

# Novel Thiophene-Derived Schiff Base as a Fluorescent Sensor for Highly Sensitive and Selective Relay Recognition of Zn<sup>2+</sup> and Fe<sup>2+</sup> ions

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## Abstract:

Chemosensors based on Schiff bases are pivotal in environmental and biological applications, serving to identify specific metal ions at trace levels. Despite the distinctive importance of thiophene-based molecules in medicinal contexts, the number of reported chemosensors utilizing these moieties remains limited.

In this study, we present the synthesis and characterization of a novel Schiff base sensor (TBH), derived from thiophene-2-carboxaldehyde and benzil. We investigate its application as a selective relay probe for the detection of Zn<sup>2+</sup> and Fe<sup>2+</sup> ions.

The introduction of Zn<sup>2+</sup> to TBH, resulted in a significant enhancement of fluorescent intensity, attributed to the formation of a 1:1 TBH–Zn<sup>2+</sup> complex, with no response observed for other cations, including Mg<sup>2+</sup>, Ba<sup>2+</sup>, Cd<sup>2+</sup>, Cu<sup>2+</sup>, Co<sup>2+</sup>, Mn<sup>2+</sup>, Cr<sup>3+</sup>, Hg<sup>2+</sup>, Sn<sup>2+</sup>, La<sup>3+</sup>, Ca<sup>2+</sup>, Na<sup>+</sup>, K<sup>+</sup>, and particularly Fe<sup>2+</sup>. Furthermore, Fe<sup>2+</sup> induced fluorescence quenching in the TBH–Zn<sup>2+</sup> system, forming a 1:1 MY–Fe<sup>2+</sup> complex. The TBH–Zn<sup>2+</sup> solvento-complex demonstrates potential as a secondary sensor for Fe<sup>2+</sup> ions. The sensor's signal change is based on the chelation-enhanced fluorescence (CHEF) effect of TBH–Zn<sup>2+</sup>, coupled with the inhibition of photoinduced electron transfer (PET).

Moreover, the rapid and selective features of the proposed sensor make it promising for the precise monitoring of Zn<sup>2+</sup> and Fe<sup>2+</sup> in biological and environmental research.

**Keywords:** Schiff-base Derivatives, Thiophene, Fluorescent chemosensor, Ion Detection.