

# Synthesis and Characterization of $\text{In}_2\text{O}_3\text{-ZnO}$ Nanostructures via the Precipitation Method

Alina MATEI\*, Oana BRÎNCOVEANU, Cosmin ROMANIȚAN, Vasilica ȚUCUREANU\*

National Institute for Research and Development in Microtechnologies IMT-Bucharest

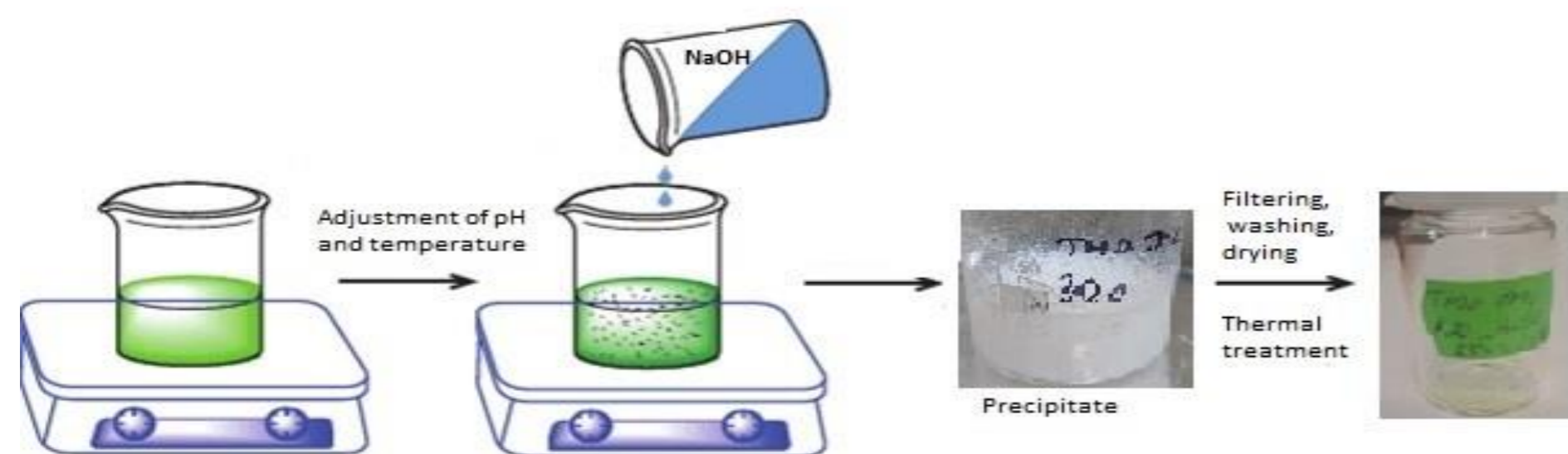
\*Correspondence: [alina.matei@imt.ro](mailto:alina.matei@imt.ro); [vasilica.tucureanu@imt.ro](mailto:vasilica.tucureanu@imt.ro)

## INTRODUCTION & AIM

Nanocomposites, through their synergetic interaction between components, lead to integrated compatible structures, which opens new horizons for their application as materials in the realization of sensors.  $\text{In}_2\text{O}_3\text{-ZnO}$  nanostructures exhibit a large bandgap, high optical transparency, electrical conductivity, uniform surface, chemical and thermal stability in different environments, and excellent photoelectrocatalytic performance, which can be attributed to the enhanced absorption of photons in the visible range and the effective separation of charge carriers at the interface, which makes them interesting for biomedical applications. To obtain these types of materials, the synthesis methods play a fundamental role, influencing the characteristics of the individual components, the bonds formed between them, size, degree of distribution, interface interactions, performance criteria, etc.

In the present paper, for the synthesis of  $\text{In}_2\text{O}_3\text{-ZnO}$  composites, the wet chemical method was used, followed by the steps of maturation, aging, filtration, drying, and, finally, heat-treatment steps. To obtain the desired properties of the synthesized composites, it was ensured that the process parameters (reaction temperature, rate of addition of reactants, concentration, etc.) were precisely controlled because they have a direct effect on the size and morphology of the particles.

## METHOD



Steps in the precipitation process for the obtaining of  $\text{In}_2\text{O}_3\text{-ZnO}$  powder

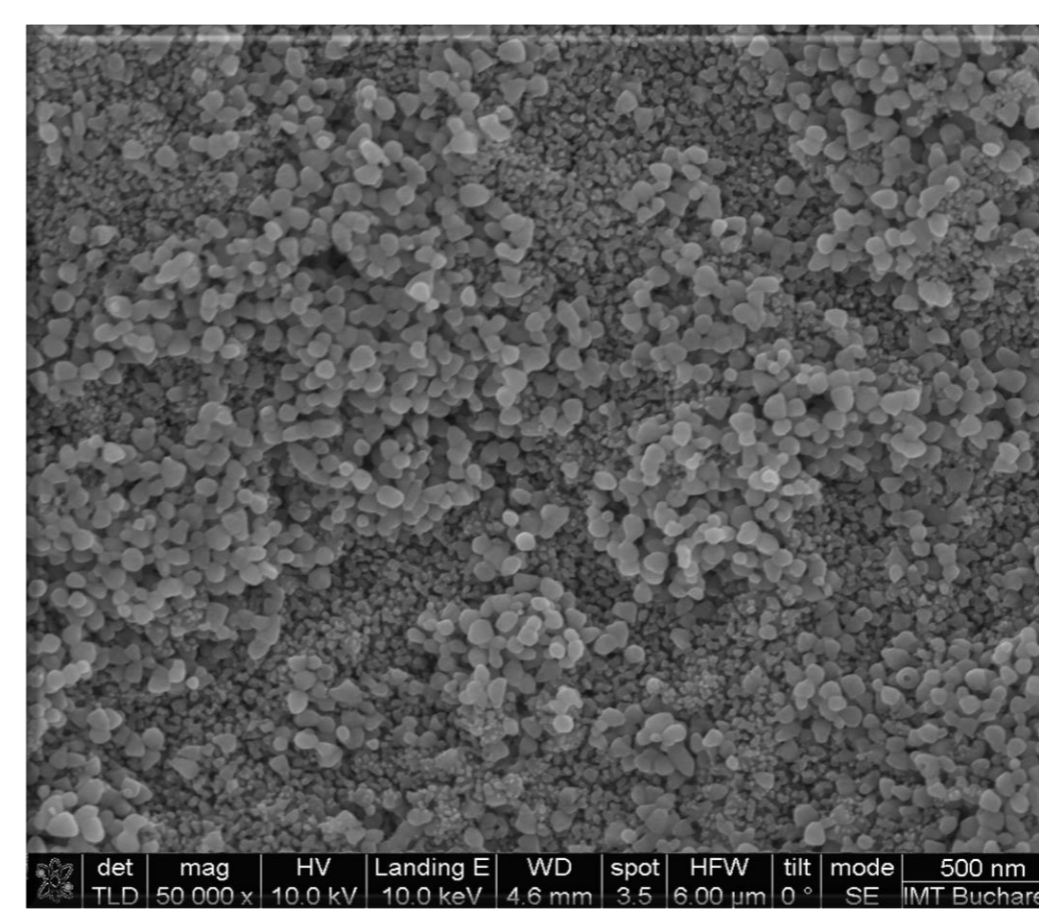
## CONCLUSIONS

- ❖ Methods for investigating the structure, morphology, and wetting capacity are consistent with the physical-chemical properties.
- ❖ The SEM images showed the formation of some spherical particles, with a range of average particle size below 50 nm.
- ❖ The XRD patterns confirm the hexagonal wurtzite-type crystalline structure (Card No. 00-036-1451) and  $\text{In}_2\text{O}_3$  (Card No. 00-006-0416). The average crystallite size of the  $\text{In}_2\text{O}_3\text{-ZnO}$  NPs was found of 20 nm.
- ❖ The EDX peaks indicated that zinc, indium and oxygen exist in sample and no other elements were observed in the spectra, which confirms the purity of the nanoparticles.
- ❖ FTIR spectrum reveals the presence of the characteristic In-O and Zn-O bonds, indicating that the precursors used transform to the desired nanostructures.
- ❖ The wetting capacity exhibits a strong hydrophilic character and good percolation properties, with contact angles varying between  $26\text{-}12^\circ$ .

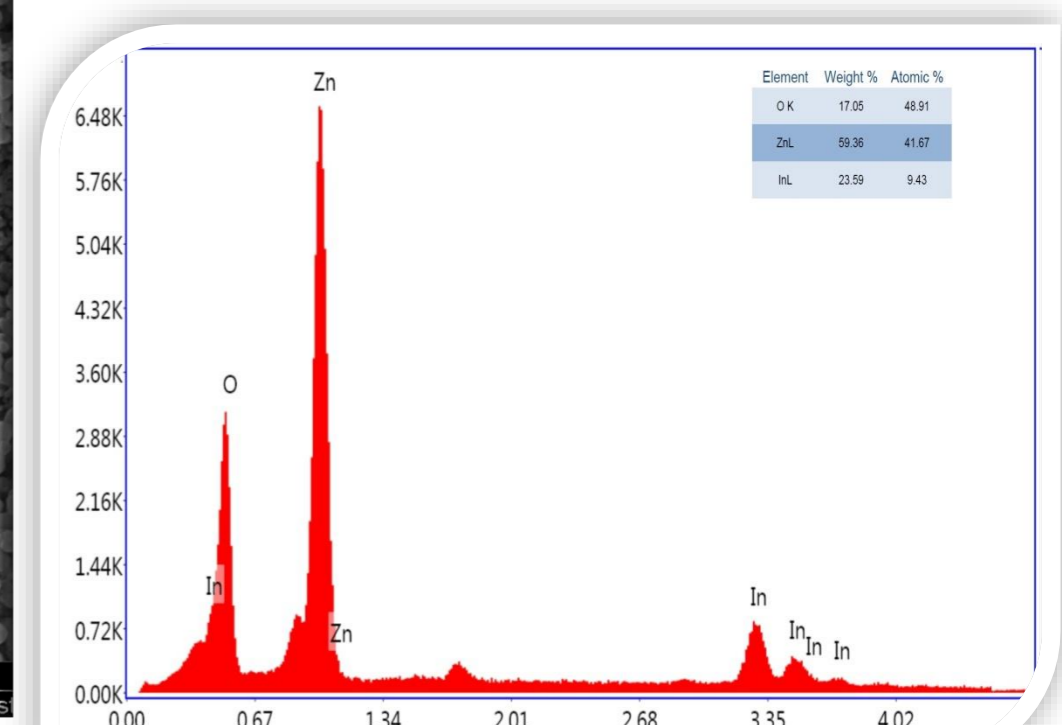
This work was supported by Core Program within the National Research Development and Innovation Plan 2022-2027, project no. 2307 ( $\mu\text{NanoEI}$ ).

- [1] M. Gholami, A.A. Khodadadi, A.A. Firooz, Y. Mortazavi, *In<sub>2</sub>O<sub>3</sub>-ZnO nanocomposites: High sensor response and selectivity to ethanol*, Sensors and Actuators B: Chemical 212 [2015]
- [2] G. Tatrai, M. Ahmed, F.U. Shah, *Synthesis, thermoelectric and energy storage performance of transition metal oxides composites*, Coordination Chemistry Reviews 498 [2024]
- [3] Z. Wang, B. Huang, Y. Dai, X. Qin, X. Zhang, P. Wang, H. Liu, J. Yu, *Highly Photocatalytic ZnO/In<sub>2</sub>O<sub>3</sub> Heteronanostructures Synthesized by a Coprecipitation Method*, The Journal of Physical Chemistry C 113(11) [2009]
- [4] [https://ebrary.net/191955/engineering/liquid\\_phase\\_reactions](https://ebrary.net/191955/engineering/liquid_phase_reactions)

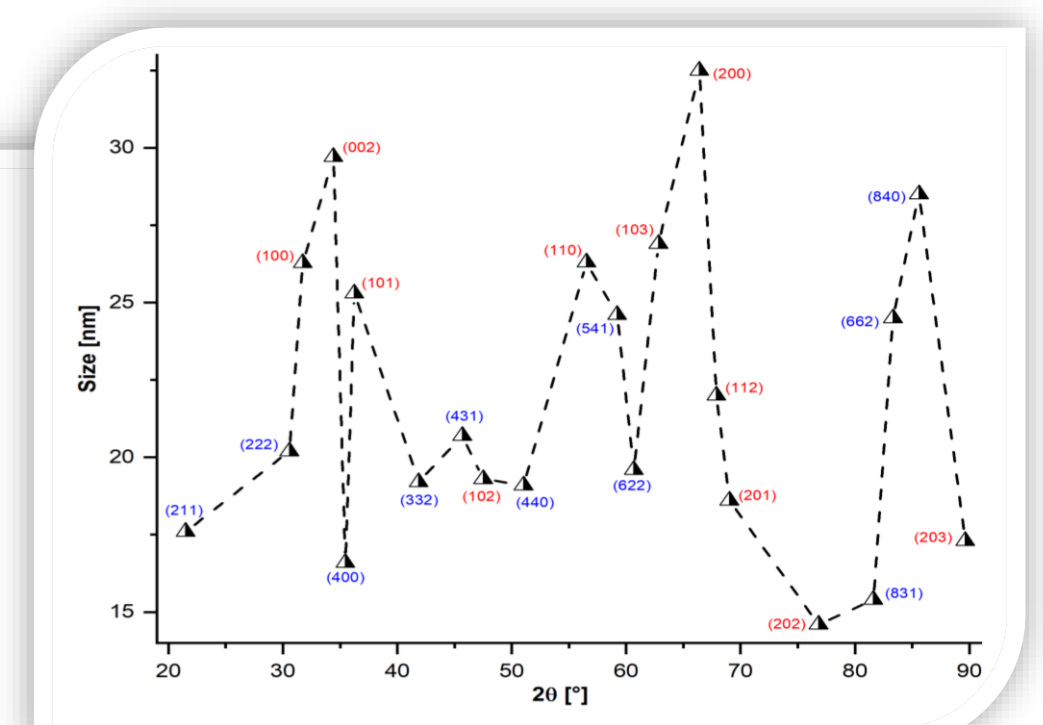
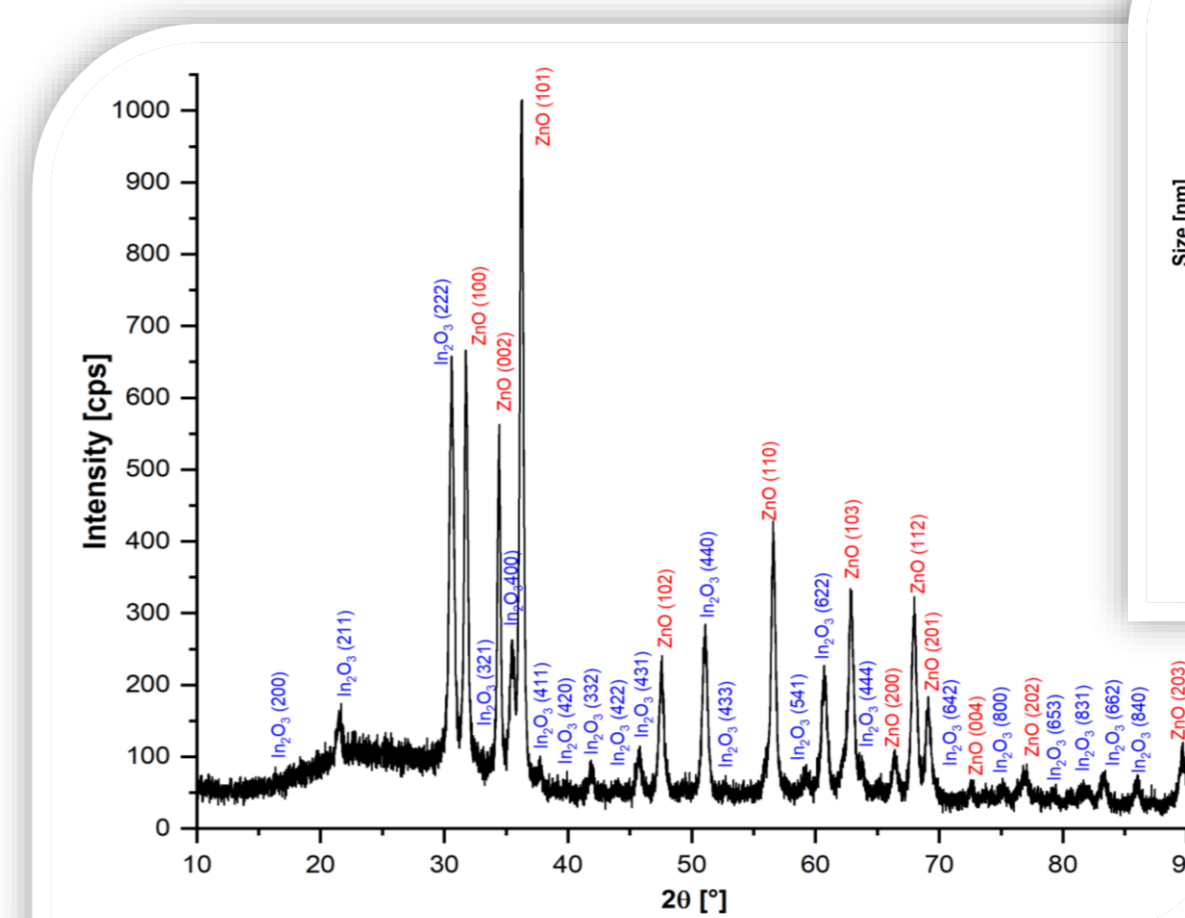
## RESULTS & DISCUSSION



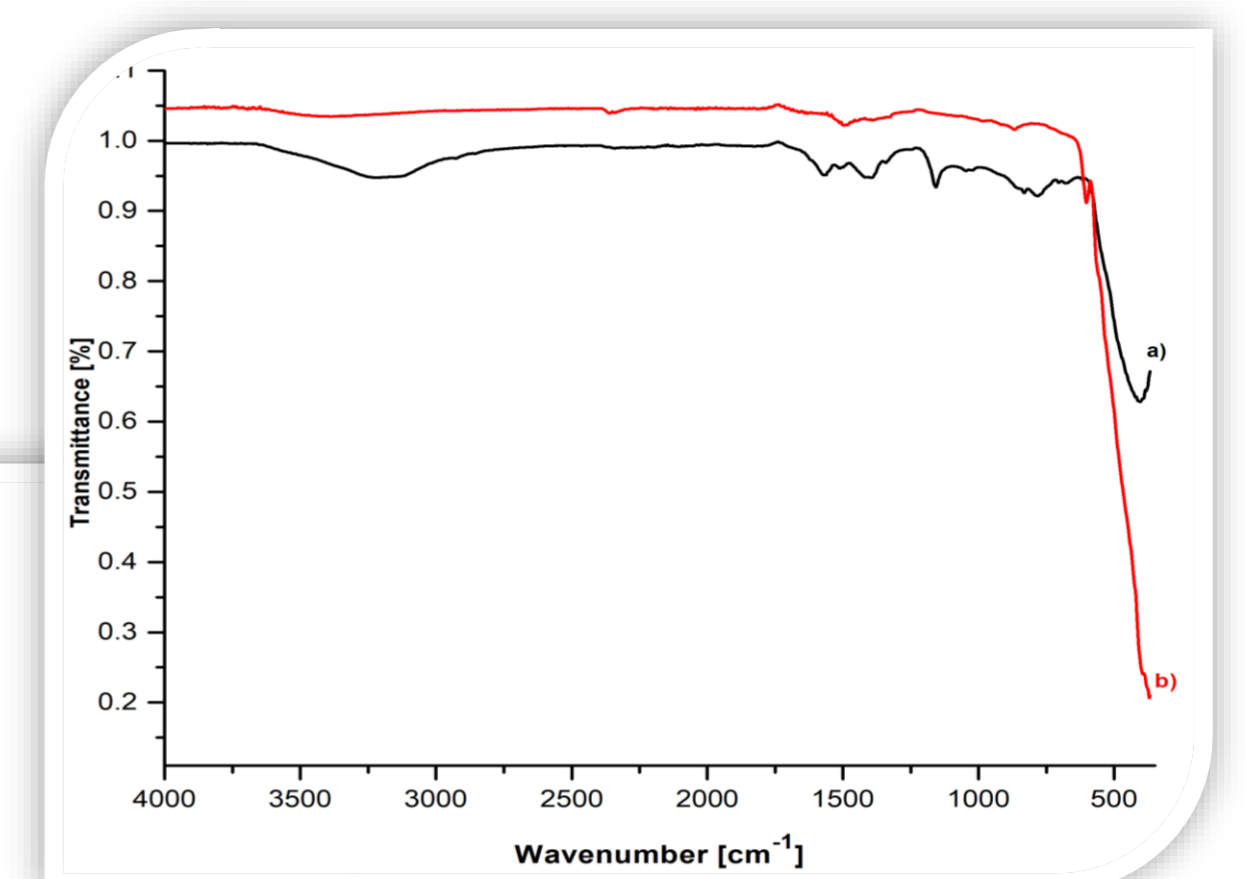
SEM image of  $\text{In}_2\text{O}_3\text{-ZnO}$  samples



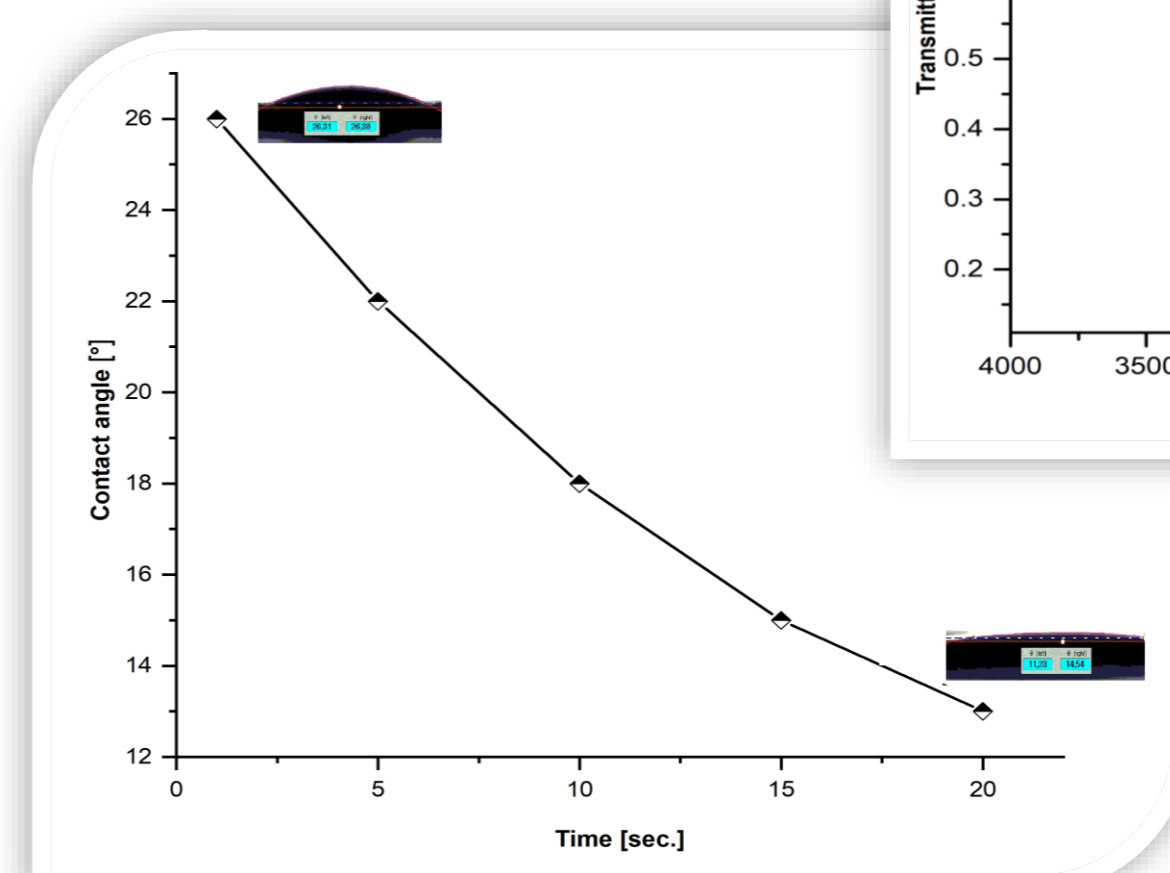
EDX spectra of  $\text{In}_2\text{O}_3\text{-ZnO}$  samples



XRD diffraction patterns and size for the  $\text{In}_2\text{O}_3\text{-ZnO}$  samples



ATR-FTIR spectra for the  $\text{In}_2\text{O}_3\text{-ZnO}$  samples, sintered at a)  $100^\circ\text{C}$  and b)  $550^\circ\text{C}$



The variation of contact angle depending on time at the contact of the water droplet with the surface of the  $\text{In}_2\text{O}_3\text{-ZnO}$  samples

## ACKNOWLEDGEMENTS / REFERENCES