



The 9th International Electronic Conference on Medicinal Chemistry (ECMC 2023)

01–30 November 2023 | Online

Optimization and Synthesis of Perfluorocarbon Nanoemulsion with Fluorous Photosensitizer for Photodynamic Therapy

Chaired by **Dr. Alfredo Berzal-Herranz**
and **Prof. Dr. Maria Emília Sousa**



pharmaceuticals



Minh Tuan Nguyen ^{1,*}, Elizaveta Guseva ², Andrey Sigan ², Ivan Burtsev ¹, Anton Egorov ¹, Anna Shibaeva ¹, Vladimir Kuzmin ¹, Alexander Shtil ³ and Alina Markova ¹

¹ Emanuel Institute of Biochemical Physics, Russian Academy of Sciences, 4 Kosygin Street, 119334 Moscow, Russia;

² A.N. Nesmeyanov Institute of Organoelement Compounds, Russian Academy of Sciences, 28 Vavilov Street, 119991 Moscow, Russia;

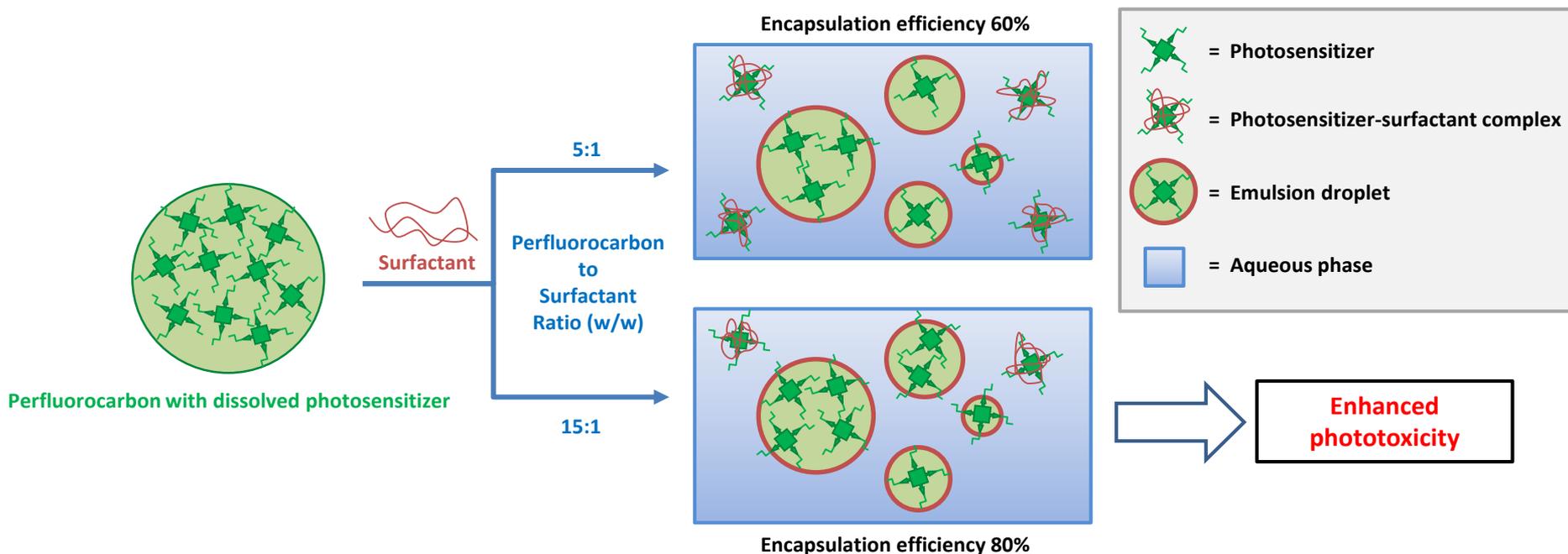
³ Institute of Cyber Intelligence Systems, National Research Nuclear University MEPhI, 31 Kashirskoe Shosse, 115409 Moscow, Russia.

* Corresponding author: tuantonyx@yahoo.com



Optimization and Synthesis of Perfluorocarbon Nanoemulsion with Fluorous Photosensitizer for Photodynamic Therapy

Graphical Abstract





Abstract: Perfluorocarbon-based nanoemulsions (PFC-NEs) have been shown to be an effective tool for oxygen delivery in various therapeutic modalities, including photodynamic therapy (PDT) in cancer treatment. PFC droplets are able to transport and supply oxygen to hypoxic cells as well as enhance the singlet oxygen generation by the photosensitizer (PS), functions which make them a promising platform for PDT. To further enhance PDT efficacy, we synthesized PFC-NEs with fluorous PS that is soluble in the perfluorocarbon phase, creating a closer proximity of the PS to the dissolved oxygen in PFC. However, the fluorous PS used in this work had a high tendency to leak into the aqueous phase with a surfactant, leading to less than desirable encapsulation efficiency (EE%). Therefore, the main goal of this study is to develop a formulation to ensure high EE% of fluorous PS in PFC droplets. PFC-NEs were prepared by ultrasonic emulsification and were characterized using dynamic light scattering, UV-Vis and fluorescence spectroscopy. The optimization of the PFC-NE formulation did not significantly affect the nanoemulsion properties, such as hydrodynamic diameter, polydispersity index and colloidal stability, and successfully increased EE%, as well as improved dark cytotoxicity profile and enhanced photoinduced cytotoxicity of PFC-NE.

Keywords: drug formulation; encapsulation efficiency; nanoemulsion; perfluorocarbon; photodynamic therapy.



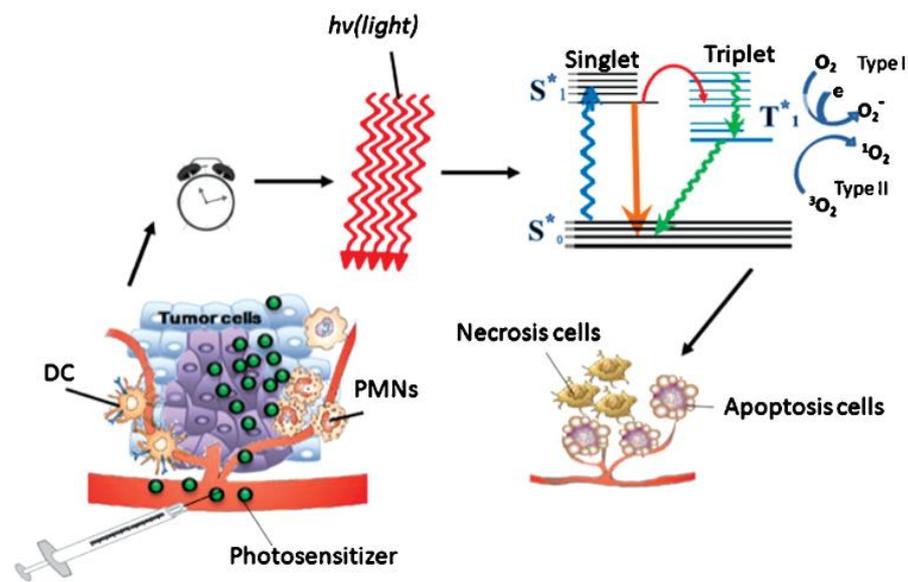
Introduction

Photodynamic therapy (PDT) is a minimally invasive method of cancer treatment

Photodynamic processes are mostly **oxygen-dependent**

Hypoxia leads to tumor progression and resistance to oxygen-dependent therapies (radiotherapy, PDT)

Perfluorocarbons are able to dissolve 20- to 40-fold larger amounts of oxygen compared to water and temporarily alleviate hypoxia

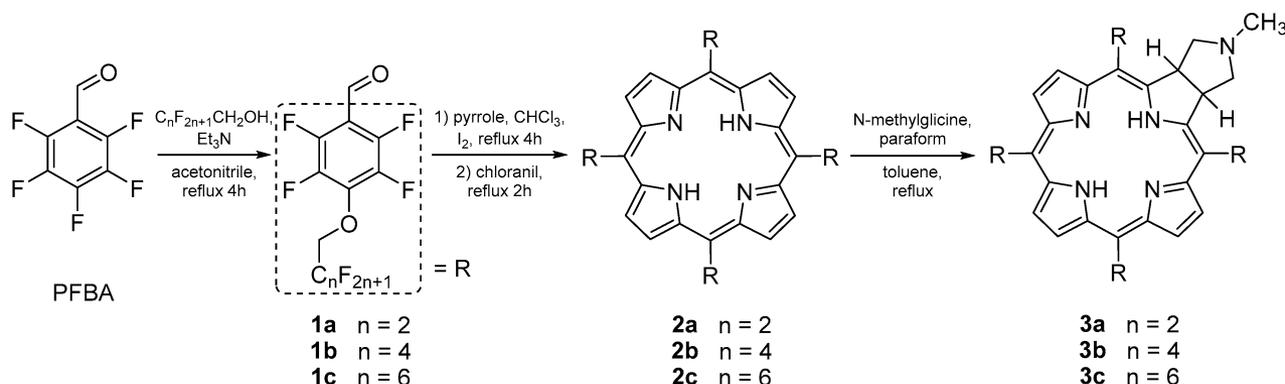


The mechanisms of PDT

Source: Denis, T. G.; Aziz, K.; Waheed, A. A.; Huang, Y.-Y.; Sharma, S. K.; Mroz, P.; Hamblin, M. R. Combination approaches to potentiate immune response after photodynamic therapy for cancer. *Photochemical & Photobiological Sciences* **2011**, *10*, 792–801. <https://doi.org/10.1039/c0pp00326c>



Previous work: Synthesis of Fluorous Photosensitizers and Preparation of Perfluorocarbon Nanoemulsions

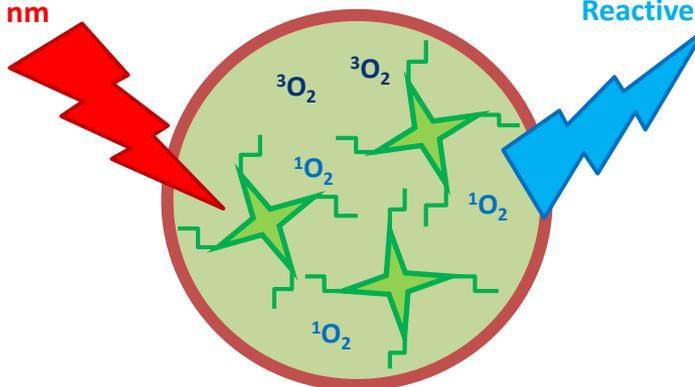


Chlorin-based PSs absorb
red light (600-700 nm)

=

Better and deeper
tissue penetration

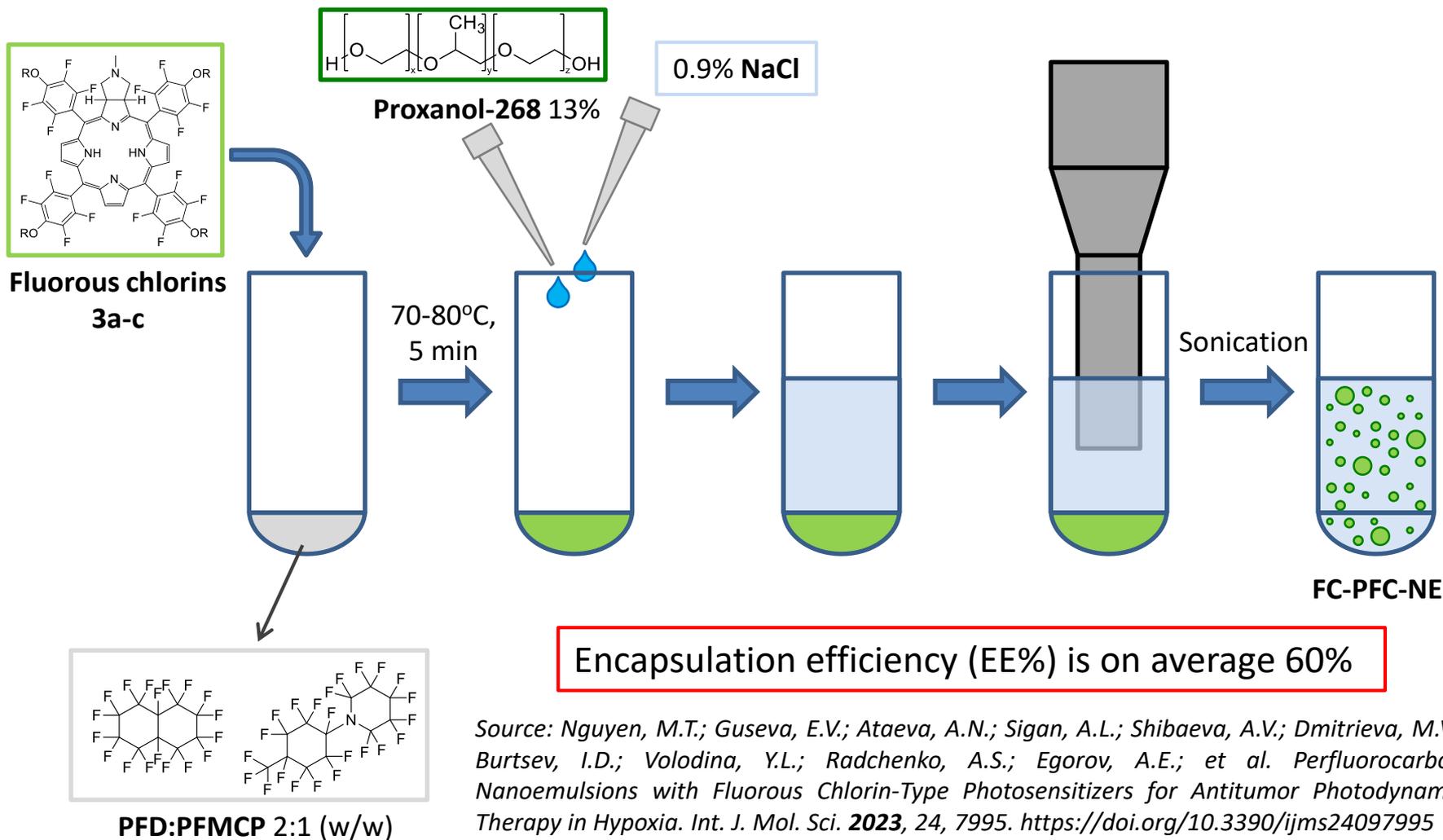
650 nm



Proximity of fluorous PS
to molecular oxygen in PFC

=

Optimized conditions for
singlet oxygen generation



Source: Nguyen, M.T.; Guseva, E.V.; Ataeva, A.N.; Sigan, A.L.; Shibaeva, A.V.; Dmitrieva, M.V.; Burtsev, I.D.; Volodina, Y.L.; Radchenko, A.S.; Egorov, A.E.; et al. Perfluorocarbon Nanoemulsions with Fluororous Chlorin-Type Photosensitizers for Antitumor Photodynamic Therapy in Hypoxia. *Int. J. Mol. Sci.* **2023**, *24*, 7995. <https://doi.org/10.3390/ijms24097995>



Previous work:

| Components | Functions | Components |
|--|--|--|
| Fluorous chlorin 3b (FC-3b) <i>0.6507 mg</i> | Photosensitizer | Fluorous chlorin 3b <i>1.952 mg</i> |
| PFD <i>0.867 g</i> | Oxygen carrier, PS solvent | PFD <i>2.6 g</i> |
| PFMCP <i>0.433 g</i> | Oxygen carrier, PS solvent, droplet stabilizer | PFMCP <i>1.3 g</i> |
| Proxanol-268 <i>0.2665 g</i> | Surfactant | Proxanol-268 <i>0.2665 g</i> |
| 0.9% NaCl solution <i>Up to 10 ml</i> | Dispersion medium, buffer solution | Phosphate Buffer Saline (PBS) <i>Up to 10 ml</i> |

PFC/Proxanol-268 5:1

PFC/Proxanol-268 15:1



Results and discussion

PFC/Proxanol-268 5:1



Translucent, slightly opalescent
Grayish green o/w nanoemulsion

Sonication settings:

Mode: pulsed
Duty cycle: 50%
Power: 5 (50%)
Duration: 4x2 min
Rest: 1-2 min
Ice bath

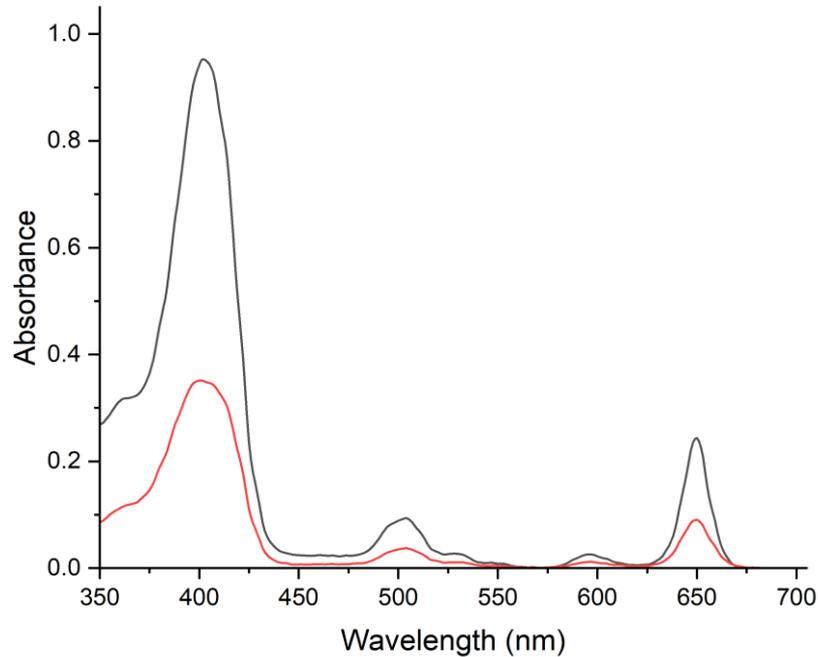
PFC/Proxanol-268 15:1



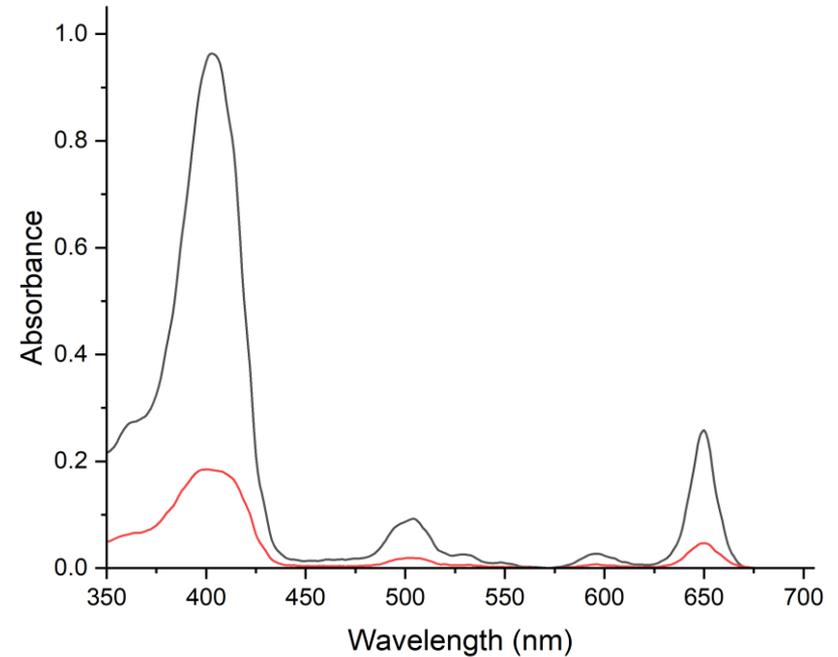
Opaque
Green o/w nanoemulsion



UV-Vis spectra of FC-3b in PFC-NEs and their supernatant



— PFC/Proxanol-268 5:1 — Supernatant



— PFC/Proxanol-268 15:1 — Supernatant

Less absorbance in the supernatant = higher EE%

$$EE\% = (AUC(\text{emulsion}) - AUC(\text{supernatant})) / AUC(\text{emulsion}) \times 100\%$$



Characterization of PFC-NEs

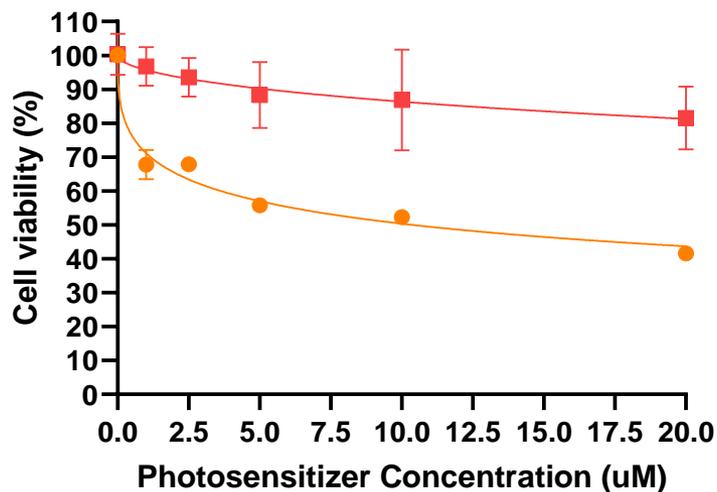
| Emulsion | Storage Temp., °C | z-average hydrodynamic diameter, nm | | | Polydispersity index (PDI) | | | EE% | |
|--------------------------|-------------------|-------------------------------------|-------------|-------------|----------------------------|--------------|--------------|-------|--------|
| | | Day 1 | Day 7 | Day 30 | Day 1 | Day 7 | Day 30 | Day 1 | Day 30 |
| PFC/Proxanol-268 5:1 | +4 | 200.6 ± 6.1 | 208.7 ± 5.6 | 221.1 ± 1.4 | 0.096 ± 0.01 | 0.027 ± 0.02 | 0.060 ± 0.04 | 63.8 | 44.1 |
| | -20 | | 214.9 ± 4.8 | 192.9 ± 2.5 | | 0.120 ± 0.03 | 0.089 ± 0.04 | | 69.6 |
| PFC/Proxanol-268 15:1 | +4 | 214.5 ± 8.5 | 221.5 ± 4.9 | 236.7 ± 0.3 | 0.070 ± 0.02 | 0.068 ± 0.02 | 0.025 ± 0.02 | 81.9 | 70.5 |
| | -20 | | 217.4 ± 5.9 | 206.7 ± 0.7 | | 0.083 ± 0.02 | 0.091 ± 0.02 | | 83.2 |

- Resulting **PFC/Proxanol-268 15:1** formulation is a **nanoemulsion**;
- Changes in **droplet size** and **PDI** are **insignificant**;
- Both formulations are stable during storage;
- **Encapsulation efficiency** has **improved**.



Improved dark and photoinduced cytotoxicity profile

Dark cytotoxicity

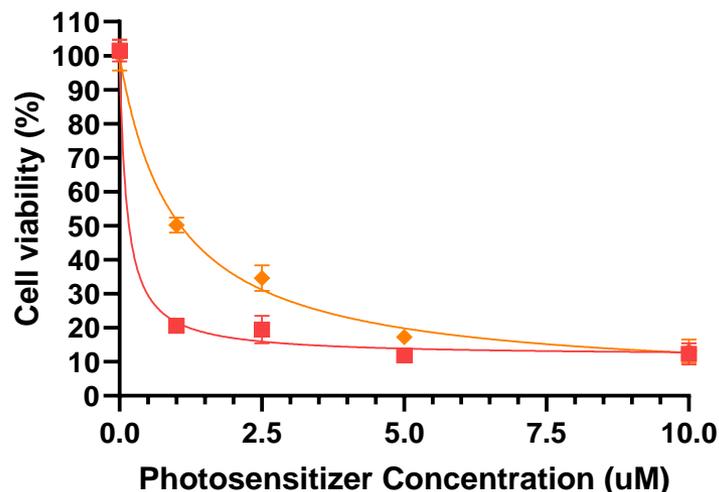


—●— PFC/Proxanol-268 5:1 —■— PFC/Proxanol-268 15:1

↓ Decreased dark cytotoxicity

HCT116 cells, 48h incubation with PFC-NEs

Photoinduced cytotoxicity



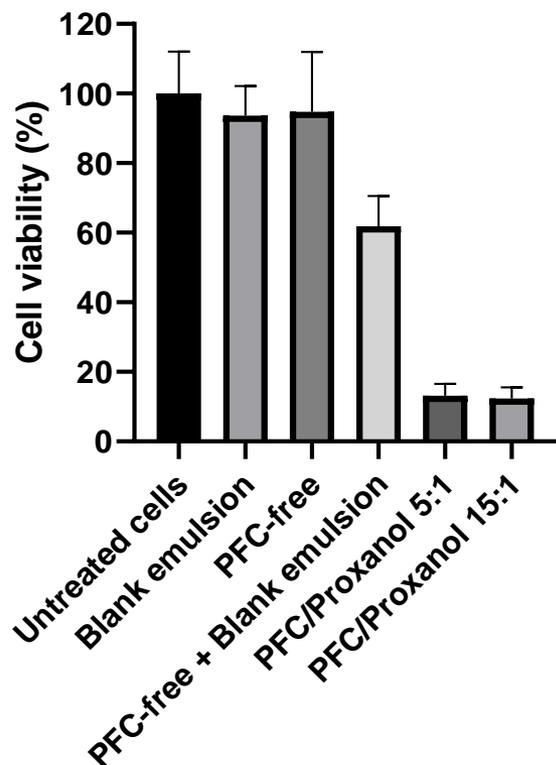
—◆— PFC/Proxanol-268 5:1 —■— PFC/Proxanol-268 15:1

↑ Increased phototoxicity

HCT116 cells, 24h incubation + 24h after irradiation
Irradiation: 660 nm, 8.3 mW/cm², 45 J/cm²



Control experiment



HCT116 cells, 24h incubation + 24h after irradiation

Irradiation: 660 nm, 8.3 mW/cm², 45 J/cm²

FC-3b concentration: 10 μM

Treatment groups:

- 1) Blank emulsion (PFC/Proxanol-268 15:1 without FC-3b) – no cell death
- 2) PFC-free (PFC/Proxanol-268 15:1 without PFC) – no cell death
- 3) PFC-free + Blank emulsion – 40% cell death
- 4) PFC/Proxanol-268 5:1 – 85% cell death
- 5) PFC/Proxanol-268 15:1 – 85% cell death

Encapsulated FC-3b is the main **photodynamically active** component of PFC-NEs



Conclusions

Optimization of PFC-NE formulation resulted in:

- Higher encapsulation efficiency;
- Improved dark cytotoxicity profile;
- Enhanced photoinduced cytotoxicity.

Optimization process did not affect:

- Particle size and polydispersity index – resulting o/w emulsion remained nanoscale;
- Colloidal stability – optimized formulation is stable during storage.

Further research should and will be focused on:

- Optimizing PS concentration in PFC;
- Studying difference in physical and chemical properties of encapsulated and non-encapsulated PS in PFC-NEs;
- Effect of surfactant on photochemistry of PFC-NEs.



Acknowledgments

The study was carried out using the equipment of the Core Facility “New Materials and Technologies” at the Emanuel Institute of Biochemical Physics, Russian Academy of Sciences.