Proceeding paper

# Determination of critical storage conditions for spray-dried habanero pepper (*Capsicum chinense*) extracts by coupling water adsorption isotherms and glass transition temperature

Ubaldo Richard Marín Castro<sup>1\*</sup>, Fernando Cansino Jácome<sup>2</sup>, J. Arturo Olguín-Rojas<sup>3\*</sup>, Guadalupe del Carmen Rodríguez-Jimenes<sup>2</sup>, María Teresa González Arnao<sup>1</sup>, Enrique Flores Andrade<sup>1</sup>, Martha Paola Rascón Díaz<sup>4</sup>

<sup>1</sup> Facultad de Ciencias Químicas, Universidad Veracruzana, Zona Universitaria, C.P. 91090, Xalapa, Veracruz, México; ibtrichard.castro@gmail.com

<sup>2</sup> Tecnológico Nacional de México/Instituto Tecnológico de Veracruz/Unidad de Investigación y Desarrollo en Alimentos (UNIDA). M.A. de Quevedo 2779, Col. Formando Hogar, Veracruz, Ver. C.P. 91860, México. guadalupe.rj@veracruz.tecnm.mx

<sup>3</sup> Ingeniería en Procesos Bioalimentarios, Universidad Tecnológica de Tecamachalco. Avenida, Universidad Tecnológica 1, 75483 Tecamachalco, Puebla, México j.a.olguin.rojas@personal.uttecama.edu.mx

<sup>4</sup> Centro de Investigación y Desarrollo en Alimentos (CIDEA), Universidad Veracruzana, Dr. Luis, Dr. Castellazo Ayala s/n, Col. Industrial Ánimas, 91190 Xalapa-Enríquez, Veracruz, México; mrascon@uv.mx

\* Correspondence: ibtrichard.castro@gmail.com, j.a.olguin.rojas@personal.uttecama.edu.mx

**Abstract:** This study aimed to determinate storage conditions for microparticles containing habanero pepper extracts with maltodextrin (MD) and a 95:5 w/w mixture with precipitated silica (MDSP) as wall materials. State diagrams (SD) using water adsorption isotherms and glass transition temperatures were created. Monolayer values were 6.17 g (MD) and 6.76 g (MDSP) of water/100 g d.s. Critical water activity values ( $a_wC$ ) were 0.49 for MD and 0.41 for MDSP. When stored at  $a_w > a_wC$ , both samples underwent physical transformations, with significant color change ( $\Delta E > 8$ ). Conversely, storage below  $a_wC$  resulted in minimal changes ( $\Delta E < 4$ ), consistent with the SD.

**Keywords:** microencapsulation, physical stability, critical water activity

## 1. Introduction

The ethanolic extract of habanero peppers contains two main groups of bioactive compounds: carotenoids and capsaicinoids, which are responsible for the characteristic color and pungency, respectively [1,2]. However, carotenoids are highly sensitive to heat, light, and oxidation due to their polymeric structure. To preserve and recover these bioactive compounds effectively, encapsulation processes offer a promising solution. Microencapsulation involves creating easily manageable particles with a protective polymeric coating, effectively shielding bioactive compounds from environmental factors [3]. This encapsulation technique enables precise dosing of the active agent and has widespread applications in various industries. In pharmaceuticals, it is used for controlled drug release, while in the food industry, it is employed to manage sensory attributes like taste, color, aroma, and texture. Moreover, it allows for the incorporation of health-beneficial compounds [4,5]. Powders formed through spray drying should be able to be stored for extended periods without compromising their stability. However, structural changes in microparticles, such as stickiness, agglomeration, and caking, can occur when stored under conditions exceeding their critical storage parameters [6,7]. Understanding the water adsorption characteristics is crucial for predicting shelf life and determining the critical moisture content and water activity required for product acceptability, especially for products prone to deterioration due to increased humidity. Additionally, it plays a

**Citation:** To be added by editorial staff during production.

Academic Editor: Firstname Last-name

Published: date



**Copyright:** © 2023 by the authors. Submitted for possible open access publication under the terms and conditions of the Creative Commons Attribution (CC BY) license (<https://creativecommons.org/licenses/by/4.0/>).

significant role in drying, packaging, and storage processes[8]. A commonly used methodology to assess the stability of dehydrated foods is moisture adsorption isotherms, providing valuable information about the sorption phenomenon and aiding in stability predictions[7]. Recently, the concept of water activity has been linked to the glass transition. The glass transition temperature ( $T_g$ ) serves as a reference parameter for characterizing the properties, quality, and stability of food systems, offering an integrated perspective on the role of water in foods[9]. Therefore, the objective of this research was to determine the optimal storage conditions for microparticles containing habanero pepper ethanolic extracts, using two different wall materials: maltodextrin and a mixture with precipitated silica (95:5 w/w). The study also aimed to assess the impact of storage conditions on the surface color of the microparticles.

## 2. Materials and Methods

### 2.1. Microparticles of habanero pepper ethanolic extract

Microparticles from red habanero pepper ethanolic extract were obtained using a spray dryer equipped with a heat pump and a dehumidifier (Büchi, Mod. B-290, Flawil, Switzerland). The system operated at an inlet temperature of 140°C and an outlet temperature of 60°C, with nitrogen utilized as the drying gas. **The ethanolic extract was derived from red habanero peppers through maceration at 50°C (20 g of chili pepper with 100g of 70% w/w ethanol as the solvent). This extract was directly mixed with maltodextrin DE10 (MD) at a 4:1 ratio. Additionally, a mixture of the extract with precipitated silica (95:5) (MDSP) was used as supporting materials. The resulting microparticles were stored under vacuum in laminated bags at -20°C for subsequent evaluations.**

### 2.2. Water vapor adsorption isotherms

The microparticles of habanero extract with MD and MDSP were placed in vacuum desiccators containing **20 g of phosphorous pentoxide (P<sub>2</sub>O<sub>5</sub>)** for 20 days at room temperature. Moisture adsorption was determined by equilibrium moisture content at several water activities were determined by the static gravimetric method at 35 °C. Eight saturated salt solutions were prepared (LiCl, CH<sub>3</sub>COOK, MgCl<sub>2</sub>, K<sub>2</sub>CO<sub>3</sub>, Mg(NO<sub>3</sub>)<sub>2</sub>, KI, NaCl and KCl) [10]. For data analysis, three models were applied to assess water adsorption: GAB, OSWIN and LEWICKI (Eq. 1 – 3, respectively) [11–13].

$$M = \frac{M_0 C_{GAB} K_{GAB} a_w}{(1 - K_{GAB} a_w)(1 - K_{GAB} a_w + C_{GAB} K_{GAB} a_w)} \quad \text{Eq. 1}$$

$$M = A \left[ \frac{a_w}{1 - a_w} \right]^B \quad \text{Eq. 2}$$

$$M = A \left( \frac{1}{a_w} - 1 \right)^{B-1} \quad \text{Eq. 3}$$

Where:  $a_w$  is the water activity;  $M$  is the moisture content of the sample on a dry basis (g of water/100 g dry weight);  $M_0$  is the monolayer moisture content (g of water/100 g dry weight);  $C_{GAB}$  and  $K_{GAB}$  are constants related to temperature effect; and  $A$  and  $B$  are constants specific to the model. Isotherm modeling and graph construction were carried out using Kaleida Graph 4.0 software. Goodness of fit of the data was assessed using the relative mean deviation modules,  $E\%$ , according to Eq. 4 [14].

$$E\% = \frac{100}{n} \sum_{i=1}^n \frac{|M_i - M_{pi}|}{M_i} \quad \text{Eq. 4}$$

Where:  $M_i$  is the experimental moisture content;  $M_{pi}$  is the model-predicted moisture content, and  $n$  is the number of observations.

### 2.3. Calorimetric analysis

$T_g$  was determined using a Differential Scanning Calorimeter (MDSC Q2000, TA INSTRUMENTS, New Castle, Del., U.S.A.). Samples (5 mg) stored in a  $a_w = 0.3$  were transferred to aluminum pans and hermetically sealed. Initially, the samples were cooled to  $-40\text{ }^\circ\text{C}$ , then an isothermal was performed by 10 min and finally samples were heated at  $5\text{ }^\circ\text{C}/\text{min}$  until reach a temperature of  $120\text{ }^\circ\text{C}$ , using the amplitude of  $1.272\text{ }^\circ\text{C}$  and a period of 60 s.  $T_g$  was determined as the onset point of the step change on the heat flow curve. The experimentally obtained  $T_g$  data were modeled using the Gordon-Taylor equation and water adsorption data. The plasticizing effect of water on the transition was described by the Gordon-Taylor model [15], where was taken as  $-138\text{ }^\circ\text{C}$  (Eq. 5).

$$T_g = \frac{x_1 T_{g1} + K x_2 T_{g2}}{x_1 + K x_2} \quad \text{Eq. 5}$$

where  $T_g$ ,  $T_{g1}$ , and  $T_{g2}$  are the glass transition temperatures of the binary mixture, dry microcapsule, and water ( $-137\text{ }^\circ\text{C}$ ), respectively,  $x_1$  and  $x_2$  are the molar fraction or weight fraction of dry microcapsule and water, respectively, and  $K$  is the arithmetic average of a series of  $K$  values that are obtained by solving the equation for a series of binary systems at different ratios of dry food and water.

### 2.3. Changes in surface color

The color determination was determinate employed a Hunter-Lab colorimeter (Hunter Lab, Reston, USA) the colour of a sample is denoted by the three dimensions,  $L^*$ ,  $a^*$  and  $b^*$ . Total color change ( $\Delta E$ ) was determinate whit Eq. 6, lower  $\Delta E$  value represents better colour retention [16].

$$\Delta E = \sqrt{(L^* - L_0^*)^2 + (a^* - a_0^*)^2 + (b^* - b_0^*)^2} \quad \text{Eq. 6}$$

where  $L^*$  represents the brightness of the color,  $a^*$  is the range in the red (+), and green (-),  $b^*$  is the range in the yellow (+) and blue (-) after 4 weeks of storage.  $L_0^*$ ,  $a_0^*$ , and  $b_0^*$  are the values of microcapsules at time zero.

## 3. Results and discussions

### 3.1. Adsorption isotherms and critical storage conditions

In the Table 1 is presented the parameters of experimental data fitted different models to water sorption isotherms of microparticles of habanero extract with MD and MDSP as wall material. GAB model shown the best fitted (E%: 4.57%) for MD and MDSP, a model is considered acceptable when the value of E% is less than 10% and  $R^2$  is greater than 0.9 [17]. **The constant monolayer ( $M_0$ ) predicted by GAB is an important stability parameter, because at this point a product should be stable against microbial spoilage [9].** The isotherm exhibits a Type II behavior, as per the Brunauer-Emmet-Teller classification [18].

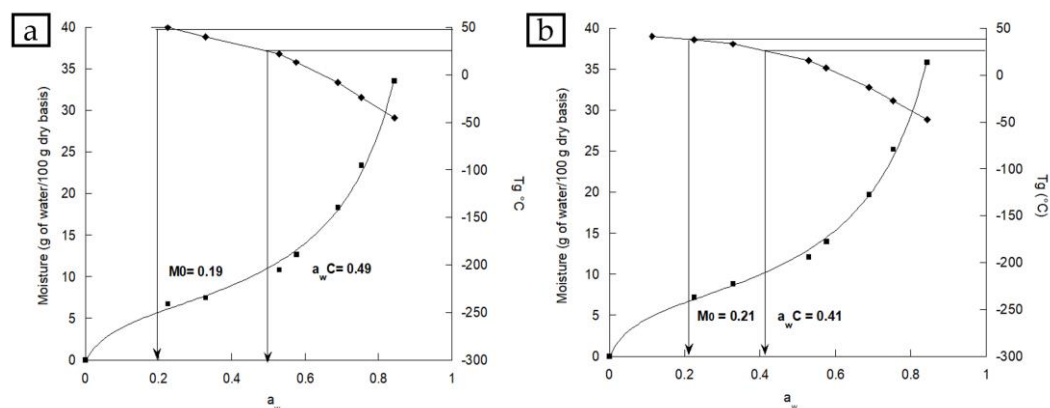
The glass transition temperature of the capsules is dependent on both moisture content and water activity within the food matrix, serving as predictive indicators for stability during storage. The combined influence of temperature and water content serves as a plasticizing agent within food matrices [6]. **The critical water activity ( $a_wC$ ) values signify the point at which a product's glass transition temperature matches room temperature.** When the temperature exceeds this threshold, amorphous powders become vulnerable to detrimental transformations, such as collapse, stickiness, and caking, leading to a degradation in product quality [6,19]. The critical water activity value was determined to be 0.49 for MD (Figure 1).

Similar results were previously reported for paprika powder produced via spray drying with maltodextrin as the encapsulating material, yielding an  $a_wC$  of 0.496 [20], as well as for acai microparticles, with an  $a_wC$  of 0.574 [19]. In contrast, the incorporation of

precipitated silica (5% w/w) leads to a reduction in the glass transition temperature and an augmentation in the monolayer adsorption capacity on the particle surface. Consequently, the critical water activity values decrease from 0.49 to 0.41 for MDSP, indicating reduced stability of the microparticles. This reduction in stability increases the likelihood of the microparticles transitioning into a rubbery state, causing physical transformations in the samples, ultimately resulting in collapse and caking.

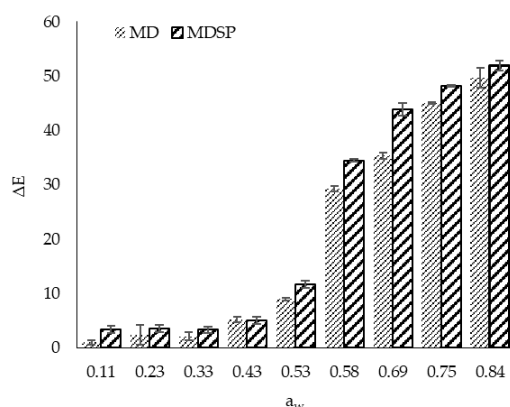
**Table 1.** Estimated parameters from GAB, OSWIN and LEWICKI models for microparticles of habanero extract with MD and MDSP as wall material.

Model	Parameter	MD	MDSP
GAB	$M_0$ (g of H <sub>2</sub> O/100 g) d.s.)	6.17	6.79
	$C_{GAB}$	12.21	14.64
	$K_{GAB}$	0.97	0.96
	R <sup>2</sup>	0.99	0.99
	E%	4.57	3.17
LEWICKI	A	11.07	12.28
	B	0.34	0.36
	R <sup>2</sup>	0.99	0.99
	E%	21.74	7.75
	OSWIN	A	11.07
	B	0.65	0.63
	R <sup>2</sup>	0.99	0.99
	E%	7.86	21.24



**Figure 1.** Variation of glass transition temperature and moisture content with water activity for microparticles of habanero extract with MD (a) and MDSP (b) as wall material.

In Figure 2, total color variation observed during storage at different water activity values, is present. The addition of precipitated silica within the evaluated range (5%, w/w) had no significant effect on color preservation. Minimal color variation ( $\Delta E$ : 1.0 to 5.0) was observed at  $a_w$  levels ranging from 0.11 to 0.43. According to Obon *et al.* [21] when  $\Delta E < 5.0$ , the human eye can only perceive minimal differences. The most significant variation in color retention occurs when particles are stored under conditions exceeding the  $a_wC$  [20].  $\Delta E$  increased with the increasing storage  $a_w$  of microparticles containing habanero pepper extract. This behavior is consistent with reported for paprika powder [22], pumpkin [23] and borojó powder [24]. As mentioned earlier, at  $a_w$  values greater than  $a_wC$ , the capsules tend to collapse and cake, leading to the dilution of reactants within the capsule and, consequently, an increase in color change.



**Figure 2.** Variation of  $\Delta E$  for microparticles of habanero extract with MD (a) and MDSP (b) as wall material stored to different  $a_w$  values.

#### 4. Conclusion

The GAB, OSWIN and LEWICKI models accurately describe the adsorption of water onto microparticles containing habanero extract with MD and MDSP as the wall material. **Optimal color retention was achieved when the particles were stored below the critical water activity level (0.49 for MD and 0.41 for MDSP). This data enabled the determination of the critical water activity level for both materials, which was found to be 0.49 for MD and 0.41 for MDSP. Maintaining particles below the critical water activity level ensured optimal color retention. Although, the moisture content corresponding to the monolayer (6.17 and 6.79 g of H<sub>2</sub>O/100 g d.s., for MD and MDSP, respectively) is suggested as a point of maximum stability, to complete the present study, it is essential to evaluate the occurrence of chemical reactions, such as the degradation of capsaicinoids, during storage.**

**Supplementary Materials:** Not applicable

**Author Contributions:** Conceptualization, U.R.M.C. and J.A.O.R.; methodology, U.R.M.C., MPRD, MTGA, J.A.O.R. and F.C.J.; software, U.R.M.C., MTGA, EFA.; validation, U.R.M.C., MPRD, J.A.O.R., F.C.J. and G.C.R.J.; formal analysis, U.R.M.C., J.A.O.R., EFA; investigation, U.R.M.C. and J.A.O.R. and F.C.J.; resources, U.R.M.C., J.A.O.R. and F.C.J.; data curation, U.R.M.C., J.A.O.R., MTGA; writing—original draft preparation, U.R.M.C. and J.A.O.R.; writing—review and editing, U.R.M.C. and J.A.O.R.; visualization, U.R.M.C., MPRD, J.A.O.R. and G.C.R.J.; supervision, G.C.R.J. and MPRD; project administration, G.C.R.J.; funding acquisition, G.C.R.J. All authors have read and agreed to the published version of the manuscript.

**Funding:** This research received no external funding

**Informed Consent Statement:** Not applicable

**Data Availability Statement:** Not applicable

**Conflicts of Interest:** The authors declare no conflict of interest.

#### References

1. Olgún Rojas, J.A.; Vázquez-León, L.A.; Salgado-Cervantes, M.A.; Fernandez-Barbero, G.; Díaz-Pacheco, A.; García-Alvarado, M.A.; Rodríguez-Jimenes, G.C. Water and Phytochemicals Dynamic during Drying of Red Habanero Chili Pepper (*Capsicum chinense*) Slices. *Rev Mex Ing Quim* 2019, 18, 851–864.
2. Fabela-Morón, M.F.; Cuevas-Bernardino, J.C.; Ayora-Talavera, T.; Pacheco, N. Trends in Capsaicinoids Extraction from Habanero Chili Pepper (*Capsicum chinense* Jacq.): Recent Advanced Techniques. *Food Reviews International* 2020, 36, 105–134, doi:10.1080/87559129.2019.1630635.
3. Ezhilarasi, P.N.; Karthik, P.; Chhanwal, N.; Anandharamakrishnan, C. Nanoencapsulation Techniques for Food Bioactive Components: A Review. *Food Bioproc Tech* 2013, 6.
4. Rollyson, W.D.; Stover, C.A.; Brown, K.C.; Perry, H.E.; Stevenson, C.D.; McNeese, C.A.; Ball, J.G.; Valentovic, M.A.; Dasgupta, P. Bioavailability of Capsaicin and Its Implications for Drug Delivery. *Journal of Controlled Release* 2014, 196, 96–105.

5. Gharsallaoui, A.; Roudaut, G.; Chambin, O.; Voilley, A.; Saurel, R. Applications of Spray-Drying in Microencapsulation of Food Ingredients: An Overview. *Food research international* 2007, 40, 1107–1121. 1  
2
6. Roos, Y.H. Water Activity and Physical State Effects on Amorphous Food Stability. *J Food Process Preserv* 1993, 16, 433–447. 3
7. Pascual-Pineda, L.A.; Rascón, M.P.; Quintanilla-Carvajal, M.X.; Castillo-Morales, M.; Marín, U.R.; Flores-Andrade, E. Effect of Porous Structure and Spreading Pressure on the Storage Stability of Red Onion Microcapsules Produced by Spray Freezing into Liquid Cryogenic and Spray Drying. *J Food Eng* 2019, 245, 65–72. 4  
5  
6
8. Rahman, M.S.; Labuza, T.P. Water Activity and Food Preservation. In *Handbook of food preservation*; CRC Press, 2007; pp. 465–494. 7  
8
9. Flores-Andrade, E.; Bonilla, E.; Luna-Solano, G.; Marín, U.R.; González-Arno, M.T.; Rascón, M.P. Effect of the Microstructure on the Stability of Red Onion Microcapsules. *Drying Technology* 2019, 37, 223–231. 9  
10
10. Lang, K.W.; McCune, T.D.; Steinberg, M.P. A Proximity Equilibration Cell for Rapid Determination of Sorption Isotherms. *J Food Sci* 1981, 46, 936–938, doi:<https://doi.org/10.1111/j.1365-2621.1981.tb15386.x>. 11  
12
11. Staudt, P.B.; Kechinski, C.P.; Tessaro, I.C.; Marczak, L.D.F.; Soares, R. de P.; Cardozo, N.S.M. A New Method for Predicting Sorption Isotherms at Different Temperatures Using the BET Model. *J Food Eng* 2013, 114, 139–145. 13  
14
12. Oswin, C.R. The Kinetics of Package Life. III. The Isotherm. *Journal of the Society of Chemical Industry* 1946, 65, 419–421. 15
13. Lewicki, P.P. A Three Parameter Equation for Food Moisture Sorption Isotherms. *J Food Process Eng* 1998, 21, 127–144. 16
14. Lomauro, C.J.; Bakshi, A.S.; Labuza, T.P. Evaluation of Food Moisture Sorption Isotherm Equations Part II: Milk, Coffee, Tea, Nuts, Oilseeds, Spices and Starchy Foods. *LWT-Food Science and Technology* 1985, 18, 118–124. 17  
18
15. Gordon, M.; Taylor, J.S. Ideal Copolymers and the Second-order Transitions of Synthetic Rubbers. I. Non-crystalline Copolymers. *Journal of Applied Chemistry* 1952, 2, 493–500. 19  
20
16. Maskan, M. Kinetics of Colour Change of Kiwifruits during Hot Air and Microwave Drying. *J Food Eng* 2001, 48, 169–175. 21
17. Kaymak-Ertekin, F.; Gedik, A. Sorption Isotherms and Isotheric Heat of Sorption for Grapes, Apricots, Apples and Potatoes. *LWT-Food Science and Technology* 2004, 37, 429–438. 22  
23
18. Brunauer, S.; Deming, L.S.; Deming, W.E.; Teller, E. On a Theory of the van Der Waals Adsorption of Gases. *J Am Chem Soc* 1940, 62, 1723–1732. 24  
25
19. Tonon, R. V.; Baroni, A.F.; Brabet, C.; Gibert, O.; Pallet, D.; Hubinger, M.D. Water Sorption and Glass Transition Temperature of Spray Dried Açai (*Euterpe oleracea* Mart.) Juice. *J Food Eng* 2009, 94, 215–221, doi:10.1016/J.JFOODENG.2009.03.009. 26  
27
20. Díaz, D.I.; Lugo, E.; Pascual-Pineda, L.A.; Jiménez-Fernández, M. Encapsulation of Carotenoid-Rich Paprika Oleoresin through Traditional and Nano Spray Drying. *Italian Journal of Food Science* 2019, 31. 28  
29
21. Obón, J.M.; Castellar, M.R.; Alacid, M.; Fernández-López, J.A. Production of a Red–Purple Food Colorant from *Opuntia stricta* Fruits by Spray Drying and Its Application in Food Model Systems. *J Food Eng* 2009, 90, 471–479. 30  
31
22. Shirkole, S.S.; Sutar, P.P. Modeling Sorption Phenomena and Moisture Migration Rates in Paprika (*Capsicum Annuum* L.) Using Physicochemical Characteristics. *J Food Sci Technol* 2018, 55, 678–688. 32  
33
23. Al-Ghamdi, S.; Hong, Y.K.; Qu, Z.; Sablani, S.S. State Diagram, Water Sorption Isotherms and Color Stability of Pumpkin (*Cucurbita Pepo* L.). *J Food Eng* 2020, 273, 109820, doi:10.1016/J.JFOODENG.2019.109820. 34  
35
24. Mosquera, L.H.; Moraga, G.; de Córdoba, P.F.; Martínez-Navarrete, N. Water Content–Water Activity–Glass Transition Temperature Relationships of Spray-Dried Borjón as Related to Changes in Color and Mechanical Properties. *Food Biophys* 2011, 6, 397–406, doi:10.1007/s11483-011-9215-2. 36  
37  
38